#### Appendix 2 Food additives specifications

Category 1 Preservatives

§ 01001

Sorbic Acid **Synonyms INS No. 200** Definition Chemical names Sorbic acid, 2,4-hexadienoic acid, 2-propenylacrylic acid CAS NUMBER 110-44-1 Molecular formula  $C_6H_8O_2$ Molecular weight 112.12 Not less than 99.0% on the dried basis. Assay Appearance Colorless needle-shaped or white liquid powder with a characteristic odor. Characteristics Identification Solubility Slightly soluble in water, soluble in ethanol. 132~135°C Melting range Spectral analysis The product's isopropanol solution (2.5 µg/mL) has a maximum absorbance at a wavelength of  $254 \pm 2$  nm. Take 0.02 g of this product and shake it with 1 mL of bromine test solution. The Double bond test color of the solution disappears. Purity Water Not more than 0.5% (Karl Fischer Method). Sulfated ash Not more than 0.2%. Aldehydes Not more than 0.1% as formaldehyde Lead Not more than 2 mg/kg. Food Additives Category 1 Category Uses Preservative

Potassium Sorbate		
Synonyms	INS No. 202	
Definition		
Chemical names	Potassium sorbate, potassium salt of trans, trans-2,4-hexadienoic acid	
CAS NUMBER	24634-61-5	
Molecular formula	C <sub>6</sub> H <sub>7</sub> KO <sub>2</sub>	
Molecular weight	150.22	
Assay	Not less than 98% and not more than 102% on the dried basis.	
Appearance	White or yellowish-white crystals or crystalline powder or granules.	
Characteristics		
Identification		
Solubility	Soluble in water, slightly soluble in ethanol.	
Test for potassium	Passes test	
Melting range of sorbic acid	132~135°C	
derived from the sample		
Test for unsaturation	To 2 mL of a 1 in 10 solution of the sample, add a few drops of bromine TS.	
	The color of the bromine disappears.	
Purity		
Loss on drying	Not more than 1% (105°, 3 h).	
Acidity or alkalinity	Not more than 1% (as sorbic acid or potassium carbonate)	
Aldehydes	Not more than 0.1% as formaldehyde	
Lead	Not more than 2 mg/kg.	
Category	Food Additives Category 1	
Uses	Preservative	

2

#### Sodium Sorbate

Cher	nical formula: C <sub>6</sub> H <sub>7</sub> C	D2NaMolecular weight: 134.11
1.	Assay	: 98~102% (vacuum desiccator over sulfuric acid, 4 h)
2.	Appearance	: White to light yellow brown scaly crystal or crystalline powder, odorless or slightly
		stinky.
3.	Melting range of	: 132°C~135°C.
	sorbic acid	Acidify a solution of the sample with dilute hydrochloric acid TS. Collect the
	derived from the	precipitated sorbic acid on a filter paper, wash free of chloride with water and dry.
	sample	
4.	Solution	: When 0.2 g of this product is dissolved in 5 mL water, the liquid color should not be
		thicker than the Matching Fluid F.
5.	Free alkali	: 1 g of this product is dissolved in 20 mL of fresh boiled cooling water, red should
		appear when adding 2 drops of phenolphthalein test solution. However, when 0.1 mL
		of 0.1 N sulfuric acid is added, the red color should disappear.
6.	Chloride	: Not more than 0.015% (as Cl).
7.	Sulfate	: Not more than 0.04% (as SO <sub>4</sub> ).
8.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
9.	Heavy metals	: Not more than 10 ppm of (as Pb).
10.	Loss on drying	: Not more than 1% (vacuum desiccator over sulfuric acid, 4 h)
11.	Residue on	: 52.0~53.5%.
	ignition	
12.	Category	: Food Additives Category 1
13.	Uses	: Preservative

	Calcium Propionate
Synonyms	Calcium propanoate, INS No. 282
Definition	
Chemical names	Calcium propionate
CAS NUMBER	4075-81-4
Molecular formula	$C_6H_{10}CaO_4$
Molecular weight	186.22
Assay	Not less than 98.0% on the dried basis
Appearance	White crystals, powder or granules with not more than a faint odor of
	propionic acid.
Characteristics	
Identification	
Solubility	Freely soluble in water, soluble in ethanol.
Positive test for calcium	Passes test
Positive test for propionate	Warm the sample with sulfuric acid. The propionic acid evolved may
	be recognized by its odor.
Positive test for alkali salt of organic	Ignite the sample at a relatively low temperature. The alkaline residue
acid	effervesces with acid.
Purity	
Loss on drying	Not more than 4% (105°, 2h)
pH	7.5~10.5 (1 in 10 soln)
Water-insoluble matter	Not more than 0.3%.
Fluoride	Not more of 30 mg/kg.
Iron	Not more than 50 mg/kg.
Lead	Not more than 5 mg/kg.
Category	Food Additives Category 1
Uses	Preservative

	Sodium Propionate
Synonyms	Sodium propanoate, INS No. 281
Definition	
Chemical names	Sodium propionate
CAS NUMBER	137-40-6
Molecular formula	C <sub>3</sub> H <sub>5</sub> NaO <sub>2</sub>
Molecular weight	96.06
Assay	Not less than 99.0% on the dried basis.
Appearance	White or colorless, hygroscopic crystals with not more than a faint
	characteristic odor.
Characteristics	
Identification	
Solubility	Freely soluble in water, soluble in ethanol.
Positive test for sodium	Passes test
Positive test for propionate	Warm the sample with sulfuric acid. The propionic acid evolved may
	be recognized by its odor.
Positive test for alkali salt of organic	Ignite the sample at a relatively low temperature. The alkaline residue
acid	effervesces with acid.
Purity	
Loss on drying	Not more than 4% (105°, 2h)
pH	7.5~10.5 (1 in 10 soln)
Water-insoluble matter	Not more than 0.1%.
Iron	Not more than 50 mg/kg.
Lead	Not more than 5 mg/kg.
Category	Food Additives Category 1
Uses	Preservative

### Dehydroacetic Acid

Chen	nical formula: C <sub>8</sub> H <sub>8</sub> O <sub>4</sub>	Molecular weight: 168.15
1.	Assay	: Not less than 98.0%
2.	Appearance	: Colorless to white needle, plate crystal or crystalline powder; odorless, or slightly
		stinky.
3.	Identification	: (1) 0.1 g of this product is added with 1 mL water, 3 to 5 drops of salicylaldehyde-
		alcohol solution (salicylaldehyde 1 mL dissolved in alcohol 5 mL) and 0.5 mL of
		sodium hydroxide solution (sodium hydroxide 1 g dissolved in water 2 mL).
		When heated in a water, the solution appears red.
		(2) The 1 mL alcohol solution (1 g of this product dissolved in 100 mL alcohol), water
		1 mL, 3 drops of potassium sodium tartrate solution (7 g of potassium sodium
		tartrate dissolved in 50 mL water) and 2 drops of strong copper acetate test
		solution are mixed to form a purple precipitate with white color.
4.	Melting range	: 109~112°C.
5.	Color of the	: When 0.5 g of this product is dissolved in 10 mL of acetone, the solution should be
	solution	colorless.
6.	Arsenic	: Not more than 4 ppm(as $As_2O_3$ ).
7.	Heavy metals	: Not more than 10 ppm of Pb.
8.	Readily	: When 0.30 g of this product is dissolved in 5 mL of sulfuric acid, the color of the
	carbonizable	solution can not be deeper than the matching fluid C.
	substance	
9.	Residue on	: Not more than 0.10%.
	ignition	
10.	Loss on drying	: Not more than 1%.
11.	Category	: Food Additives Category 1
12.	Uses	: Preservative

#### Sodium Dehydroacetate

Chen	nical formula: C <sub>8</sub> H <sub>7</sub> O	Molecular weight: 208.15
1.	Assay	: Not less than 98.0% on the dried basis.
2.	Appearance	: Colorless crystalline powder, odorless or with a slightly characteristic odor.
3.	Identification	: (1) 0.1 g of this product is added with 1 mL water, 3 to 5 drops of salicylaldehyde-
		alcohol solution (salicylaldehyde 1 mL dissolved in alcohol 5 mL) and 0.5 mL of
		sodium hydroxide solution (sodium hydroxide 1 g dissolved in water 2 mL).
		When heated in a water, the solution appears red.
		(2) The 1 mL alcohol solution (1 g of this product dissolved in 100 mL alcohol), water
		1 mL, 3 drops of potassium sodium tartrate solution (7 g of potassium sodium
		tartrate dissolved in 50 mL water) and 2 drops of strong copper acetate test
		solution are mixed to form a purple precipitate with white color.
		(3) Passes test for sodium.
4.	Color of the	: When 0.5 g of this product is dissolved in water 10 mL, the solution should be colorless
	solution	or almost clear.
5.	Dehydroacetic	: Acidify a solution of the sample (0.5 g in 10 mL of water) with 1 mL of dilute
	acid	hydrochloric acid TS (1 mL of hydrochloric acid to 4 mL). Collect the precipitation on
		a filter paper, wash free of chloride with water and dry at 105°C for an hour. The
		melting range of the filtrate is 109~112°C.
6.	Free alkali	: Dissolve 1.0 g of the sample in 20 mL of freshly boiled and cooled water, and add 2
		drops of phenolphthalein TS. If the solution is red, the red color is discharged by
		adding 0.30 mL of 0.1 N sulfuric acid.
7.	Chloride	: Not more than of 0.01% (as Cl).
8.	Sulfate	:Not more than $0.015\%$ of (as SO <sub>4</sub> ).
9.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
10.	Heavy metals	: Not more than 10 ppm of Pb.
11.	Readily	: When 0.30 g of this product is dissolved in 5 mL of sulfuric acid, the color of the
	carbonizable	solution can not be deeper than the matching fluid C.
	substance	
12.	Water	: 8.3~10.0% (Karl Fischer Method).
13.	Category	: Food Additives Category 1
14.	Uses	: Preservative

Synonyms

# Benzoic Acid INS No. 210

Definition	
Chemical names	Benzoic acid, benzenecarboxylic acid, phenylcarboxylic acid
CAS NUMBER	65-85-0
Molecular formula	$C_7H_6O_2$
Molecular weight	122.12
Assay	Not less than 99.5% (on the dried basis).
Appearance	White crystalline solid, usually in the form of scales or needles, having not
	more than a faint characteristic odor.

#### Characteristics

Solubility       Slightly soluble in water, freely soluble in ethanol.         Melting range       121~123°C         Test for benzoate       Passes test (Use 0.1 g of the sample with 0.1 g of calcium carbonate and 5 mL of water)         pH of the solution       About 4.0 (solution in water)         Purity       Item test that 0.5% (over sulfuric acid, 3 h)         Sublimation test       Place a small amount of the sample in a dry test tube. Wrap the test tube about 4 cm from the bottom with moistened filter paper. Heat the test tube over a low flame. Benzoic acid sublimes and crystals deposit in the colder part of the test tube leaving no residue at the bottom.         Sulfated ash       Not more than 0.05%.         Lead       Not more than 2 mg/kg.         Readily carbonizable       Dissolve 0.5 g of the sample, weighed to the nearest mg, in 5 mL of sulfuric acid TS (94.5~95.5%). The color produced should not be darker than the Matching Fluid Q.         Readily oxidizable matter       Add 1.5 mL of sulfuric acid to 100 mL of water, heat to boiling and add 0.1N potassium permanganate in drops, until the pink color persists for 30 sec.         Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and tirate with 0.1N potassium permanganate to a pink color that persists for 30 sec.	Identification	
Test for benzoatePasses test (Use 0.1 g of the sample with 0.1 g of calcium carbonate and 5 mL of water)pH of the solutionAbout 4.0 (solution in water)PurityImage: Solution of the sample in a dry test tube. Wrap the test tube about 4 cm from the bottom with moistened filter paper. Heat the test tube over a low flame. Benzoic acid sublimes and crystals deposit in the colder part of the test tube leaving no residue at the bottom.Sulfated ashNot more than 0.05%.LeadNot more than 2 mg/kg.Readily carbonizableDissolve 0.5 g of the sample, weighed to the nearest mg, in 5 mL of sulfuric substanceReadily oxidizable matterAdd 1.5 mL of sulfuric acid to 100 mL of water, heat to boiling and add 0.1N potassium permanganate in drops, until the pink color persists for 30 sec. Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and titrate with 0.1N potassium permanganate to a pink color that persists for 30 sec.	Solubility	Slightly soluble in water, freely soluble in ethanol.
nL of water)pH of the solutionAbout 4.0 (solution in water)PurityLoss on dryingNot more than 0.5% (over sulfuric acid, 3 h)Sublimation testPlace a small amount of the sample in a dry test tube. Wrap the test tube about 4 cm from the bottom with moistened filter paper. Heat the test tube over a low flame. Benzoic acid sublimes and crystals deposit in the colder part of the test tube leaving no residue at the bottom.Sulfated ashNot more than 0.05%.LeadNot more than 2 mg/kg.Readily carbonizableDissolve 0.5 g of the sample, weighed to the nearest mg, in 5 mL of sulfuric substancesubstanceAdd 1.5 mL of sulfuric acid to 100 mL of water, heat to boiling and add 0.1N potassium permanganate in drops, until the pink color persists for 30 sec. Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and tirate with 0.1N potassium permanganate to a pink color that persists for	Melting range	121~123°C
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Sulfated ashNot more than 0.05%.LeadNot more than 2 mg/kg.Readily carbonizableDissolve 0.5 g of the sample, weighed to the nearest mg, in 5 mL of sulfuricsubstanceacid TS (94.5~95.5%). The color produced should not be darker than the Matching Fluid Q.Readily oxidizable matterAdd 1.5 mL of sulfuric acid to 100 mL of water, heat to boiling and add 0.1N potassium permanganate in drops, until the pink color persists for 30 sec. Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and titrate with 0.1N potassium permanganate to a pink color that persists for		over a low flame. Benzoic acid sublimes and crystals deposit in the colder
LeadNot more than 2 mg/kg.Readily carbonizableDissolve 0.5 g of the sample, weighed to the nearest mg, in 5 mL of sulfuricsubstanceacid TS (94.5~95.5%). The color produced should not be darker than the Matching Fluid Q.Readily oxidizable matterAdd 1.5 mL of sulfuric acid to 100 mL of water, heat to boiling and add 0.1N potassium permanganate in drops, until the pink color persists for 30 sec.Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and titrate with 0.1N potassium permanganate to a pink color that persists for		part of the test tube leaving no residue at the bottom.
Readily carbonizableDissolve 0.5 g of the sample, weighed to the nearest mg, in 5 mL of sulfuricsubstanceacid TS (94.5~95.5%). The color produced should not be darker than the Matching Fluid Q.Readily oxidizable matterAdd 1.5 mL of sulfuric acid to 100 mL of water, heat to boiling and add 0.1N potassium permanganate in drops, until the pink color persists for 30 sec. Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and titrate with 0.1N potassium permanganate to a pink color that persists for	Sulfated ash	Not more than 0.05%.
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Matching Fluid Q.Readily oxidizable matterAdd 1.5 mL of sulfuric acid to 100 mL of water, heat to boiling and add 0.1N potassium permanganate in drops, until the pink color persists for 30 sec.Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and titrate with 0.1N potassium permanganate to a pink color that persists for	Readily carbonizable	Dissolve 0.5 g of the sample, weighed to the nearest mg, in 5 mL of sulfuric
Readily oxidizable matterAdd 1.5 mL of sulfuric acid to 100 mL of water, heat to boiling and add 0.1N potassium permanganate in drops, until the pink color persists for 30 sec.Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and titrate with 0.1N potassium permanganate to a pink color that persists for	substance	acid TS ( $94.5 \sim 95.5\%$ ). The color produced should not be darker than the
potassium permanganate in drops, until the pink color persists for 30 sec. Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and titrate with 0.1N potassium permanganate to a pink color that persists for		Matching Fluid Q.
Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution, and titrate with 0.1N potassium permanganate to a pink color that persists for	Readily oxidizable matter	Add 1.5 mL of sulfuric acid to 100 mL of water, heat to boiling and add 0.1N
and titrate with 0.1N potassium permanganate to a pink color that persists for		potassium permanganate in drops, until the pink color persists for 30 sec.
		Dissolve 1 g of the sample, weighed to the nearest mg, in the heated solution,
8		and titrate with 0.1N potassium permanganate to a pink color that persists for
	8	

	15 sec. Not more than 0.5 mL should be required.
Organochloride	Not more than $0.07\%$ (as Cl <sub>2</sub> ).
Category	Food Additives Category 1
Uses	Preservative

	Sodium Benzoate
SYNONYMS	INS No. 211
DEFINITION	
Chemical names	Sodium benzoate, sodium salt of benzenecarboxylic
	acid, sodium salt of phenylcarboxylic acid
C.A.S. number	532-32-1
Chemical formula	$C_7H_5O_2Na$
Formula weight	144.11
Assay	Not less than 99.0% on the dried basis
DESCRIPTION	White, almost odourless, crystalline powder, flakes or granules
CHARACTERISTICS	
IDENTIFICATION	
Solubility	Freely soluble in water, sparingly soluble in ethanol
Test for benzoate	Passes test
	Use a 10% solution of the sample
Test for sodium	Passes test
PURITY	
Loss on drying	Not more than $1.5\%$ ( $105^{\circ}$ C, 4 h) $\circ$
Acidity or alkalinity	Dissolve 2 g of the sample, weighed to the nearest mg,
	in 20 ml of freshly boiled water. Not more than 0.5 ml
	of either 0.1N sodium hydroxide or 0.1N hydrochloric
	acid should be required for neutralization, using
	phenolphthalein TS as indicator.
Lead	Not more than 2 mg/kg
Readily carbonizable	Dissolve 0.5 g of the sample, weighed to the nearest mg,
substances	in 5 ml of sulfuric acid TS. The color produced should
	not be darker than a light pink ("Matching Fluid Q")
Readily oxidizable	Add 1.5 ml of sulfuric acid to 100 ml of water, heat to
substances	boiling and add 0.1N potassium permanganate,
	dropwise, until the pink color persists for 30 sec.
	Dissolve 1 g of the sample, weighed to the nearest mg,
	in the heated solution, and titrate with 0.1N potassium
	permanganate to a pink color that persists for 15 sec.

	Not more than 0.5 ml should be required.
Chlorinated organic	Not more than 0.07% (as chlorine)
compounds	
Category	Food additives category (1).
Functional uses	Preservatives.

### Ethyl p-Hydroxybenzoate

Cher	nical formula: C <sub>9</sub> H <sub>10</sub> G	D <sub>3</sub> Molecular weight: 166.18
1.	Assay	: Not less than 99.0%.
2.	Appearance	: Colorless crystal or white crystalline powder, odorless.
3.	Identification	: (1) To 0.5 g of the sample add 10 mL of sodium hydroxide TS (1 g of sodium
		hydroxide dissolved in 25 mL of water). Boil for 30 min and concentrate to about 5 mL.
		Cool, acidify with dilute sulfuric acid TS, collect the precipitate on a filter, and wash
		thoroughly with water. Dry in a desiccator over sulfuric acid. Determine the melting
		range of phydroxybenzoic acid so obtained. Melting range of p-hydroxybenzoic acid
		derived from the sample is 213~217°C.
		(2) Add 2 drops of acetic acid and 5 drops of sulfuric acid to $0.05$ g of the sample, and
		the odor of ethyl acetate will be produced after heating for 5 minutes.
4.	Melting range	: 115~118°C.
5.	Free acid	: Not more than 0.55% (as 4-Hydroxybenzoic acid).
6.	Sulfate	: Not more than 0.024% (as SO <sub>4</sub> ).
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
8.	Heavy metals	: Not more than 10 ppm of Pb.
9.	Loss on drying	: Not more than 0.5% (80°C , 2 h)
10.	Residue on	: Not more than 0.05%.
	ignition	
11.	Category	: Food Additives Category 1
12.	Uses	: Preservative

### Propyl p-Hydroxybenzoate

Chemical formula: C <sub>10</sub> H <sub>12</sub> G		O <sub>3</sub> Molecular weight: 180.20
1.	Assay	: Not less than 99.0%.
2.	Appearance	: Colorless crystal or white crystalline powder, odorless.
3.	Identification	: (1) To 0.5 g of the sample add 10 mL of sodium hydroxide TS (1 g of sodium
		hydroxide dissolved in 25 mL of water). Boil for 30 min and concentrate to about 5 mL.
		Cool, acidify with dilute sulfuric acid TS, collect the precipitate on a filter, and wash
		thoroughly with water. Dry in a desiccator over sulfuric acid. Determine the melting
		range of phydroxybenzoic acid so obtained. Melting range derived from the sample is
		213~217°C.
		(2) Add 2 drops of acetic acid and 5 drops of sulfuric acid to $0.05$ g of the sample. The
		odor of propyl acetate will be produced after heating for 5 minutes.
4.	Melting range	: 95~98°C.
5.	Free acid	: Not more than 0.55% (as 4-Hydroxybenzoic acid).
6.	Sulfate	: Not more than 0.024% (as SO <sub>4</sub> ).
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Loss on drying	: Not more than 0.5% (Silicone dryer, 5 h).
10.	Residue on	: Not more than 0.05%.
	ignition	
11.	Category	: Food Additives Category 1
12.	Uses	: Preservative
<ol> <li>5.</li> <li>6.</li> <li>7.</li> <li>8.</li> <li>9.</li> <li>10.</li> <li>11.</li> </ol>	Free acid Sulfate Arsenic Heavy metals Loss on drying Residue on ignition Category	<ul> <li>Not more than 0.55% (as 4-Hydroxybenzoic acid).</li> <li>Not more than 0.024% (as SO<sub>4</sub>).</li> <li>Not more than 4 ppm (as As<sub>2</sub>O<sub>3</sub>).</li> <li>Not more than 10 ppm (as Pb).</li> <li>Not more than 0.5% (Silicone dryer, 5 h).</li> <li>Not more than 0.05%.</li> </ul>

### Butyl p-Hydroxybenzoate

Chemical formula: C <sub>11</sub> H <sub>14</sub> O <sub>3</sub>		O <sub>3</sub> Molecular weight: 194.23
1.	Assay	: Not less than 99.0%.
2.	Appearance	: Colorless crystal or white crystalline powder, odorless.
3.	Identification	: (1) To 0.5 g of the sample add 10 mL of sodium hydroxide TS (1 g of sodium
		hydroxide dissolved in 25 mL of water). Boil for 30 min and concentrate to about 5 mL.
		Cool, acidify with dilute sulfuric acid TS (1 mL of sulfuric acid dissolved in water 20
		mL), collect the precipitate on a filter, and wash thoroughly with water. Dry in a
		desiccator over sulfuric acid. Determine the melting range of phydroxybenzoic acid so
		obtained. Melting range derived from the sample is 213~217°C.
		(2) Add 2 drops of acetic acid and 5 drops of sulfuric acid to $0.05$ g of the sample. The
		odor of butyl acetate will be produced after heating for 5 minutes.
4.	Melting range	: 69~72°C.
5.	Free acid	: Not more than 0.55% (as 4-Hydroxybenzoic acid.
6.	Sulfate	: Not more than 0.024% (as SO <sub>4</sub> ).
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Loss on drying	: Not more than 0.5% (Silicone dryer, 5 h)
10.	Residue on	: Not more than 0.10%.
	ignition	
11.	Category	: Food Additives Category 1
12.	Uses	: Preservative

### Isopropyl p-Hydroxybenzoate

Chemical formula: C <sub>10</sub> H <sub>12</sub> C		O <sub>3</sub> Molecular weight: 180.20
1.	Assay	: Not less than 99.0%.
2.	Appearance	: Colorless crystal or white crystalline powder, odorless.
3.	Identification	: (1) To 0.5 g of the sample add 10 mL of sodium hydroxide TS (1 g of sodium
		hydroxide dissolved in 25 mL of water). Boil for 30 min and concentrate to about 5 mL.
		Cool, acidify with dilute sulfuric acid TS (1 mL of sulfuric acid dissolved in water 20
		mL), collect the precipitate on a filter, and wash thoroughly with water. Dry in a
		desiccator over sulfuric acid. Determine the melting range of phydroxybenzoic acid so
		obtained. Melting range derived from the sample is 213~217°C.
		(2) Add 2 drops of acetic acid and 5 drops of sulfuric acid to 0.05 g of the sample, and
		the odor of isopropyl acetate will be produced after heating for 5 minutes.
4.	Melting range	: 84~86°C.
5.	Free acid	: Not more than 0.55% (as 4-Hydroxybenzoic acid).
6.	Sulfate	: Not more than 0.024% (as SO <sub>4</sub> ).
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Loss on drying	: Not more than 0.5% (Silicone dryer, 5 h).
10.	Residue on	: Not more than 0.10%.
	ignition	
11.	Category	: Food Additives Category 1
12.	Uses	: Preservative

### Isobutyl p-Hydroxybenzoate

Chemical formula: C <sub>11</sub> H <sub>14</sub>		O <sub>3</sub> Molecular weight: 194.23
1.	Assay	: Not less than 99.0%.
2.	Appearance	: Colorless crystal or white crystalline powder, odorless.
3.	Identification	: (1) To 0.5 g of the sample add 10 mL of sodium hydroxide TS (1 g of sodium
		hydroxide dissolved in 25 mL of water). Boil for 30 min and concentrate to about 5 mL.
		Cool, acidify with dilute sulfuric acid TS (1 mL of sulfuric acid dissolved in water 20
		mL), collect the precipitate on a filter, and wash thoroughly with water. Dry in a
		desiccator over sulfuric acid. Determine the melting range of phydroxybenzoic acid so
		obtained. Melting range derived from the sample is 213-217°C.
		(2) Add 2 drops of acetic acid and 5 drops of sulfuric acid to 0.05 g of the sample, and
		the taste of isobutyl acetate will be produced after heating for 5 minutes.
4.	Melting range	: 75~77°C.
5.	Free acid	: Not more than 0.55% (as 4-Hydroxybenzoic acid).
6.	Sulfate	: Not more than 0.024% (as SO <sub>4</sub> ).
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Loss on drying	: Not more than 0.5% (Silicone dryer, 5h).
10.	Residue on	: Not more than 0.10%.
	ignition	
11.	Category	: Food Additives Category 1
12.	Uses	: Preservative

#### Sodium Diacetate

### (Sodium Hydrogen Diacetate)

Chemical formula: C4H7NaO4K·xH2O		aO <sub>4</sub> K·xH <sub>2</sub> O Molecular weight (Anhydrous): 142.09
1.	Description	: A molecular compound of sodium acetate and acetic acid. White, hygroscopic
		crystalline solid with an acetic odor. The pH of 1 in 10 solution of this sample is 4.5 to
		5.0.
2.	Identification	: The 1 in 10 solution of this sample passes the test for acetate and test for sodium.
3.	Assay	: CH <sub>3</sub> COOH 39.0~41.0%; CH <sub>3</sub> COONa 58.0~60.0%.
4.	Water	: Not more than 2%.
5.	Readily	: Not more than 0.2% (as formic acid).
	oxidizable	
	substances	
6.	Arsenic	: Not more than 3 ppm (as As).
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 1
9.	Uses	: Preservative

#### Potassium Benzoate

Chemical formula: C <sub>7</sub> H <sub>5</sub> O <sub>2</sub> K		Molecular weight: 160.11
1.	Assay	: Not less than 99% (drying at 110°C for 4 hours)
2.	Appearance	: White granular or crystalline powder, odorless.
3.	Solution	: When 1 g of the sample is dissolved in 5 mL of water, the solution should be clear.
4.	Free alkali	: Dissolve 2 g of the sample in 20 mL of hot water, and add 2 drops of
		phenolphthalein TS. If the solution is red, the red color is discharged by adding 0.2
		mL of 0.1 N sulfuric acid.
5.	Chloride compound	: Not more than 0.014%(as Cl).
6.	Sulfate	: Not more than 0.30% of (as SO <sub>4</sub> ).
7.	Benzenedicarboxylic	: The test for benzoic acid test of benzenedicarboxylic acid is applicable.
	acid	
8.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Heavy metals	: Not more than 10 ppm of (as Pb).
10.	Loss on drying	: Not more than 1% (110°C, 4 hr).
11.	Category	: Food Additives Category 1
12.	Uses	: Preservative

#### Nisin

1.	Description	: Nisin is a mixture of closely related antimicrobial polypeptides produced by
		Streptococcus lactis Lancefield Group N. White powder. Soluble in water and insoluble
		in non-polar solvents.
2.	Assay	: Not less than 900 IU/mg.
3.	Arsenic	: Not more than 1 ppm.
4.	Lead	: Not more than 2 ppm.
5.	Zinc	: Not more than 25 ppm.
6.	Total amount of	: Not more than 50 ppm.
	zinc and copper	
7.	Total plate count	: Not more than 10 CFU m.
8.	E. coli	: Negative by test/eg g.
9.	Salmonella	: Negative by test/eg g.
10.	Coagulase-positive Staphylococcus : Negative by test/eg g.	
	aureus	
11.	Category	: Food Additives Category 1
12.	Uses	: Preservative

#### Thiamine Dilaurylsulfate

Chemical formula: C <sub>36</sub> H <sub>68</sub> N <sub>4</sub> O <sub>9</sub> S <sub>3</sub> ·H <sub>2</sub> O	Molecular weight: 815.17

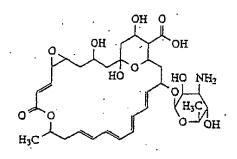
- 1. Assay : Not less than 98.0% and not more than 102.0%.
- 2. Appearance : Colorless to white crystal or white crystalline powder, odorless or slightly smelly.
- 3. Identification : (1) Dissolve 0.1 g of the sample in 20 mL of potassium chloride and hydrochloric acid TS, and boil slowly for 30 minutes. After cooling and filtration, add 1 mL of lead acetate TS and 1 mL sodium hydroxide solution to 1 mL of filtrate. The solution is yellow. Then heat in water bath. The solution turns brown, and after a period of time the dark brown precipitate appears.
  - (2) Take 1 mL of the filtrate obtained from test (1), add 2 g of sodium hydroxide TS 5 mL and 0.5 mL of potassium ferricyanide solution, then add 5 mL of isobutanol TS, shake hard after 2 minutes. When observed under UV light, blue-purple fluorescence appears on isobutanol liquid layer. When the solution is acidified, the fluorescence disappears; when the solution is alkaline, the fluorescence reappears.
  - (3) Add 30 mL of water and 15 mL of hydrochloric acid to 1 g of the sample, connect with a flow cooler, and cool down after boiling for 4 hours. Extract with 15 mL of ether twice. After the both extraction solution is combined and washed with water, the ether is removed by heating and evaporating in the water bath. The residue is dried at 100°C for 15 minutes and cooled, and its melting point should be 20°C to 28°C.
- 4. Chloride : Not more than 0.56%(as Cl).
- 5. Heavy metals : Not more than 20 ppm of (as Pb).
- 6. Loss on drying : Not more than 2.0% (Silicone dryer, 24 hr).
- 7. Residue on : Not more than 0.30%.
  - ignition
- 8. Category : Food Additives Category 1
- 9. Uses : Preservative

	Propionic Acid
Synonyms	INS No. 280; Propanoic acid, ethylformic acid, methylacetic acid
Definition	
Chemical names	Propionic acid
CAS NUMBER	79-09-4
Molecular formula	$C_3H_6O_2$
Molecular weight	74.08
Assay	Not less than 99.5% (on the dried basis)
Appearance	An oily liquid with a slightly pungent odor
Characteristics	
Identification	
Solubility	Miscible with water and ethanol
Specific gravity	D <sup>20</sup> <sub>20</sub> : 0.993~0.997
Purity	
Distillation range	138.5~142.5°C
Non-volatile residue	Not more than 0.01% when dried at 140°C to constant weight.
Formic acid	Not more than 0.1%.
Aldehydes	Not more than 0.2% (as propionaldehyde)
Lead	Not more than 2 mg/kg.
Category	Food Additives Category 1
Uses	Preservative

#### Natamycin (Pimaricin)

Chemical formula: C33H47NO13

Molecular weight: 665.74



- Description : White to ivory crystalline powder, almost tasteless, odorless. May contain up to three mole of water. Melts and discomposes at about 280°C. Insoluble in water, fat and mineral oil; slightly soluble in methanol; soluble in acetic acid and dimethylformamide.
- 2. Identification : Transfer 50 mg, accurately weighed, to a 200 mL volumetric flask, add 5.0 mL of water, and moisten the specimen. Add 100 mL of a 1 in 1000 solution of glacial acetic in methanol, and shake by mechanical means in the dark until dissolved. Dilute with the acetic acid-methanol solution to volume, and mix. Transfer 2.0 mL of this solution to a 100 mL volumetric flask, dilute with the acetic acid-methanol solution to volume, and mix; the UV absorption spectrum of the solution so obtained exhibits maxima and minima at the same wavelengths as that of a similar solution of USP Natamycin RS, concomitantly measured.

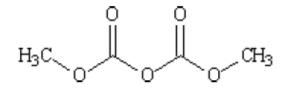
3. Assay : Not less than 97.0% and not more than 102.0% of  $C_{33}H_{47}NO_{13}$  on the dried basis.

- 4. Heavy metals : Not more than 20 mg/kg of Pb.
- 5. pH : 5.0~7.5.

6.

- Specific rotation :  $(\alpha) \frac{20}{D}$ :  $+276^{\circ} +280^{\circ}$ .
- 7. Water : 6.0%~9.0%.
- 8. Category : Food Additives Category 1
- 9. Uses : Preservative

### Dimethyl Dicarbonate



Synonyms : DMDC, Dimethyl Pyrocarbonate

INS No. 242

CAS No. 004-525-33-1

Chemical formula: C<sub>4</sub>H<sub>6</sub>O<sub>5</sub>

Molecular weight: 134.09

1.	Assay	: Not less than 99.8%.
2.	Appearance	: Colorless liquid.
3.	Solubility	: Soluble in water; miscible with toluene.
4.	Identification	: The infrared spectrum of the sample corresponds with the reference infrared spectrum
		and standard spectrum.
5.	Dimethyl	: Not more than 0.2%.
	carbonate	
6.	Lead	: Not more than 2 mg/kg.
7.	Category	: Food Additives Category 1
8.	Uses	: Preservative

### Hydrogen Peroxide

Chemical formula: H <sub>2</sub> O <sub>2</sub>		Molecular weight: 34.01
1.	Description	: Colorless, nearly odorless liquid. Miscible with water. Concentration suitable for food
		use is 30 ~ 50%.
2.	Identification	: Add 10 mL of water containing 1 drop of dilute sulfuric acid to 1 mL of the sample,
		shake. Add 2 mL of ether, then add 1 drop of potassium dichromate TS to the water
		layer to form a dissipative blue color. After being shaken and placed, the blue color
		will enter the ether layer.
3.	Assay	: Not less than the labelled concentration or within the range stated on the label.
4.	Acidity	: Not more than 0.03% of $H_2SO_4$ .
5.	Phosphate	: Not more than 0.005%.
6.	Iron	: Not more than 0.5 ppm.
7.	Tin	: Not more than 10 ppm.
8.	Arsenic	: Not more than 3 ppm (as As).
9.	Heavy metals	: Not more than 10 ppm (as Pb).
10.	Non-volatile	: Not more than 0.006%.
	residue	
11.	Category	: Food Additives Category 2
12.	Uses	: Sanitizing Agents

### Dibutyl Hydroxy Toluene

Chemical formula: C15H24		O Molecular weight: 220.35
1.	Assay	: Not less than 99.0%.
2.	Appearance	: Colorless crystal or white crystalline powder or particle, odorless or with a slightly
		characteristic odor.
3.	Identification	: To 10 mL of a 1 in 100,000 solution of the sample in methanol add 10 mL of water, 2
		mL of sodium nitrite solution (3 in 1000) and 5 mL of dianisidine dihydrochloride
		solution (200 mg of 3,3-dimethoxy-benzidine dihydrochloride dissolved in a mixture of
		40 mL of methanol and 60 mL of 1 N hydrochloric acid). An orange red color develops
		within 3 min. Add 5 mL of chloroform, and shake. The chloroform layer exhibits a
		purple or magenta color that fades when exposed to light.
4.	Melting range	: 69~72°C.
5.	Turbidity	: When 1.0 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.
6.	p-Cresol	: Not more than 0.1%.
7.	Sulfate	: Not more than 0.02% (as SO <sub>4</sub> ).
8.	Arsenic	: Not more than 3 ppm(as As).
9.	Heavy metals	: Not more than 10 ppm (as Pb).
10.	Residue on	: Not more than 0.05%.
	ignition	
11.	Category	: Food Additives Category 3
12.	Uses	: Antioxidants.

### Butyl Hydroxy Anisole

1.Assay: Not less than 98.5% (as C11H16O2)2.Description: Mixture mainly consists of 3-tert-butyl-4-hydroxyanisole(3-BHA), and, secondly, 2- tert-butyl-4-hydroxyanisole(2-BHA). White or slightly yellow crystals or waxy solid, with a faint characteristic odor. Insoluble in water; freely soluble in ethanol and propane- 1,2-diol.3.Identification: To 5 mL of a 1 in 10,000 solution of the sample in 72% ethanol, add 2 mL of sodium borate TS and 1 mL of a 1 in 10,000 solutions of 2,6- dichloroquinonechlorimide in absolute ethanol, and mix. A blue color appears.4.Melting range: 57~65°C5.Turbidity: When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.6.Sulfate: Not more than 0.02% (as SO4).7.Arsenic: Not more than 10 ppm (as As).8.Heavy metals: Not more than 0.05%. ignition10.Category: Food Additives Category 3	Cher	nical formula: C <sub>11</sub> H <sub>16</sub>	Molecular weight: 180.25
<ul> <li>tert-butyl-4-hydroxyanisole(2-BHA). White or slightly yellow crystals or waxy solid, with a faint characteristic odor. Insoluble in water; freely soluble in ethanol and propane-1,2-diol.</li> <li>Identification : To 5 mL of a 1 in 10,000 solution of the sample in 72% ethanol, add 2 mL of sodium borate TS and 1 mL of a 1 in 10,000 solutions of 2,6- dichloroquinonechlorimide in absolute ethanol, and mix. A blue color appears.</li> <li>Melting range : 57~65°C</li> <li>Turbidity : When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.</li> <li>Sulfate : Not more than 0.02% (as SO<sub>4</sub>).</li> <li>Arsenic : Not more than 3 ppm (as As).</li> <li>Heavy metals : Not more than 10 ppm (as Pb).</li> <li>Residue on : Not more than 0.05%. ignition</li> </ul>	1.	Assay	: Not less than 98.5% (as $C_{11}H_{16}O_2$ )
<ul> <li>with a faint characteristic odor. Insoluble in water; freely soluble in ethanol and propane- 1,2-diol.</li> <li>Identification : To 5 mL of a 1 in 10,000 solution of the sample in 72% ethanol, add 2 mL of sodium borate TS and 1 mL of a 1 in 10,000 solutions of 2,6- dichloroquinonechlorimide in absolute ethanol, and mix. A blue color appears.</li> <li>Melting range : 57~65°C</li> <li>Turbidity : When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.</li> <li>Sulfate : Not more than 0.02% (as SO<sub>4</sub>).</li> <li>Arsenic : Not more than 3 ppm (as As).</li> <li>Heavy metals : Not more than 10 ppm (as Pb).</li> <li>Residue on : Not more than 0.05%. ignition</li> </ul>	2.	Description	: Mixture mainly consists of 3-tert-butyl-4-hydroxyanisole(3-BHA), and, secondly, 2-
1,2-diol.3.Identification: To 5 mL of a 1 in 10,000 solution of the sample in 72% ethanol, add 2 mL of sodium borate TS and 1 mL of a 1 in 10,000 solutions of 2,6- dichloroquinonechlorimide in absolute ethanol, and mix. A blue color appears.4.Melting range: 57~65°C5.Turbidity: When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.6.Sulfate: Not more than 0.02% (as SO4).7.Arsenic: Not more than 3 ppm (as As).8.Heavy metals: Not more than 10 ppm (as Pb).9.Residue on ignition: Not more than 0.05%.			tert-butyl-4-hydroxyanisole(2-BHA). White or slightly yellow crystals or waxy solid,
<ol> <li>Identification : To 5 mL of a 1 in 10,000 solution of the sample in 72% ethanol, add 2 mL of sodium borate TS and 1 mL of a 1 in 10,000 solutions of 2,6- dichloroquinonechlorimide in absolute ethanol, and mix. A blue color appears.</li> <li>Melting range : 57~65°C</li> <li>Turbidity : When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.</li> <li>Sulfate : Not more than 0.02% (as SO<sub>4</sub>).</li> <li>Arsenic : Not more than 3 ppm (as As).</li> <li>Heavy metals : Not more than 10 ppm (as Pb).</li> <li>Residue on : Not more than 0.05%. ignition</li> </ol>			with a faint characteristic odor. Insoluble in water; freely soluble in ethanol and propane-
<ul> <li>borate TS and 1 mL of a 1 in 10,000 solutions of 2,6- dichloroquinonechlorimide in absolute ethanol, and mix. A blue color appears.</li> <li>Melting range : 57~65°C</li> <li>Turbidity : When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.</li> <li>Sulfate : Not more than 0.02% (as SO<sub>4</sub>).</li> <li>Arsenic : Not more than 3 ppm (as As).</li> <li>Heavy metals : Not more than 10 ppm (as Pb).</li> <li>Residue on : Not more than 0.05%. ignition</li> </ul>			1,2-diol.
<ul> <li>absolute ethanol, and mix. A blue color appears.</li> <li>Melting range : 57~65°C</li> <li>Turbidity : When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.</li> <li>Sulfate : Not more than 0.02% (as SO<sub>4</sub>).</li> <li>Arsenic : Not more than 3 ppm (as As).</li> <li>Heavy metals : Not more than 10 ppm (as Pb).</li> <li>Residue on : Not more than 0.05%. ignition</li> </ul>	3.	Identification	: To 5 mL of a 1 in 10,000 solution of the sample in 72% ethanol, add 2 mL of sodium
<ul> <li>4. Melting range : 57~65°C</li> <li>5. Turbidity : When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.</li> <li>6. Sulfate : Not more than 0.02% (as SO<sub>4</sub>).</li> <li>7. Arsenic : Not more than 3 ppm (as As).</li> <li>8. Heavy metals : Not more than 10 ppm (as Pb).</li> <li>9. Residue on : Not more than 0.05%. ignition</li> </ul>			borate TS and 1 mL of a 1 in 10,000 solutions of 2,6- dichloroquinonechlorimide in
<ol> <li>Turbidity : When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.</li> <li>Sulfate : Not more than 0.02% (as SO<sub>4</sub>).</li> <li>Arsenic : Not more than 3 ppm (as As).</li> <li>Heavy metals : Not more than 10 ppm (as Pb).</li> <li>Residue on : Not more than 0.05%. ignition</li> </ol>			absolute ethanol, and mix. A blue color appears.
<ul> <li>6. Sulfate : Not more than 0.02% (as SO<sub>4</sub>).</li> <li>7. Arsenic : Not more than 3 ppm (as As).</li> <li>8. Heavy metals : Not more than 10 ppm (as Pb).</li> <li>9. Residue on : Not more than 0.05%. ignition</li> </ul>	4.	Melting range	: 57~65°C
<ol> <li>Arsenic : Not more than 3 ppm (as As).</li> <li>Heavy metals : Not more than 10 ppm (as Pb).</li> <li>Residue on : Not more than 0.05%. ignition</li> </ol>	5.	Turbidity	: When 0.5 g of the sample is dissolved in 10 mL of ethanol, the solution should be clear.
<ul> <li>8. Heavy metals : Not more than 10 ppm (as Pb).</li> <li>9. Residue on : Not more than 0.05%.</li> <li>ignition</li> </ul>	6.	Sulfate	: Not more than $0.02\%$ (as SO <sub>4</sub> ).
9. Residue on : Not more than 0.05%. ignition	7.	Arsenic	: Not more than 3 ppm (as As).
ignition	8.	Heavy metals	: Not more than 10 ppm (as Pb).
-	9.	Residue on	: Not more than 0.05%.
10. Category   : Food Additives Category 3		ignition	
	10.	Category	: Food Additives Category 3
11. Uses : Antioxidants.	11.	Uses	: Antioxidants.

### L-Ascorbic Acid (Vitamin C)

Chen	nical formula: C <sub>6</sub> H <sub>8</sub> C	D <sub>6</sub> Molecular weight: 176.13
1.	Assay	: Not less than 99.0%.
2.	Description	: White to slightly yellow, odorless crystalline powder; having a sour taste. Soluble in
		water and ethanol; insoluble in chloroform, ether and benzene.
3.	Identification	: (1) Dissolve 0.1 g of the sample in 100 mL of metaphosphoric acid TS (1 in 50
		solution of metaphosphoric acid). Add iodine TS dropwise to 5 mL of this
		solution until the solution was slightly yellow. Then, add a drop of copper sulfate
		solution (1 in 1000 solution of copper sulphate) and a drop of pyrrole and heat in
		$50 \sim 60$ °C water bath for 5 minutes. The solution should be blue or green.
		(2) Add 1 to 2 drops of Sodium 2,6-dichlorophenolindophenol TS to 10 mL of 1 in
		100 solution of the sample. The blue color should disappear immediately.
4.	Melting range	: 187~192°C.
5.	Specific rotation	: $(\alpha) \frac{20}{D}$ : +20.5°~+21.5° (10% (w/v) aqueous solution)
6.	Arsenic	: Not more than 4 ppm(as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 20 ppm (as Pb).
8.	Loss on drying	: Not more than 0.4% (reduced pressure, silica gel, 3 h)
9.	Residue on	: Not more than 0.10%.
	ignition	
10.	Category	: Food Additives Category 3, 8.
11.	Uses	: Antioxidants; Nutritional additives

#### Sodium L-Ascorbate

Chemical formula: C <sub>6</sub> H <sub>7</sub> O <sub>6</sub> Na		Molecular weight: 198.11
1.	Assay	: Not less than 99.0%.
2.	Appearance	: White or yellowish-white, odorless and salty crystalline powder.
3.	Identification	: (1) Solubility: Soluble in water; sparingly soluble in ethanol
		(2) Passes test for ascorbate.
		(3) Passes test for sodium.
		(4) A solution of the sample will decolorize a solution of 2,6-dichlorophenolindophenol
		TS.
4.	рН	: 6.5 ~ 8.0 (1 in 10 soln)
5.	Specific rotation	: $(\alpha)_{D}^{20}$ : +103.0°~+108.0° (10% (w/v) aqueous solution)
6.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 20 ppm (as Pb).
8.	Loss on drying	: Not more than 0.25% (vacuum desiccator over sulfuric acid, 24 h)
9.	Category	: Food Additives Category 3, 8.
10.	Uses	: Antioxidants; Nutritional additives

### L-Ascorbyl Stearate

Chemical formula: C <sub>24</sub> H <sub>42</sub>		2O <sub>7</sub> Molecular weight: 442.59
1.	Assay	: Not less than 95%
2.	Appearance	: White or yellowish-white crystalline powder or powder.
3.	Identification	: (1) Dissolve 0.1 g of the sample in 100 mL of metaphosphoric acid TS (1 in 50 solution
		of metaphosphoric acid) by heating. Cool and add iodine TS dropwise to 5 mL of
		this solution until the solution was slightly yellow. Then, add a drop of copper
		sulfate solution (1 in 1000 solution of copper sulphate) and a drop of pyrrole and
		heat at $50 \sim 60$ °C for 5 minutes. The solution should be blue or green.
		(2) 10 mL of the 1 in 100 solution of the sample in ethanol will decolorize a blue
		solution of 2,6- dichlorophenol-indophenol TS.
4.	Melting range	: 114~119°C.
5.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Residue on	: Not more than 0.10%.
	ignition	
8.	Category	: Food Additives Category 3, 8.
9.	Uses	: Antioxidants; Nutritional additives

### L-Ascorbyl Palmitate

Chemical formula: C <sub>22</sub> H <sub>38</sub>		O <sub>7</sub> Molecular weight: 414.54
1.	Assay	: Not less than 95% (vacuum oven, 56 - 60°C, 1 h)
2.	Appearance	: White or yellowish-white powder, with a citrus-like odor.
3.	Melting range	: 107~117°C.
4.	Specific rotation	: $(\alpha) \frac{25}{D} = +21^{\circ} \sim +24^{\circ}$ (vacuum oven, 56 - 60°C, 1 h, then make 10% (w/v)
		methanol solution)
5.	Arsenic	: Not more than 3 ppm (as As).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Loss on drying	: Not more than 2% (vacuum oven, 56 - 60°C, 1 h)
8.	Residue on	: Not more than 0.1%.
	ignition	
9.	Category	: Food Additives Category 3, 8.
10.	Uses	: Antioxidants; Nutritional additives

#### Erythorbic Acid

Chemical form	ula: C <sub>6</sub> H <sub>8</sub> O <sub>6</sub>
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Molecular weight: 176.13

Assay : Not less than 99.0%.
 Description : White or yellowish-white, odorless, sour powder. Soluble in water and in ethanol,

slightly soluble in glycerin.

- 3. Identification : (1) Dissolve 0.1 g of the sample in 100 mL of metaphosphoric acid TS (1 in 50 solution of metaphosphoric acid). Add iodine TS dropwise to 5 mL of this solution until the solution was slightly yellow. Then, add a drop of copper sulfate solution (1 in 1000 solution of copper sulphate) and a drop of pyrrole and heat in 50 ~ 60 °C water bath for 5 minutes. The solution should be blue or green.
  - (2) Add 10 mL of potassium permanganate TS (1 g of potassium permanganate dissolved in 300 mL of water) to 10 mL of the 1 in 100 solution of the sample, and the pink color of the solution disappears immediately.
- 4. Melting range  $: 166 \sim 172^{\circ}$ C.
- 5. Specific rotation :  $[\alpha] \frac{20}{D} = -16.2^{\circ} \sim -18.2^{\circ} (10\% \text{ (w/v) aqueous solution)}.$
- 6. Arsenic : Not more than 4 ppm (as  $As_2O_3$ ).
- 7. Heavy metals : Not more than 20 ppm (as Pb).
- 8. Loss on drying : Not more than 0.40% (reduced pressure, silica gel, 3 h)
- 9. Residue on : Not more than 0.30%.

ignition

- 10. Category : Food Additives Category 3
- 11. Uses : Antioxidants.

Sodium Erythorbate		
Synonyms	INS No. 316; Sodium isoascorbate	
Definition		
Chemical names	Sodium isoascorbate, sodium D-isoascorbic acid, sodium salt of 2,3-	
	didehydro-D-erythro-hexono-1,4-lactone, 3-keto-D-gulofurano-lactone	
	sodium enolate monohydrate	
CAS NUMBER	6381-77-7	
Molecular formula	$C_6H_7O_6Na\cdot H_2O$	
Molecular weight	216.13	
Assay	Not less than 98.0% on the dried basis	
Appearance	White crystalline powder, almost odorless.	
Characteristics		
Identification		
Solubility	Freely soluble in water, very slightly soluble in ethanol.	
Reducing activity	A solution of the sample will decolorize a solution of 2,6-	
	dichlorophenolindophenol TS.	
Test for ascorbate	Passes test	
Test for sodium	Passes test	
Purity		
Loss on drying	Not more than $0.25\%$ (in vacuum over sulfuric acid, 24 h)	
Specific rotation	$[\alpha] 25, D: +95.5^{\circ} \rightarrow +98.0^{\circ} (10\% (w/v) \text{ solution})$	
pH	5.5~8.0 (10% solution)	
Oxalate	To a solution of 1 g in 10 mL of water add 2 drops of glacial acetic acid and	
	5 mL of 10% calcium acetate solution. The solution should remain clear.	
Lead	Not more than 2 mg/kg.	
Category	Food Additives Category 3	
Uses	Antioxidants.	

### dl-a-Tocopherol (Vitamin E)

Chemical formula: C <sub>29</sub> H <sub>50</sub> C		D2Molecular weight: 430.71
1.	Assay	: Not less than 96.0%
2.	Description	: Slightly yellow to amber, nearly odorless, clear, viscous oil. Insoluble in water, freely
		soluble in ethanol, miscible with acetone, chloroform, ether and vegetable oils.
3.	Identification	: Dissolve about 10 mL of the sample in 10 mL of absolute ethanol. Add 2 mL of nitric
		acid and heat at about 75°C for 15 min. A bright red to orange color develops.
4.	Spectrophotometry	: E (1%, 1 cm) (292 nm): 71.0~76.0 (Dilute 5 mL of 0.1 g in 100 mL of absolute ethanol
		to 100 mL by absolute ethanol.)
5.	Index of refraction	$n_{\rm D}^{20} = 1.503 \sim 1.507.$
6.	Turbidity	: When 0.10 g of the sample is dissolved in 10 mL of ethanol, the solution should be
		clear.
7.	Arsenic	: Not more than 4 ppm (as As2O3).
8.	Heavy metals	: Not more than 20 ppm (as Pb).
9.	Category	: Food Additives Category 3, 8.
10.	Uses	: Antioxidants; Nutritional additives

### Propyl Gallate

Cher	nical formula: C <sub>10</sub> H <sub>12</sub>	O <sub>5</sub> Molecular weight: 212.20
1.	Assay	: 98.0~102.0%.
2.	Description	: White or creamy-white, crystalline odorless powder. Odorless, with little bitter taste.
		Slightly soluble in water; freely soluble in ethanol, ether and propane-1,2- diol.
3.	Identification	: (1) Dissolve 0.5 g of the sample in 10 mL of sodium hydroxide solution (1 in 25
		solution of sodium hydroxide). After distillation, take 4 mL of the initial distillation,
		which should be clear. The odor of propanol should be produced after heating.
		(2) Add 1 drop of ferric chloride solution (1 g of ferric chloride in 500mL of water) to 5
		mL of 1 in 50 ethanol solution of the sample. The solution is purple.
4.	Melting range	: 146~150°C (after drying at 105°C for 2 hr).
5.	Color of the	: Dissolve 0.5 g of the sample in 10 mL of ethanol, and the color of the solution shall not
	solution	be darker the Matching Fluid C.
6.	Chloride	: Not more than 0.028% (as Cl).
7.	Sulfate	: Not more than 0.048% (SO <sub>4</sub> ).
8.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Loss on drying	: Not more than 1.5% (105°C, 2 h)
11.	Residue on	: Not more than 0.10%.
	ignition	
12.	Category	: Food Additives Category 3
13.	Uses	: Antioxidants.

#### Guaiac Resin

1.	Description	: The resin from the wood of Guajacum officinale L., or of Guajacum sanctum L.
		Externally, it is brownish black to dusky brown, acquiring a greenish color on long
		exposure, the fractured surface having a glassy lustre, the thin pieces being transparent
		and varying in color from brown to yellowish orange; the powder is moderate yellow
		brown, becoming olive brown on exposure to air. It has a mild balsamic odor. Dissolves
		readily but incompletely in ethanol, ether, chloroform and solutions of alkalis; slightly
		soluble in carbon disulfide and benzene.
2.	Identification	: (1) Add 1 drop of ferric chloride TS to 5 mL of an ethanolic solution of the sample (1 in
		100). A blue color is produced which gradually changes to green, finally becoming
		greenish yellow.
		(2) A mixture of 5 mL of an ethanolic solution of the sample (1 in 100) and 5 mL of
		water becomes blue upon shaking with 20 mg of lead peroxide. Filter the solution, and
		boil a portion of the filtrate. The color disappears but may be restored by the addition of
		lead peroxide and shaking. Add a few drops of diluted hydrochloric acid TS to a second
		portion of the filtrate. The color is immediately discharged.
3.	Melting range	: 85~90°C.
4.	Alcohol-insoluble	: Not more than 15%.
	solid	
5.	Acid-insoluble ash	: Not more than 15%.
6.	Total ash	: Not more than 5%.
7.	Lead	: Not more than 10 ppm.
8.	Arsenic	: Not more than 3 ppm(as As).
9.	Heavy metals	: Not more than 40 ppm (as Pb).
10.	Category	: Food Additives Category 3
11.	Uses	: Antioxidants.

### L-Cysteine Monohydrochloride

Chemical names	L-2-Amino-3-mercaptopropanoic Acid Monohydrochloride
C.A.S. number	Monohydrate: 7048-04-6
	Anhydrous: 52-89-1
Chemical formula	Monohydrate: C <sub>3</sub> H <sub>7</sub> NO <sub>2</sub> S·HCl·H <sub>2</sub> O
	Anhydrous: C <sub>3</sub> H <sub>7</sub> NO <sub>2</sub> S·HCl
Structural formula	нз он · нсі · н <sub>2</sub> о
Formula weight	Monohydrate: 175.63
	Anhydrous: 157.62
Assay	98.0% - 101.5% C <sub>3</sub> H <sub>7</sub> NO <sub>2</sub> S·HCl, on the dried basis
Description	White, crystalline powder. It is freely soluble in water and in alcohol. The anhydrous
	form melts with decomposition at about 175°C.
Identification	The spectrum of the sample exhibits maxima at the same wavelengths as those in the
	spectrum of the Reference standard.
Lead	Not more than 5 mg/kg
Loss on drying	8.0%~12.0%
	Room temperature for 24 h in a vacuum desiccator using a suitable desiccant and
	maintaining a pressure of not more than 5 mmHg
Specific rotation	$[\alpha]_{\rm D}{}^{20}\!\!=\!\!+5.0^{\circ}\!\sim\!+8.0^{\circ}\!,$ calculated on the dried basis
	$[\alpha]_{D}{}^{25}\!\!=\!\!+4.9^{\circ}\sim\!+7.9^{\circ},$ calculated on the dried basis
Residue on ignition	Not more than 0.1%
Category	Food additives category (03) (07) (10)
Functional uses	Antioxidants; Food quality improvement, fermentation and food processing agents;
	Flavoring Agents.

### Tertiary-Butyl Hydroquinone

Chemical formula: C <sub>10</sub> H <sub>34</sub>		4O <sub>2</sub> Molecular weight: 166.22
1.	Assay	: Not less than 99.0%.
2.	Description	: White, crystalline solid having a characteristic odor. Practically insoluble in water;
		soluble in ethanol and ether.
3.	Identification	: Dissolve several mg of the sample in 1 mL of methanol, and add several drops of
		dimethylamine solution (1 in 4). A red to pink color is produced.
4.	Melting range	: 126.5~128.5°C.
5.	Toluene	: Not more than 25 ppm.
6.	Hydroxyquinone	: Not more than 0.1%.
7.	2,5-Di-t-butyl hydr	roquinone : Not more than 0.2%.
8.	t-Butyl-p-benzoquinone : Not more than 0.2%.	
9.	Arsenic	: Not more than 3 ppm(as As).
10.	Heavy metals	: Not more than 10 ppm (as Pb).
11.	Polycyclic	:
	aromatic	
	hydrocarbons	
	When me	easured at a wavelength of $280 \sim 289$ nm, its UV absorbance should be below 0.15.
	When me	easured at a wavelength of 290 $\sim$ 299 nm, its UV absorbance should be below 0.12.
	When me	easured at a wavelength of 300 to 359 nm, the UV absorbance should be below 0.08.
	When me	easured at a wavelength of $360 \sim 400$ nm, its UV absorbance should be below 0.02.
12.	Category	: Food Additives Category 3.
13.		

#### Calcium Ascorbate

Chemical formula: C <sub>12</sub> H <sub>14</sub> CaO <sub>12</sub> ·2H <sub>2</sub> O		CaO <sub>12</sub> ·2H <sub>2</sub> O Molecular weight: 426.25
1.	Assay	: $C_{12}H_{14}CaO_{12}$ ·2H <sub>2</sub> O Not less than 98.0%
2.	Description	: White to slightly yellow odorless crystalline powder. Soluble in water; slightly soluble
		in ethanol and insoluble in ether.
3.	Identification	: The 1 in 10 solution of the sample passes the test for calcium. A solution of the sample
		will decolorize a solution of dichlorophenol-indophenol TS.
4.	pН	: 6.8~7.4 (1 in 10 soln)
5.	Specific rotation	: $[\alpha] \frac{25}{D} = +95^{\circ} \sim +97^{\circ} (5\% \text{ (w/w) soln})$
6.	Oxalate	: Dissolve 1 g of the sample in 10 mL of water, add 2 drops of glacial acetic acid and
		5mL of calcium acetate solution (1 in 10 solution of calcium acetate). The solution will
		remain clear after placed for 5 minutes.
7.	Fluoride	: Not more than 10 ppm.
8.	Arsenic	: Not more than 3 ppm (as As).
9.	Heavy metals	: Not more than 10 ppm (as Pb).
10.	Category	: Food Additives Category 3, 8.
11.	Uses	: Antioxidants; Nutritional additives

- 1. Description : Mixed Tocopherol Concentrate contains 2 types: high- $\alpha$  type and low- $\alpha$  type. Bothe are obtained by the vacuum steam distillation of edible vegetable oil products, comprising concentrated tocopherols of different assay. High– $\alpha$  type contains relatively higher content of d-alpha-tocopherols, so it can be deemed as a form of Vitamin E and an antidoxant. Low– $\alpha$  type contains higher content of d- $\beta$ -, d-gamma-, d-delta-tocopherols and less of d-alpha-tocopherols, so it can only be considered as an antioxidant but not a form of Vitamin E. Both types may contain an edible vegetable oil added to adjust the required amount of total tocopherols, and the tocopherol forms may be adjusted by suitable physical and chemical means. Brownish red to red, clear, viscous oil having a mild, characteristic odor; may show a slight separation of wax like constituents in microcrystalline form. It oxidizes and darkens slowly in air and on exposure to light, particularly when in alkaline media. Insoluble in water; soluble in ethanol; miscible in acetone, chloroform, ether and vegetable oil. 2. Identification A. Dissolve about 50 mg of the sample in 10 mL of absolute ethanol. Add, with
  - A. Dissolve about 50 mg of the sample in 10 mL of absolute ethanol. Add, with swirling, 2 mL of nitric acid and heat at about 75°C for 15 min. A bright red to orange color develops.
    - B. High–α type: The retention time of the major peak in the chromatogram of the Assay Preparation is the same as that of the Standard Preparation, both relative to the internal standard.

Low– $\alpha$  type: The retention time of the third major peak (i.e. the peak occurring just before that of the internal standard) in the chromatogram of the Assay Preparation is the same as that of the Standard Preparation, both relative to the internal standard.

- 3. Assay : High-α type: Not less than 50.0% of total tocopherols. d-alpha-tocopherols (C29H50O2) should be not less than 50.0% of of total tocopherols. Total of d-β-, d-gamma- (C28H48O2), d-delta-tocopherols (C27H46O2) should not be less than 20.0% of total tocopherols.
  Low-α type: Not less than 50.0% of total tocopherols. Total of d-β-, d-gamma- (C28H48O2), d-delta-tocopherols (C27H46O2) should not be less than 80.0% of total tocopherols.
- 4. Acidity : Moderate.

5.	Lead	: Not more than 10 ppm.
6.	Heavy metals	: Not more than 0.004% (as Pb).
7.	Specific rotation	: $[\alpha] \frac{25}{D}$ : Not less than + 24° for high- $\alpha$ type; not less than + 20° for low- $\alpha$ type.
8.	Category	: High-a type: Food Additives Category 3, 8.
		Low-a type: Food Additives Category 3.
9.	Uses	: High-a type: Antioxidants; Nutritional additives
		Low-a type: Antioxidants

### d-a-Tocopherol Concentrate

1.	Description	: d-Alpha-Tocopherol, concentrate is a form of Vitamin E obtained by the vacuum steam
		distillation of edible vegetable oil products, comprising a concentrated form of d-alpha-
		tocopherol. It may contain an edible vegetable oil added to adjust the required amount of
		total tocopherols, and the content of d-alpha-tocopherol may be adjusted by suitable
		physical and chemical means.
		Brownish red, nearly odorless, clear viscous oil, which oxidizes and darkens slowly in air
		and on exposure to light.
		Insoluble in water; soluble in ethanol; miscible with ether, acetone, chloroform and
		vegetable oil.
2.	Identification	A. Dissolve about 50 mg of the sample in 10 mL of absolute ethanol. Add, with
		swirling, 2 mL of nitric acid and heat at about 75°C for 15 min. A bright red to
		orange color develops.
		B. The retention time of the major peak in the chromatogram of the sample
		solution is the same as that of the standard solution, both relative to the internal
		standard, as obtained in the assay preparation.
3.	Assay	: Not less than 40.0% of total tocopherols, of which not less than 95.0% consists of d-
		alpha-tocopherol ( $C_{29}H_{50}O_2$ ).
4.	Acidity	: Moderate.
5.	Lead	: Not more than 10 ppm.
6.	Heavy metals	: Not more than 0.004% (as Pb).
7.	Specific rotation	: Not less than ( $\alpha$ ) $\frac{25}{D}$ : +24°
8.	Category	: Food Additives Category 3, 8.
9.	Uses	: Antioxidants; Nutritional additives

### Disodium Ethylenediaminetetraacetate (EDTA Na<sub>2</sub>)

Chemical formula: C <sub>10</sub> H <sub>14</sub> N <sub>2</sub> Na <sub>2</sub> O <sub>8</sub> ·2H <sub>2</sub> O		N <sub>2</sub> Na <sub>2</sub> O <sub>8</sub> ·2H <sub>2</sub> O Molecular weight: 372.24
1.	Description	: White crystalline powder. Miscible with water.
2.	Identification	A. The 1 in 20 solution of this sample passes test for sodium.
		B. To 5 mL of water in a test tube add 2 drops of ammonium to thiocyanate TS and
		2 drops of ferric chloride TS. A deep red solution develops. Add about 50 mg of
		the sample and mix. The deep red color disappears.
3.	Assay	: $C_{10}H_{14}N_2Na_2O_8 \cdot 2H_2O$ not less than 99.0%.
4.	pH of 1% solution	: 4.3~4.7.
5.	Nitrilotriacetic acid	: Passes test
6.	Cyanide	: Not more than 1 ppm(as CN).
7.	Arsenic	: Not more than 3 ppm (as As).
8.	Lead	: Not more than 10 ppm.
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Category	: Food Additives Category 3, 7.
11.	Uses	: Antioxidants; Food quality improvement, fermentation and food processing agents

### Calcium Disodium Ethylenediaminetetraacetate (EDTA CaNa2)

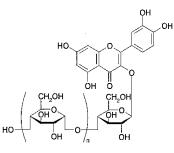
Chemical formula: C10H12CaN2Na2O8·2H2O		2CaN2Na2O8.2H2O Molecular weight: 410.30
1.	Description	: White, odorless crystalline granules or a white to nearly white powder; slightly
		hygroscopic and salty. Stable in air. Freely soluble in water.
2.	Identification	: A. The 1 in 20 solution of this sample passes test for calcium and test for sodium.
		B. To 5 mL of water in a test tube add 2 drops of ammonium thiocyanate TS and 2 drops
		of ferric chloride TS. A deep red solution develops. Add about 50 mg of the sample
		and mix. The deep red color disappears.
3.	Assay	: Not less than 97% and not more than the equivalent of 102% calculated on the
		anhydrous basis.
4.	рН	: 6.5~7.5 (1 in 100 soln)
5.	Water	: Not more than 13%.
6.	Magnesium chelat	ing : Transfer 1 g of the sample, accurately weighed, to a small beaker, and dissolve it in
	substances	5 mL of water. Add 5 mL of buffer solution prepared by dissolving 67.5 g of
		ammonium chloride in 200 mL of water, adding 570 mL of strong ammonia TS, and
		diluting with water to 1000 mL. To the buffered solution add 5 drops of eriochrome
		black TS, and titrate with 0.1 M magnesium acetate to the appearance of a deep
		wine-red color. Not more than 2.0 mL should be required.
7.	Arsenic	: Not more than 3 ppm (as As).
8.	Lead	: Not more than 10 ppm.
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Category	: Food Additives Category 3, 7.
11.		

### α– Glycosyl–isoquercitrin

Synonyms

### : Enzymatically modified isoquercitrin; isoquercetin; EMIQ

Molecular weight: Approximately 800



The number of glucose units may vary from 1 (n=0) to 11.

Definition	: Isoquercitrin is obtained by enzymatic hydrolysis of rutin. After mixed with starch or
	dextrin, it is reacted with cyclodextrin glucanotransferase.
Assay	: Not less than 60% of rutin ( $C_{27}H_{30}O_{16}$ ) on the dried basis.
Appearance	: Yellow to orange powder, lumps or mush, with a slightly special odor.
Identification	: (1) Dissolve 5 mg of the sample in 10 mL of water and add 1 to 2 drops of 1 in 50
	ferric chloride solution. A dark brown color appears.
	(2) Dissolve 5 mg of the sample in 5 mL of water, add 2 mL of hydrochloric acid and
	0.05 g of magnesium powder. The solution is orange to red.
	(3) Dissolve 0.1g of the sample in 100 mL of 1 N sulfuric acid, boil for 2 hours.
	Yellow precipitates appear after cooling.
	(4) Spectrophotometry: Dissolve 10 mg of the sample in 500 mL of 1 in 1,000
	phosphoric acid solution. The solution has the maximum absorption value around the
	wavelength of 255 nm and 350 nm.
	(5) TLC: Analyze the test solution and control solution(rutin for assay)in methanol by
	TLC. Several brown spots are observed: one having an Rf value greater than that of
	the main spot of rutin for assay and others having Rf values the same as or smaller
	than that of the main spot of rutin for assay.
Quercetin	: Not more than 1%.
Lead	: Not more than 2 mg/kg
Arsenic	: Not more than 1.5 mg/kg
Loss on drying	: Not more than 50% (135°C, 2 h).
Category	: Food Additives Category 3
Functional uses	: Antioxidants

### **Extracts of Rosemary**

	Extracts of Rosemary
SYNONYMS	Extracts of Rosemary Leaf (Antioxidant)
DEFINITION	Extracts of rosemary contain several components, which have been
	proven to exert antioxidative functions. These components belong
	mainly to the classes of phenolic acids, flavonoids, diterpenoids.
	Besides the antioxidant compounds, the extracts can also contain
	triterpenes and organic solvent extractable material specifically
	defined in the following specification. Rosemary leaf extract
	antioxidant is prepared by extraction of the leaves of Rosmarinus
	officinalis using a food approved solvent system. Extracts may then
	be deodorized, decolorized and standardized.
Chemical names	Rosemary extract (Rosmarinus officinalis)
CHARACTERISTICS	
IDENTIFICATION	
Reference antioxidative	: Carnosic acid( $C_{20}H_{28}O_4$ ) and Carnosol ( $C_{20}H_{26}O_4$ ) (which comprise
compounds: phenolic	not less than 90% of the total phenolic diterpenes)
diterpenes	Acetone extraction: not less than 10% w/w, expressed as the total of
	carnosic acid and carnosol
	Supercritical carbon dioxide extraction: not less than 13% w/w,
	expressed as the total of carnosic acid and carnosol
	Deodorized ethanolic extract: not less than 5% w/w, expressed as the
	total of carnosic acid and carnosol
	Two-step extraction using hexane and ethanol: not less than 5% w/w,
	expressed as the total of carnosic acid and carnosol
Antioxidants/Volatiles	Total % of carnosic acid and carnosol / Total % of reference
Ratio	volatiles: (-)-borneol, (-)bornyl acetate, (-)-camphor, 1,8-Cineole
	(eucalyptol) and verbenone: not less than 15
Density	not less than 0.25 g/mL
Solubility	Insoluble in water
Residual solvents	Acetone extraction: Acetone: not more than 500 mg/kg
	Supercritical carbon dioxide extraction: Ethanol: not more than 2%
	Deodorized ethanolic extract: Ethanol: not more than 500 mg/kg
	Two-step extraction using hexane and ethanol: Hexane: not more than
	25 mg/kg, Ethanol: not more than 500 mg/kg

45

### Purity

Loss on drying	: Not more than 5%
Arsenic	: Not more than 3 mg/kg
Lead	: Not more than 2 mg/kg
Category	: Food additives category (03)
Functional uses	: Antioxidants

# Category 4 Bleaching AgentsColor Fasting Agents

### § 04001

### Potassium Sulfite

Chemical formula: K <sub>2</sub> SO <sub>3</sub>		Molecular weight: 158.25
1.	Assay	: Not less than 90.0%.
2.	Description	: White, odorless, granular powder. Easily oxidize in air. Freely soluble in water; slightly
		soluble in ethanol.
3.	Identification	: The solution of this sample (1 in 20) passes test for potassium and test for sulfite.
4.	Alkalinity	: Between 0.25 and 0.45% as K <sub>2</sub> CO <sub>3</sub> .
5.	Selenium	: Not more than 30 ppm.
6.	Arsenic	: Not more than 4 ppm (as As2O3).
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 4
9.	Uses	: Bleaching agents.

### Sodium Sulfite

Cher	nical formula: Na <sub>2</sub> SC	D <sub>3</sub> ·7H <sub>2</sub> O Molecular weight: 252.16
1.	Assay	: Not less than 95% on the dried basis.
2.	Description	: White or colorless, odorless powder. Easily oxidizes in air. Freely soluble in water;
		sparingly soluble in ethanol.
3.	Identification	: The solution of this sample (1 in 20) passes test for sodium and test for sulfite.
4.	Solution	: Dissolve 0.5g of the sample in 10 mL of water. The solution should be "almost clear"
		and alkaline.
5.	Selenium	: Not more than 30 ppm (calculated as Se on the anhydrous basis)
6.	Arsenic	: Not more than 4 ppm (calculated as As <sub>2</sub> O <sub>3</sub> on the anhydrous basis)
7.	Heavy metals	: Not more than 10 ppm (calculated as Pb on the anhydrous basis)
8.	Category	: Food Additives Category 4
9.	Uses	: Bleaching agents.

### Sodium Sulfite, Anhydrous

Chemical formula: Na <sub>2</sub> SO <sub>3</sub>		Molecular weight: 126.04
1.	Assay	: Not less than 95%
2.	Description	: Colorless, white or light pink and odorless. Easily oxidizes in air. Freely soluble in
		water; sparingly soluble in ethanol.
3.	Identification	: The solution of this sample (1 in 20) passes test for sodium and test for sulfite.
4.	Solution	: Dissolve 0.5 g of the sample in 10 mL of water. The solution should be less than
		"almost clear" and colorless.
5.	Selenium	: Not more than 30 ppm.
6.	Arsenic	: Not more than 4 ppm (as As2O3).
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 4
9.	Uses	: Bleaching agents.

### Sodium Bisulfite

1.	Assay	: Between 58.5 to 67.4% of SO <sub>2</sub> .	
2.	Description	: A mixture of $NaHSO_3$ and $Na_2S_2O_5$ . White or yellowish-white crystals or granular	
		powder having an odor of sulfur dioxide. Unstable in air. Freely soluble in water; slightly	
		soluble in ethanol.	
3.	Identification	: The solution of this sample (1 in 10) passes test for sodium and test for sulfite.	
4.	Iron	: Not more than 50 ppm.	
5.	Selenium	: Not more than 30 ppm.	
6.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).	
7.	Heavy metals	: Not more than 10 ppm (as Pb).	
8.	Category	: Food Additives Category 4	
9.	Uses	: Bleaching agents.	

### Sodium Hydrosulfite

Chemical formula: Na <sub>2</sub> S <sub>2</sub> C		D <sub>4</sub> Molecular weight: 174.11		
1.	Assay	: Not less than 85.0%.		
2.	Description	: White or off-white crystals or granular powder, odorless or having an odor of sulfur		
		dioxide.		
3.	Identification	: (1) Add 2 mL of 1 in 20 copper sulfate to 10 mL of 1 in 100 solution of the sample. The		
		solution is grayish black.		
		(2) Add 10 mL of potassium permanganate TS (1 g of potassium permanganate dissolved		
		in 300 mL of water) to 10 mL of the 1 in 100 solution of the sample, and the		
		color of the solution disappears immediately.		
		(3) Passes test for sodium.		
4.	Solution	: Add 10 mL of water to 10 mL of formaldehyde solution, neutralize it with 1 in 25		
		sodium hydroxide solution. Dissolve 0.5 g of the sample in 10 Ml of the previous		
		solution, and then put it for 5 minutes. Its turbidity should be below "slightly turbid".		
5.	Sodium diamine vi	nyl : Dissolve 0.5 g of the sample in 5 mL of water, add 2 mL of 1 in 200 solution		
	acetate	of potassium chromate and 2 mL of arsenite acid TS. After heated in a		
		water bath for 2 minutes, color purple does not appear.		
6.	Formate	: Not more than 0.05% of HCHO.		
7.	Zinc	: Not more than 80 ppm.		
8.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).		
9.	Heavy metals	: Not more than 10 ppm (as Pb).		
10.	Category	: Food Additives Category 4		
11.	Uses	: Bleaching agents.		

### Potassium Metabisulfite

Chemical formula: K <sub>2</sub> S <sub>2</sub> O <sub>5</sub>		Molecular weight: 222.33	
1.	Assay	: Not less than 93.0%.	
2.	Description	: Colorless free-flowing crystals, crystalline powder, or granules, usually having an odor	
		of sulfur dioxide. Easily oxidizes in air. Soluble in water; insoluble in ethanol.	
3.	Identification	: The solution of this sample (1 in 10) passes test for potassium and test for sulfite.	
4.	Solution	: Dissolve 1 g of the sample in 10 mL of water. The solution should be clear with only a	
		trace of suspended matter while the solution being acid.	
5.	Iron	: Not more than 10 ppm.	
6.	Selenium	: Not more than 30 ppm.	
7.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
8.	Heavy metals	: Not more than 10 ppm (as Pb).	
9.	Category	: Food Additives Category 4	
10.	Uses	: Bleaching agents.	

### Sodium Metabisulfite

Chemical formula: Na <sub>2</sub> S <sub>2</sub> O		D <sub>5</sub> Molecular weight: 190.11	
1.	Assay	: Not less than 93.0%.	
2.	Appearance	: White or light yellow crystals or crystalline powder having an odor of sulfur dioxide.	
		Freely soluble in water; slightly soluble in ethanol.	
3.	Identification	: The solution of this sample (1 in 20) passes test for sodium and test for sulfite.	
4.	Solution	: Dissolve 0.5g of the sample in 10 mL of water. The solution should be "almost clear"	
		and acidic.	
5.	Iron	: Not more than 20 ppm.	
6.	Selenium	: Not more than 30 ppm.	
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).	
8.	Lead	: Not more than 10 ppm.	
9.	Heavy metals	: Not more than 10 ppm (as Pb).	
10.	Category	: Food Additives Category 4	
11.	Uses	: Bleaching agents.	

§ 07079

# Benzoyl Peroxide

Chemical formula: C <sub>14</sub> H <sub>10</sub> O <sub>4</sub>		O <sub>4</sub> Molecular weight: 242.23
1.	Assay	: Not less than 96.0%.
2.	Description	: Colorless, crystalline solid having a faint odor of benzaldehyde. Insoluble in water,
		slightly soluble in ethanol and soluble in benzene, chloroform and ether. 1 g of this
		product can be dissolved in 40 mL of carbon disulfide. Melts between 103°C and 106°C
		with decomposition. (Caution: Benzoyl peroxide, especially in the dry form, is a
		dangerous, highly reactive, oxidizing material and has been known to explode
spontaneously)		spontaneously)
3.	Identification	: To 500 mg of the sample add 50 mL of 0.5 N ethanolic potassium hydroxide, heat
		gradually to boiling and continue boiling for 15 min. Cool and dilute with 200 mL of
		water. Add sufficient 0.5 N hydrochloric acid to make strongly acidic and extract with
		ether. Dry the ether solution over anhydrous sodium sulfate, and then evaporate to
		dryness on a steam bath. The benzoic acid so obtained melts between 121°C and 123°C.
4.	Arsenic	: Not more than 3 ppm (as As).
5.	Heavy metals	: Not more than 0.004% (as Pb).
6.	Lead	: Not more than 10 ppm.
7.	Category	: Food Additives Category 4
8.	Uses	: Bleaching agents.

# Category 5 Color Fasting Agents

### § 05001

### Potassium Nitrite

Chemical formula: KNO <sub>2</sub>		Molecular weight: 85.10	
1.	Assay	: Not less than 90.0% on the dried basis	
2.	Description	: White or slightly yellow, deliquescent granules or rods. Freely soluble in water,	
		sparingly soluble in ethanol. The 1 in 10 solution is alkaline on litmus paper.	
3.	Identification	: The solution of this sample (1 in 10) passes test for potassium and test for nitrite.	
4.	Arsenic	: Not more than 4 ppm (as As2O3).	
5.	Lead	: Not more than 10 ppm.	
6.	Heavy metals	: Not more than 20 ppm (as Pb).	
7.	Category	: Food Additives Category 5	
8.	Uses	: Color Fasting Agents.	

### Sodium Nitrite

Chemical formula: NaNO <sub>2</sub>		Molecular weight: 69.00	
1.	Assay	: Not less than 97.0% on the dried basis.	
2.	Description	: Clear, colorless, odorless, transparent crystals, or white granules or powder;	
		deliquescent in air. The solution is alkaline on litmus paper. Freely soluble in water;	
		slightly soluble in ethanol.	
3.	Identification	: Passes test for sodium and test for nitrate.	
4.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
5.	Lead	: Not more than 10 ppm.	
6.	Heavy metals	: Not more than 20 ppm (as Pb).	
7.	Loss on drying	: Not more than 3.0% (100°C, 5 h).	
8.	Category	: Food Additives Category 5	
9.	Uses	: Color Fasting Agents.	

### Potassium Nitrite

Chemical formula: KNO <sub>3</sub>		Molecular weight: 101.10	
1.	Assay	: Not less than 99.0% on the dried basis.	
2.	Description	: White, odorless, transparent prisms, or white granular or crystalline powder, salty and	
		having a cool taste. Slightly deliquescent in moist air. The aqueous solution was neutral	
		on litmus paper. 1 g of the sample is soluble in 3 mL of water at 25°C, 0.5 mL of boiling	
		water, 620 mL, or in ethanol.	
3.	Identification	: The solution of this sample (1 in 10) passes test for potassium and test for nitrite.	
4.	Chloride	: Not more than 0.021% (as Cl).	
5.	Arsenic	: Not more than 4 ppm (as As2O3).	
6.	Lead	: Not more than 10 ppm.	
7.	Heavy metals	: Not more than 20 ppm (as Pb).	
8.	Loss on drying	: Not more than 1.0% (105°C, 4 h).	
9.	Category	: Food Additives Category 5	
10.	Uses	: Color Fasting Agents.	

### Sodium Nitrite

Chemical formula: NaNO <sub>3</sub>		Molecular weight: 84.99	
1.	Assay	: Not less than 99.0% on the dried basis.	
2.	Description	: White crystal or white crystalline powder, odorless, salty. Deliquescent in moist air.	
		The aqueous solution was neutral on litmus paper. Soluble in water; sparingly soluble	
		in ethanol.	
3.	Identification	: The solution of this sample (1 in 5) passes test for potassium and test for nitrite.	
4.	Chloride	: Not more than 0.02% (as Cl).	
5.	Arsenic	: Not more than 4 ppm (as As2O3).	
6.	Heavy metals	: Not more than 20 ppm (as Pb).	
7.	Loss on drying	: Not more than 1.0% (105°C, 4 h).	
8.	Category	: Food Additives Category 5	
9.	Uses	: Color Fasting Agents.	

### Category 6 Leavening agents

# § 06001

#### Potassium Alum

### (Aluminum Potassium Sulfate)

Chemical formula: AlK(SO <sub>4</sub> ) <sub>2</sub> ·12H <sub>2</sub> O		O <sub>4</sub> ) <sub>2</sub> ·12H <sub>2</sub> O Molecular weight: 474.38	
1.	Assay	: Not less than 99.5% of AlK(SO <sub>4</sub> ) <sub>2</sub> ·12H <sub>2</sub> O.	
2.	Description	: Colorless, transparent crystals or crystalline fragments, or white crystall	line powder;
		odorless and slightly sweet and astringent. 1 g of the sample is soluble in	7.5 mL of 25°C
		of water or 0.3 mL of boiling water, soluble in glycerin; insoluble in etha	nol. The
		aqueous solution is acidic on litmus paper.	
3.	Identification	: The solution of this sample (1 in 20) passes test for potassium, test for a	luminum and
		test for nitrite.	
4.	Solution	: Dissolve 1 g of the sample in 10 mL of water. The solution should be "almost clear"	
		and colorless.	
5.	Fluoride	: Not more than 30 ppm.	
6.	Ammonium salt	: Heat 1 g of the sample with 10 mL of sodium hydroxide TS on a steam bath for 1 min.	
		The odor of ammonia is not perceptible.	
7.	Iron	: Not more than 190 ppm (as Fe on anhydrous basis)	
8.	Lead	: Not more than 10 ppm.	
9.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> on anhydrous basis)	
10.	Selenium	: Not more than 30 ppm.	
11.	Heavy metals	: Not more than 40 ppm (as Pb on anhydrous basis)	
12.	Category	: Food Additives Category 6	
13.	Uses	: Leavening agents	

#### Sodium Alum

Chemical formula: AlNa(SO <sub>4</sub> ) <sub>2</sub> ·12H <sub>2</sub> O		SO <sub>4</sub> ) <sub>2</sub> ·12H <sub>2</sub> O Molecular weight: 458.26		
1.	Assay	: Not less than 99.5% of $AlNa(SO_4)_2$ on the dried basis.		
2.	Description	: Colorless crystals, white particles or powder. Odorless, slightly salty and astringent.		
		Soluble in water; insoluble in ethanol.		
3.	Identification	: Passes test for aluminum, test for sulfate and sodium flame test.		
4.	Fluoride	: Not more than 30 ppm (Fluoride, on anhydrous basis)		
5.	Ammonium salt	: Heat 1 g of the sample with 10 mL of sodium hydroxide TS on a steam bath for 1 min.		
		The odor of ammonia is not perceptible.		
6.	Selenium	: Not more than 30 ppm (Se, on anhydrous basis)		
7.	Lead	: Not more than 10 ppm.		
8.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> , on anhydrous basis)		
9.	Heavy metals	: Not more than 20 ppm (as Pb, on anhydrous basis)		
10.	Loss on drying	: Not more than 47.2%.		
11.	Category	: Food Additives Category 6		
12.	Uses	: Leavening agents		

#### Burnt Potassium Alum

### (Aluminum Potassium Sulfate, Anhydrous)

Chemical formula: AlK(SO <sub>4</sub> ) <sub>2</sub> ·12H <sub>2</sub> O			Molecular weight: 258.21
1.	Assay	: Not less than 96.5% on the drie	d basis (200°C,4 h)
2.	Description	: Large, transparent crystals or cr	rystalline fragments, or white crystalline powder;
		odorless with a slightly sweet an	d astringent taste.
3.	Identification	: The solution of the sample (1 ir	n 20) passes test for potassium, test for aluminum and
		test for sulfate.	
4.	Solution	: Dissolve 1 g of the sample in 10	0 mL of water. The solution should be "almost clear"
		and colorless.	
5.	Water-insoluble	: Not more than 2%.	
	matter		
6.	Fluoride	: Not more than 30 ppm.	
7.	Iron	: Not more than 190 ppm (as Fe)	
8.	Lead	: Not more than 10 ppm.	
9.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub>	3).
10.	Selenium	: Not more than 30 ppm.	
11.	Heavy metals	: Not more than 40 ppm (as Pb).	
12.	Category	: Food Additives Category 6	
13.	Uses	: Leavening agents	

#### Ammonium Alum

### (Aluminum Ammonium Sulfate)

Chemical formula: AlNH4(SO4)2·12H2O		(SO <sub>4</sub> ) <sub>2</sub> ·12H <sub>2</sub> O	Molecular weight: 453.32
1.	Assay	: Not less than 99.5% of AlNH <sub>4</sub> (SO <sub>4</sub> ) <sub>2</sub> 12H <sub>2</sub> O.	
2.	Description	: Large, colorless crystals, white granules, or a powder; odorless. Sweet and very	
		astringent. 1 g of the sample is	soluble in 7.0 mL of 25°C of water or 0.3 mL of boiling
		water, slowly soluble in glycer	in; insoluble in ethanol. The aqueous solution is acidic on
		litmus paper.	
3.	Identification	: The solution of the sample (1	in 20) passes test for aluminum, test for ammonium and
		test for sulfate.	
4.	Solution	: Dissolve 1 g of the sample in	10 mL of water. The solution should be "almost clear"
		and colorless.	
5.	Fluoride	: Not more than 30 ppm.	
6.	Alkalis and alkaline	aline earths : Completely precipitate the aluminum from a boiling solution of 1 g of the	
		sample in 100 mL c	f water by the addition of enough ammonia TS to render
		the solution distinct	ly alkaline to methyl red TS, and filter with 10 mL of hot
		water. Evaporate th	e filtrate to dryness, and ignite. The weight of the residue
		does not exceed 5 n	ng.
7.	Iron	: Not more than 190 ppm of (a	s Fe, on anhydrous basis)
8.	Lead	: Not more than 10 ppm.	
9.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> , on anhydrous basis)	
10.	Selenium	: Not more than 30 ppm.	
11.	Heavy metals	: Not more than 40 ppm (as Pb, on anhydrous basis)	
12.	Category	: Food Additives Category 6	
13.	Uses	: Leavening agents	

#### Burnt Ammonium Alum

# (Aluminum Ammonium Sulfate, Anhydrous)

Chemical formula: AlNH <sub>4</sub> (SO <sub>4</sub> ) <sub>2</sub>		(SO <sub>4</sub> ) <sub>2</sub> Molecular weight: 237.15	
1.	Assay	: Not less than 96.5% on the dried basis (200°C, 4 h)	
2.	Description	: White powder or porous mass, odorless, slightly sweet, and with a strong astringent	
		taste.	
3.	Identification	: The 1 in 20 solution of the sample passes test for ammonium, test for aluminum and test	
		for sulfate.	
4.	Solution	: Dissolve 1 g of the sample in 10 mL of water. The solution should be "almost clear"	
		and colorless.	
5.	Water-insoluble	: Not more than 2%.	
	matter		
6.	Fluoride	: Not more than 30 ppm.	
7.	Iron	: Not more than 190 ppm (as Fe).	
8.	Lead	: Not more than 10 ppm.	
9.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).	
10.	Selenium	: Not more than 30 ppm.	
11.	Heavy metals	: Not more than 40 ppm (as Pb).	
12.	Category	: Food Additives Category 6	
13.	Uses	: Leavening agents	

### Ammonium Chloride

Chemical formula: NH <sub>4</sub> Cl		Molecular weight: 53.49	
1.	Assay	: Not less than 99.0% on the dried basis.	
2.	Description	: Colorless crystal or white crystalline powder or crystallized masses, salty and cool.	
		Slightly deliquescent. 1 g of the sample is soluble in 100 mL of alcohol, 8 mL of	
		glycerin, 2.6 mL of water at 25°C or 1.4 mL of boiling water.	
3.	Identification	: The 1 in 10 solution of the sample passes test for ammonium and test for chloride.	
4.	Solution	: Dissolve 2 g of the sample in 20 mL of water. The solution should be "almost clear".	
5.	pH	: 4.5~6.0 (1 in 20 soln)	
6.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
7.	Heavy metals	: Not more than 20 ppm (as Pb).	
8.	Loss on drying	: Not more than 2.0% (Silicone dryer, 4 hr).	
9.	Residue on	: Not more than 0.5%.	
	ignition		
10.	Category	: Food Additives Category 6	
11.	Uses	: Leavening agents	

#### Potassium dl-Bitartrate

Chemical formula: C <sub>4</sub> H <sub>5</sub> O <sub>6</sub> K		6K Molecular weight: 188.18
1.	Assay	: Not less than 98.5% on the dried basis (105°C, 3 h)
2.	Appearance	: Colorless crystal or white crystalline powder, with a cool and sour odor.
3.	Solution	: Dissolve 0.5 g of the sample in 3 mL of water. The solution should be less than "almost
		clear" and colorless.
4.	Sulfate	: Not more than $0.02\%$ of SO <sub>4.</sub>
5.	Ammonium salt	: Heat 0.5 g of the sample with 5 mL of sodium hydroxide TS. The odor of ammonia is
		not perceptible.
6.	Calcium	: Test for calcium of "d-potassium tartrate" is applicable.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Readily	: Dissolve 2 g of the sample in 20 mL of water 30 mL of dilute sulphuric acid. When 4
	oxidizable matter	mL of 0.1 N potassium permanganate solution is added at 20°C, the solution color
		should not disappear within 3 minutes.
9.	Loss on drying	: Not more than 0.5% (105°C, 3 h).
10.	Category	: Food Additives Category 6
11.	Uses	: Leavening agents

### Potassium d-Bitartrate

Chemical formula: C <sub>4</sub> H <sub>5</sub> O <sub>6</sub> K		D <sub>6</sub> K Molecular weight: 188.18	
1.	Assay	: Not less than 98.5% on the dried basis (105°C, 3 h)	
2.	Appearance	: Colorless crystal or white crystalline powder, having a cool and sour odor.	
3.	Solution	: Dissolve 0.5 g of the sample in 3 mL of water. The solution should be less than "almost	
		clear" and colorless.	
4.	Specific rotation	: $[\alpha] \frac{20}{D} = +32.5^{\circ} \sim +35.5^{\circ}$ [dried at 105°C for 3hours (Add 10 mL of ammonia TS	
		and water to 5 g of the sample to make a 50 mL solution) ]	
5.	Sulfate	: Not more than 0.02% of SO <sub>4</sub> .	
6.	Ammonium salt	: Heat 0.5 g of the sample with 5 mL of sodium hydroxide TS. The odor of ammonia is	
		not perceptible.	
7.	Heavy metals	: Not more than 20 ppm (as Pb).	
8.	Calcium	: Add 1 g of the sample in 5 mL of acetic acid and vibrated constantly. After placed for	
		30 minutes, add 25 mL of water. After standing, the solution is filtered with absorbent	
		cotton. The filtrate should not be turbid when 8 drops of ammonium oxalate TS are	
		added.	
9.	Loss on drying	: Not more than 0.5% (105°C, 3 h).	
10.	Category	: Food Additives Category 6	
11.	Uses	: Leavening agents	

#### Sodium Bicarbonate

Chemical formula: NaHCO <sub>3</sub>		O <sub>3</sub> Molecular weight: 84.01	
1.	Assay	: Not less than 99.0% on the dried basis.	
2.	Description	: Colorless, white, crystalline masses or crystalline powder. Stable in dry air and slowly	
		discompose in moist air. The freshly prepared cold solution of the sample is alkaline on	
		litmus paper before shaking. When the solution is placed, stirred or heated, alkalinity	
		increases. 1 g of the sample is soluble in 10 mL of water, but not in ethanol.	
3.	Identification	: The 1 in 10 solution of the sample passes test for sodium and test for carbonate.	
4.	Solution	: When 1 g of the sample is dissolved in 20 mL of water, the solution should be clear.	
5.	Chloride	: Not more than 200 ppm (as Cl).	
6.	Carbonate	: Carefully add 20 mL of freshly boiled and cooled water to 1 g of the sample. After	
	shaking at a temperature below 15°C, add 2.0 mL of 0.1 N hydrochloric acid		
		add 2 drops of phenolphthalein TS, no pink color should appear immediately.	
7.	Ammonium salt	: Heat 1 g of the sample in the tube. The odor of ammonia is not perceptible.	
8.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
9.	Heavy metals	: Not more than 10 ppm (as Pb).	
10.	Loss on drying	: Not more then 0.25% (Silicone dryer, 4 hr).	
11.	Category	: Food Additives Category 6	
12.	Uses	: Leavening agents	

#### Ammonium Carbonate

1.	Assay	: Not less than 30.0% and not more than 34.0% of $NH_3$ .	
2.	Description	: Consists of $NH_4HCO_3$ and $NH_2COONH_4$ in varying proportions. White powder or hard,	
		white or translucent masses of crystals with an odor of ammonia. On exposure to air it	
		becomes opaque and is finally converted into white porous lumps or powder (of	
		ammonium bicarbonate) due to loss of ammonia and carbon dioxide. 1 g of the sample is	
		slowly soluble in 4 mL of water.	
3.	Identification	: (1) When heated, it volatilizes without charring and the vapor is alkaline to moist	
		litmus.	
		(2) Dissolve 1 g of the sample in 20 mL of water, and bubbles will be formed when	
		adding acid.	
		(3) Passes test for ammonia and test for carbonate.	
4.	Solution	: Dissolve 2.0 g of the sample in 20 mL of water. The solution should be "almost clear".	
5.	Chloride	: Not more than 40 ppm (as Cl).	
6.	Sulfide	: Not more than 50 ppm.	
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).	
8.	Heavy metals	: Not more than 10 ppm (as Pb).	
9.	Residue on	: Not more than 0.01%.	
	ignition		
10.	Non-volatile residu	e : Not more than 0.05%.	
11.	Category	: Food Additives Category 6, 7.	
12.	Uses	: Leavening Agents; Food quality improvement, fermentation and food processing	

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#### Ammonium Bicarbonate

Chemical formula: NH4HCO3		CO <sub>3</sub> Molecular weight: 79.06	
1.	Assay	: Not less than 99.0%.	
2.	Description	: White crystals or a crystalline powder with a slight odor of ammonia. At 60°C or above,	
		it evaporates quickly and decomposes into ammonia, carbon dioxide and water. Stable at	
		room temperature. 1 g of the sample can be dissolved in 6 mL of water; insoluble in	
		ethanol.	
3.	Identification	: Passes test for ammonium and test for carbonate.	
4.	Solution	: Dissolve 2.0 g of the sample in 20 mL of water. The solution should be "almost clear".	
5.	Chloride	: Not more than 40 ppm (as Cl).	
6.	Sulfates	: Not more than 70 ppm.	
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).	
8.	Heavy metals	: Not more than 10 ppm (as Pb).	
9.	Residue on	: Not more than 0.01%.	
	ignition		
10.	Non-volatile residu	e : Not more than 0.05%.	
11.	Category	: Food Additives Category 6	
12.	Uses	: Leavening agents	

#### Potassium Carbonate

Chemical formula: K <sub>2</sub> CO <sub>3</sub>		Molecular weight: 138.21	
1.	Assay	: Not less than 99.0% on the dried basis.	
2.	Description	: This product includes anhydrous form and hydrated form containing 1.5 molecules of	
		water. Anhydrous form is white granular powder, hydrated form is small white	
		transparent crystal or particles. Odorless, with strong alkali taste, easily deliquescent. Its	
		aqueous solution is alkaline.1 g of the sample is soluble in 1 mL of 25 $^{\circ}$ C water or 0.7	
		mL of boiling water, but insoluble in alcohol.	
3.	Identification	: The 1 in 10 solution of the sample passes test for potassium and test for carbonate.	
4.	Solution	: Dissolve 1 g of the sample in 20 mL of water. The solution should be "almost clear"	
		and colorless.	
5.	Insoluble matter	: Dissolve 1 g of the sample in 20 mL of water, no residue should appear.	
6.	Chloride	: Not more than 530 ppm (as Cl).	
7.	Lead	: Not more than 10 ppm.	
8.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
9.	Heavy metals	: Not more than 20 ppm (as Pb).	
10.	Loss on drying	: Anhydrous: Not more than 1%	
		Hydrated forms: Between 10% and 16.5% (180°C, 4 h)	
11.	Category	: Food Additives Category 6, 7 and 14.	
12.	Uses	: Leavening agents; Food quality improvement, fermentation and food processing	
		agentss; chemicals for food industry.	

### Baking Powder

1.	Appearance	: White to off-white powder or fragile powder.	
2.	Solution	: Dissolve 1 g of the sample in 50 mL of water, heat in water bath until no foam appears,	
		then cool and test the pH. pH of the one dosage type and two dosage type is $5.0 \sim 8.5$ ,	
		and the pH of ammonium dosage type is $6.0 \sim 9.0$ .	
3.	Nitric acid	: Not more than 2%.	
	insoluble matter		
4.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
5.	Heavy metals	: Not more than 40 ppm (as Pb).	
6.	Gas relief	: 2 g of the sample shall produce more than 70 mL of gas.	
7.	Category	: Food Additives Category 6	
8.	Uses	: Leavening agents	

### Sodium Aluminum Phosphate, Acidic

Chemical formula: $NaAl_3H_{14}(PO_4)_8$ ·4H <sub>2</sub> O or			Molecular weight: 949.88 or
	Na <sub>3</sub> Al <sub>2</sub> H <sub>15</sub> (PO <sub>4</sub> ) <sub>8</sub>		Molecular weight: 897.82
1.	Assay	: Not less than 95.0% of NaAl <sub>3</sub> H <sub>14</sub> (PO <sub>4</sub> ) <sub>8</sub> ·4H <sub>2</sub> O	
		Not less than 95.0% of $Na_3Al_2H_{15}(PO_4)_8$	
2.	Description	: White, odorless powder. Inso	luble in water; soluble in hydrochloric acid.
3.	Identification	: Passes test for aluminum, test for sodium and test for phosphate. Test a 1 in 10 solution	
		in dilute hydrochloric acid (1 in 2).	
4.	Fluoride	: Not more than 25 ppm.	
5.	Lead	: Not more than 10 ppm.	
6.	Arsenic	: Not more than 3 ppm (as As).	
7.	Heavy metals	: Not more than 40 ppm (as Pb).	
8.	Residue on	: NaAl <sub>3</sub> H <sub>14</sub> (PO <sub>4</sub> ) <sub>8</sub> ·4H <sub>2</sub> O 19.5%~21%.	
	ignition	Na3Al2H15(PO4)8 15%~16%	
9.	Category	: Food Additives Category 6	
10.	Uses	: Leavening agents	

#### Burnt Sodium Alum

Chemical formula:	$AlNa(SO_4)_2$
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Molecular weight: 242.09

1.	Assay	: Not less than 96.5%
2.	Description	: Colorless crystals, white particles or powder. Odorless, slightly salty and astringent.
		Soluble slowly in water but insoluble in alcohol.
3.	Identification	: Passes sodium flame test, test for aluminum and test for sulfate.
4.	Fluoride	: Not more than 30 ppm(as F).
5.	Ammonium salt	: Heat 1 g of the sample with 10 mL of sodium hydroxide TS on a steam bath for 1 min.
		The odor of ammonia is not perceptible.
6.	Selenium	: Not more than 30 ppm(as Se)
7.	Lead	: Not more than 10 ppm.
8.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Loss on drying	: Not more than 10%.
11.	Category	: Food Additives Category 6
12.	Uses	: Leavening agents

Category 7 Food quality improvement, fermentation and food processing agentsFood quality improvement,

fermentation and food processing agents

§ 07001

Calcium Chloride		
Synonyms	INS No. 509	
Definition		
Chemical names	Calcium chloride	
CAS number	10043-52-4	
Chemical formula	Anhydrous: CaCl <sub>2</sub>	
	Dehydrate: CaCl <sub>2</sub> · 2H <sub>2</sub> O	
	Hexahydrate: CaCl <sub>2</sub> · 6H <sub>2</sub> O	
Molecular weight	Anhydrous: 110.99	
	Dehydrate: 147.02	
	Hexahydrate: 219.08	
Assay	Anhydrous: Not less than 93%	
	Dehydrate: Not less than 99.0% and not more than the equivalent of 107.0% of	
	CaCl <sub>2</sub> ·2H <sub>2</sub> O	
	Hexahydrate: Not less than 98.0% and not more than the equivalent of 110% of	
	CaCl <sub>2</sub> ·6H <sub>2</sub> O	
Description	Anhydrous: White, deliquescent lumps or porous masses	
	Dehydrate: White, hard, deliquescent fragments or granules Hexahydrate:	
	Colorless, very deliquescent crystals	
Characteristics		
Identification		
Solubility	Anhydrous: Freely soluble in water and ethanol	
	Dehydrate: Freely soluble in water; soluble in ethanol	
	Hexahydrate: Very soluble in water and ethanol	
Test for chloride	Passes test	
Test for calcium	Passes test	
Purity		
Free alkali	Not more than $0.15\%$ as Ca(OH) <sub>2</sub>	
	Dissolve 1 g of the sample in 20 mL of freshly boiled and cooled water, and add	
	2 drops of phenolphthalein TS. If the solution is pink, the pink color is	

## discharged by adding 2 mL of 0.02 N hydrochloric acid.

Mag	nesium and alkali	Not more than 5%
salts		
Fluo	oride	Not more than 40 mg/kg
Lead	1	Not more than 2 mg/kg
Category		Food Additives Category 7
Uses		Food quality improvement, fermentation and food processing agents

### Calcium Hydroxide

Chemical formula: Ca(OH		Molecular weight: 74.09
1.	Assay	: Not less than 95.0%
2.	Description	: Alkaline, white, bitter powder.
		1 g of this product is soluble in 630 mL of 25°C water or 1300 mL of boiling water.
		Insoluble in ethanol, soluble in glycerol and saturated sucrose solution.
3.	Identification	: (1) Mix the product with water 3 to 4 times of its weight, a muddy solution will be
		produced and its suspension is alkaline on litmus paper.
		(2) Add 1 g of the sample in 20 mL of water and add sufficient acetic acid to dissolve it.
		The solution passes test for calcium.
4.	Hydrochloric acid	: Not more than 0.5%.
	insoluble matter	
5.	Carbonate	: No bubbles shall be generated when 2 g of the sample in 50 mL of water is added with
		5 mL of dilute hydrochloric acid.
6.	Fluoride	: Not more than 50 ppm.
7.	Magnesium and	: The weight of the residue does not exceed 12 mg.
	alkali salts	
8.	Barium	: Not more than 300 ppm
9.	Lead	: Not more than 10 ppm.
10.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
11.	Heavy metals	: Not more than 40 ppm (as Pb).
12.	Category	: Food Additives Category 7
13.	Uses	: Food quality improvement, fermentation and food processing agents

#### Calcium Sulfate

Synonyms	: INS No. 516
Definition	
CAS NUMBER	7778-18-9
Chemical formula	Anhydrous: CaSO <sub>4</sub>
	Dehydrate: CaSO <sub>4</sub> ·2H <sub>2</sub> O
Molecular weight	Anhydrous: 136.14
	Dehydrate: 172.18
Assay	: Not less than 99.0% after drying
Appearance	: Fine, white to slightly yellow-white, odorless powder.
Characteristics	
Identification	
Solubility	Slightly soluble in water; insoluble in ethanol.
Test for	: Passes test.
calcium	
Test for	: Passes test.
sulfate	
Purity	
Loss on	Anhydrous: Not more than 1.5% (250° to constant weight).
drying	Dihydrate: $19 \sim 23\%$ (250° to constant weight).
Fluorine	: Not more than 30 mg/kg
Selenium	: Not more than 30 mg/kg
Lead	: Not more than 2 mg/kg
Category	: Food Additives Category 7
Uses	: Food quality improvement, fermentation and food processing agents

#### Calcium Gluconate

Chemical formula: C <sub>12</sub> H <sub>22</sub> CaO <sub>14</sub> ·		CaO <sub>14</sub> ·H <sub>2</sub> O Molecular weight: 448.39
1.	Assay	: Not less than 98% and not more than 104% (as $Cl)_2H_{22}CaO_{14}\cdot H_2O$ on the dried basis.
2.	Description	: Odorless and tasteless, white, crystalline granules or powder, stable in air. The solution
		was tested neutral on litmus paper. 1 g of this sample is slowly soluble in 30 mL of $25^{\circ}$ C
		water of 5 mL of boiling water; insoluble in ethanol and other organic solvents.
3.	Identification	: (1) Add 1 drop of ferric chloride solution (1 g of ferric chloride dissolved in 10 g of
		water) to 1 mL of calcium gluconate solution (1 g of calcium gluconate dissolved
		in 40 g of water). The solution is dark yellow.
		(2) Add 0.7 mL of glacial acetic acid and 1 mL of freshly distilled phenylhydrazine to 5
		mL of warm calcium gluconate solution (1 g of calcium gluconate dissolved in 10
		g of warm water) in a test tube. Heat on a steam bath for 30 minutes and cool.
		Scrape the inner wall of the tube with glass bar to induce crystallization and
		collect the crystal by filtration. Dissolve the crystal in 10 mL of hot water, add a
		small amount of activated carbon. The crystal is dissolved in 10 mL of hot water.
		Add a small amount of activated, mix and cool. Collect the crystal in the same
		way again. The melting point of crystal is between $192 \sim 202$ °C.
		(3) The 1 in 40 solution of the sample passes the test for calcium.
4.	Solution	: Dissolve 1 g of the sample in 20 mL of water at 60°C. The solution should be less than
		"almost clear".
5.	рН	$: 6.0 \sim 8.0 \ (1 \text{ in } 20 \text{ soln})$
6.	Chloride	: Not more than 0.07% of Cl.
7.	Sulfate	: Not more than $0.05\%$ of SO <sub>4</sub> .
8.	Lead	: Not more than 10 ppm.
9.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
10.	Heavy metals	: Not more than 20 ppm (as Pb).
11.	Sucrose and reduci	ng : Add 10 mL of hot water and 2 mL of dilute hydrochloric acid TS and boil for 2
	sugar	minutes. Cool and add 5 mL of sodium carbonate TS, stand for 5 minutes, add
		water to make a 20 mL solution. Filter, then add 2 mL of Feringin TS to 5 mL of
		clear filtrate and boil for 1 minute. Orange-yellow to red precipitates shall not be
		generated immediately.
12.	Loss on drying	: Dry it at 80° for 2 hours: it loses not more than 0.5% of its weight.

78

- 13. Category : Food Additives Category 7
- 14. Uses : Food quality improvement, fermentation and food processing agents

#### Calcium Citrate

Synonyms	INS No. 333(iii)
Definition	
Chemical names	Tricalcium citrate, tricalcium salt of 2-hydroxy-1,2,3- propanetricarboxylic acid,
	tricalcium salt of ß-hydroxy-tricarballylic acid
C.A.S. number	813-94-5
Chemical formula	$C_{12}H_{10}Ca_3O_{14}{\cdot}4H_2O$
Structural formula	
	$\begin{bmatrix} cH_2 \cdot COO^{\Theta} \\ HO_{-} - c - cOO^{\Theta} \\ - cH_2 \cdot COO^{\Theta} \\ cH_2 \cdot COO^{\Theta} \end{bmatrix}_2^2 ca_3 \cdot 4H_2O$
Formula weight	570.51
Assay	Not less than 97.5% after drying
Description	Odourless, fine white powder
Characteristics	
Identification	
Solubility	Very slightly soluble in water. Insoluble in ethanol.
Test for citrate	Passes test
Test for calcium	Passes test
Purity	
Loss on drying	Not less than 10% and not more than 14% (150°C, 4 h)
Fluorides	Not more than 30 mg/kg
Free acid and alkali	Passes test
Oxalate	Dissolve 1 g of the sample in 5 ml of warm dilute hydrochloric acid TS and filter the
	solution if necessary.
Lead	Not more than 2 mg/kg
Category	Food additives category (7) (8)
Functional uses	Food quality improvement, fermentation and food processing agents; Nutritional
	additives.

§ 08113

### Calcium Dihydrogen Phosphate

SYNONYMS	:Monobasic calcium phosphate, monocalcium orthophosphate, monocalcium
	phosphate, calcium biphosphate, acid calcium phosphate, INS No. 341(i)
DEFINITION	
Chemical names	:Calcium dihydrogen phosphate
C.A.S. number	: Anhydrous: 7758-23-8
	Monohydrate: 10031-30-8
Chemical formula	: Anhydrous: Ca(H2PO4)2
	Monohydrate: Ca(H2PO4)2·H2O
Formula weight	: Anhydrous: 234.05
	Monohydrate: 252.07
Assay	Anhydrous: Not less than 16.8% and not more than 18.3% of Ca
	Monohydrate: Not less than 15.9% and not more than 17.7% of Ca
DESCRIPTION	Hygroscopic white crystals or granules, or granular powder
CHARACTERISTICS	
IDENTIFICATION	
Solubility	: Sparingly soluble in water, insoluble in ethanol
Test for calcium	: Passes test
Test for phosphate	: Passes test
PURITY	
Loss on drying	: Monohydrate: Not more than 1% (60°C , 3 h)
Loss on ignition	: Anhydrous: Between 14.0 and 15.5% (800°C, 30 min)
Fluoride	: Not more than 50 mg/kg
Arsenic	: Not more than 3 mg/kg
Lead	: Not more than 4 mg/kg
Category	: Food additives category (7); (8).
Functional uses	: Food quality improvement, fermentation and food processing agents; Nutritional
	additives.

### Calcium Phosphate, Dibasic

Synonyms	: INS No. 341(ii); Dibasic calcium phosphate; dicalcium phosphate
Definition	
Chemical names	: Calcium monohydrogen phosphate, calcium hydrogen orthophosphate, secondary calcium
	phosphate
CAS NUMBER	: 7757-93-9
Chemical formula	Anhydrous: CaHPO <sub>4</sub>
	Dihydrate: CaHPO <sub>4</sub> · 2H <sub>2</sub> O
Molecular weight	Anhydrous: 136.06
	Dihydrate: 172.09
Assay	: 98.0 ~ 102.0% after drying
Appearance	: A white, odorless powder.

### Characteristics

	Identification	
	Solubility	Practically insoluble in water; insoluble in ethanol.
	Test for phosphate	Passes test.
	Test for calcium	Passes test.
	Purity	
	Loss on drying	Anhydrous: Not more than 2% (200°C, 3 hr).
		Dihydrate: 18% ~ 22% (200°C, 3 hr).
	Fluorine	: Not more than 50 mg/kg.
	Arsenic	: Not more than 3 mg/kg.
	Lead	: Not more than 4 mg/kg.
Category		: Food Additives Category 7, 8.
Use	°S	: Food quality improvement, fermentation and food processing agents; Nutritional additives

### Calcium Phosphate, Tribasic

Synonyms	: INS No. 341(iii); Tricalcium phosphate; precipitated calcium phosphate
Definition	: Consists of a variable mixture of calcium phosphates having and approximate composition of 10
	$CaO^{\cdot}3P_2O_5^{\cdot}H_2O_{\cdot}$
Assay	: Not less than the equivalent of 90% of Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> , calculated on the ignited basis
Appearance	: White, odorless and tasteless powder which is stable in air.
Characteristics	
Identification	
Solubility	Practically insoluble in water; insoluble in ethanol, soluble in dilute hydrochloric and nitric acid.
Test for	Passes test.
phosphate	
Test for	Passes test.
calcium	
Purity	
Loss on	: Not more than 10% (825° to constant weight).
ignition	
Fluorine	: Not more than 50 mg/kg.
Lead	: Not more than 4 mg/kg.
Category	: Food Additives Category 7, 8.
Uses	: Food quality improvement, fermentation and food processing agents; Nutritional additives

#### Calcium Dihydrogen Pyrosphate

Chemical formula:	$CaH_2P_2O_7$
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#### Molecular weight: 216.04

- 1. Assay : Not less than 90%
- 2. Description : A white powder or granules.
- 3. Identification : (1) Shake 0.5 g of the sample and 10 mL of water evenly. The solution is acid.
  - (2) Heat and dissolve 5 mL of dilute nitric acid (1 mL of nitric acid dissolved in 9 mL of water) to 0.2 g of the sample. Add 2 mL of ammonium molybdate TS. After heating, yellow precipitate forms.
    - (3) Dissolve 9 mL of water and 1 mL of dilute hydrochloric acid (1 mL of hydrochloric acid in 3 mL of water) to 0.3 of the sample by heating. Cool and filter. Add 3 mL of ammonium oxalate solution (1 g of ammonium oxalate in 30 mL of water) to the filtrate, and white precipitate forms. Add 5 mL of dilute hydrochloric acid (1 mL of hydrochloric acid in 30 mL of water), and the precipitate dissolves.
- Hydrochloric acid : Not more than 0.4%.
   insoluble matter
- Test for : Add 2 to 3 drops of silver nitrate TS (1 in 50 solution of silver nitrate) to 1.0 g of the sample. An obvious yellow color should not appear.
   Arsenic : Not more than 4 ppm (as As<sub>2</sub>O<sub>3</sub>).
- 7. Heavy metals : Not more than 20 ppm (as Pb).
- 8. Loss on drying : Dry it at  $150^{\circ}$  for 4 hours: it loses not more than 5% of its weight.
- 9. Category : Food Additives Category 7
- 10. Uses : Food quality improvement, fermentation and food processing agents

### Calcium Glycerophosphate

Chen	nical formula: C <sub>3</sub> H <sub>7</sub> C	aO <sub>6</sub> P Molecular weight: 210.14
1.	Assay	: Not less than 98.0%
2.	Description	: White odorless, powder. Almost tasteless or slightly bitter, slightly hygroscopic. At low
		temperature, it is more soluble in water and citric acid increases its water solubility.
		Insoluble in alcohol.
3.	Identification	: (1) Mix and heat 100 mg of the sample with 500 mg of potassium bisulfate, and irritant
		propylene aldehyde gas will be produced.
		(2) Mix 1 g of the sample with 10 mL of cold water at $5^{\circ}$ C or below. After shaking
		evenly, perform the following tests:
		(a) When the test solution is boiled, white scaly crystals form.
		(b) Add 2 to 3 drops of lead acetate TS to 3 mL of the sample, and a white curd-
		like precipitate appears. Add 3 mL of nitric acid, and the precipitate
		disappears.
		(c) Saturated solution of the sample passes the test for calcium.
4.	Solution	: Add 1 g of this product is to 50 mL of water, and the solution should be below "slightly
		turbid".
5.	Ethanol-soluble	: Not more than 1%.
	substances	
6.	Free alkali	: Add 5 drops of phenolphthalein TS to 1 in 60 solution of the sample. Not more than 1.5
		mL of 0.1 N sulfuric acid should be required to titration.
7.	Chloride	: Not more than 0.07% of Cl.
8.	Sulfate	: Not more than $0.048\%$ of SO <sub>4</sub> .
9.	Phosphate	: Not more than 0.04% of PO <sub>4</sub> .
10.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
11.	Heavy metals	: Not more than 40 ppm (as Pb).
12.	Lead	: Not more than 10 ppm.
13.	Loss on drying	: Dry it at 150° for 4 hours: it loses not more than 13.0% of its weight.
14.	Category	: Food Additives Category 7
15.	Uses	: Food quality improvement, fermentation and food processing agents

#### Calcium Lactate

Chemical formula: $C_6H_{10}CaO_6 \cdot 0 \sim 5H_2O$ Molecular weight: 218.22		$CaO_6 \cdot 0 \sim 5H_2O$ Molecular weight: 218.22	
1.	Assay	: Not less than 97.0% and not more than 101.0% of on the dried basis as $C_6H_{10}CaO_{6.}$	
2.	Description	: White to cream colored, almost odorless, crystalline powder or granules. The	
		pentahydrate is somewhat efflorescent. Soluble in water, practically insoluble in ethanol.	
3.	Identification	: Dissolve 1 g of this product in 20 mL of water. The solution passes the test for calcium	
		and lactate.	
4.	Solution	: Dissolve 1 g of the sample in 20 mL of water in water bath. The solution should be	
		clear.	
5.	pH	: 6.0~8.0. 6.0-8.0 (1 in 20 soln)	
6.	Lead	: Not more than 10 ppm (as Pb).	
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).	
8.	Heavy metals	: Not more than 20 ppm (as Pb).	
9.	Magnesium and	: Not more than 1%.	
	alkali salts		
10.	Volatile fatty acid	: Add 0.5 g of the sample in 1 mL of sulfuric acid. When heated in a water bath,	
		butyric acid-like odor should not occur.	
11.	Acidity	: Not more than 0.55% of lactic acid.	
12.	Fluoride	: Not more than 15 ppm.	
13.	Loss on drying	: Dry it at 120° for 4 hours: it loses not more than 30.0% of its weight.	
14.	Category	: Food Additives Category 7	
15.	Uses	: Food quality improvement, fermentation and food processing agents	

1.	Description	: White or slightly yellowish powder or brittle solid with a characteristic odor. Slightly
		soluble in hot water.
2.	Identification	: (1) Ignite 1 g of the sample at 500°C for 1 hour and dissolve it in 5 mL of dilute
		hydrochloric acid (1+3). The solution passes test for calcium.
		(2) Add 10 mL of dilute hydrochloric acid TS to 2 g of the sample, heat for 5 min in a
		water bath, filter and neutralize the filtrate with ammonia TS. Dissolve the residue from
		the filter in 30 mL of sodium hydroxide solution $(1+25)$ and heat in 95°C or hotter water
		bath for 30 min. Add 20 mL of dilute hydrochloric acid (1+3) to it after cooling, extract
		twice with 30 mL of diethyl ether, wash the ether solution with 20 mL of water, dry with
		anhydrous sodium sulfate and evaporate the ether. The residue melts between 54 and
		69°C.
		(3) Passes the test for lactate.
3.	Acid value	: 50~86.
4.	Ester value	: 125~164.
5.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Total lactic acid	: 32~38%.
8.	Calcium content	: 4.2~5.2%
9.	Residue on	: 14.3~17.7% (800°C).
	ignition	
10.	Category	: Food Additives Category 7
11.	Uses	: Food quality improvement, fermentation and food processing agents

#### Calcium Carbonate

Chen	nical formula: CaCO <sub>3</sub>	Molecular weight: 100.09
1.	Assay	: Not less than 98.0% after drying at 200°C for 4 hours.
2.	Description	: An odorless, tasteless powder or crystal. Stable in air. Practically insoluble in water and
		in alcohol.
3.	Identification	: Dissolve 1.0 g of the sample in 10 mL of water and 7 mL of dilute acetic acid (1 mL of
		acetic acid in 3 mL of water). Neutralize the solution with ammonia after boiling. The
		solution passes test for calcium.
4.	Hydrochloric acid	: Not more than 0.2%.
	insoluble matter	
5.	Free alkali	: Add 3 g of the sample to 30 mL of freshly boiled and cooled water, stir for 3 min. and
		filter. To 20 mL of the filtrate add 2 drops of phenolphthalein TS. Though a red color
		is produced, it should disappear when 0.2 mL of 0.1 N hydrochloric acid is added.
6.	Heavy metals	: Not more than 30 ppm (as Pb).
7.	Magnesium and	: Not more than 1%.
	alkali salts	
8.	Barium	: Not more than 0.03%.
9.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
10.	Fluoride	: Not more than 0.005%.
11.	Lead	: Not more than 10 ppm.
12.	Loss on drying	: Dry it at 200° for 4 hours: it loses not more than 2.0% of its weight.
13.	Category	: Food Additives Category 7
14.	Uses	: Food quality improvement, fermentation and food processing agents

#### Ammonium Carbonate

The same as § 06009

#### Potassium Carbonate

The same as § 06011

### Sodium Carbonate; Sodium Carbonate, Anhydrous

Chemical names	: Sodium carbonate; sodium salt of carbonic acid	
Synonyms	: Soda ash; INS No. 500 (i); CAS No. 497-19-8.	
Chemical formula	: Na <sub>2</sub> CO <sub>3</sub> (anhydrous)	Molecular weight:
	Na <sub>2</sub> CO <sub>3</sub> ·xH <sub>2</sub> O (hydrated)	106.00

1.	Assay	: Not less than 99.0% after drying.	
2.	Appearance	: Colorless crystals or white, granular or crystalline powder; the anhydrous form is	
		hygroscopic; hydrated forms available include the monohydrate and the decahydrate; the	
		latter is efflorescent.	
3.	Solubility	: Freely soluble in water; insoluble in ethanol.	
4.	Identification	: Passes test for sodium and test for carbonate.	
5.	Loss on drying	: Anhydrous: Not more than 2%	
		Monohydrate: Not more than 15%	
		Decahydrate: 55 - 65%	
		For all forms, heat the sample first at about 70o, then gradually raise the temperature and	
		finally dry at 250-300°C to constant weight.	
6.	Lead	: Not more than 2 mg/kg.	
7.	Heavy metals	: 10 mg/kg (as Pb).	
8.	Category	: Food Additives Category 7, 14.	
9.	Uses	: Food quality improvement, fermentation and food processing agents; chemicals for food	
		industry.	

	Magnesium Carbonate	
SYNONYMS	:INS No. 504(i), Magnesium subcarbonate (light or	
	heavy), hydrated basic magnesium carbonate,	
	magnesium carbonate hydroxide; INS No. 504(ii)	
DEFINITION	A basic hydrated or a normal hydrated magnesium	
	carbonate or a mixture of the two	
Chemical names	: Magnesium carbonate, Magnesium carbonate	
	hydroxide hydrated	
C.A.S. number	: Magnesium Carbonate : 546-93-0	
Assay	Magnesium Carbonate : $24.0\% \sim 26.4\%$ (As Mg)	
	Magnesium Hydroxide Carbonate : 40.0% ~ 45.0% (As MgO)	
DESCRIPTION	Odourless, light, white friable masses or as a bulky	
	white powder	
CHARACTERISTICS		
IDENTIFICATION		
Solubility	: Practically insoluble in water; insoluble in ethanol	
Test for carbonate	Magnesium Carbonate: Passes test	
	Magnesium Hydroxide Carbonate: -	
Test for magnesium	Passes test	
Alkalinity	Magnesium Carbonate: -	
	Magnesium Hydroxide Carbonate: Slurry shows	
	slight alkalinity	
PURITY		
Acid insoluble	Not more than 0.05%	
substances		
Water insoluble	Magnesium Carbonates: Not more than 1%	
substances	Magnesium Hydroxide Carbonate: -	
Soluble salts	Magnesium Carbonates: -	
	Magnesium Hydroxide Carbonate: Not more than	
	1.0%	

<u>Calcium</u>	Magnesium Carbonates: Not more than 0.4%	
	Magnesium Hydroxide Carbonate:1.0%以下	
Lead	Not more than 2 mg/kg	
Arsenic	Not more than $4 \frac{\text{mg/kg}}{\text{kg}}$ (As As <sub>2</sub> O <sub>3</sub> )	
Category	: Food additives category (7).	
Functional uses	: Food quality improvement, fermentation and food processing agents.	

#### Ammonium Sulfate

Chemical formula: (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>		2SO <sub>4</sub> Molecular weight: 132.14	
1.	Assay	: Not less than 99.0%.	
2.	Description	: Colorless crystals, white granule; odorless. Decomposes above 280 °C. Freely soluble	
		in water; insoluble in ethanol.	
3.	Identification	: Passes test for ammonium and test for sulfate.	
4.	Solution	: Dissolve 1 g of the sample in 20 mL of water. The solution should be less than "almost	
		clear" and colorless.	
5.	pН	: 4.5 - 6.0 (0.1 M solution)	
6.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
7.	Heavy metals	: Not more than 20 ppm (as Pb).	
8.	Selenium	: Not more than 4 ppm.	
9.	Residue on	: Not more than 0.25%.	
	ignition		
10.	Category	: Food Additives Category 7	
11.	Uses	: Food quality improvement, fermentation and food processing agents	

#### Sodium Sulfate

Chemical formula: Na <sub>2</sub> SO <sub>4</sub> ·nH <sub>2</sub> O (n=10 or 0)		₄·nH <sub>2</sub> O (n=10 or 0)	Molecular weight: 322.20 (Decahydrate); 142.04
			(Anhydrous)
1.	Assay	: Not less than 99.0% on the dr	ried basis
2.	Description	: Colorless crystal or white, fin	e crystalline powder. The hydrated form is deliquescent.
		Freely soluble in water; practic	cally insoluble in ethanol. The 1 in 20 solution of the
		sample is neutral or weak base	on litmus paper.
3.	Identification	: Passes test for sodium and tes	st for sulfate.
4.	Solution	: Dissolve 1 g of the sample in	10 mL of water. The solution should be less than "almost
		clear" and colorless.	
5.	Chloride	: Not more than 0.11% of Cl.	
6.	Heavy metals	: Not more than 10 ppm (as Pb	).
7.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub>	O <sub>3</sub> ).
8.	Selenium	: Not more than 0.003%.	
9.	Loss on drying	: $Na_2SO_4$ 10H <sub>2</sub> O: 51.0~57.0	0% (105°C, 4hr).
		Na <sub>2</sub> SO <sub>4</sub> : 5.0% (105°C, 4hr).	
10.	Category	: Food Additives Category 7	
11.	Uses	: Food quality improvement, fo	ermentation and food processing agents

	Magnesium Stearate
SYNONYMS	: Magnesium distearate, dibasic magnesium
	stearate, INS No. 470(iii)
DEFINITION	: Magnesium stearate is a mixture of
	magnesium salts of fatty acids obtained from
	edible fats and oils. The product consists
	mainly of magnesium stearate and palmitate in
	varying proportions. It is manufactured by one
	of the two following processes: a) direct
	process wherein fatty acids are directly reacted
	with a magnesium source, such as magnesium
	oxide to form magnesium salts of the fatty
	acids; b) indirect process where a sodium soap
	is produced by the reaction of fatty acids with
	sodium hydroxide in water and the product is
	precipitated by adding magnesium salts to the
	soap.
Chemical names	: Magnesium stearate, magnesium
	octadecanoate, fatty acids C <sub>16</sub> -C <sub>18</sub> magnesium
	salts
C.A.S number	: 557-04-0 (magnesium stearate)
	91031-63-9 (fatty acids $C_{16-18}$ magnesium salts)
Chemical formula	: $Mg(C_{18}H_{35}O_2)_2$ (magnesium distearate)
Formula weight	: 591.27 (magnesium distearate)
Assay	: Magnesium: Not less than 4.0% and not more
	than 5.0%, on dried basis.
	Fatty acids: Not less than 40.0% stearic acid in
	the fatty acid fraction; and not less than 90.0%
	as the sum of stearic acid and palmitic acid in the fatty acid fraction
DESCRIPTION	the fatty acid fraction.
DESCRIPTION	Off-white to white, very fine powder; greasy to the touch
CHARACTERISTICS	
IDENTIFICATION	
IDENTICATION	

Solubility	Practically insoluble in water
Magnesium	Presence of magnesium in the
	sample
Fatty acid	Identify the individual fatty acids in the sample
composition	
PURITY	
Loss on drying	Not more than 6% (105°C, constant weight, use
	1 g of sample)
Acidity or alkalinity	Passes test
Unsaponifiable	Not more than 2%
matter	
Cadmium	Not more than 1 mg/kg
Lead	Not more than 2 mg/kg
Nickel	Not more than 3 mg/kg
Category	: Food additives category (7).
Functional uses	: Food quality improvement, fermentation and food processing agents.

### Magnesium Sulfate

Synonyms	Epsom salt (heptahydrate) ; INS No. 518
Definition	Magnesium sulfate occurs naturally in sea water, mineral springs and in minerals such as
	kieserite and epsomite. It is recovered from them or by reacting sulfuric acid and
	magnesium oxide. It is produced with one or seven molecules of water of hydration or in
	a dried form containing the equivalent of between 2 and 3 waters of hydration.
Chemical names	Magnesium sulfate
C.A.S. number	Monohydrate: 14168-73-1
	Heptahydrate: 10034-99-8
	Dried: 15244-36-7
Chemical formula	Monohydrate: MgSO <sub>4</sub> ·H <sub>2</sub> O
	Heptahydrate: MgSO <sub>4</sub> ·7H <sub>2</sub> O
	Dried: MgSO <sub>4</sub> $\cdot$ xH <sub>2</sub> O, where x is the average hydration value (between 2 and 3)
Formula weight	Monohydrate: 138.38
	Heptahydrate: 246.47
Assay	Not less than 99.0 % and not more than 100.5% on the ignited basis
Description	Colourless crystals, granular crystalline powder or white powder. Crystals effloresce in
	warm, dry air.
Characteristics	
Identification	
Solubility	Freely soluble in water, very soluble in boiling water, and sparingly soluble in ethanol.
Test for magnesium	Passes test
Test for sulfate	Passes test
Purity	
Loss on ignition	Monohydrate: between 13.0 and 16.0 %,
	Heptahydrate: between 40.0 and 52.0 %,
	Dried: between 22.0 and 32.0 %
	(105°C, 2 h, then 400°C to constant weight)

рН	Between 5.5 and 7.5 (1 in 20 solution)
Chloride	Not more than 0.03%
Arsenic	Not more than 3 mg/kg
Iron	Not more than 20 mg/kg
Selenium	Not more than 30 mg/kg
Lead	Not more than 2 mg/kg
Category	Food additives category (07) (08)
Functional uses	Food quality improvement, fermentation and food processing agents; Nutritional
	additives.

### Magnesium chloride

Chen	nical formula: MgCl <sub>2</sub>	·6H <sub>2</sub> O Molecular weight: 203.30
1.	Assay	: Not less than 95.0% of MgCl <sub>2</sub> ·6H <sub>2</sub> O
2.	Description	: Colorless, odorless flakes, granules, lumps or crystals; very soluble in water; freely
		soluble in ethanol.
3.	Identification	: Passes the test for magnesium and test for chloride.
4.	Solution	: 1 g of this product is dissolved in water 10 mL, and the solution should be below
		"slightly turbid".
5.	Heavy metals	: Not more than 20 ppm (as Pb).
6.	Zinc	: Not less than 0.07 mg/g
7.	Calcium	: 0.5 g of the sample is dissolved in water to make a 50 mL solution. Take 5 mL of the
		sample, add 1 mL of ammonium oxalate test solution (1 in 25 solution of ammonium
		oxalate) and place it for 5 minutes. The turbidity of the solution should be below
		"slightly turbid".
8.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Ammonium salt	: Not more than 0.005%.
10.	Category	: Food Additives Category 7
11.	Uses	: Food quality improvement, fermentation and food processing agents

#### Ammonium Phosphate, Monobasic

Chem	nical formula: NH <sub>4</sub> H <sub>2</sub>	PO <sub>4</sub> Molecular weight: 115.03
1.	Assay	: 96.0∼102.0%.
2.	Description	: Colorless to white crystal or white crystalline powder or granules; odorless. Very
		soluble in water.
3.	Identification	: The 1 in 20 solution of this product passes the test for ammonium and test for
		phosphate.
4.	Solution	: Dissolve 1 g of the sample in 20 mL of water. The solution should be less than "almost
		clear" and colorless.
5.	pH	: 4.1~5.0 (1 in 100 soln)
6.	Chloride	: Not more than 0.035% of Cl.
7.	Fluoride	: Not more than 10 ppm.
8.	Sulfate	: Not more than $0.039\%$ of SO <sub>4</sub> .
9.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
10.	Heavy metals	: Not more than 20 ppm (as Pb).
11.	Category	: Food Additives Category 7
12.	Uses	: Food quality improvement, fermentation and food processing agents

### Ammonium Phosphate, Dibasic

Chen	nical formula: (NH <sub>4</sub> )	2HPO <sub>4</sub> Molecular weight: 132.06
1.	Assay	: 96.0∼102.0%.
2.	Description	: Colorless to white crystal or white crystalline powder or granule, having an odor of
		ammonium. Soluble in water.
3.	Identification	: The 1 in 20 solution of this product passes the test of ammonium and test of phosphate.
4.	Solution	: Dissolve 1.0 g of the sample in 20 mL of water. The solution should be less than
		"almost clear" and colorless.
5.	pН	$: 7.6 \sim 8.4 \ (1 \text{ in } 100 \text{ soln})$
6.	Chloride	: Not more than 0.035% of Cl
7.	Fluoride	: Not more than 10 ppm.
8.	Sulfate	: Not more than $0.039\%$ of SO <sub>4</sub> .
9.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
10.	Heavy metals	: Not more than 20 ppm (as Pb).
11.	Category	: Food Additives Category 7
12.	Uses	: Food quality improvement, fermentation and food processing agents

Potassium Dihydrogen Phosphate	
Synonyms	Monobasic potassium phosphate, monopotassium monophosphate potassium acid
	phosphate, potassium biphosphate; INS No. 340(i)
Definition	
Chemical names	Potassium dihydrogenphosphate, monopotassium dihydrogenorthophosphate,
	monopotassium dihydrogen monophosphate
C.A.S. number	7778-77-0
Chemical formula	KH <sub>2</sub> PO <sub>4</sub>
Formula weight	136.09
Assay	Not less than 98.0% after drying
Description	Odourless, colourless crystals or white granular or crystalline powder
Characteristics	
Identification	
Solubility	Freely soluble in water; insoluble in ethanol
рН	4.2 - 4.7 (1 in 100 soln)
Test for potassium	Passes test
Test for phosphate	Passes test
Test for orthophosphate	To 5 ml of a 1 in 100 soln of the sample, add silver nitrate TS. A yellow precipitate is
	obtained.
Purity	
Loss on drying	Not more than 2% (105°C, 4 h)
Water insoluble substances	Not more than 0.2%
Fluoride	Not more than 10 mg/kg
Arsenic	Not more than 3 mg/kg
Lead	Not more than 4 mg/kg
Category	Food additives category (7) (8) (13)
Functional uses	Food quality improvement, fermentation and food processing agents; Nutritional
	additives; Coagulating Agents.

### Potassium Phosphate, Dibasic

Chen	nical formula: K <sub>2</sub> HPC	D <sub>4</sub> Molecular weight: 174.18
1.	Assay	: Not less than 98.0% on the dried basis (105°C, 4 h)
2.	Description	: Colorless or white granular powder, crystals or masses; deliquescent. 1 g of this product
		is soluble in 3 mL of water, insoluble in ethanol.
3.	Identification	: The 1 in 20 solution of this product passes the test for potassium and test for phosphate.
4.	Solution	: Dissolve 1.0 g of the sample in 20 mL of water. The solution should be colorless and
		the turbidity is below "slightly turbid".
5.	pH	: 8.7~9.3 (1 in 100 soln)
6.	Chloride	: Not more than 0.011% of Cl.
7.	Fluoride	: Not more than 10 ppm.
8.	Sulfate	: Not more than 0.019% of SO <sub>4</sub> .
9.	Arsenic	: Not more than 3 ppm (as $As_2O_3$ ).
10.	Heavy metals	: Not more than 20 ppm (as Pb).
11.	Lead	: Not more than 5 ppm.
12.	Water-insoluble	: Not more than 0.2%.
	matter	
13.	Loss on drying	: Dry it at 105°C for 4 hours: it loses not more than 5% of its weight.
14.	Category	: Food Additives Category 7, 13.
15.	Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

### Potassium Phosphate, Tribasic

### Chemical formula: $K_3PO_4 \cdot 0 \sim 3H_2O$

1.	Assay	: Not less than 97.0% of K <sub>3</sub> PO <sub>4</sub> , calculated on the ignited basis.
2.	Description	: Colorless or white, odorless hygroscopic crystals or granules; freely soluble in water;
		insoluble in ethanol.
3.	Identification	: The 1 in 20 solution of this product passes the test for potassium and test for phosphate.
4.	Solution	: Dissolve 1.0 g of the sample in 20 mL of water. The solution should be colorless and
		the turbidity is below "slightly turbid".
5.	pH	: 11.5~12.5 (1 in 100 soln)
6.	Chloride	: Not more than 0.011% of Cl
7.	Fluoride	: Not more than 10 ppm.
8.	Sulfate	: Not more than $0.019\%$ of SO <sub>4</sub> .
9.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
10.	Heavy metals	: Not more than 20 ppm (as Pb).
11.	Lead	: Not more than 5 ppm.
12.	Water-insoluble	: Not more than 0.2%.
	matter	
13.	Loss on ignition	: Not more than 23.0% (120°C, 2 h, then 800°C , 1 h)
14.	Category	: Food Additives Category 7, 13.
15.	Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

Sodium Dihydrogen Phosphate		
Synonyms	Monobasic sodium phosphate, monosodium monophosphate sodium acid	
	phosphate, sodium biphosphate; INS No. 339(i)	
Definition		
Chemical names	Sodium dihydrogenphosphate, monosodium dihydrogenortho- phosphate, monosodium	
	dihydrogen monophosphate	
C.A.S. number	7558-80-7	
Chemical formula	Anhydrous: NaH <sub>2</sub> PO <sub>4</sub>	
	Monohydrate: NaH <sub>2</sub> PO <sub>4</sub> · H <sub>2</sub> O	
	Dihydrate: NaH <sub>2</sub> PO <sub>4</sub> · 2H <sub>2</sub> O	
Formula weight	Anhydrous: 119.98	
	Monohydrate: 138.00	
	Dihydrate: 156.01	
Assay	Not less than 97% after drying	
Description	White odourless, slightly deliquescent powder, crystals, or granules	
Characteristics		
Identification		
Solubility	Freely soluble in water; insoluble in ethanol, ether or chloroform	
рН	4.2 - 4.6 (1 in 100 soln)	
Test for sodium	Passes test	
Test for phosphate	Passes test	
Test for orthophosphate	To a 1% solution of the sample add silver nitrate TS; the yellow precipitate formed is	
	soluble in dilute nitric acid TS.	
Purity		
Loss on drying	Anhydrous: Not more than $2\%$ ( $60^{\circ}$ , 1 h, then $105^{\circ}$ , 4 h)	
	Monohydrate: Not more than 15% (60°, 1 h, then 105°, 4 h)	
	Dihydrate: Not more than $25\%$ ( $60^{\circ}$ , 1 h, then $105^{\circ}$ , 4 h)	

Free acid and disodium	2.00 g of the sample dissolved in 40 ml of water require for neutralization not more than
phosphate	0.3 ml of either N sodium hydroxide or N sulfuric acid, using methyl orange TS as
	indicator.
Fluoride	Not more than 10 mg/kg
Arsenic	Not more than 3 mg/kg
Lead	Not more than 4 mg/kg
Category	Food additives category (07) (08) (13)
Functional uses	Food quality improvement, fermentation and food processing agents; Nutritional
	additives; Coagulating Agents.

#### Sodium Phosphate, Monobasic, Anhydrous

Cher	nical formula: NaH <sub>2</sub> P	PO <sub>4</sub> Molecular weight: 119.98
1.	Assay	: Not less than 98.0% and not more than 103.0% of $\rm NaH_2PO_4$ after drying at 105°C for 4
		hours.
2.	Description	: White odorless, slightly deliquescent powder, crystals, or granules. Freely soluble in
		water; insoluble in ethanol.
3.	Identification	: The 1 in 20 solution of this product passes the test for sodium and test for phosphate.
4.	Solution	: Dissolve 2.0 g of the sample in 20 mL of water. The solution should be colorless and
		the turbidity is below "slightly turbid".
5.	pH	: 4.3~4.9 (1 in 100 soln)
6.	Chloride	: Not more than 0.11% of Cl
7.	Fluoride	: Not more than 0.005%.
8.	Sulfate	: Not more than $0.048\%$ of SO <sub>4</sub> .
9.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
10.	Heavy metals	: Not more than 20 ppm (as Pb).
11.	Water-insoluble	: Not more than 0.2%.
	matter	
12.	Loss on drying	: Not more than 2.0% (60°C, 1 h, then 105°C, 4h)
13.	Category	: Food Additives Category 7, 13.
14.	Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

# Sodium Phosphate, Dibasic

# Chemical formula: Na<sub>2</sub>HPO<sub>4</sub>·2~12H<sub>2</sub>O

1.	Assay	: Not less than 98.0% of $Na_2HPO_4$ after drying at 105°C for 4 hours.
2.	Description	: White to colorless, odorless powder, crystals, or granules. Freely soluble in water;
		insoluble in ethanol.
3.	Identification	: The 1 in 20 solution of this product passes the test for sodium and test for phosphate.
4.	Solution	: Dissolve 0.5 g of the sample in 20 mL of water. The solution should be less than
		"almost clear" and colorless.
5.	pH	: 9.0~9.6 (1 in 100 soln)
6.	Chloride	: Not more than 0.21% of Cl
7.	Fluoride	: Not more than 0.005%.
8.	Sulfate	: Not more than $0.039\%$ of SO <sub>4</sub> .
9.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
10.	Heavy metals	: Not more than 20 ppm (as Pb).
11.	Water-insoluble	: Not more than 0.2%.
	matter	
12.	Loss on drying	: Not more than 61.0%.(40°C, 3h, then 120°C, 4 h).
13.	Category	: Food Additives Category 7, 13.
14.	Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

### Sodium Phosphate, Dibasic, Anhydrous

Chem	ical formula: Na <sub>2</sub> HPO	D <sub>4</sub> Molecular weight: 141.96
1.	Assay	: Not less than 98.0% of $Na_2HPO_4$ after drying at 105°C for 4 hours.
2.	Description	: White, hygroscopic powder Freely soluble in water; insoluble in ethanol.
3.	Identification	: The 1 in 20 solution of the sample passes the test for sodium and the test for phosphate.
4.	Solution	: Dissolve 0.5 g of the sample in 20 mL of water. The solution should be less than
		"almost clear" and colorless.
5.	pH	: 9.0~9.6 (1 in 100 soln)
6.	Chloride	: Not more than 0.21% of Cl
7.	Fluoride	: Not more than 0.005%.
8.	Sulfate	: Not more than 0.039% of SO <sub>4</sub> .
9.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
10.	Heavy metals	: Not more than 20 ppm (as Pb).
11.	Water-insoluble	: Not more than 0.2%.
	matter	
12.	Loss on drying	: Dry it at $120^{\circ}$ for 4 hours: it loses not more than 5% of its weight.
13.	Category	: Food Additives Category 7, 13.
14.	Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

### Trisodium Phosphate

Synonyms	Tribasic sodium phosphate, sodium phosphate; INS No. 339(iii)
Definition	
Chemical names	Trisodium orthophosphate, trisodium phosphate, trisodium monophosphate
C.A.S. number	7601-54-9
Chemical formula	Anhydrous: Na <sub>3</sub> PO <sub>4</sub>
	Hydrated: $Na_3PO_4 \cdot xH_2O$
Formula weight	Anhydrous: 163.94
Assay	Anhydrous, hemihydrate and monohydrate: Not less than 97.0% calculated on the dried
	basis
	Dodecahydrate: Not less than 92.0% calculated on the ignited basis
Description	White odourless crystals, granules or a crystalline powder; hydrated forms available
	include hemi- and monohydrates, hexahydrate, octahydrate, decahydrate and
	dodecahydrate; the dodecahydrate contains 1/4 mol of sodium hydroxide.
Characteristics	
Identification	
Solubility	Freely soluble in water; insoluble in ethanol
pH	11.5 - 12.5 (1 in 100 soln)
Test for sodium	Passes test.
Test for phosphate	To 5 ml of a 1 in 100 solution of the sample add 1 ml of concentrated nitric acid and 5
	ml of ammonium molybdate TS and warm. A bright canary-yellow precipitate is
	obtained.
Test for orthophosphate	Dissolve 0.1 g of the sample in 10 ml water, acidify slightly with dilute acetic acid TS,
	and add 1 ml of silver nitrate TS. A yellow precipitate is formed.
Purity	
Loss on ignition	Anhydrous: Not more than 2% (120°C, 2 h, then 800°C, 30 min)
	Monohydrate: Not more than 11% (120°C, 2 h, then 800°C, 30 min)
	Dodecahydrate: 45-58% (120°C, 2 h, then 800°C, 30 min)

Water insoluble substances	Not more than 0.2%
Fluoride	Not more than 50 mg/kg
Arsenic	Not more than 3 mg/kg
Lead	Not more than 4 mg/kg
Category	Food additives category (07) (13)
Functional uses	Food quality improvement, fermentation and food processing agents; Coagulating
	Agents.

### Sodium Phosphate, Tribasic, Anhydrous

Chemical formula: Na <sub>3</sub> PO	4 Molecular weight: 163.94
1. Assay	: Not less than 97.0% of $Na_3PO_4$ on the dried basis.
2. Description	: White, odorless powder or granules. Freely soluble in water; insoluble in ethanol .
3. Identification	: The 1 in 20 solution of this product passes the test for sodium and test for phosphate.
4. Solution	: Dissolve 0.5 g of the sample in 20 mL of water. The solution should be colorless and
	the turbidity is below "slightly turbid".
5. pH	: 11.5~12.5 (1 in 100 soln)
6. Chloride	: Not more than 0.71% of Cl
7. Fluoride	: Not more than 0.005%.
8. Sulfate	: Not more than 0.058% of SO <sub>4</sub> .
9. Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
10. Heavy metals	: Not more than 20 ppm (as Pb).
11. Water-insoluble	: Not more than 0.2%.
matter	
12. Loss on drying	: Not more than 5.0% (200°C, 5hr).
13. Category	: Food Additives Category 7, 13.
14. Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

### Potassium Metaphsphate

1.	Assay	: 56~80% of P <sub>2</sub> O <sub>5</sub> on the dried basis (110°C,4 h)
2.	Appearance	: Odorless, colorless or white glassy masses, fragments, crystals or powder.
3.	Solution	: Heat 1 g of the sample with 50 mL of water in water bath, stir and add 50 mL of sodium
		hydroxide TS slowly. After 10 min, when the solution is cooled to $35 \sim 45$ °C, it should
		be colorless and the turbidity should be below "slightly turbid".
4.	Chloride	: Not more than 0.1% of Cl.
5.	Sulfate	: Not more than 0.1% of SO <sub>4</sub> .
6.	Phosphate	: Add 2 to 3 drops of silver nitrate TS to 1 g of the sample. An obvious yellow color
		should not appear.
7.	Carbonate	: Boil 2 g of the sample in 5 mL of water, cool, and add 2 mL of hydrochloric acid. No
		bubbles are generated.
8.	Arsenic	: Not more than 3 ppm (as $As_2O_3$ ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Loss on drying	: Dry it at 110° for 4 hours: it loses not more than 5% of its weight.
11.	Category	: Food Additives Category 7, 13.
12.	Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

### Sodium Metaphsphate

1.	Assay	: Not less than 65% and not more than 83% of $P_2O_5$ on the dried basis (110°C, 4 h)
2.	Appearance	: Colorless or white glassy masses, fragments, crystals or powder.
3.	Solution	: Dissolve 1 g of the sample in 20 mL of water, and the solution should be below
		"slightly turbid".
4.	Chloride	: Not more than 0.2% of Cl.
5.	Sulfate	: Not more than $0.03\%$ of SO <sub>4</sub> .
6.	Phosphate	: Add 2 to 3 drops of silver nitrate TS to 1 g of the sample. An obvious yellow color
		should not appear.
7.	Carbonate	: Boil 2 g of the sample in 5 mL of water, cool, and add 2 mL of hydrochloric acid. No
		bubbles are generated.
8.	Arsenic	: Not more than 3 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Loss on drying	: Dry it at $110^{\circ}$ for 4 hours: it loses not more than 5% of its weight.
11.	Category	: Food Additives Category 7, 13.
12.	Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

### Potassium Polyphosphate

1.	Assay	: $P_2O_5 = 43 \sim 76\%$ on the dried basis (110°C, 4 h)
2.	Appearance	: Colorless or white glassy masses, fragments, crystals or powder.
3.	Solution	: Dissolve 1 g of the sample in 4 g of sodium acetate and 100 mL of water. The solution
		should be colorless and the turbidity is below "slightly turbid".
4.	Chloride	: Not more than 0.1% of Cl.
5.	Sulfate	: Not more than $0.1\%$ of SO <sub>4</sub> .
6.	Test for	: Add 2 to 3 drops of silver nitrate TS to 1 g of the sample. An obvious yellow color
	orthophosphate	should not appear.
7.	Carbonate	: Boil 2 g of the sample in 5 mL of water, cool, and add 2 mL of hydrochloric acid. Not
		more than a few bubbles are generated.
8.	Arsenic	: Not more than 3 ppm (as $As_2O_3$ ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Loss on drying	: Dry it at 110° for 4 hours: it loses not more than 5% of its weight.
11.	Category	: Food Additives Category 7, 13.
12.	Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

### Sodium Polyphosphate

1.	Assay	: $P_2O_5 = 53 \sim 80\%$ on the dried basis (110°C, 4 h)
2.	Appearance	: Colorless or white glassy masses, fragments, crystals or powder.
3.	Solution	: Heat and dissolve 1 g of the sample in 20 mL of water. The solution should be colorless
		and the turbidity should be below "slightly turbid".
4.	Chloride	: Not more than 0.2% of Cl.
5.	Sulfate	: Not more than $0.05\%$ of SO <sub>4.</sub>
6.	Test for	: Add 2 to 3 drops of silver nitrate TS to 1 g of the sample. An obvious yellow color
	orthophosphate	should not appear.
7.	Carbonate	: Boil 2 g of the sample in 5 mL of water, cool, and add 2 mL of hydrochloric acid. Not
		more than a few bubbles are generated.
8.	Arsenic	: Not more than 3 ppm (as $As_2O_3$ ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Loss on drying	: Dry it at 110° for 4 hours: it loses not more than 5% of its weight.
11.	Category	: Food Additives Category 7, 13.
12.	Uses	: Food quality improvement, fermentation and food processing agents; coagulating agent

### Sodium Acetate

Chemical formula: C <sub>2</sub> H <sub>3</sub> NaO <sub>2</sub> ·3H <sub>2</sub> O		aO <sub>2</sub> ·3H <sub>2</sub> O Molecular weight: 136.08
1.	Assay	: Not less than 98.5% of $C_2H_3NaO_2$ on the dried basis.
2.	Identification	: (1) The 1 in 20 solution of the sample passes test for sodium and test for acetate.
		(2) When heating the sample slowly, it liquefies. Then water evaporates and later
		decomposes evolving an odor of acetone. A solution of the residue gives alkaline
		reaction with litmus paper.
3.	Description	: Colorless, transparent crystals or a granular crystalline powder, odorless or with a faint,
		acetic odor. Effloresces in warm, dry air. Very soluble in water; soluble in ethanol.
4.	Solution	: When 1.0 g of the sample is dissolved in 20 mL of water, the solution should be clear
		and colorless.
5.	Free Acid and free	alkali : Dissolve 2.0 g of the sample in 20 mL of freshly boiled and cooled water. Add 2
		drops of phenolphthalein TS, and keep the solution at 10°C. If a colorless solution
		is produced, not more than 0.1 mL of 0.1 N sodium hydroxide should be required
		to give a red color. If a red color is produced, not more than 0.1 mL of 0.1 N
		hydrochloric acid should be required to discharge it.
6.	Alkalinity	: Not more than $0.05\%$ of Na <sub>2</sub> CO <sub>3</sub> .
7.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Test for	: Mix 5 mL of the saturated solution of the sample with a few drops of acid sodium
	potassium	hydrogen tartrate TS. No turbidity occurs within 5 minutes.
10.	Loss on drying	: 36.0~42.0% (120°C, 4 h).
11.	Category	: Food Additives Category 7
12.	Uses	: Food quality improvement, fermentation and food processing agents

### Sodium Acetate (Anhydrous)

Chemical formula: C <sub>2</sub> H <sub>3</sub> N <sub>8</sub>		aO <sub>2</sub> Molecular weight: 82.03
1.	Assay	: Not less than 98.5% of $C_2H_3NaO_2$ on the dried basis.
2.	Identification	: (1) Dissolve 1 g of the sample in 20 mL of water. The solution passes the test for
		sodium and test for acetate.
		(2) When heating the sample slowly, it first fuses gradually and boils, and later
		decomposes evolving an unpleasant odor of acetone. A solution of the residue gives
		alkaline reaction with litmus paper.
3.	Description	: White, odorless, granular, hygroscopic powder. Miscible with water.
4.	Solution	: When 1.0 g of the sample is dissolved in 20 mL of water, the solution should be clear.
5.	Free acid and free a	Ikali : Dissolve 1.2 g of the sample in 20 mL of freshly boiled and cooled water. Add 2
		drops of phenolphthalein TS, and keep the solution at 10o. If a colorless solution is
		produced, not more than 0.1 mL of 0.1 N sodium hydroxide should be required to
		give a pink color. If a pink color is produced, not more than 0.1 N hydrochloric
		acid should be required to discharge it.
6.	Alkalinity	: Not more than $0.05\%$ of Na <sub>2</sub> CO <sub>3</sub> .
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Test for	: Mix 5 mL of the saturated solution of the sample with a few drops of acid sodium
	potassium	hydrogen tartrate TS. No turbidity occurs within 5 minutes.
10.	Loss on drying	: Dry it at 120° for 4 hours: it loses not more than 2.0% of its weight.
11.	Category	: Food Additives Category 7
12.	Uses	: Food quality improvement, fermentation and food processing agents

### Glycerol

Chemical formula: C <sub>3</sub> H <sub>8</sub> O <sub>3</sub>		Molecular weight: 92.10
1.	Assay	: Not less than 95%
2.	Identification	: Add 2 to 3 drops of the sample to 0.5 g of potassium hydrogen sulfate. After heating,
		a pungent acrolein odor is produced.
3.	Description	: Clear, colorless, hygroscopic, syrupy liquid, having a not more than a slight
		characteristic odor, which is neither harsh nor disagreeable. Miscible with water and
		with ethanol; immiscible with chloroform, ether and grease.
4.	рН	: The aqueous solution of the sample should be neutral.
5.	Density	: 1.250~1.264.
6.	Chloride	: Not more than 0.003% of Cl.
7.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
8.	Heavy metals	: Not more than 5 ppm (as Pb).
9.	Fatty acids and este	ers : Not more than 0.1% (as butyric acid)
10.	Acrolein, glucose a	and ammonium : Mix and heat 5mL of the sample and 5 mL of potassium hydroxide
	salt	TS (1 g of potassium hydroxide dissolved in 10 mL of water) at
		60°C for 5 minutes. No yellow color or an odor of ammonia is
		produced.
11.	Residue on	: Not more than $0.01\%$ ( $800\pm25^{\circ}$ to constant weight).
	ignition	
12.	Category	: Food Additives Category 7, 15.
13.	Uses	: Food quality improvement, fermentation and food processing agents; carrier.

#### Sodium Stearyl 2-Lactylate

- Description : A mixture of sodium salts of stearoyl lactylic acids and minor proportions of other salts of related acids, formed by the esterification of stearic acid with lactic acid and neutralized to the sodium salts. White or slightly yellowish powder or brittle solid with a characteristic odor. Slightly hygroscopic, having a mild, caramel-like odor. Soluble in hot grease or dispersed in warm water. The fatty acids used should comply with the food additives specifications.
- Identification : A. Add 1 g of the sample to the mixture of 25 mL of water and 5 mL of hydrochloric acid. When heated, the fatty acids are released and a layer of oil appears on the surface of the liquid.

The water layer passes test for sodium.

B. Take 25 g of the sample to triangular flask, add 15% potassium hydroxide alcohol solution 50g. The mixture was heated to a minimum of 1 hour or until saponification was completed. After cooling, add 150 mL of water and mix.

After the soap dissolves completely, add 60 mL dilute sulfuric acid TS. Heat the mixture and keep stirring until the fatty acids separate completely to form a transparent layer. The fatty acids were washed with boiling water until there was no sulfate reaction. Collect the fatty acid in a small beaker. Heat the water in a steam bath until the fatty acid is separated from the water. After fatty acid is cooled and solidified, dispose water, and then dissolve fatty acid. Filter to a dry beaker, and dry at 105°C for 20 minutes. The fatty acids obtained has a freezing point that is above 54 °C.

- 3. Acid value  $:60 \sim 80.$
- 4. Ester value  $: 150 \sim 190.$
- 5. Sodium content  $: 3.5 \sim 5.0\%$ .
- 6. Total lactic acid  $: 31.0 \sim 34.0\%$ .
- 7. Arsenic : Not more than 3 ppm (as As).
- 8. Heavy metals : Not more than 10 ppm (as Pb).
- 9. Category : Food Additives Category 7

10. Uses : Food quality improvement, fermentation and food processing agents

§ 07043

Aluminum silicate

- 1. Synonyms : Kaolin
- 2. Definition : A native hydrated aluminum silicate, freed from most of its impurities by elutriation and dried.
- 3. Appearance : A soft, whitish powder free from gritty particles; odorless.
- 4. Identification : (1) Test for silicate: Mix about 500 mg of the sample with about 200 mg of anhydrous sodium carbonate and 2 g of anhydrous potassium carbonate, and heat the mixture in a platinum or nickel crucible until it melts completely. Cool, add 5 mL of water, and allow to stand for 3 min. Heat the bottom of the crucible gently, detach the melt, and transfer it to a beaker with the aid of about 50 mL of water. Add gradually hydrochloric acid until no effervescence is observed, then add 10 mL more of the acid, and evaporate the mixture on a steam bath to dryness. Cool, add 20 mL of water, boil and filter the mixture through an ash-free filter paper. An insoluble residue of silica remains. (Note: Retain the filtrate for the test for aluminum). Transfer the gelatinous residue into a platinum dish, and cautiously add 5 mL of hydrofluoric acid (Warning: toxic, corrosive, must not contact skin; work under fume hood). The precipitate dissolves. (If it does not dissolve, repeat the evaporation with hydrofluoric acid.) Heat and hold in the vapors a glass stirring rod with a drop of water on the tip. The drop becomes turbid. (2) Test for aluminum: Add ammonia TS to the filtrate obtained in the test for silicate. A white gelatinous precipitate is formed which is insoluble in excess ammonia but dissolves in sodium hydroxide TS. 5. Solubility : Insoluble in water, ethanol and mineral acids. 6. Plasticity : To 8 g of the sample add 5 mL of water and mix well. The mixture is plastic. 7. Water soluble : Not more than 13%.
- 8. Acid-soluble : Not more than 2%. matter
- 9. Asbestos : Absent

substances

- 10. Lead : Not more than 5 mg/kg.
- 11. Category : Food Additives Category 7
- 12. Uses : Food quality improvement, fermentation and food processing agents

### Diatomaceous Earth

1.	Identification	: (1) Place 0.2 g of the sample in a platinum crucible, and add 5 mL of hydrofluoric acid.
		The sample is dissolved. When the solution is heated, almost all of it volatilizes.
		(2) When examined with 100x to 200x microscope, typical diatom shapes are observed.
2.	Description	: White, light grey or reddish brown powder. Insoluble in water, in acids (except
		hydrofluoric), and in dilute alkalis.
3.	Water soluble	: Boil 10 g of sample with 100 mL of water for 2 h and filter through a
	substances and pH	fineporosity sintered-glass filter (aperture: 0.45 $\mu$ m, diameter: 47mm). Dilute
		filtrate to 100 mL. pH range of the solution is 5.0 to 11.0. Take 50 mL of the
		filtrate and evaporate it. Dry the residue is dried at 105 °C for 2 hours, and the
		amount of residue should be less than 25 mg.
4.	Hydrochloric	: Not more than 3%.
	acid-soluble	
	substances	
5.	Arsenic	: Not more than 4 ppm (as As).
6.	Heavy metals	: Not more than 50 ppm (as Pb).
7.	Lead	: Not more than 10 ppm.
8.	Hydrofluoric acid-	: Not more than 25%.
	insoluble substance	s
9.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 10% of its weight.
10.	Loss on ignition	: Not more than 7.0% on the dry basis (1000°C, 30 min)
11.	Category	: Food Additives Category 7, 17.
12.	Uses	: Food quality improvement, fermentation and food processing agents; others.

Talo	2

- Identification : Mix 0.2 g sample with 0.9 g of anhydrous sodium carbonate and 1.3 g of anhydrous potassium carbonate, and heat the mixture in a platinum crucible until fusion is complete. Cool, and transfer the fused mixture to a dish or beaker with the aid of about 5 mL of hot water. Add hydrochloric acid to the liquid until effervescence ceases, then add 20 mL more of the acid, and evaporate the mixture on a steam bath to dryness. Cool, add 20 mL of water, boil and filter the mixture, an insoluble residue of gel remains. The filtrate passes test for magnesium.
- 2. Description : Odorless, very fine, white or grayish white, crystalline powder. Do not dissolve in water and alkali metal hydroxide solution, but slightly soluble in dilute mineral acid.
- Water-soluble : Boil a 10 g sample with 100 mL of water for 2 hours and from time to time, add substances
   Water to maintain approximately the original volume, and filter through a fineporosity sintered-glass filter (aperture: 0.45 μm, diameter: 47mm). Dilute filtrate to 100 mL. pH range of the solution is 6.0 to 8.0. Take 50 mL of the filtrate and evaporate it. Dry the residue is dried at 105 °C for 2 hours, and the amount of residue should be less than 10 mg. (less than 0.2%)
- 4. Hydrochloric : Not more than 2.5% of SO<sub>4</sub>.

acid-soluble

substances

- 5. Arsenic : Not more than 4 ppm (as  $As_2O_3$ ).
- 6. Heavy metals : Not more than 40 ppm (as Pb).
- 7. Lead : Not more than 10 ppm.
- 8. Free alkali : Not more than 1% (as NaOH)
- 9. Extractable : Not more than 0.002%.

amount of

fluorine

- 10.
   Water-soluble
   : Slightly acidify with hydrochloric acid the remaining half of the 20 mL filtrate obtained

   iron
   in the test for water-soluble substances (3.) and add 1 mL of potassium ferrocyanide TS.

   The solution does not turn blue.
- 11. Water-soluble : Not more than 0.2%.

substances

12. Loss on drying : Not more than 0.5% (105°C, 1 h).

- 13. Loss on ignition : Not more than 6.0% (550°C, to a constant weight).
- 14. Category : Food Additives Category 7
- 15. Uses : Food quality improvement, fermentation and food processing agents

### L-Cystein Monohydrochloride

The same as § 03012.

### Sodium Ferrocyanide

Chemical	: Sodium ferrocyanide, Sodium hexacyanoferrate (II)		
names			
Synonyms	: Yellow prussiate of soda; hexacyanoferrate of sodium; IN	S No. 535; CAS	No. 13601-19-9
Chemical	: Na <sub>4</sub> Fe(CN) <sub>6</sub> $\cdot$ 10H <sub>2</sub> O	Molecular	: 484.1
formula		weight	

1.	Assay	: Not less than 99.0%.
2.	Appearance	: Yellow crystalline powder.
3.	Solubility	: Soluble in water, insoluble in ethanol.
4.	Identification	: (1) Test for ferrocyanide: To 10 mL of a 1% solution of the sample add 1 mL of ferric
		chloride TS. A dark blue precipitate is formed.
		(2) Passes test for calcium.
5.	Cyanide	: Not detectable.
6.	Prussiate	: Not detectable.
7.	Arsenic	: Not more than 3 mg/kg.
8.	Lead	: Not more than 5 mg/kg.
9.	Category	: Food Additives Category 7
10.	Uses	: Food quality improvement, fermentation and food processing agents

Disodium Ethylenediaminetetraacetate (EDTA Na2)

The same as § 03017.

Calcium Disodium Ethylenediaminetetraacetate (EDTA CaNa2)

The same as § 03017.

### Silicon Dioxide (Synthetic Amorphous Silica)

Chemical formula: SiO <sub>2</sub>		Molecular weight: 60.08
1.	Description	<ul> <li>Silicon dioxide is an amorphous substance when scanned with Xray diffraction, which is produced synthetically by either a thermal process, yielding pyrogenic (fumed) silica, or by a wet process, yielding hydrated silica, precipitated silica and silica gel. Pyrogenic silica is produced in an essentially anhydrous state, whereas the wet process products are obtained as hydrates or contain surface absorbed water.</li> <li>Pyrogenic (fumed) silica: A pyrogenic silicon dioxide occurring as a fine, white amorphous powder or granules. Hydrated silica (precipitated silica and silica gel): A precipitated, hydrated silicon dioxide occurring as a fine, white amorphous powder or granules. Deliquescent or capable of adsorbing unequal amounts of moisture from the air. All cerium oxide products are insoluble in water and in organic solvents, but soluble in hydrofluoric acid and in high concentration of hot lye.</li> </ul>
2.	Identification	<ul> <li>: A. Mix about 5 mg of the sample with 200 mg of anhydrous potassium carbonate in a platinum crucible. Ignite for 10 minutes until red and hot, cool down. Dissolve the melt in 2 mL of fresh distilled water (heat if necessary). Then slowly add 2 mL of ammonium molybdate TS. The solution is dark yellow.</li> <li>B. Take 1 drop of the solution of the Test A on the filter paper. After evaporation of the solvent, add 1 drop of o-toluidine/ent, add 1 drop of o-toluidineththen put the filter paper on the ammonia TS. Blue-green spots will appear.</li> </ul>
3.	Assay	<ul> <li>Pyrogenic (fumed) silica: Not less than 99.0% of SiO<sub>2</sub> on the ignited basis.</li> <li>Hydrated silica (precipitated silica and silica gel): Not less than 94.0% of SiO<sub>2</sub> on the ignited basis.</li> </ul>
4.	Arsenic	: Not more than 3 ppm (as As).
5.	Lead	: Not more than 10 ppm.
6.	Heavy metals	: Not more than 0.003% (as Pb).
7.	Loss on drying	: Pyrogenic (fumed) silica: Not more than 2.5%.
		Precipitated silica and silica gel: Not more than 7%
		Hydrated silica Not more than 70%
8.	Loss on ignition	: Pyrogenic (fumed) silica: Not more than 2.5% (on dried sample)
		Hydrated silica (precipitated silica and silica gel): Not more than 8.5% (on dried sample)
9.	Dissolvable salts	: Precipitated silica, silica geland hydrated silica: Not more than 5% (as
		Na <sub>2</sub> SO <sub>4</sub> ).

- 10. Category : Food Additives Category 7
- 11. Uses : Food quality improvement, fermentation and food processing agents

### Calcium Oxide

Chemical formula: CaO	Molecular weight: 56.08
1. Description	: Odorless, white or greyish white masses or granules, or white to greyish white powder.
	1 g of the sample is soluble in 840 mL of 25°C water and 1740 mL of boiling water;
	insoluble in ethanol, soluble in glycerol.
2. Identification	: Shake 1 g of the sample with 20 mL of water, add acetic acid until it dissolves. The
	solution passes test for calcium.
3. Assay	: Not less than 95.0% after ignition.
4. Acid insoluble	: Not more than 1%.
matter	
5. Magnesium and	: Not more than 3.6%.
alkali salts	
6. Fluoride	: Not more than 0.005%.
7. Arsenic	: Not more than 3 ppm (as As).
8. Lead	: Not more than 10 ppm (as Pb).
9. Heavy metals	: Not more than 0.004%.
10. Loss on ignition	: Not more than 10%.
11. Category	: Food Additives Category 7, 8.
12. Uses	: Food quality improvement, fermentation and food processing agents; Nutritional
	additives

### Potassium Bicarbonate

Chemical formula: KHCO		Molecular weight: 100.12
1.	Description	: Odorless, colorless crystals or white powder or granules, stable in air. 1 g of the sample
		is soluble in 2.8 mL of water; insoluble in ethanol. The solution has a neutral or alkaline
		reaction to the phenolphthalein TS. 1 g of the sample is soluble in 2.8 mL of water and is
		insoluble in alcohol.
2.	Identification	: The 1 in 10 solution of the sample passes test for potassium and test for carbonate.
3.	Assay	: Not less than 99.0% and not more than 101% calculated on the dried basis.
4.	Normal carbonate	: Moderate.
5.	Arsenic	: Not more than 3 ppm (as As).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Loss on drying	: Not more than 0.25%.
8.	Category	: Food Additives Category 7
9.	Uses	: Food quality improvement, fermentation and food processing agents

### Glycerol Ester of Wood Rosin

1.	Description	: The sample is obtained by esterifying of wood rosin with food grade glycerin, and then
		purifying it by steam distillation (when used as chewing gum and bubble gum base) or
		steam convection distillation (when used as beverage stabilizer). Hard, yellow to pale
		amber-colored solid. (If measured by ASTM D509, the chroma is N or slightly lighter)
		Soluble in acetone and benzene, but insoluble in water.
2.	Identification	: The infrared absorbance spectrum of the sample melted and prepared on a caesium or
		potassium bromide plate corresponds to the infrared absorption spectrum and the
		standard map.
3.	Acid value	: 3~9.
4.	Drop Softening Poi	nt : $88 \sim 96^{\circ}$ C.
5.	Arsenic	: Not more than 3 ppm (as As).
6.	Lead	: Not more than 3 ppm.
7.	Heavy metals	: Not more than 0.004% (as Pb).
8.	Category	: Food Additives Category 7
9.	Uses	: Food quality improvement, fermentation and food processing agents

Patroleum Wax (Refined Paraffin Wax, Microcrystalline Wax)

- Description : This product is a paraffin solid hydrocarbon mixture from refined paraffin, including refined paraffin and microcrystalline paraffin. The refined paraffin wax is usually obtained from the low molecular weight of petroleum. The molecular weight, ignition point and melting point are lower than that of microcrystalline paraffin, and the viscosity is smaller after melting The refined paraffin wax is usually obtained from the low molecular weight of petroleum. The molecular weight, ignition point and melting point are lower than that of microcrystalline paraffin, and melting point are lower than that of microcrystalline paraffin, and melting point are lower than that of microcrystalline paraffin, and the viscosity is smaller after melting. Colorless or white, translucent, tasteless and odorless wax. Melting point is between 48 ~ 93°C (120 ~ 200°F). The melting point and color are used as the basis for classification commercially.
   Identification : The infrared absorbance spectrum of the sample melted and prepared on a caesium or
- potassium bromide plate corresponds to the infrared absorption spectrum and the standard map.
- 3.Ultraviolet absorbance:  $280 \sim 289$  nm, not more than 0.15.limits $290 \sim 299$  nm, not more than 0.12.

 $300 \sim 359$  nm, not more than 0.08.

 $360 \sim 400$  nm, not more than 0.02.

- 4. Arsenic : Not more than 3 ppm (as As).
- 5. Lead : Not more than 3 ppm.
- 6. Heavy metals : Not more than 0.002% (as Pb).
- Color, melting point and : Should meet the specifications set by the seller.
   odor
- 8. Category : Food Additives Category 7, 17.
- 9. Uses : Food quality improvement, fermentation and food processing agents; others.

### Rice Bran Wax

1.	Description	: Refined wax from rice bran. Yellowish brown to light tan, hard and having
		microcrystalline. Soluble in chloroform and benzene, but insoluble in water.
2.	Identification	: The infrared absorbance spectrum of the sample melted and prepared on a caesium or
		potassium bromide plate corresponds to the infrared absorption spectrum and the
		standard map.
3.	Free fatty acid	: Not more than 10%.
4.	Iodine value	: Not more than 20.
5.	Saponification	: 75∼120.
	value	
6.	Melting range	: 75~80°C.
7.	Arsenic	: 3 ppm (As)
8.	Lead	: Not more than 3 ppm.
9.	Heavy metals	: Not more than 0.004% (as Pb).
10.	Category	: Food Additives Category 7
11.	Uses	: Food quality improvement, fermentation and food processing agents

### Stearic Acid (Octadecanoic Acid)

Chemical formula: C <sub>18</sub> H <sub>36</sub>	<sub>6</sub> O <sub>2</sub> Molecular weight: 284.48
1. Description	: Mixture of solid organic acids obtained from fat, mainly containing stearic acid
	$(C_{18}H_{36}O_2)$ and palmitic acid $(C_{16}H_{32}O_2)$ . White or yellow, hard and slightly smooth
	crystal solid or powder that has a characteristic odor of butter. Insoluble in water. 1 g
	of the sample is soluble in about 20 mL of ethanol, in 2 mL of chloroform or in 3 mL
	of ether.
2. Acid value	: 196~211.
3. Iodine value	: Not more than 7
4. Saponification	: 197~212.
value	
5. Unsaponifiable	: Not more than 1.5%.
matter	
6. Water	: Not more than 0.2%.
7. Titer	: 54.5~69°C.
8. Arsenic	: 3 ppmu (as As).
9. Heavy metals	: Not more than 10 ppm (as Pb).
10. Residue on	: Not more than 0.1%.
ignition	
11. Category	: Food Additives Category 7
12. Uses	: Food quality improvement, fermentation and food processing agents

### Adipic Acid

# (Hexanedioic Acid; 1,4-Butanedicarboxylic Acid)

# HOOC(CH<sub>2</sub>)<sub>4</sub> COOH

Chemical formula: C <sub>6</sub> H <sub>10</sub> O		D <sub>4</sub> Molecular weight: 146.14
1.	Description	: White crystals or crystalline powder that is not deliquescent; slightly soluble in water;
		freely soluble in ethanol; soluble in acetone.
2.	Assay	: Not less than 99.6% and not more than 101.0%.
3.	Water	: Not more than 0.2%.
4.	Melting range	: 151.5∼154.0°C.
5.	Arsenic	: 3 ppmu (as As).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Residue on	: Not more than 0.002%.
	ignition	
8.	Category	: Food Additives Category 7
9.	Uses	: Food quality improvement, fermentation and food processing agents

### Aluminum Sulfate

Chemical formula: Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> ·xH <sub>2</sub> O		D <sub>4</sub> ) <sub>3</sub> ·xH <sub>2</sub> O Molecular weight (anhydrous): 342.14	
1.	Description	: Odorless, white powder, shining plates, or crystalline fragments, with a slightly	
		astringent taste. The molecular structure can be in anhydrous form or hydrated form that	
		contains 18 molecules of water. Due of the weathering of the hydrated sample, its	
		molecular formula may be $Al_2(SO_4)_3$ ·14H <sub>2</sub> O. 1 g of hydrated form is soluble in about 2	
		mL of water. The solubility of anhydrous sample is similar to that of hydrated sample.	
		However, the dissolution rate is slow that it seems insoluble at the beginning. The pH of	
		1 in 20 solution of the sample is 2.9 or higher.	
2.	Identification	: The 1 in 10 solution of the sample passes the test for aluminum and test for sulfate.	
3.	Assay	: $Al_2(SO_4)_3$ (anhydrous) – Not less than 99.5% on the ignited basis	
		$Al_2(SO_4)_3 \cdot 18H_2O$ (hydrated) $-99.5 \sim 114.0\%$ . If calculate by $Al_2(SO_4)_3 \cdot 14H_2O$ , the	
		limit is about 101.7%.	
4.	Alkali metal and al	kaline earth : Moderate.	
	metal		
5	Ammonium salt	: Moderate.	
6.	Fluorine ion	: Not more than 0.003%.	
7.	Selenium	: Not more than 0.003%.	
8.	Arsenic	: 3 ppm (as As)	
9.	Lead	: Not more than 10 ppm.	
10.	Heavy metals	: Not more than 0.004% (as Pb).	
11.	Loss on ignition A	$I_2(SO_4)_3$ (anhydrous) : Not more than 5%.	
[Note: Not applicable to Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> ·18H <sub>2</sub> O ]			
12.	Category	: Food Additives Category 7	
10	T		

13. Uses : Food quality improvement, fermentation and food processing agents

### Perlite (Expanded Perlite)

1.	Description	: Natural perlite is gray to brown, thick, glassy igneous rock. Primary component is
		sodium potassium aluminum silicate. The sample contains 3% to 5% of water. After
		being crushed and heated to a temperature of 900 to 1100°C, it pops like a popcorn
		(due to the sorption of water inside), and the volume expands more than 20 times. This
		product is white and non-hygroscopic powder obtained by crushing this expanded
		material. The bulk density is $32 \sim 400 \text{ kg/m}^3$ ( $2 \sim 25 \text{ lb/ft}^3$ ). The particle size ranges
		from less than 1 $\mu$ m to hundreds of $\mu$ m. The sample is used as a filter aid in food
		processing. Slightly soluble in water and almost insoluble in dilute acids and bases.
2.	Identification	: A. Take about 1 g of the sample and place it in a beaker. Mix with 25 mL of dilute
		hydrochloric acid TS, cover the mouth of the cup with a surface glass. After
		filtration, the filtrate was neutralized by ammonia solution until it is neutral on
		litmus paper. The neutralized filtrate passes test for aluminum, test for potassium
		and test for sodium.
		B. Place some ammonium phosphate crystal on the platinum ring, and dissolve into
		beads with flame. Contact this hot, transparent bead with the sample, reheat it
		with flame to dissolve. The silica will float inside the bead. When cooled, it a
		reticular, opaque bead forms.
3.	pН	: $5 \sim 9$ (filtrate of 10% suspension)
4.	Arsenic	: Not more than 3 ppm (as As).
5.	Lead	: Not more than 10 ppm.
6.	Loss on drying	: Not more than 3% (powder)
7.	Loss on ignition	: Not more than 7% (glass)
8.	Category	: Food Additives Category 7
9.	Uses	: Food quality improvement, fermentation and food processing agents

### Hydroxypropyl Cellulose

1.	Description	: The sample contains hydroxypropyl cellulose ether compound. White powder. Soluble
		in water and some organic solvents. May contain some anti-adhesive.
2.	Identification	: A. Prepare a 0.1% solution of the sample. A layer of foam appears after oscillation.
		(distinguished from sodium carboxymethyl cellulose)
		B. Take an appropriate amount of the sample to prepare a 0.5% aqueous solution. Add
		5 mL of 5% copper sulfate or aluminum sulfate solution to 5 mL of the solution.
		No precipitation appears. (distinguished from sodium carboxymethyl cellulose)
3.	Assay	: Not more than 80.5% of hydroxypropyl (-OCH2CHOHCH3) on the dried basis, which
		is equivalent to 4.6 hydroxypropyl/anhydrous glucose units or less.
4.	рН	: $5.0 \sim 8.0$ (1% solution)
5.	Viscosity of 10% s	olution : Not less than 145 centipoises
6.	Arsenic	: Not more than 3 ppm (as As).
7.	Lead	: Not more than 10 ppm.
8.	Heavy metals	: Not more than 0.004% (as Pb).
9.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 5% of its weight
10.	Residue on	: Not more than 0.5%.
	ignition	
11.	Category	: Food Additives Category 7, 12, 16.
12.	Uses	: Food quality improvement, fermentation and food processing agents; pasting (binding)
		agent.; emulsifier
<ol> <li>4.</li> <li>5.</li> <li>6.</li> <li>7.</li> <li>8.</li> <li>9.</li> <li>10.</li> <li>11.</li> </ol>	pH Viscosity of 10% s Arsenic Lead Heavy metals Loss on drying Residue on ignition Category	<ul> <li>is equivalent to 4.6 hydroxypropyl/anhydrous glucose units or less.</li> <li>: 5.0~8.0 (1% solution)</li> <li>olution : Not less than 145 centipoises</li> <li>: Not more than 3 ppm (as As).</li> <li>: Not more than 10 ppm.</li> <li>: Not more than 0.004% (as Pb).</li> <li>: Dry it at 105°C for 3 hours: it loses not more than 5% of its weight</li> <li>: Not more than 0.5%.</li> </ul> Food Additives Category 7, 12, 16. Food quality improvement, fermentation and food processing agents; pasting (binding)

### Hydroxypropyl Methylcellulose

# (Propylene Glycol Ether of Methylcellulose)

1.	Description	: The sample is propylene glycol methyl cellulose, in which both hydroxypropyl and
		methyl ether bond on anhydrous glucose ring of cellulose. There are several types
		depending on the content and combination of methoxyl and hydroxypropyl. Hygroscopic
		white or off-white powder, or granules or fine fibers. Swells in water and some organic
		solvent, producing a clear to opalescent, viscous colloidal solution; insoluble in ethanol.
2.	Identification	: A. When 1 g of this product is added to 100 mL water, it absorbs water and expands to
		form a clear milky adhesive solution (depending on its inherent viscosity), which
		can coexist with most electrolytes.
		B. Add 1 g of the sample into 100 mL of boiling water, which appears mushy after
		stirring. When cooled to 20°C, the solution becomes a clear or milky viciousness
		solution.
		C. If the solution prepared in the above identification test B is put on the glass plate for
		the moisture to vaporize, a self-sustaining thin film forms.
3.	Assay	: Not less than 19.0% and not more than 30.0% of methyl groups (-OCH_3) and not less
		than 3.0% and not more than 12.0% hydroxypropoxy groups (-OCH <sub>2</sub> CHOHCH <sub>2</sub> ).
4.	Viscosity	: For products with a viscosity value of 100 centipoises or less, when 2 g is dissolved into
		a 100 g solution, the viscosity should be between 80% and 120% of the indicated
		value. For those with a value greater than 100 centipoises, the viscosity of the above
		solution should be between 75% and 140% of the indicated value.
5.	Arsenic	: Not more than 3 ppm (as As).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 5% of its weight.
8.	Residue on	: Not more than 1.5% for products with viscosities of 50 centipoise or above, and not
	ignition	more than 3.0% for products with viscosities below 50 centipoise.
9.	Category	: Food Additives Category 7, 12, 16.
10.	Uses	: Food quality improvement, fermentation and food processing agents; pasting (binding)
		agent.; emulsifier

### Polydextrose

- Description : Polydextrose is obtained by melting and condensation of the ingredients which consist of approximately 89 parts D-glucose, 10 parts sorbitol and up to 1-part citric acid. White to light tan-colored solid, odorless and a little sour. Very soluble in water but slightly soluble or insoluble in most organic solvents.
- Polymer of molecular weight greater than : Negative 22,000
   Arsenic : Not more than 3 ppm (as As).
- 4. Heavy metals : Not more than 10 ppm (as Pb).
- 5. Residue on : Not more than 13%. ignition
- 6. Category : Food Additives Category 7, 12.
- 7. Uses : Food quality improvement, fermentation and food processing agents; pasting (binding) agent.

### Food Gypsum

Chen	nical formula: CaSO4	Molecular weight: 172.18
1.	Assay	: Not less than 93% on the dried basis.
2.	Description	: This product is made from bittern. White powder.
3.	Solution	: 0.2 g of the sample is heated and dissolved in 10 mL of dilute hydrochloric acid, and
		solution turbidity should be below "almost clear".
4.	Free alkali	: Dissolve 0.5 g of the sample in 100 mL of freshly boiled and cooled water, mix and
		filter. Take 10 mL of the solution and add a drop of phenolphthalein TS. The red color
		should not appear.
5.	Chloride	: Not more than 3% of Cl.
6.	Carbonate	: No bubbles shall be generated when 0.5 g of the sample is added with 5 mL of dilute
		hydrochloric acid.
7.	Arsenic	: Not more than 1 ppm (as As).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Loss on ignition	: Not more than 40%.
10.	Category	: Food Additives Category 7
11.	Uses	: Food quality improvement, fermentation and food processing agents

## Acid Clay

Definition	Acid Clay is obtained by purifying clay(such as montmorillonite clay, bentonite clay,	
	etc.) . It consists mainly of hydrous aluminum silicate.	
Description	Acid Clay occurs as a grayish-white to yellow-brown powder or as granules.	
Identification	(1) Mix 1.0 g of Acid Clay with 3.0 g of sodium carbonate and 0.4 g of boric acid, and	
	heat the mixture in a platinum or nickel crucible until it melts completely. Cool, and	
	then add hydrochloric acid until no effervescence is observed. Add an additional 10	
	mL of hydrochloric acid, heat on a water bath until the mixture becomes gelatinous,	
	cool, and then filter. The filtrate obtained responds to all the tests for Aluminum Salt	
	as directed in the Qualitative Tests.	
	(2) To a 100 mL measuring cylinder containing 100 mL of water, add 2.0 g of Acid	
	Clay in small portions, and allow to stand for 24 hours. The precipitate formed is not	
	more than 15 mL.	
pH	$4.0 \sim 10.0$	
Purity		
Water-soluble substances	Not more than 0.50%	
Lead	Not more than 40 mg/kg	
Arsenic	Not more than 3 mg/kg	
Loss on Ignition	Not more than 35.0% (at 110°C for 3 hours, then at 550°C for 3 hours).	
Category	Food additives category 7	
Functional uses	Food quality improvement, fermentation and food processing agents	

## Disodium Dihydrogen Pyrophosphate

Chemical formula: Na <sub>2</sub> H <sub>2</sub> P <sub>2</sub> O <sub>7</sub>		P2O7Molecular weight: 221.95
1.	Assay	: Not less than 95% (110°C, 4 h)
2.	Appearance	: White crystalline powder .
3.	Water-insoluble	: Dissolve 5 g of the sample in 100 mL of water, shake occasionally for an hour, and
	matter	filter through a tared filtering crucible (1G4). Wash the insoluble residue with 30 mL of
		water, dry at 110°C for 2 h, cool and weigh. The weight should not be more than 40 mg.
4.	pH	: 3.8~4.5 (1 in 100 soln)
5.	Chloride	: Not more than 0.057% (as Cl).
6.	Sulfate	: Not more than 0.038% of SO <sub>4</sub> .
7.	Test for	: Add 2 to 3 drops of 2% silver nitrate TS to 1 g of the sample. An obvious yellow color
	orthophosphate	should not appear.
8.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Loss on drying	: Dry it at 110° for 4 hours: it loses not more than 5% of its weight.
11.	Category	: Food Additives Category 7
12.	Uses	: Food quality improvement, fermentation and food processing agents

## Carnauba Wax

1.	Description	Description : The refined wax obtained from the fronds of the Brazilian tropical palm tree Copernicia	
		cerifera (Arruda) Mart. A pale yellow to light brown, hard and brittle solid, having a	
		clean fracture. The density is around 0.997. Insoluble in water; partially soluble in	
		boiling ethanol; soluble in ether.	
2.	Acid value	: 2~7.	
3.	Arsenic	: Not more than 3 ppm (as As).	
4.	Ester value	: Between 75 and 85.	
5.	Heavy metals	: Not more than 40 ppm (as Pb).	
6.	Lead	: Not more than 10 ppm.	
7.	Melting range	: 82~86°C.	
8.	Unsaponifiable	: Between 50 % and 55 %.	
	matter		
9.	Category	: Food Additives Category 7	
10.	Uses	: Food quality improvement, fermentation and food processing agents	

## Calcium Chloride Anhydrous

nical formula: CaCl <sub>2</sub>	Molecular weight: 110.99
Assay	: Not less than 93.0%.
Description	: White, deliquescent, odorless lumps or porous masses. Freely soluble in water and
	ethanol.
Identification	: The 1 in 20 solution of the sample passes test for calcium and test for chloride.
Solution	: The turbidity of the 1 in 20 solution of the sample should be "slightly turbid".
pН	: 4.5~8.5 (1 in 20 soln)
Free Acid and free	alkali : Dissolve 1 g of the sample in 20 mL of freshly boiled and cooled water, and
	add 2 drops of phenolphthalein TS. If the solution is colorless, the pink color
	appears after adding 2.0 mL of 0.02 N sodium hydroxide. If the solution is pink,
	the pink color is discharged by adding 2 mL of 0.02 N hydrochloric acid.
Fluoride	: Not more than 40 ppm.
Magnesium and	: The weight of the residue does not exceed 25 mg.
alkali salts	
Lead	: Not more than 10 ppm.
Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).
Heavy metals	: Not more than 20 ppm (as Pb).
Category	: Food Additives Category 7
Uses	: Food quality improvement, fermentation and food processing agents
	Description Identification Solution pH Free Acid and free Fluoride Magnesium and alkali salts Lead Arsenic Heavy metals Category

## Sodium Trimetaphosphate

Chen	nical formula: (NaPO	Molecular weight: 305.92
1.	Assay	: Not less than 68.0% and not more than 70.0% of $P_2O_5$ .
2.	Description	: White crystal or crystalline powder. Soluble in water, The 1 in 100 soultion of the
		sample is around 6.0.
3.	Identification	: (1) The 1 in 20 solution of the sample passes test for sodium.
		(2) Dissolve 100 mg of the sample in 5 mL of hot dilute nitric acid TS, heat the solution
		in a steam bath for 10 minutes, then cool, and add sodium hydroxide TS to make
		it a neutral on litmus paper. Then add silver nitrate TS, a yellow precipitate
		forms. This precipitate can be dissolved in dilute nitric acid TS.
4.	Fluoride	: Not more than 0.005%.
5.	Arsenic	: Not more than 3 ppm (as As).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Insoluble	: Not more than 0.1%.
	substances	
8.	Category	: Food Additives Category 7
9.	Uses	: Food quality improvement, fermentation and food processing agents

## Azodicarbonamide

Chemical formula: C <sub>2</sub> H <sub>4</sub> N <sub>4</sub> O <sub>2</sub>		M <sub>4</sub> O <sub>2</sub> Molecular weight: 116.08
1.	Assay	: Not less than 98.6% on the dried basis.
2.	Description	: Yellow to orange-red, odorless, crystalline powder. Practically insoluble in water and in
		most organic solvents; slightly soluble in dimethyl sulfoxide. Melts with decomposition
		when the temperature is above 180°C.
3.	Identification	: Dissolve 35 mg of the sample in 1000 mL of water. The solution has the maximum
		absorption value of 245 nm.
4.	Arsenic	: Not more than 3 ppm (as As).
5.	Heavy metals	: Not more than 0.003% (as Pb).
6.	Lead	: Not more than 10 ppm.
7.	Loss on drying	: Not more than 0.5%.
8.	Nitrogen	: 47.2%~48.7%.
9.	pН	: Not less than 5.0 (1 in 50 suspension)
10	Residue on	: Not more than 0.15%.
	ignition	
11.	Category	: Food Additives Category 7
12.	Uses	: Maturing agent for flour

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# Benzoyl Peroxide

Chemical formula: C <sub>14</sub> H <sub>10</sub> O <sub>4</sub>		O <sub>4</sub> Molecular weight: 242.23
1.	Assay	: Not less than 96.0%.
2.	Description	: Colorless, crystalline solid having a faint odor of benzaldehyde. Insoluble in water,
		slightly soluble in ethanol and soluble in benzene, chloroform and ether. Melts with
		decomposition between 103°C and 106°C. (Caution: Benzoyl peroxide, especially in the
		dry form, is a dangerous, highly reactive, oxidizing material and has been known to
		explode spontaneously. Please read the safety instructions on the package before use.)
3.	Identification	: To 500 mg of the sample add 50 mL of 0.5 N ethanolic potassium hydroxide, heat
		gradually to boiling and continue boiling for 15 min. Cool and dilute with 200 mL of
		water. Add sufficient 0.5 N hydrochloric acid to make strongly acidic and extract with
		ether. Dry the ether solution over anhydrous sodium sulfate, and then evaporate to
		dryness on a steam bath. The benzoic acid so obtained melts between 121°C and 123°C.
4.	Arsenic	: Not more than 3 ppm (as As).
5.	Heavy metals	: Not more than 0.004% (as Pb).
6.	Lead	: Not more than 10 ppm.
7.	Category	: Food Additives Category 4, 7
0	<b>T</b> T	

8. Uses : Bleaching agents; Food quality improvement, fermentation and food processing agents.

#### Cross-Linked Sodium Carboxymethyl Cellulose

Synonyms: Cross–linked sodium CMC, croscarmellose sodium, INS No.468 Chemical formula: –  $(C_6H_7O_2(OR_1)(OR_2)(OR_3))_n -$ 

where R<sub>1</sub>, R<sub>2</sub>, and R<sub>3</sub> represent the following groups, present in varying proportions:

#### -H

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-CH2COONa
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-CH2COOH

- 1. Definition : Cross-linked sodium carboxymethyl cellulose is the sodium salt of thermally crosslinked partly O-carboxymethylated cellulose. It is manufactured by acidifying an aqueous suspension of sodium carboxymethyl cellulose and heating the suspension to achieve cross-linking. The product is then washed and dried. It is also produced during the manufacture of sodium carboxymethyl cellulose by lowering the pH and heating to achieve crosslinking.
- 2. Description : A slightly hygroscopic, white to greyish-white, odorless powder.
- 3. Identification : (1) Solubility: Practically insoluble in acetone, in ethanol and in toluene.

(2) Color reaction: Add 0.5 g of the powdered sample to 50 mL water, while stirring to produce a uniform dispersion. Dilute 1 mL of this mixture with 1 mL of water in a small test tube and add 5 drops of 1-naphthol TS. Incline the test tube, and carefully introduce down the side of the tube 2 mL of sulfuric acid so that it forms a lower layer. A red-purple color develops at the interface.

(3) Precipitate formation: Mix 1 g of the powdered sample with 100 mL of solution containing 4 mg/kg of methylene blue in water and allow to settle. The substance absorbs methylene blue and settles as a blue, fibrous mass.

(4) Test for sodium: Passes test

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4. pH : Not less than 5.0 and not more than 7.0 (1 in 100 suspension in water).
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- 5. Loss on drying : Dry it at 105°C for 3 hours: it loses not more than 6% of its weight.
- 6. Sulfated ash : Not less than 14.0 % and not more than 28.0 % on the dried basis (2 g of sample).
- 7. Water soluble : Not more than 10%. substances
- 8. Degree of : Not less than 0.2 and not more than 1.50 carboxymethyl groups (CH<sub>2</sub>COOH) per substitution anhydroglucose unit on the dried basis.

9. Sodium chloride and sodium glycolate

: Not more than 0.5%, in combination (on

the dried basis)

- 10. Lead : Not more than 2 mg/kg.
- 11. Category : Food Additives Category 7
- 12. Uses : Food quality improvement, fermentation and food processing agents

## Sodium γ-Polyglutamate

Chemical formula: (C5H6NNaO3)n

	H <sub>2</sub> N CC	$H_{2}$ $H_{2$
1.	Extent of	: About 100 to 20,000 glutamic acid molecules.
	polymerization	
2.	Assay	: Not less than 70% on the dried basis.
3.	Description	: White or off-white granular or powder, odorless, tasteless. Soluble in water; insoluble
		in alcohol or other organic solvents. 1 g of the sample is soluble in 5 mL of water.
4.	Identification	: (1) Add 6 mol/L hydrochloric acid to 5 mL of sodium polyglutarate aqueous solution (1
		g of sodium polyglutamate dissolved in 200 mL of water), heat at 110°C for 24
		hours for acid hydrolysis, and adjust the solution to neutral with NaOH. Add 1
		mL of ninhydrin reagent (1 in 1000 solution). After heating for 5 minutes, the
		solution is blue-purple.
		(2) The FT–IR analysis of sodium glutamate 1–2 mg showed a weak absorption near
		1600 $\text{cm}^{-1}$ and a strong absorption near 1410 $\text{cm}^{-1}$ .
5.	pН	: 4.0~7.0.
6.	Heavy metals	: Not more than 15 ppm (as Pb).
7.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
8.	Loss on drying	: Not more than 5% (Infrared moisture analyzer, 105°C, to constant weight)
9.	Ash	: Not more than 30% (2 $\sim$ 3 g of sample, 650°C).
10.	Category	: Food Additives Category 7
11.	Uses	: Food quality improvement, fermentation and food processing agents

#### Polyvinylpyrrolidone

General name : Povidone, PVP

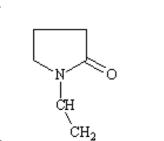
Chemical names : Polyvinylpyrrolidone, poly-[1-(2-oxo-1-pyrrolidinyl)-ethylene]

Chemical formula: (C<sub>6</sub>H<sub>9</sub>NO)<sub>n</sub>

Molecular weight: About 40,000 for low molecular

n

weight product, and about 360,000 for high molecular weight product.



- 1. Assay
   : Not less than 12.2 % and not more than 13.0% nitrogen calculated on the anhydrous basis.
- 2. Appearance : White to tan powder.

3. Solubility : Soluble in water, alcohol and chloroform; insoluble in ether.

- 4. pH : 3.0-7.0 (5% soln)
- 5. 沉澱物形態 : (1) Add 5 mL of dilute hydrochloric acid liquid, 5 mL of water and 2 mL of potassium dichromate solution (1 g of potassium dichromate dissolved in 10 mL of water) to 5 mL of solution of the sample (1 g of the sample in 50 mL of water). Yellow sediment forms.
  - (2) Add 2 mL of cobalt nitrate and ammonium rhodanide solution (75 mg of cobalt nitrate and 0.3 g of ammonium rhodanide dissolved in 2 mL of water) to 5 mL of the solution of the sample (1 g of the sample dissolved in 50 mL of water). After mixing, acidify the solution with dilute hydrochloric acid, generating light blue precipitate.
  - (3) Add 1 mL of 25% hydrochloric acid, 5 mL of 5% of lanthanum chloride aqueous solution and 1 mL of 5% phosphonium acid aqueous solution to 5 mL of the solution of the sample (1 g of the sample dissolved in 50 mL of water). Bulky white precipitate forms and gradually turns blue under sunlight. The characteristics of the becoming blue under illumination can be used to distinguish polyvinylpyrrolidone and polyethylene oxide. The latter becomes white by the same treatment.

155

6.	Water	: Not more than 5% (Karl Fischer Method)	
7.	Relative viscosity : Low molecular weight products: 1.188-1.325; High molecular weight		
		3.225-5.662 (Measured by Cannon-Fenske capillary viscometer)	
8.	Total ash	: Not more than 0.02% (Test 10 g of the sample)	
9.	Aldehydes	: Not more than 0.2% (as aldehydes)	
10.	Monomer content	: Not more than 1% of vinylpyrrolidone.	
11.	Hydrazine	: Not more than 1 mg/kg.	
12.	Lead	: Not more than 2 mg/kg (Determine using an atomic absorption technique)	
13.	Category	: Food Additives Category 7	
14.	Uses	: Food quality improvement, fermentation and food processing agents	

#### Calcium Stearate

1.	Assay	: Not less than 95% on the dried basis.
2.	Description	: White or white-yellow, slightly glossy crystalline solid, semi-solid or powder.
3.	Solubility	: Insoluble in water, ethanol and ether.
4.	Identification	: Passes test for cation and test for fatty acid.
5.	Free fatty acid	: Not more than 3%.
6.	Unsaponifiable	: Not more than 2%.
	matter	
7.	Lead	: Not more than 2 mg/kg.
8.	Loss on drying	: Not more than 4.0% (105°, constant weight, then 2 h)
9.	Category	: Food Additives Category 7
10.	Uses	: Food quality improvement, fermentation and food processing agents

## Potassium Ferrocyanide

Chemical	: Potassium ferrocyanide, Potassium hexacyanoferrate (II)		
names			
Synonyms	: Yellow prussiate of potash; hexacyanoferrate of potassium	n; INS No. 536; (	CAS No. 13943-58-3
Chemical	: $K_4Fe(CN)_6 \cdot 3H_2O$	Molecular	: 422.4
formula		weight	

1.	Assay	: Not less than 99.0%.
2.	Appearance	: Yellow crystalline powder.
3.	Solubility	: Soluble in water, insoluble in ethanol.
4.	Identification	: (1) Test for ferrocyanide: To 10 mL of a 1% solution of the sample add 1 mL of ferric
		chloride TS. A dark blue precipitate is formed.
		(2) Passes test for potassium.
5.	Cyanide	: Not detectable.
6.	Prussiate	: Not detectable.
7.	Arsenic	: Not more than 3 mg/kg.
8.	Lead	: Not more than 5 mg/kg.
9.	Category	: Food Additives Category 7
10.	Uses	: Food quality improvement, fermentation and food processing agents

## Calcium Ferrocyanide

Chemical	: Calcium ferrocyanide,		
names	Calcium hexacyanoferrate (II)		
Synonyms	: Yellow prussiate of lime; hexacyanoferrate of calcium; IN	S No. 538; CAS I	No. 1327-39-5
Chemical	: $Ca_2Fe(CN)_6 \cdot 12H_2O$	Molecular	: 508.3
formula		weight	

1.	Assay	: Not less than 99.0%.
2.	Appearance	: Yellow crystalline powder.
3.	Solubility	: Miscible with water.
4.	Identification	: (1) Test for ferrocyanide: To 10 mL of a 1% solution of the sample add 1 mL of ferric
		chloride TS. A dark blue precipitate is formed.
		(2) Passes test for calcium.
5.	Cyanide	: Not detectable.
6.	Prussiate	: Not detectable.
7.	Arsenic	: Not more than 3 mg/kg.
8.	Lead	: Not more than 5 mg/kg.
9.	Category	: Food Additives Category 7
10.	Uses	: Food quality improvement, fermentation and food processing agents

§ 11-1-001

#### **D-Sorbitol Synonyms** INS No. 420(i); D-Glucitol, D-sorbitol, sorbit, sorbol Definition Chemical names **D-Sorbitol** CAS NUMBER 50-70-4 $C_6H_{14}O_6$ Molecular formula 182.17 Molecular weight Not less than 97.0% of C<sub>6</sub>H<sub>14</sub>O<sub>6</sub> of total glycitols and not less than 91.0% of Assay D-sorbitol on the anhydrous basis. The term glycitols refers to compounds with the structural formula CH<sub>2</sub>OH-(CHOH)<sub>n</sub>-CH<sub>2</sub>OH, where n is an integer less than or equal to 4. Appearance White hygroscopic powder, crystalline powder, flakes or granules. Characteristics Identification Solubility Very soluble in water, slightly soluble in ethanol. 88~102°C Melting range Thin layer chromatography Passes test Purity Water Not more than 1% (Karl Fischer Method). Sulfated ash Not more than 0.1%. Chloride Not more than 50 mg/kg. Sulfate Not more than 100 mg/kg. Not more than 2 mg/kg. Nickel Reducing sugar Not more than 0.3%. Not more than 1% (as glucose). Total sugars Lead Not more than 1 mg/kg. Food Additives Category 11-1, 7. Category Sweetener, Food quality improvement, fermentation and food processing Uses

agents

Xylitol

## D-Mannitol

Maltitol

Maltitol Syrup

## Isomalt (Hydrogenated Palatinose)

Lactitol

Erythritol

	Triethyl citrate
	$CH_2 - COOC_2H_5$
	$CH_2 - COOC_2H_5$ HO $-C - COOC_2H_5$ $H_2 - COOC_2H_5$ $CH_2 - COOC_2H_5$
	$L_{\rm CH_2-COOC_2H_5}$
Synonyms	: Ethyl citrate;
	INS No. 1505;
	CAS No 77-93-0
Chemical names	Triethyl 2-hydroxypropan-1,2,3-tricarboxylate
Chemical formula:	C <sub>12</sub> H <sub>20</sub> O <sub>7</sub> Molecular weight: 276.29
1. Description	: Odorless, practically colorless, oily liquid. Slightly soluble in water; miscible with
	ethanol and ether.
2. Assay	: Not less than 99% w/w.
3. Index of ref	raction : $n_{\rm D}^{20} = 1.439 - 1.441$ .
4. Density	: 1.135-1.139.
5. Water	: Not more than 0.25% (Karl Fischer Method).
6. Acidity	: Not more than 0.02% w/w (as citric acid)
7. Lead	: Not more than 2 mg/kg.
8. Category	: Food Additives Category 7
9. Uses	: Food quality improvement, fermentation and food processing agents

### Nitrous oxide

SYNONYMS	Dinitrogen oxide; Dinitrogen monoxide; INS No. 942
Definition	Nitrous oxide, a colourless and non-flammable gas, commonly known as laughing gas, is
	manufactured by thermal decomposition processes of ammonium nitrate. The hot,
	corrosive mixture of gases are cooled to condense the steam and filtered to remove higher
	oxides of nitrogen. The gas is further purified in a train of three gas washes with base, acid
	and base again. Nitric oxide impurity, if present, is chelated out with ferrous sulfate, or
	reduced with iron metal, or oxidised and absorbed in a base as a higher oxide.
Chemical names	Nitrous oxide
C.A.S. number	9000-69-5
Chemical formula	N <sub>2</sub> O
Formula weight	44.01
Assay	Not less than 99 % (v/v)
DESCRIPTION	Colourless, odourless gas
Characteristics	
IDENTIFICATION	
Solubility	1 volume dissolves in 1.4 volumes of water (20° 760 mm Hg). Freely soluble in alcohol;
	soluble in ether and in oils.
Infrared absorption or	Corresponds with the nitrous oxide standard
Chromatography	
PURITY	
Carbon dioxide	Not more than $0.03\%(v/v)$
Carbon monoxide	Not more than 10 µl/l
Nitric oxide	Not more than 1 µl/l
Nitrogen dioxide	Not more than 1 µl/l
Halogens (as Cl)	Not more than 5 $\mu$ l/l
Ammonia	Not more than 25 $\mu$ l/l
Category	Food additives category (07).
Functional uses	Food quality improvement, fermentation and food processing agents.

## **Carbon Dioxide**

SYNONYMS	INS No. 290
Definition	
C.A.S. number	124-38-9
Chemical formula	CO <sub>2</sub>
Formula weight	44.01
Assay	Not less than 99.5% $(v/v)$
DESCRIPTION	Colourless, odourless gas, 1 litre of which weighs about 1.98 g at 0°C and 760 mm of
	mercury. Under a pressure of about 59 atmospheres it may be condensed to a liquid, a
	portion of which forms a white solid (Dry Ice) upon rapid evaporation. Solid carbon
	dioxide evaporates without melting upon exposure to air. Commercial carbon dioxide is
	shipped and handled in pressurized cylinders or low pressure bulk liquid systems, or in
	solid blocks.
Characteristics	The following specifications apply to gaseous carbon dioxide as produced from its
	condensed liquid or solid phase by evolution to the gas phase at normal environmental
	conditions.
IDENTIFICATION	
Precipitate formation	When a stream of the sample is passed through a solution of barium hydroxide, a white
	precipitate is produced which dissolves with effervescence in dilute acetic acid.
Detector tube test	Passes test
PURITY	
Acidity	Transfer 50 ml of water, previously boiled and cooled to room temperature, into a
	Nessler tube. Introduce 1,000 ml of the sample into the water through a tube (1 mm
	internal diameter) keeping the opening of the tube within 2 mm from the bottom of the
	vessel. Add 0.1 ml of methyl orange TS. The red colour produced is not darker than the
	colour of an identical control solution to which has been added 1.0 ml of 0.01 N
	hydrochloric acid instead of the carbon dioxide.

170

Phosphine, hydrogen sulfide,	Transfer 25 ml of silver ammonium nitrate TS and 3 ml of ammonia TS into a Nessler
and other organic reducing	tube. In the absence of light, introduce 1,000 ml of the sample in the same manner as in
substances	the test of Acidity. No brown colour is produced.
Carbon monoxide	Not more than $10 \ \mu L/L$
Non-volatile hydrocarbons	Not more than 10 mg/kg
Volatile hydrocarbons	Not more than 50 $\mu$ L/L
Water	Passes test
Category	Food additives category (07).
Functional uses	Food quality improvement, fermentation and food processing agents.

## Nitrogen

Synonyms	INS No. 941
Definition	
Chemical name	Nitrogen
C.A.S. number	7727-37-9
Chemical formula	N <sub>2</sub>
Formula weight	28.0
Assay	Not less than 99.0% v/v
Description	Colourless, odourless gas or liquid
Characteristics	
Identification	
Flame test	A flame is extinguished in an atmosphere of the sample.
Purity	
Oxygen	Not more than 1% v/v
Carbon monoxide	Not more than 10 µl/l
Category	Food Additives Category 7
Uses	Food quality improvement, fermentation and food processing agents

## Activated Acid Clay

Definition	Activated Acid Clay is obtained by treating acid clay with sulfuric acid. Its principal
	constituent is hydrous aluminum silicate.
Description	Activated Acid Clay occurs as a whitish to gray powder or as granules.
Identification	Mix 1.0 g of Activated Acid Clay with 3.0 g of sodium carbonate and 0.4 g of boric acid,
	and heat the mixture in a platinum or nickel crucible until it melts completely. Cool, and
	add hydrochloric acid until no effervescence is observed. Add an additional 10 mL of
	hydrochloric acid, and heat on a water bath until the mixture becomes gelatinous, cool,
	and filter. The filtrate obtained responds to all the tests for Aluminum Salt as directed in
	the Qualitative Tests.
pН	2.0 ~ 6.0
Purity	
Water-soluble substances	Not more than 1.6%
Lead	Not more than 40 mg/kg
Arsenic	Not more than 3 mg/kg
Loss on Ignition	Not more than 35% (at 110°C for 3 hours, then at 550°C for 3 hours).
Category	Food additives category 7
Functional uses	Food quality improvement, fermentation and food processing agents

# Category 8 Nutritional additivesNutritional additives

## § 08001

# Dry Formed Vitamin A

1.	Assay	: Powder processed from vitamin A oil, fatty acid or vitamin A ester oil. 1 g of the
		sample contains 6 to 150 mg Vitamin A. The content should be 100% to 120% of the
		indicated content. (Vitamin A 150 mg=500,000 I.U.)
2.	Appearance	: Light yellow to light auburn powder.
3.	Spoilage	: This product must not have a strange smell.
4.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
5.	Heavy metals	: Not more than 20 ppm (as Pb).
6.	Loss on drying	: Not more than 5% (vacuum desiccator over sulfuric acid, 4 h)
7.	Residue on	: Not more than 5%.
	ignition	
8.	Category	: Food Additives Category 8
9.	Uses	: Nutritional additivesNutritional additives

## Vitamin A Oil

1.	Assay	: 1 g of the sample contains 30 to 300 mg of Vitamin A. The content should be 100% to
		110% of the indicated content. (Vitamin A 300 mg=1,000,000 I.U.)
2.	Appearance	: Yellow to slightly reddish orange liquid, having a characteristic odor.
3.	Acid value	: Not more than 2.8
4.	Chloroform-	: No insoluble matter appears when 0.5 g of the sample is dissolved in 3 mL of
	insoluble matter	chloroform.
5.	Category	: Food Additives Category 8
6.	Uses	: Nutritional additivesNutritional additives

#### Vitamin A Fatty Acid Ester, in Oil

- 1. Appearance : Colorless to slightly reddish light yellow grease, having a peculiar smell.
- 2. Acid value : Not more than 1.96
- : Dissolve 100 mg of the sample in petroleum ether, making 1 mL of it equivalent to 3. Alcohol type Vitamin Α 100-200 IU to be used as a test liquid. The coloring layer of about 5 cm is suspended in a chromatography tube with a suspension of petroleum ether. Note that the aluminum glue should be immersed in petroleum ether usually. After a small piece of cotton is placed on the aluminum layer, charge 10 mL of petroleum ether. The petroleum ether is discharged at the rate of 30 drops per minute. When the layer is about 1 cm above the aluminum adhesive layer, add 5~10 mL of the test solution to the chromatographic tube. When the liquid level of the liquid to be tested reaches 1 cm above the aluminum adhesive layer, the ester type vitamin A is discharged at the same speed with petroleum ether containing 5% ether. In the same way, the alcoholic vitamin A was dissolved in petroleum ether containing 50% diethyl ether. The movement and dissolvent of the ester type vitamin A and the alcohol type vitamin A in the coloring layer can be observed by ultraviolet light having a wavelength of about 350 nm in a short time. If the chromatographic layer is cracked or bubbled, the experiment should be re-operated. This test should be carried out below normal temperature.
  - Continuously distill the alcoholic vitamin A solution obtained by the above method by nitrogen or carbon dioxide gas into a water bath at about 70°C to evaporate the petroleum ether.
  - Dissolve the residue in isopropanol immediately to formulate a 1 mL, which corresponds to about 10 IU. Measure the absorbance at wavelengths of 310 nm, 325 nm and 334 nm. When the content of alcoholic Vitamin A was calculated according to the following formula, the content should be below 10%.
    Alcohol type Vitamin A content = a / b × 100 (%)

Where a: International unit of alcoholic Vitamin A obtained by quantitative method (I.U.);

b: International unit of total Vitamin A obtained by quantitative method (I.U.).All the petroleum ether, ether, aluminum gel and isopropyl alcohol used are required to be tested with Vitamin A.

176

- 4. Category : Food Additives Category 8
- 5. Uses : Nutritional additives Nutritional additives

## Thiamine Hydrochloride (Vitamin B1)

Chen	nical formula: C <sub>12</sub> H <sub>17</sub>	ON <sub>4</sub> ClS·HCl Molecular weight: 337.29
1.	Assay	: Not less than 98% and not more than 102% on the dried basis (105°C, 2 h)
2.	Appearance	: Small white to yellowish crystals or crystalline powder; odorless or with a slight
		characteristic odor.
3.	Solution	: Dissolve 1 g of the sample in water to make a 10 mL solution, The color of the solution
		should not be thicker than the solution made from diluting $1.5 \text{ mL pf } 0.1 \text{ N}$ potassium
		dichromate solution to 1,000 mL.
4.	pH of the solution	: 2.7~3.4.
5.	Sulfate	: Not more than $0.05\%$ of SO <sub>4</sub> .
6.	Nitrate	: Dissolve 1 g of the sample in 50 mL of water. Oscillate 2 mL of sulfuric acid with 2 mL
		of the previous solution. After cool and add 2 mL of horizon ferrous sulfate solution,
		the interface shall not show a brown ring.
7.	Hydrobromide	: Dissolve 7 mg of the sample in 0.1 mL of water. Add 1 drop of the fluorescent red TS
		to 1 drop of the solution made in the previous step, 1 drop of the same amount of
		glacial acetic acid and hydrogen peroxide, stir and then dry on the water bath. The
		residue should not be red.
8.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 5% of its weight.
9.	Residue on	: Not more than 0.2%.
	ignition	
10.	Category	: Food Additives Category 8
11.	Uses	: Nutritional additivesNutritional additives

### Thiamine Mononitrate

Chemical formula: C <sub>12</sub> H <sub>17</sub> O <sub>4</sub> N <sub>5</sub> S		O <sub>4</sub> N <sub>5</sub> S Molecular weight: 327.37
1.	Assay	: Not less than 98% and not more than 102% on the dried basis (105°C, 2 h)
2.	Appearance	: White crystalline powder, odorless or with a slight characteristic odor.
3.	pН	: 6.5~8.0 (1 in 50 soln)
4.	Chloride	: Not more than 0.05% of Cl.
5.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 1% of its weight.
6.	Residue on	: Not more than 0.2%.
	ignition	
7.	Category	: Food Additives Category 8
8.	Uses	: Nutritional additivesNutritional additives

## Riboflavin (Vitamin B<sub>2</sub>)

Chemical formula: C <sub>17</sub> H <sub>20</sub> O <sub>6</sub> N <sub>4</sub>		Molecular weight: 376.37
1.	Assay	: Not less than 98% after drying at 105°C for 2 hours.
2.	Appearance	: Yellow to orange-yellow crystalline powder, with slight odor.
3.	Specific rotation	: $(\alpha)_{D}^{20} = -115 \sim -140^{\circ}$ Dry the sample at 100°C for 4 h. Dissolve 50.0 mg in 0.05 N
		sodium hydroxide free from carbonate and dilute to 10.0 mL with the same solvent.
		Measure the optical rotation within 30 min of dissolution.
4.	Lumiflavin	: Prepare the standard for this test for the absence of lumiflavin by diluting 3 mL of 0.1 N $$
		potassium dichromate with water to 1000 mL. Pour some chloroform through an alumina
		column to remove any ethanol. To 10 mL of this chloroform add 35 mg of the sample,
		shake for 5 min and filter. The color of the filtrate should be no more intense than that of
		10 mL of the standard when viewed in identical containers.
5.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 1.5% of its weight.
6.	Residue on	: Not more than 13%.
	ignition	
7.	Category	: Food Additives Category 8, 9.
8.	Uses	: Nutritional additivesNutritional additives; Colors

### Riboflavin Phosphate Sodium

Chemical formula:  $C_{17}H_{20}O_9N_4NaP-0 \sim 2H_2O$ 

1.	Assay	: Not less than 98% (vacuum desiccator over sulfuric acid, 4 h)
2.	Appearance	: Yellow to orange crystalline hygroscopic powder, with slight odor and a bitter taste.
3.	Solution	: When 0.2 g of the sample is dissolved in 10 mL of water, the solution should be clear.
4.	Specific rotation	: $(\alpha)_{D}^{20} = +38.0 \sim +43.0^{\circ}$ (Dissolve 0.3 g of the sample in 5 N hydrochloric acid to
		make a 20 mL solution)
5.	Absorbance	: Dissolve 1 g of the sample in 100,000 mL of solution. The ratio of absorbance at 260
		nm to 375 nm is $2.60\pm0.15.260$ nm, the ratio to 445 nm is $2.21\pm0.15.375$ nm, and the
		ratio of 445 nm is 0.85± 0.05.
6.	Lumiflavin	: Prepare the standard for this limit test for the absence of lumiflavin by diluting 3 mL of
		0.1 N potassium dichromate with water to 1000 mL. Pour some chloroform through an
		alumina column to remove any ethanol. To 10 mL of this chloroform add 35 mg of the
		sample, shake for 5 min and filter. The color of the filtrate should be no more intense
		than that of 10 mL of the standard when viewed in identical containers.
7.	Water	: Not more than 12% (Karl Fischer Method)
8.	Category	: Food Additives Category 8, 9.
9.	Uses	: Nutritional additivesNutritional additives; Colors

## Pyridoxine Hydrochloride (Vitamin B<sub>6</sub>)

Chemical formula: C <sub>8</sub> H <sub>11</sub> O <sub>3</sub>		Ŋ₃N·HCl	Molecular weight: 205.64
1.	Assay	: Not less than 98% (vacuum de	esiccator over sulfuric acid, 4h)
2.	Appearance	: White to yellowish crystals or	crystalline powder; odorless.
3.	Heavy metals	: Not more than 30 ppm (as Pb)	).
4.	Loss on drying	: Not more than 0.5% (vacuum	desiccator over sulfuric acid, 4 h)
5.	Residue on	: Not more than 0.1%.	
	ignition		
6.	Category	: Food Additives Category 8	
7.	Uses	: Nutritional additivesNutrition	al additives

### Cyanocobalamin (Vitamin B<sub>12</sub>)

1.	Assay	: Not less than 95%.
2.	Appearance	: Dark red crystals or crystalline powder; odorless and tasteless.
3.	Solubility	: Soluble in water and ethanol; insoluble in acetone, chloroform or ether.
4.	Loss on drying	: Not more than 12% (105°C using vacuum, approx. 5 mm Hg, 2 h)
5.	Category	: Food Additives Category 8
6.	Uses	: Nutritional additivesNutritional additives

L-Ascorbic Acid (Vitamin C)

#### Sodium L-Ascorbate

## L-Ascorbyl Stearate

## L-Ascorbyl Palmitate

## Calciferol (Vitamin D<sub>2</sub>)

Chemical formula: C <sub>28</sub> H <sub>44</sub>		O Molecular weight: 396.66
1.	Appearance	: White, odorless crystal.
2.	Melting range	: 115 $\sim$ 118°C (vacuum desiccator over sulfuric acid, 3 h)
3.	Specific rotation	: $(\alpha)_{D}^{20} = +102.0 \sim +107.0^{\circ}$ (Dissolve 0.3 g of the sample in ethanol to make 20 mL
		solution.)
4.	Specific	: Dissolve the sample in Idehyde-free ethanol. The absorbance at a wavelength of 265
	absorbance	nm, E(1%, 1cm) is 445 to 485.
5.	Ergosterol	: Dissolve 10 mg of the sample in 2 mL of 90 v/v% ethanol, add the solution that
		contains 20 mg of and 2 mL of 90 v/v% ethanol. After 18 hours of standing, no
		precipitation forms.
6.	Category	: Food Additives Category 8
7.	Uses	: Nutritional additivesNutritional additives

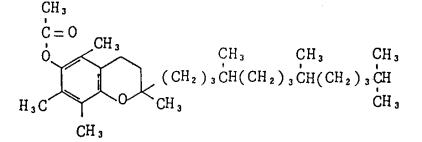
*dl*-α-Tocopherol (Vitamin E)

Tocopherols Concentrate, Mixed

## *d*-α-Tocopherol Concentrate

#### Chemical formula: C<sub>31</sub>H<sub>52</sub>O<sub>3</sub>

Molecular weight: 472.75



- Description : A form of Vitamin E that is made from edible vegetable oil by vacuum distillation and acetylation. Colorless to yellow, transparent and viscous oil, almost odorless. Coagulate easily when it is standing. Melts at about 25°C. Insoluble in water; soluble in ethanol, and miscible with ether, acetone, chloroform and vegetable oil; instable under alkaline condition.
- Identification : A. Prepare a test solution of the sample according to "specific optical rotation measurement". Take 10 mL of the solution, add 2 mL of nitric acid while stirring, and heat at about 75°C for 15 minutes. The solution is bright red to orange.
  - B. The retention time of the major peak in the chromatogram of the Assay Preparation is the same as that of the Standard Preparation, both relative to the internal standard.
- 3. Assay :  $C_{31}H_{52}O_3$  96.0~102.0%.
- 4. Acidity : Moderate.
- 5. Lead : Not more than 10 ppm.
- 6. Heavy metals : Not more than 0.004% (as Pb).
- 7. Specific rotation : Not less than  $(\alpha)_{D}^{25} = +24^{\circ}$
- 8. Category : Food Additives Category 8
- 9. Uses : Nutritional additives Nutritional additives

Chemical formula: C <sub>31</sub> H <sub>52</sub> O <sub>3</sub>		<sub>52</sub> O <sub>3</sub> Molecular weight: 472.75
	1	H <sub>3</sub> = 0 CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> CH
1.	Description	: A form of Vitamin E. Colorless to yellowish green, transparent and sticky oil, nearly
		odorless. Insoluble in water; soluble in ethanol, and miscible with ether, acetone,
		chloroform and vegetable oil; instable under alkaline condition.
2.	Identification	: A. Prepare a test solution of the sample according to "specific optical rotation
		measurement". Take 10 mL of the solution, add 2 mL of nitric acid while stirring,
		and heat at about 75°C for 15 minutes. The solution is bright red to orange.
		B. The retention time of the major peak in the chromatogram of the Assay Preparation is
		the same as that of the Standard Preparation, both relative to the internal standard.
		C. The specific optical rotation of 1 in 10 chloroform solution of the sample is about
		$\pm 0.05^{\circ}$ (almost unmeasurable).
3.	Assay	: Not less than 96.0% and not more than 102.0% of $C_{31}H_{52}O_{3.}$
4.	Acidity	: Moderate.
5.	Lead	: Not more than 10 ppm.
6.	Heavy metals	: Not more than 0.004% (as Pb)
7.	Category	: Food Additives Category 8
8.	Uses	: Nutritional additivesNutritional additives

### *d*-α-Tocopheryl Acetate Concentrate

1.	Description	: This product is a form of Vitamin E, which is prepared from edible vegetable oil by
		vacuum distillation and acidification. The content of d- $\alpha$ -tocopheryl acetate can be
		adjusted by appropriate physical or chemical methods. Light yellowish brown,
		transparent and viscous oil, almost odorless. Insoluble in water; soluble in ethanol, and
		miscible with ether, acetone, chloroform and vegetable oil; instable under alkaline
		condition.
2.	Identification	: A. Prepare a test solution of the sample according to "specific optical rotation
		measurement". Take 10 mL of the solution, add 2 mL of nitric acid while stirring,
		and heat at about 75°C for 15 minutes. The solution is bright red to orange.
		B. The retention time of the major peak in the chromatogram of the Assay Preparation
		is the same as that of the Standard Preparation, both relative to the internal
		standard.
3.	Assay	: Not less than 40.0% <i>d</i> - $\alpha$ -tocopheryl acetate (C <sub>31</sub> H <sub>52</sub> O <sub>3</sub> ).
4.	Acidity	: Moderate.
5.	Lead	: Not more than 10 ppm.
6.	Heavy metals	: Not more than 0.004% (as Pb).
7.	Specific rotation	: Not less than $\left( \alpha \right) {}_{D}^{25} = +24^{\circ}$
8.	Category	: Food Additives Category 8
9.	Uses	: Nutritional additivesNutritional additives

## d-α-Tocopheryl Acid Succinate

Chemical formula: C <sub>33</sub> H <sub>54</sub>		O <sub>5</sub> Molecular weight: 530.79
		$\begin{array}{c} \text{COOH} \\ (\text{CH}_2)_2 \\ \text{C} = 0 \end{array}$
		O CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub>
		$H_{3}C \qquad (CH_{2})_{3}CH(CH_{2})_{3}CH(CH_{2})_{3}CH \\ CH_{3} \qquad CH_{3}$
1.	Description	: A type of <b>OHt</b> amin E, which is made from edible vegetable oil by vacuum distillation
		and acylation of succinic acid. White to off-white crystalline powder, almost odorless,
		tasteless. Melts at about 75°C. Insoluble in water; soluble in ethanol, ether, acetone
		and vegetable oil; very soluble in chloroform. Stable in air, but not stable under
		alkaline conditions or heating.
2.	Identification	:A. Prepare a test solution of the sample according to "specific optical rotation
		measurement". Take 10 mL of the solution, add 2 mL of nitric acid while stirring,
		and heat at about 75°C for 15 minutes. The solution is bright red to orange.
		B. The retention time of the major peak in the chromatogram of the Assay Preparation is
		the same as that of the Standard Preparation, both relative to the internal standard.
3.	Assay	: Not less than 96.0% and 102.0% of $C_{33}H_{54}O_5$
4.	Acidity	: Moderate.
5.	Lead	: Not more than 10 ppm.
6.	Heavy metals	: Not more than 0.004% (as Pb).
7.	Specific rotation	: Not less than $\left( \alpha \right)_{D}^{25} = +24^{\circ}$
8.	Category	: Food Additives Category 8
9.	Uses	: Nutritional additivesNutritional additives

§ 08025

## Nicotinic Acid

Chemical formula: C <sub>6</sub> H <sub>5</sub> O <sub>2</sub> N		2N Molecular weight: 123.11
1.	Assay	: Not less than 99.5% on the dried basis (105°C, 1 h)
2.	Appearance	: White crystal or crystalline powder, odorless, with slight sour taste.
3.	Melting point	: 234~237°C.
4.	Chloride	: Not more than 0.02% of Cl.
5.	Sulfate	: Not more than 0.02% of SO <sub>4.</sub>
6.	Heavy metals	: Not more than 20 ppm (as Pb).
7.	Loss on drying	: Dry it at 105°C for 1 hour: it loses not more than 1% of its weight.
8.	Residue on	: Not more than 0.1%.
	ignition	
9.	Category	: Food Additives Category 8
10.	Uses	: Nutritional additivesNutritional additives

#### Nicotinamide

Chemical formula: C <sub>6</sub> H <sub>6</sub> ON <sub>2</sub>		N <sub>2</sub> Molecular weight: 122.13
1.	Assay	: Not less than 98.5% on the dried basis (over sulfuric acid, 4 h)
2.	Appearance	: White crystalline powder, odorless, bitter.
3.	Melting range	: 128~131°C.
4.	Heavy metals	: Not more than 30 ppm (as Pb).
5.	Readily	: When 0.2 g of the sample is tested by readily carbonized substance measurement, the
	carbonizable	color of the solution should not be thicker than Matching Fluid A.
	substance	
6.	Loss on drying	: Not more than 0.5% (over sulfuric acid, 4 hr)
7.	Residue on	: Not more than 0.1%.
	ignition	
8.	Category	: Food Additives Category 8
9.	Uses	: Nutritional additivesNutritional additives

## Folic Acid

Chemical formula: C19H19O6N7		O <sub>6</sub> N <sub>7</sub> Molecular weight: 441.40
1.	Assay	: Not less than 95.0% and not more than 102.0% (By high performance liquid
		chromatography)
2.	Appearance	: Yellow to orange crystalline powder, odorless.
3.	Water	: Not more than 8.5% (Karl Fischer Method)
4.	Residue on	: Not more than 0.5%.
	ignition	
5.	Category	: Food Additives Category 8
6.	Uses	: Nutritional additivesNutritional additives

Calcium Oxide

#### Calcium Carbonate

### Iron, Reduced

Chemical formula: Fe		Molecular weight: 55.85
1.	Assay	: Not less than 96.0%
2.	Description	: This product is an elemental iron obtained by chemical methods. Grayish black, matt to
		slightly lustrous powder. Can pass through the 100 mesh screen. When viewed under a
		microscope at 100x magnification, the sample is an amorphous powder that does not
		have a crystalline structure.
3.	Identification	: When dissolved in dilute mineral acid, the sample releases hydrogen chloride and
		produces corresponding iron salt solution. Passes test for ferrous salts.
4.	Acid insoluble	: Not more than 1.25%
	matter	
5.	Arsenic	: Not more than 8 ppm (as As).
6.	Lead	: Not more than 0.0025%.
7.	Mercury	: Not more than 5 ppm.
8.	Category	: Food Additives Category 8
9.	Uses	: Nutritional additivesNutritional additives

## Ferric Pyrophosphate (Iron Pyrophosphate)

Chemical formula: Fe <sub>4</sub> (P <sub>2</sub> O <sub>7</sub> ) <sub>3</sub> ·xH <sub>2</sub> O		O <sub>7</sub> ) <sub>3</sub> ·xH <sub>2</sub> O Molecular weight: 745.22 (Anhydrous)
1.	Description	: Tan or yellowish-white, odorless powder. Insoluble in water but soluble in mineral
		acids.
2.	Identification	: Dissolve 500 mg of the sample in 5 mL of dilute hydrochloric acid $(1\rightarrow 2)$ , add an
		excess of sodium hydroxide TS to form a reddish brown precipitate. After allowing the
		solution to stand for a few minutes, filter it and discard the first few milliliters of the
		filtrate. Take 5 mL of clear filtrate, add 1 drop of bromophenol blue TS, titrate with 1
		N hydrochloric acid until it becomes green. Then add 10 mL of zinc sulphate solution
		(1en aand adjust the pH to 3.8 (green) to form a white precipitate.
3.	Assay	: Not less than 24.0% and not more than 26.0% of Fe.
4.	Arsenic	: Not more than 3 ppm (as As).
5.	Lead	: Not more than 10 ppm.
6.	Mercury	: Not more than 3 ppm.
7.	Loss on ignition	: Not more than 20%.
8.	Category	: Food Additives Category 8
9.	Uses	: Nutritional additivesNutritional additives

## Iron, Carbonyl

Cher	nical formula: Fe	Molecular weight: 55.85
1.	Assay	: Not less than 98.0%
2.	Description	: An elemental iron obtained by decomposition of iron pentacarbonyl. Dark gray powder.
		When viewed under a microscope at a magnification of 500 times or more, it is a
		spherical particle with a concentric shell. The sample can pass through 200 mesh
		sieve; more than 95% of the sample can pass 325 mesh sieve. Stable in dry air.
3.	Identification	: Soluble in dilute mineral acid and releases hydrogen, and generate corresponding ferric
		salt solution. The solution passes test for ferrous salts.
4.	Acid insoluble	: Not more than 0.2%.
	matter	
5.	Arsenic	: Not more than 4 ppm (as As).
6.	Lead	: Not more than 0.002%.
7.	Mercury	: Not more than 2 ppm.
8.	Category	: Food Additives Category 8
9.	Uses	: Nutritional additivesNutritional additives

## Iron, Electrolytic

Chemical formula: Fe		Molecular weight: 55.85
1.	Assay	: Not less than 97%
2.	Description	: The sample is an elemental iron obtained by electrolytic process. Amorphous, matt
		grayish black powder. The sample can pass through 100 mesh, and more than 95% of
		the sample can pass through 325 mesh sieve. Stable in dry air.
3.	Identification	: Soluble in dilute mineral acid and releases hydrogen, generating corresponding ferric
		salt solution. The solution passes test for ferrous salts.
4.	Acid insoluble	: Not more than 0.2%.
	matter	
5.	Arsenic	: Not more than 4 ppm (as As).
6.	Lead	: Not more than 0.002%.
7.	Mercury	: Not more than 2 ppm.
8.	Category	: Food Additives Category 8
9.	Uses	: Nutritional additivesNutritional additives

#### Ferric Ammonium Citrate

1.	Assay	: Not less than 16.5% and not more than 21.1% of iron (Fe).	
2.	Appearance	: Thin, transparent brown, reddish brown, or garnet red scales or granules, or a brownish	
		yellow powder; odorless or has a slight ammonia and salty odor.	
3.	Solubility	: Very soluble in water; insoluble in ethanol.	
4.	Tartrate	: Heat to dissolve 4 mL of potassium hydroxide TS to 10 mL of 1 in 10 solution of the	
		sample and filter. Acidify 5 mL of the filtrate with acetic acid.	
		Add 2 mL of glacial acetic acid and stand the solution for 24 hours. No white crystalline	
		precipitate should be formed.	
5.	Ferric citrate	: Add a drop of potassium ferrocyanide TS to 10 mL 1 in 100 solution of the sample. No	
		blue precipitation forms.	
6.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
7.	Lead	: Not more than 20 ppm.	
8.	Category	: Food Additives Category 8	
9.	Uses	: Nutritional additivesNutritional additives	

## Ferric Chloride

Cher	nical formula: FeCl <sub>3</sub> .	6H <sub>2</sub> O Molecular weight: 270.32
1.	Assay	: 98.5∼102.0%.
2.	Appearance	: Deliquescent tan crystal or solid block.
3.	Solution	: Dissolve 1 g of the sample in 0.1 mL of hydrochloric acid and 10 mL of water by
		heating. The turbidity should be below "slightly turbid".
4.	Free acid	: Near a glass rod moistened with ammonia solution to a $2\rightarrow 5$ solution of the sample. No
		smoke forms.
5.	Free chlorine	: Heat a $2 \rightarrow 5$ solution of the sample and near a filter paper wetted by the zinc iodide
		starch TS to it. The filter paper should not be blue.
6.	Nitrate	: Dissolve 0.5 g of the sample in 25 mL of water, boil, add 50 mL of ammonia TS
		$(1\rightarrow 2)$ , then add water to make a 100 mL solution. Cool and filter. Add 5 mL of water,
		0.1 mL of Indigo Carmine TS and 10 mL of sulfuric acid to 5 mL of the filtrate. Color
		blue should exist for more than 5 minutes.
7.	Sulfate	: Not more than 0.016% of SO <sub>4</sub> .
8.	Heavy metals	: Not more than 30 ppm (as Pb).
9.	Lead	: Not more than 10 ppm.
10.	Zinc	: Neutralize 20 mL of the filtrate obtained in "6.Nitrate" with hydrochloric acid, and add
		water to make it 30 mL. Then add 3 mL of dilute hydrochloric acid and 0.2 mL of
		potassium ferrocyanide TS. When stood for 15 minutes, the turbidity should not be
		thicker than that of 3 mL of zinc standard solution (1 mL = $Zn 0.01$ mg).
11.	Arsenic	: Not more than 3.3 ppm (as As <sub>2</sub> O <sub>3</sub> ).
12.	Category	: Food Additives Category 8
13.	Uses	: Nutritional additivesNutritional additives

### Ferric Citrate

1.	Assay	: Not less than 16.5% and not more than 18.5% of iron (Fe).
2.	Appearance	: Thin, transparent brown, reddish brown, or garnet red scales, or a brownish yellow
		powder.
3.	Solution	: Dissolve 1 g of the sample in water and heat. The turbidity of the solution should be
		below "almost clear".
4.	Ammonium salt	: Heat 1 g of the sample with 10 mL of water and 5 mL of sodium hydroxide TS on a
		steam bath for 1 min. The odor of ammonia is not perceptible.
5.	Tartrate	: Filter the solution used in the test "4. Ammonium salt". Make 4 mL of the filtrate to
		make weakly acidic with acetic acid, add 2 mL of glacial acetic acid. After standing
		for 24 hours, no white crystalline precipitate should be produced.
6.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Lead	: Not more than 20 ppm.
8.	Alkalis and alkaline	e earths : Mix 0.5 g of the ignition residue of the sample with 1 mL of water. The
		solution should not be alkaline.
9.	Category	: Food Additives Category 8
10.	Uses	: Nutritional additivesNutritional additives

## Ferrous Sulfate

Chemical formula: FeSO4·7H2O		·7H <sub>2</sub> O Molecular	weight: 278.03
1.	Assay	: Not less than 98% and not more than 104%	<b>/o.</b>
2.	Appearance	: Whitish green, odorless crystals, crystallin	e powder or granules.
3.	Solution	: Dissolve 1 g of the sample in 20 mL of wa	ter and 1 mL of dilute sulfuric acid. The
		turbidity of the solution should be below	"almost clear".
4.	pH	: Not less than 3.7 (1 in 10 soln)	
5.	Arsenic	: Not more than 4 ppm (as $As_2O_3$ ).	
6.	Heavy metals	: Not more than 25 ppm (as Pb).	
7.	Mercury	: Not more than 3 ppm.	
8.	Category	: Food Additives Category 8	
9.	Uses	: Nutritional additivesNutritional additives	

# Ferrous Lactate

Synonyms	: INS No. 585.
Chemical form	ula: $C_6H_{10}FeO_6 \cdot xH_2O$ , (x
= 2  or  3)	

Molecular weight: 270.02 (Dihydrate ); 288.03 (Trihydrate)

$$\begin{bmatrix} coo^{\Theta} \\ I \\ CHOH \\ I \\ CH_3 \end{bmatrix}_2$$
 Fe<sup>2⊕</sup>

1.	Assay	: Not less than 96% on the dried basis.
2.	Appearance	: Greenish white crystals or light green powder having a
		weak, characteristic smell.
3.	Solubility	: Soluble in water; practically insoluble in ethanol.
4.	pН	: 5.0~6.0 (1 in 50 solution)
5.	Identification	: (1) Test for lactate: Passes test.
		(2) Test for ferrous salts: Passes test.
6.	Loss on drying	: Not more than 18% (100°C using vacuum, 209pprox 700
		mm Hg)
7.	Sulfate	: Not more than 0.1%
8.	Chloride	: Not more than 0.1%
9.	Iron (III)	: Not more than 0.6%
10.	Lead	: Not more than 1 mg/kg.
11.	Category	: Food Additives Category (8)
12.	Uses	: Nutritional additives

#### Sodium Ferrous Citrate

#### (Iron and Sodium Succinate Citrate)

- 1. Assay :  $10.0 \sim 11.0\%$  of iron.
- 2. Description : White green to yellowish green powder, odorless, with a weak iron taste.
- 3. Identification : (1) Add 1 mL of dilute hydrochloric acid (1 g of the sample dissolved in 3 g of water), freshly prepared potassium ferricyanide solution (1 g of this product is 10 g of water) to 5 mL of 1 in 100 solution of the sample. Blue color appears.
  - (2) Add 2 mL of ammonia solution to 5 mL of the solution of the sample (1 g of the sample dissolved in 100 g of water). Reddish brown color appears, but no precipitation will occur.
  - (3) Ignite 3 g of the sample at  $500 \sim 600^{\circ}$ C for 3 hours. The residue passes test for sodium.
  - (4) Add 5 mL of water and 10 mL of potassium hydroxide solution (1 g of potassium hydroxide in 25 g of water) to 0.5 g of the sample. Heat in water bath for 10 minutes, stir, cool and filter. Neutralize a part of filtrate with dilute acetic acid (1 mL of acetic acid in 1 mL of water), add an excess of calcium chloride solution (3 g of calcium chloride in 40 g of water) and boil. White crystalline precipitate appears. Collect the residue. Add sodium hydroxide solution (1 g of sodium hydroxide in 25 g of water) to part of residue. Precipitation will not dissolve. Add dilute hydrochloric acid (1 mL of hydrochloric acid in 3 mL of water) to another part of residue. Precipitate will dissolve.
- Iron (III) : Place 2.0 g of the sample in a glass bolt triangle flask. Add 5 mL of hydrochloric acid and 30 mL of water, add 4 g of potassium iodide, and place it in dark for 1 minute. Afterward, add 2 mL of starch TS 2, and mix. The color appears. However, if 1 mL of 0.1 N sodium thiosulfate solution is added, the color should disappear immediately.
- 5. Tartrate : Add 5 mL of water and 10 mL of potassium hydroxide TS (add water to 1 g of potassium hydroxide to make a 15 mL solution) to 1.0 g of the sample, stir in a water bath for 10 minutes, cool and filter. Add dilute acetic acid (add water to 1 mL of acetic acid to make a 4 mL solution) to make 5 mL of the filtrate weakly acidic, then add 2 mL of acetic acid. After standing for 24 hours, no white crystalline precipitate should appear.
- 6. Sulfate : Not more than 0.48% of SO<sub>4</sub>.

- 7. Arsenic : Not more than 4 ppm (as  $As_2O_3$ ).
- 8. Heavy metals : Not more than 20 ppm (as Pb).
- 9. Category : Food Additives Category 8
- 10. Uses : Nutritional additivesNutritional additives

### Potassium Iodide

Chemical formula: KI			Molecular weight: 166.0
1.	Assay	: Not less than 99% on the	dried basis (105°C,4 h)
2.	Appearance	: Colorless transparent, white	opaque hexagonal crystal, or white granular or powder,
		odorless, salty and bitter.	
3.	Solubility	: 1 g of the sample is soluble	in 0.7 mL, 0.5 mL of boiling water, 22 Ml of ethanol, or 2
		mL of glycerol.	
4.	Alkalinity	: Add 0.1 mL of 0.1 N sulfur	ic acid and 1 drop of phenolphthalein TS to 1 g of the
		sample in 10 mL of freshly	v boiled and cooled water. No pink color is produced.
5.	Iodate, nitrite, thios	sulfate and barium salt	: Dissolve 0.5 g of the sample in 10 mL of freshly
			boiled and cooled water. When adding 2 drops
			of dilute sulfuric acid, it should not show
			obvious yellow color in 30 seconds. It should
			not be turbid in one minute.
6.	Nitrate, nitrite or a	nmonium salt : Place 1	g of the sample in a roughly 40 mL test tube, dissolve the
		sample	with 5 mL of water, add 5 mL of sodium hydroxide TS and
		about 2	200 mg of aluminum wire. Fill the test tube with refined
		cotton,	and put a wet red litmus test paper at the nozzle. Heat the
		test tub	e in water bath for 15 minutes. The test paper shall not
		appear	blue.
7.	Arsenic	: Not more than 2 ppm (as As	5 <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 10 ppm (as F	'b).
9.	Positive test for	: The flame test of the 1 in 20	) solution of the sample should not be yellow.
	sodium		
10.	Loss on drying	: Dry it at 105°C for 4 hours:	it loses not more than 1% of its weight.
11.	Category	: Food Additives Category 8	
12.	Uses	: Nutritional additivesNutritie	onal additives

0		Potassium Iodate	
Cher	mical formula: KIO <sub>3</sub>	Molecular weight: 214.01	
1.	Appearance	: White crystalline powder.	
2.	Water-insoluble matter	: Not more than 50 ppm.	
3.	Acidity or	: Dissolve 3 g of the sample in 40 mL of warm water, add	
	alkalinity	3 drops of phenolphthalein TS. The solution should not	
		be red. Then add 0.25 mL of 0.02 N hydrochloric acid.	
		The red color appears.	
4.	Chloride and	: Not more than 0.02% of Cl.	
	bromide		
5.	Chlorate	: Add 2 mL of sulfuric acid to 2 g of the sample. The sample should stay white and generate no smell or gas.	
6.	Iodide	: Dissolve 1 g of the sample in 20 mL of water, add 1 mL	
		of chloroform and 0.5 mL of 1 N sulfuric acid. The	
		chloroform layer should not appear violet in 1 minute.	
7.	Nitrogen	: Not more than 0.025% of N.	
	compound		
8.	Sulfate	: Not more than 50 ppm of $SO_4$ .	
9.	Heavy metals	: Not more than 10 ppm (as Pb).	
10.	Iron	: Not more than 10 ppm.	
11.	Positive test for	: The flame test of 1 in 10 solution of the sample should	
	sodium	not show an obvious yellow.	
12.	Category	: Food Additives Category 8	
13.	Uses	: Nutritional additives	

## Methyl Hesperidin (Vitamin P)

1.	Assay	: Not less than 90% (over sulfuric acid, 24 h)	
2.	Appearance	: Yellow to orange powder, odorless or nearly odorless.	
3.	Solution	: Dissolve 1 g of the sample in 10 mL of water. The solution should be less than "almost	
		clear".	
4.	Sulfate	: Not more than 0.02% of $SO_{4}$ .	
5.	Heavy metals	: Not more than 20 ppm (as Pb).	
6.	Loss on drying	: Not more than 3% (vacuum desiccator over sulfuric acid, 24 h)	
7.	Residue on	: Not more than 0.5%.	
	ignition		
8.	Category	: Food Additives Category 8	
9.	Uses	: Nutritional additivesNutritional additives	

# Menadione (Vitamin K<sub>3</sub>)

Chen	nical formula: C <sub>11</sub> H <sub>8</sub> O	2 Molecular weight: 172.18
1.	Assay	: Not less than 98.5% (over sulfuric acid, 4 h)
2.	Appearance	: Bright yellow crystalline powder, nearly odorless.
3.	Melting range	: 105~107°C.
4.	Solubility	: 1 g of the sample is soluble in about 60 mL of ethanol or 10 mL of benzene; less soluble
		in chloroform, carbon tetrachloride or vegetable oil; almost insoluble in water.
5.	Loss on drying	: Not more than 0.3% (over sulfuric acid, 4 hr)
6.	Residue on	: Not more than 0.1%.
	ignition	
7.	Category	: Food Additives Category 8
8.	Uses	: Nutritional additivesNutritional additives

## L-Histidine Monohydrochloride

Chem	nical formula: C <sub>6</sub> H <sub>9</sub> O <sub>2</sub>	$N_3$ ·HCl·H <sub>2</sub> O Molecular weight: 209.64
1.	Assay	: Not less than 98% on the dried basis (98°C, 3 h)
2.	Appearance	: White, odorless crystal or crystalline powder, with a sour and bitter taste.
3.	Solution	: Dissolve 1 g of the sample in 10 mL of water. The solution should be less than "almost
		clear" and colorless.
4.	pH	$: 3.5 \sim 4.5 (1 \text{ in } 10 \text{ soln})$
5.	Specific rotation	: ( $\alpha$ ) $_{D}^{20}$ = +8.5 ~+10.5° (Dry at 98°C for 3 hours. Dissolve 5.5 g of the sample in 6 N
		hydrochloric acid to make a 50 mL solution.)
6.	Ammonium salt	: Not more than 0.02% of NH <sub>4</sub> .
7.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 20 ppm (as Pb).
9.	Other amino acids	: The test method of other amino acid for "L-diaminohexanoic acid hydrochloride" is
		applicable.
10.	Loss on drying	: Dry it at $98^{\circ}$ for 3 hours: it loses not more than 0.2% of its weight.
11.	Residue on	: Not more than 0.05%.
	ignition	
12.	Category	: Food Additives Category 8
13.	Uses	: Nutritional additivesNutritional additives

#### L-Isoleucine

Chemical formula: C <sub>6</sub> H <sub>13</sub> O <sub>2</sub> N		D <sub>2</sub> N Molecular weight: 131.18
1.	Assay	: Not less than 98.5% (105°C, 3 h)
2.	Appearance	: White crystal or crystalline powder, odorless, with a slight bitter taste.
3.	Solution	: Dissolve 0.5 g of the sample in 20 mL of water. The solution should be "almost clear"
		and colorless.
4.	pН	: 5.5~7.0 (1 in 100 soln)
5.	Specific rotation	: $(\alpha)_{D}^{20} = +39.5 \sim +41.5^{\circ}$ (Dry at 105 °C for 3 hours. Dissolve 2 g of the sample in 6 N
		hydrochloric acid to make a 50 mL solution.)
6.	Chloride	: Not more than 0.02% of Cl.
7.	Ammonium salt	: Not more than 0.02% of NH <sub>4</sub> .
8.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Other amino acids	: Test for other amino acid of "L-aspartate" is applicable.
11.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 0.2% of its weight.
12.	Residue on	: Not more than 0.1%.
	ignition	
13.	Category	: Food Additives Category 8
14.	Uses	: Nutritional additivesNutritional additives

### DL-Tryptophan

Chemical formula: C <sub>11</sub> H <sub>12</sub> C		D2N2Molecular weight: 204.23
1.	Assay	: Not less than 98.5% on the dried basis (105°C, 3 h)
2.	Appearance	: White to slightly yellow crystal or crystalline powder, odorless or slightly smelly, a
		little sweet.
3.	Solution	: Dissolve 0.5 g of the sample in 10 mL of 0.5 N sodium hydroxide solution. The color of
		the solution should not be darker than Matching Fluid C.
4.	pH	: 5.5~7.0 (1 in 500 soln)
5.	Chloride	: Not more than 0.02% (as Cl).
6.	Ammonium salt	: Not more than 0.03% of NH <sub>4</sub> .
7.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 20 ppm (as Pb).
9.	Other amino acids	: Test for other amino acid of "L-aspartate" is applicable.
10.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 0.3% of its weight.
11.	Residue on	: Not more than 0.1%.
	ignition	
12.	Category	: Food Additives Category 8
13.	Uses	: Nutritional additivesNutritional additives

### L-Tryptophan

Chemical formula: C <sub>11</sub> H <sub>12</sub>		O <sub>2</sub> N <sub>2</sub> Molecular weight: 204.23
1.	Assay	: Not less than 98.5% on the dried basis (105°C,3 h)
2.	Appearance	: White to yellowish white crystal or crystalline powder, odorless or with slight odor,
		slightly bitter.
3.	Solution	: Dissolve 0.5 g of the sample in 10 mL of 0.5 N sodium hydroxide solution. The color of
		the solution should not be darker than Matching Fluid C and the turbidity should be
		below "almost clear".
4.	pH	: 5.5~7.0 (1 in 100 soln)
5.	Specific rotation	: $(\alpha)_{D}^{20} = -30 \sim -33^{\circ}$ (Dry at 105°C for 3 hours. Dissolve 0.5 g of the sample in
		water to make a 50 mL solution.)
6.	Chloride	: Not more than 0.02% of Cl.
7.	Ammonium salt	: Not more than 0.03% of NH <sub>4</sub> .
8.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Other amino acids	: Test for other amino acid of "L-aspartate" is applicable.
11.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 0.3% of its weight.
12.	Residue on	: Not more than 0.1%.
	ignition	
13.	Category	: Food Additives Category 8
14.	Uses	: Nutritional additivesNutritional additives

L-Valine

Chemical names	(2S)-2-Amino-3-methylbutanoic acid
C.A.S. number	72-18-4
Chemical formula	$C_5H_{11}NO_2$
Structural formula	

Formula weight	117.15
Content	L-Valine, when calculated on the dried basis, contains 98.0%-102.0% of L-valine
Description	L-Valine occurs as white crystals or crystalline powder. It is odorless or has a slight,
	characteristic odor, and has a slight, characteristic taste.
Identification	To 5 mL of a solution of L-Valine (1 in 1000), add 1 mL of ninhydrin solution (1 in
	1000), and heat for 3 minutes. A purple color develops.
Specific Rotation	$\left[ \alpha \right]_{D}^{20}$ : +26.5~+29.0° (Dry at 105 °C for 3 hours. Dissolve 4 g of the sample in 6 N
	hydrochloric acid to make a 50 mL solution.)
pH	$5.5 \sim 7.0 \ (0.5 \text{ g}, \text{ water } 20 \text{ mL})$
Purity	
Clarity and color	Colorless and clear (0.50 g, water 20 mL)
of solution	
Chloride	Not more than 0.021% as Cl
Arsenic	Not more than 2 mg/kg as As <sub>2</sub> O <sub>3</sub>
Lead	Not more than 2 mg/kg
Loss on Drying	Not more than 0.3% (105°C, 3 hours)
Residue on Ignition	Not more than 0.1%
Category	Food additives category 8
Functional uses	Nutritional additives

220

### L – Lysine L – Glutamate

Chemical formula: C <sub>11</sub> H <sub>23</sub> O <sub>6</sub> N <sub>3</sub> ·2H <sub>2</sub> O		O <sub>6</sub> N <sub>3</sub> ·2H <sub>2</sub> O Molecular weight: 329.30
1.	Assay	: 98~102% (105°C, 5 h).
2.	Appearance	: White powder, odorless or with a slightly characteristic odor; having a specific taste.
3.	Solution	: Dissolve 1 g of the sample in 20 mL of water. The solution should be less than "almost
		clear" and colorless.
4.	pН	: 6.0~7.5 (1 in 10 soln)
5.	Specific rotation	: $(\alpha)_{D}^{20} = +27.5 \sim +29.5^{\circ}$ (Dry at 105 °C for 5 hours. Dissolve 4 g of the sample in 6 N
		hydrochloric acid to make a 50 mL solution.)
6.	Chloride	: Not more than 0.04% of Cl.
7.	Ammonium salt	: Not more than 0.02% of NH <sub>4</sub> .
8.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Heavy metals	: Not more than 30 ppm (as Pb).
10.	Loss on drying	: Dry it at 105°C for 5 hours: it loses not more than 11.4% of its weight.
11.	Residue on	: Not more than 13%.
	ignition	
12.	Other amino acid	: Dissolve 0.1 g of the sample in water to make a 50 mL solution. Take 5 $\mu$ l for filter
		paper chromatography test, and stop when the fluid reach 30 cm from the origin. After
		drying the filter paper, dry at 100°C for 20 minutes. Spray coloring solution on the
		filter paper and dry at 100°C for 5 minutes. In addition to spots of the diaminocaproic
		acid and glutamate, no other spots should appear.
		Developing solvent: Mixture of n-butyl, glacial acetic acid and water (5: 1: 2).
		Coloring solution: 1 g of ninhydrin is dissolved in 500 mL of n-butanol that is
		saturated with water.
		Filter paper: Chromatography filter paper No.2
13.	Category	: Food Additives Category 8
14.	Uses	: Nutritional additivesNutritional additives

### L – Lysine Monohydrochloride

Chemical formula: C <sub>6</sub> H <sub>14</sub> O <sub>2</sub> N <sub>2</sub> ·HCl		D <sub>2</sub> N <sub>2</sub> ·HCl Molecular weight: 182.66
1.	Assay	: Not less than 98.5% (105°C, 3 h).
2.	Appearance	: White powder, odorless or with a slightly characteristic odor.
3.	Solution	: When 0.5 g of the sample is dissolved in 10 mL of water, the solution should be clear.
4.	pH	: 5.0~6.0 (1 in 10 soln)
5.	Specific rotation	: $(\alpha)_{D}^{20} = +19.0 \sim +21.5^{\circ}$ (Dry at 105 °C for 3 hours. Dissolve 4 g of the sample in 6 N
		hydrochloric acid to make a 50 mL solution.)
6.	Ammonium salt	: Test for ammonium salt of "L-aspartate" is applicable.
7.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Other amino acid	: Dissolve 0.3 g of the sample in water to make a 50 mL solution. Take 5 $\mu l$ for filter
		paper chromatography test, and stop when the fluid reach 30 cm from the origin. After
		drying the filter paper, dry at 100°C for 20 minutes. Spray coloring solution on the
		filter paper and dry at 100°C for 5 minutes. Only one spot should appear.
		Developing solvent: Mixture of n-butanol, acetone, dicyclohexylamine and water (10:
		10: 2: 5).
		Coloring solution: Dissolve 0.2 g of ninhydrin in 100 mL of the mixture of n-butanol,
		glacial acetic acid and water (5: 1: 2)
		Filter paper: Chromatography filter paper No.2
9.	Heavy metals	: Not more than 10 ppm (as Pb).
10.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 1% of its weight.
11.	Residue on	: Not more than 13%.
	ignition	
12.	Category	: Food Additives Category 8
13.	Uses	: Nutritional additivesNutritional additives

#### DL-Methionine

Chemical formula: C <sub>5</sub> H <sub>11</sub> O <sub>2</sub> NS		D <sub>2</sub> NS Molecular weight: 149.22
1.	Assay	: Not less than 98.5% (105°C, 4 h).
2.	Appearance	: White flake crystals or crystalline powder with characteristic odor and slightly sweet
		taste.
3.	Solution	: When 0.5 g of the sample is dissolved in 20 mL of water, the solution should be clear.
4.	Chloride	: Not more than 0.024% of Cl.
5.	Sulfate	: Not more than $0.05\%$ of SO <sub>4</sub> .
6.	Ammonium salt	: Not more than 0.02% of NH <sub>4</sub> .
7.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 20 ppm (as Pb).
9.	Loss on drying	: Not more than 0.5% (105°C, 4 h)
10.	Residue on	: Not more than 0.1%.
	ignition	
11.	Category	: Food Additives Category 8
12.	Uses	: Nutritional additivesNutritional additives

#### L-Methionine

Chemical formula: C <sub>5</sub> H <sub>11</sub> O <sub>2</sub> NS		D <sub>2</sub> NS Molecular weight: 149.22
1.	Assay	: Not less than 98.5% on the dried basis (105°C, 4 h)
2.	Appearance	: White flake crystal or crystalline powder with characteristic odor and slightly bitter
		taste.
3.	Solution	: When 0.5 g of the sample is dissolved in 20 mL of water, the solution should be clear.
4.	Specific rotation	: $(\alpha)_{D}^{20} = +21.0 \sim +25.0^{\circ}$ (Dissolve 1 g of the sample in 6 N hydrochloric acid to make
		the solution into 50 mL after drying at 105°C for 4 hours.)
5.	Chloride	: Not more than 0.024% of Cl.
6.	Sulfate	: Not more than $0.05\%$ of SO <sub>4</sub> .
7.	Ammonium salt	: Not more than 0.02% of NH <sub>4</sub> .
8.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Loss on drying	: Dry it at 105°C for 4 hours: it loses not more than 0.5% of its weight.
11.	Residue on	: Not more than 0.1%.
	ignition	
12.	Category	: Food Additives Category 8
13.	Uses	: Nutritional additivesNutritional additives

### L-Phenylalanine

Chemical formula: C <sub>9</sub> H <sub>11</sub> O <sub>2</sub> N		D <sub>2</sub> N Molecular weight: 165.20
1.	Assay	: Not less than 98.5% on the dried basis (105°C, 3 h)
2.	Appearance	: White crystal or crystalline powder, a little bitter.
3.	Solution 及溶性	: Dissolve 1 g of the sample in 100 mL of water. The solution should be less than "almost
		clear" and colorless. The pH should be $5.4 \sim 6.0$ .
4.	Specific rotation	: $(\alpha)_{D}^{20} = -33 \sim -35^{\circ}$ (Dry at 105°C for 3 hours. Dissolve 1 g of the sample in water
		to make a 50 mL solution.)
5.	Chloride	: Not more than 0.02% of Cl.
6.	Ammonium salt	: Not more than 0.02% of NH <sub>4</sub> .
7.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 20 ppm (as Pb).
9.	Other amino acids	: Test for other amino acid of "L-aspartate" is applicable.
10.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 0.3% of its weight.
11.	Residue on	: Not more than 0.1%.
	ignition	
12.	Category	: Food Additives Category 8
13.	Uses	: Nutritional additivesNutritional additives

#### DL-Threonine

Chemical formula: C <sub>4</sub> H <sub>9</sub> O <sub>2</sub>		3N Molecular weight: 119.12
1.	Assay	: Not less than 98% on the dried basis (105°C,3 h)
2.	Appearance	: White crystalline powder, odorless, with a slight sweet taste.
3.	Solution	: Dissolve 1 g of the sample in 20 mL of water, and the solution should be less than
		"almost clear" and colorless.
4.	рН	: 5.0~6.5 (1 in 20 soln)
5.	Chloride	: Not more than 0.02% of Cl.
6.	Ammonium salt	: Not more than $0.02\%$ of NH <sub>4</sub> .
7.	Arsenic	: Not more than 3 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 20 ppm (as Pb).
9.	Other amino acids	: Not detectable.
10.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 0.2% of its weight.
11.	Residue on	: Not more than 0.1%.
	ignition	
12.	Category	: Food Additives Category 8
13.	Uses	: Nutritional additivesNutritional additives

#### $L\!-\!Threonine$

Chemical formula: C <sub>4</sub> H <sub>9</sub> O <sub>3</sub> N		Molecular weight: 119.12
1.	Assay	: Not less than 98% on the dried basis (105°C, 3 h)
2.	Appearance	: White crystal or crystalline powder, odorless, with a slight sweet taste.
3.	Specific rotation	: $(\alpha)_{D}^{20} = -26 \sim -29^{\circ}$ (Dry at 105°C for 3 hours. Dissolve 3 g of the sample in water
		to make a 50 mL solution.)
4.	Solubility, liquid, c	chloride, ammonium, arsenic, heavy : Test 3.~11. of the DL-hydroxybutyric acid is
	metals, other amino	o acids, loss on drying, residue on applicable.
	ignition	
5.	Category	: Food Additives Category 8
6.	Uses	: Nutritional additivesNutritional additives

#### Sodium Pantothenate

Chemical formula: C <sub>9</sub> H <sub>16</sub> O <sub>5</sub> NNa		D <sub>5</sub> NNa Molecular weight: 241.23
1.	Assay	: N=5.6 $\sim$ 6.0%;Na=9.4 $\sim$ 9.8% (vacuum desiccator over sulfuric acid, 24 h)
2.	Appearance	: White crystal or crystalline powder, odorless, with a slight sour taste.
3.	Specific rotation	:: $(\alpha)_{D}^{25} = +25 \sim +30^{\circ}$ (Vacuum desiccator over sulfuric acid, 24 h. Dissolve 1.25 g
		the sample in 25 mL of water)
4.	Heavy metals	: Not more than 20 ppm (as Pb).
5.	Calcium	: Dissolve 1 g of the sample in 10 mL of water, add 0.5 mL of dilute acetic acid and 0.5
		mL of ammonium oxalate TS. No precipitate occurs.
6.	Alkaloid	: Dissolve 0.2 g of the sample in 5 mL of water, add 1 mL of dilute hydrochloric acid and
		2 drops of mercury iodide TS. No turbidity should be generated within 1 minute.
7.	Loss on drying	: Not more than 5% (vacuum desiccator over sulfuric acid, 24 h)
8.	Category	: Food Additives Category 8
9.	Uses	: Nutritional additivesNutritional additives

#### Calcium Pantothenate

Chemical formula: C <sub>18</sub> H <sub>32</sub> O <sub>10</sub> N <sub>2</sub> Ca		2O <sub>10</sub> N <sub>2</sub> Ca Molecular weight: 476.55
1.	Assay	: N=5.7 $\sim$ 6.0%; Ca=8.2 $\sim$ 8.6% on the dried basis (105°C, 3 h)
2.	Appearance	: White, odorless powder, having a bitter taste.
3.	Specific rotation	:: $(\alpha)_{D}^{20} = +25.0 \sim +28.5^{\circ}$ (dried at 105°C for 3 hours, and dissolve 1.25 g of the
		sample in water to male a 25 mL solution)
4.	Heavy metals	: Not more than 20 ppm (as Pb).
5.	Alkaloid	: Dissolve 0.2 g of the sample in 5 mL of water, add 1 mL of dilute hydrochloric acid and
		2 drops of mercury iodide TS. No turbidity should be generated within 1 minute.
6.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 5% of its weight.
7.	Category	: Food Additives Category 8
8.	Uses	: Nutritional additivesNutritional additives

#### Potassium Chloride

Chemical formula: KCl		Molecular weight: 74.55
1.	Assay	: Not less than 99.0% on the dried basis.
2.	Description	: Colorless, elongated, prismatic, or cubital crystals, or white granular powder; odorless,
		having a salty taste and stable in the air. Freely soluble in water; insoluble in ethanol.
		The solution is neutral on litmus paper. 1 g of the sample is soluble in 2.8 mL of $25^{\circ}$ C
		water and 2 mL of boiling water.
3.	Identification	: The 1 in 20 solution of the sample passes test for potassium and test for chloride.
4.	Acidity or	: To a solution of 5 g of the sample in 50 mL of recently boiled and cooled water add 3
	alkalinity	drops of phenolphthalein TS. No pink color is produced. Then add 0.3 mL of 0.02 N $$
		sodium hydroxide. A pink color is produced.
5.	Iodide or bromide	: Dissolve 2 g of the sample in 6 mL of water, add 1 mL of chloroform, and then
		add, dropwise and with constant agitation, 5 mL of a mixture of equal parts of
		chlorine TS and water. The chloroform is free from even a transient violet or
		permanent orange color.
6.	Test for sodium	: Take the 1 in 20 solution of the with platinum wire, and perform flame test on non-glow
		flame. No obvious yellow color should appear.
7.	Arsenic	: Not more than 3 ppm (as As).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 1% of its weight.
10.	Category	: Food Additives Category 8, 9.
11.	Uses	: Nutritional additivesNutritional additives; Flavoring Agents agent.

### Magnesium Sulfate

#### Choline Bitartrate

Chemical formula: C <sub>9</sub> H <sub>19</sub> NO <sub>7</sub>		NO <sub>7</sub> Molecular weight: 253.25
1.	Description	: The sample is a white, hygroscopic crystalline powder with a sour taste, odorless or
		with a slight trimethylamine odor. Completely soluble in water; slightly soluble in
		alcohol; insoluble in ether, chloroform and benzene.
2.	Identification	: A. Dissolve 500 mg of the sample in 2 mL of water, add 3 mL of sodium hydroxide TS,
		heated to boiling. An odor of trimethylamine appears.
		B. Dissolve 500 mg of the sample in 2 mL of iodine TS. Reddish brown precipitate
		immediately appears. After adding 5 mL of sodium hydroxide TS, the precipitate
		is dissolved and the solution becomes clear and yellow. This solution is heated to
		produce a pale yellow precipitate and iodoform odor.
		C. Add 2 mL of cobalt chloride TS and 2 mL of 1 in 50 potassium ferrocyanide
		solution to 1 mL of 1 in 100 solution of the sample. Color emerald green will
		appear immediately.
3.	Assay	: Not less than 98.0% on the dried basis.
4.	Arsenic	: Not more than 3 ppm (as As).
5.	1,4-Dioxane	: Passes test
6.	Heavy metals	: Not more than 20 ppm (as Pb).
7.	Lead	: Not more than 10 ppm.
8.	Residue on	: Not more than 0.1%.
	ignition	
9.	Water	: Not more than 0.5%.
10.	Category	: Food Additives Category 8
11.	Uses	: Nutritional additivesNutritional additives

#### Choline Chloride

Chemical formula: C <sub>5</sub> H <sub>14</sub> ClNO		CINO Molecular weight: 139.62
1.	Description	: Colorless or white, hygroscopic crystals or crystalline powder, usually with an odor of
		trimethylamine. Very soluble in water and alcohol.
2.	Identification	: A. The sample passes Test A, B and C for choline bitartrate.
		B. The 1 in 20 solution of the sample passes test for chloride.
3.	Assay	: Not less than 98% on the dried basis.
4.	Arsenic	: Not more than 3 ppm (as As).
5.	1,4-Dioxane	: Passes test
6.	Heavy metals	: Not more than 20 ppm (as Pb).
7.	Lead	: Not more than 10 ppm.
8.	Residue on	: Not more than 0.05%.
	ignition	
9.	Water	: Not more than 0.5%.
10.	Category	: Food Additives Category 8
11.	Uses	: Nutritional additivesNutritional additives

#### Ferrous Gluconate

Chemical formula: C12H22FeO14·2H2O

Molecular weight: 482.17

$$\begin{bmatrix} OH OH H OH \\ HOH_2C - C - C - C - C - C \\ H H OH H \end{bmatrix}_2 Fe^{2\Theta}$$

- 1. Assay : Not less than 95% on the dried basis.
- 2. Appearance : Fine yellowish-grey or pale greenish-yellow powder or granules having a slight odor resembling that of burnt sugar.
- 3. Solubility : Soluble with slight heating in water; practically insoluble in ethanol.
- 4. Loss on drying  $: 6.5 \sim 10.0\% (105^{\circ}C, 16 h).$
- 5. : Dissolve 0.5 g of the sample in 10 mL of water; warm, and make the solution alkaline Reducing sugar with 1 mL of ammonia TS. Pass hydrogen sulfide gas into the solution to precipitate the iron, and allow the mixture to stand for 30 min to coagulate the precipitate. Filter, and wash the precipitate with two successive 5 mL portions of water. Acidify the combined filtrate and washings with hydrochloric acid, and add 2 mL of dilute hydrochloric acid TS in excess. Boil the solution until the vapors no longer darken lead acetate paper, and continue to boil, if necessary, until concentrated to about 10 mL. Allow to cool, add 5 mL of sodium carbonate TS and 20 mL of water; filter, and adjust the volume of the filtrate to 100 mL. To 5 mL of the filtrate add 2 mL of alkaline cupric tartrate TS and boil for 1 min. No red precipitate should be formed within 1 min. : Not more than 2%. 6. Iron 7. Lead : Not more than 2 ppm. 8. : Food Additives Category 8 Category
- 9. Uses : Nutritional additivesNutritional additives

#### Magnesium Oxide

Chemical formula: MgO		Molecular weight: 40.31
1.	Assay	: Not less than 96.0% after ignition at about 800°C.
2.	Description	: Very bulky white powder, known as light magnesium oxide, or as a relatively dense,
		white powder, known as heavy magnesium oxide. 5 g of light magnesium oxide occupy a
		volume of 40 to 50 mL, while 5 g of heavy magnesium oxide occupy a volume of 10 to
		20 mL. The sample is alkaline to moistened litmus paper.
3.	Solubility	: Practically insoluble in water; insoluble in ethanol.
4.	Loss on drying	: Not more than 5% after ignition at 800°C to 825°C to constant weight.
5.	Alkali (free) and so	luble : Boil 2 g of the sample, weighed to the nearest mg, with 100 mL of water for
	salts	5 min in a covered beaker and filter while hot. Add methyl red TS and titrate
		50 mL of the cooled filtrate with 0.1 N sulfuric acid. Not more than 2 mL of
		the acid should be consumed. Evaporate 25 mL of the filtrate to dryness and
		dry at 105°C for 1 h. Not more than 10 mg of residue should remain.
6.	Calcium oxide	: Not more than 1.5%
		(Weigh 400 mg of the sample to the nearest 0.1 mg, and dissolve in a mixture of 3 mL of
		sulfuric acid and 22 mL of water. Add 50 mL of ethanol, and allow the mixture to stand
		overnight. If crystals of magnesium sulfate separate, warm the mixture to about 50°C to
		dissolve. Filter through a tared, previously ignited, porcelain filter crucible, and wash the
		precipitate several times with a mixture of 2 volumes of ethanol and 1 volume of dilute
		sulfuric acid TS. Ignite the crucible and contents at a dull red heat, cool and weigh. The
		weight of calcium sulfate obtained, multiplied by 0.4119, gives the equivalent of calcium
		oxide in the sample taken for the test.)
7.	Arsenic	: Not more than 3 ppm.
8.	Lead	: Not more than 10 ppm.
9.	Heavy metals	: Not more than 40 ppm (as Pb).
10.	Category	: Food Additives Category 8
11.	Uses	: Nutritional additives

#### L-Carnitine

Chemical names	4-Amino-3-hydroxybutyric Acid Trimethylbetaine; Levocarnitine; 4-
	Trimethylamino-3-hydroxybutyrate; (R)-3-Carboxy-2-hydroxy-N,N,N-trimethyl-1-
	propanaminium Hydroxide, Inner Salt
C.A.S. number	541-15-1
Chemical formula	C7H15NO3
Structural formula	
	$H_3C$ $CH_3$ $OH$ $O$ $H_3C$ $P$ $O^-$
Formula weight	161.20
Assay	$97.0\% \sim 103.0\%$ , calculated on the anhydrous basis
Description	L-Carnitine occurs as white crystals or as a white, crystalline, hygroscopic powder.
Identification	Dissolve 1 g of sample in 10 mL of water and 10 mL of 1 N hydrochloric acid, and add 5
	mL of sodium tetraphenylborate test solution. It forms a white precipitate.
Solubility	It is freely soluble in water, in alcohol, in alkaline solutions, and in dilute mineral acids.
	It is practically insoluble in acetone and in ethyl acetate. It decomposes without melting
	at about 185°C to 195°C.
Specific Rotation	$\left( \ \alpha \ \right) \ _{D}^{20}$ $\ : \ -29.0 \sim -32.0^{\circ},$ calculated on the anhydrous basis
рН	5.5 ~ 9.5
Chloride	Not more than 0.4%
Water	Not more than 4.0%
Lead	Not more than 1 mg/kg
Potassium	Not more than 0.2%
Sodium	Not more than 0.1%
Residue on Ignition	Not more than 0.5%
Category	Food additives category 8
Functional uses	Nutritional additives

#### Manganese Sulfate

Synony	m : CAS	5 No. 7785-87-7	
Chemic	cal formula : MnSO4	·H <sub>2</sub> O	Molecular weight : 169.02
1.	Assay	: 98.0 ~ 102.0% °	
2.	Description	: Light pink granular powder	
3.	Solubility	Freely soluble in water, inso	luble in ethanol
4.	Arsenic	: Not more than 3 mg/kg	
5.	Lead	: Not more than 4 mg/kg	
6.	Selenium	: Not more than 0.003%	
7.	Loss on drying	: 10.0 ~ 13.0% (400 ~ 500°C	, heat to constant weight)
8.	Category	: Food Additives Category 8	
9.	Uses	: Nutritional additives	

#### Magnesium Gluconate

#### Chemical formula: C12H22MgO14

Formula weight: 414.60 (Anhydrous); 450.63

(Dihydrate)

- 1. Assay : Not less than 98.0% and not more than 102.0% on the anhydrous basis.
- 2. Description : White to off-white, odorless, fine powder.
- 3. Solubility : Soluble in water; sparingly soluble in ethanol.
- 4. Water : Between 3.0% and 12.0% (Karl Fischer Method).
- 5. Reducing : Not more than 1.0% calculated as D-glucose.

substances

- 6. Lead : Not more than 2 ppm.
- 7. Category : Food Additives Category 8
- 8. Uses : Nutritional additivesNutritional additives

#### Magnesium Hydroxide

Chemical	formula:	$Mg(OH)_2$
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#### Molecular weight: 58.32

- 1. Assay : Not less than 95%
- 2. Description : Odorless, white bulky powder. The sample is alkaline to moistened litmus paper.
- 3. Solubility : Practically insoluble in water and in ethanol.
- 4. Loss on drying : Not more than 2% (105°C, 2 h)
- 5. Loss on ignition : Not more than 30 33% (approx. 800°C to constant weight)
- 6. Alkalis (free) and soluble : Boil 2 g of the sample with 100 mL of water for 5 min in a covered beaker and filter while hot. Add methyl red TS and titrate 50 mL of the cooled filtrate with 0.1 N sulfuric acid. Not more than 2 mL of the acid is required to reach the endpoint. Evaporate 25 mL of the filtrate to dryness and dry at 105°C for 3 h. Not more than 10 mg of residue remains.
- 7. Calcium oxide : Not more than 1.5%

(Dissolve about 500 mg of the sample, accurately weighed, in a mixture of 3 mL of concentrated sulfuric acid and 22 mL of water. Add 50 mL of ethanol and allow the mixture to stand overnight. If crystals of magnesium sulfate separate, warm the mixture to about 500 to dissolve. Filter through a Gooch crucible containing an asbestos mat previously washed with dilute sulfuric acid TS, water, and ethanol and ignited and weighed. Wash the crystals on the mat several times with a mixture of 3 volumes of ethanol and 1 volume of water. Ignite the crucible and contents at a dull red heat, cool and weigh. The weight of calcium sulfate obtained, multiplied by 0.4119, gives the equivalent of calcium oxide in the sample taken for the test.)

- 8. Arsenic : Not more than 3 ppm.
- 9. Lead : Not more than 10 ppm.
- 10. Heavy metals : Not more than 40 ppm.
- 11. Category : Food Additives Category 8
- 12. Uses : Nutritional additivesNutritional additives

#### Lactoferrin

Description	Obtained by degreasing, separating and refining the milk. White to
	pink powder.
Crude protein	Not less than 93% (TN×6.38, on the dried basis).
Lactoferrin	Not less than 95% on the protein basis.
Iron	30 mg/100 g
Water	Not more than 4.5%.
Solubility	Dissolve 2 g of the sample in 100 mL of water. The solution should
	be "transparent".
pH	5.2-7.2 (20°C) (2% soln)
Ash	Not more than 1%
Heavy metals	Not more than 20 mg/kg as Pb
Arsenic	Not more than 2 mg/kg as As
Category	Food Additives Category 8
Uses	Nutritional additives

### Calcium Dihydrogen Phosphate

### Calcium Phosphate, Dibasic

### Calcium Phosphate, Tribasic

#### Iron Lactate

1.	Assay	: Not less than 15.5% and not more 20.0% of Fe.
2.	Appearance	:带綠白至黃褐色粉末或塊狀,略具特異臭.
3.	Identification	:(1)本品 0.5g在 450~550℃高溫下加熱一小時,取其殘渣,加入稀釋鹽酸(鹽酸 1
		mL 溶於水 1 mL)3 mL, 並加熱溶解之, 其溶液之鐵離子試驗呈陽性反應.
		(2) Passes test for lactate.
4.	Solution	: Dissolve 1 g of the sample in 20 mL of water in water bath. The solution should be
		"almost clear".
5.	Chloride	: Not more than 0.07% of Cl.
6.	Sulfate	: Not more than 0.48% of SO <sub>4</sub> .
7.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 50 ppm (as Pb).
9.	易碳化物及酪酸鹽	:本品 0.5 g 加硫酸 1 mL 混合時,不得產生顏色或類似酪酸之臭味.
10.	Category	: Food Additives Category 8
11.	Uses	: Nutritional additivesNutritional additives

#### Calcium Lactate

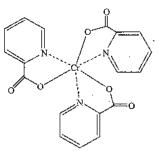
#### Calcium Gluconolactate

Chemical formula: $Ca_5(C_3H_5O_3)_6(C_6H_{11}O_7)_4$ ·2H <sub>2</sub> O		$(C_{6}H_{11}O_{7})_{4} \cdot 2H_{2}O$ Molecular weight: 1551.5
1.	Assay	: Not less than 98.0% and not more than 101.0%.
2.	Description	: White or slightly gray or yellow powder.
3.	Identification	: The 100 mg in 10 mL solution of the sample passes the test for calcium. In the thin
		layer chromatography analysis, the sample passes test for lactic acid and test for
		luconic acid.
4.	Total sugars	: No red precipitate exists.
5.	pН	: Dissolve about 10 g of the sample in 1000 mL of boiled water, and the pH of the
		solution is $5.6 \sim 8.6$ .
6.	Heavy metals	: Not more than 5 ppm (as Pb).
		The total amount of Fe, Ni, Cu, Zn, Pb and Cd is not more than 50 ppm.
		Not more than 40 ppm of Fe.
		The total amount of Ni, Cu, Zn, Pb and Cd is not more than 10 ppm.
		Not more than 1 ppm (as Pb) and Cd.
7.	Arsenic	: Not more than 1 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Chloride	: Not more than 400 ppm of $Cl^{-}$ .
9.	Sulfate	: Not more than 600 ppm of $SO_4^{-2}$ .
10.	Phosphate	: Not more than 400 ppm of $PO_4^{-3}$ .
11.	Loss on drying	: Not more than 4.0% (85°C, more than 30 min)
12.	Category	: Food Additives Category 8
13.	Uses	: Nutritional additivesNutritional additives

#### Chromium Picolinate

### Chemical formula: C18H12N3O6Cr

Molecular weight: 418.31

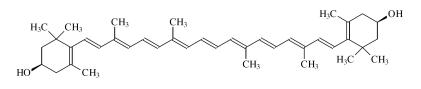


1.	Assay	: Not less than 98% and not more than 102% on the dried basis.
2.	Appearance	: Red to pink powder.
3.	Identification	: The infrared absorption spectrum measured by the sample should be consistent with the
		standard product.
4.	Chromium	: 12~12.6%.
5.	Chloride	: Not more than 0.06%.
6.	Bulk density	: Not less than 0.45 g/mL.
7.	Loss on drying	: Dry it at 105°C for 4 hours: it loses not more than 4% of its weight.
8.	Sulfate	: Not more than 0.2%.
9.	Arsenic	: Not more than 1 ppm.
10.	Lead	: Not more than 10 ppm.
11.	Mercury	: Not more than 1 ppm.
12.	Particle size	: Not more than 60 Mesh.
13.	Category	: Food Additives Category 8
14.	Uses	: Nutritional additivesNutritional additives

#### Synthetic Zeaxanthin

Chemical names	: (all-E)-1,1'-(3,7,12,16-Tetramethyl-1,3,5,7,9,11,13,15,17- octadecan	onaene-1,18-
	diyl)bis[2,6,6-trimethylcyclohexene-3-ol]; 3R,3'R- $\beta$ , $\beta$ -Carotene-3	,3'-diol
Synonyms	: Zeaxanthol; anchovyxanthin; INS 161h(i); CAS No. 144-68-3	
Chemical formula	$: C_{40}H_{56}O_2$	Molecular weight:

568.9



1. Assay	: Not less than 96%.
----------	----------------------

- 2. Description : Orange-red crystalline powder, with little or no odor.
- 3. Solubility : Sparingly soluble in chloroform, practically insoluble in water and ethanol.
- Test for carotenoid : The color of the solution of the sample in acetone disappears after successive additions of a 5 % solution of sodium nitrite and 1N sulfuric acid.
- 5. Spectrophotometry : An ethanol solution of the sample shows maximum absorption between 450 and 454 nm.
- 6. Loss on drying : Not more than 0.2%.
- 7. Sulfated ash : Not more than 0.1%.
- 8. cis-Zeaxanthins : Not more than 2.0%.
- 9. 12' Apo-zeaxanthinal, diatoxanthin, parasiloxanthin : Not more than 1.1 % combined.
- 10. Triphenyl phosphine oxide (TPPO) : Not more than 0.01%.
- 11. Heavy metals : Not more than 20 mg/kg (as Pb).
- 12. Lead : Not more than 2 mg/kg.
- 13. Category : Food Additives Category 8
- 14. Uses : Nutritional additivesNutritional additives

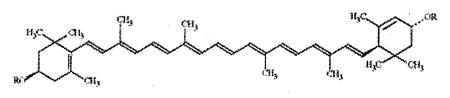
§ 09033

Lutein

Chemical names : 3,3'-dihydroxy-d-carotene

Chemical formula: C40H56O2

Molecular weight: 568.88



Lutein: R = H

- 1. Definition
   : Crystalline lutein is obtained by extracting with solvent, saponifying, centrifuging marigold flower. The crystal contains a small amount of zeaxanthin.
- 2. Assay : Not less than 75% of lutein.
- 3. Description : Yellowish brown crystalline powder.
- 4. Identification : (1) Solubility: Insoluble in water; soluble in hexane.

(2) Positive carotenoid test: The color disappears after continuously adding 5% sodium nitrite solution and 0.5 M sulfuric acid to the acetone solution of the sample.

- 5. Residual solvent : Hexane: Not more than 50 mg/kg.
- 6. Lead : Not more than 2 mg/kg.
- 7. Category : Food Additives Category 8
- 8. Uses : Nutritional additivesNutritional additives

#### Calcium L-Threonate

Molecular weight: 310.27 Chemical formula: (C<sub>4</sub>H<sub>7</sub>O<sub>5</sub>)<sub>2</sub>Ca CH<sub>2</sub>OH нфон Ca HOCH 2 COO : Not less than 95% on the dried basis. 1. Assay 2. Appearance : White powder 3. Melting range : Not less than 265°C. 4. Solubility : Soluble in water. The solution is colorless and clear. 5. Loss on drying : Dry it at 105°C for 4 hours: it loses not more than 0.40% of its weight. : 0.15~0.25%. 6. Water 7.  $: 8.0 \sim 9.0$  (saturated solution) pН 8. Ascorbic acid : Not more than 18 ppm. 9. Arsenic : 2 ppm (as As<sub>2</sub>O<sub>3</sub>). 10. Heavy metals : Not more than 20 ppm (as Pb). 11. Ascorbic acid : Not detectable. 12. Sulfate : Not more than 0.05%. 13. Chloride : Not more than 0.05%. 14. Category : Food Additives Category 8 : Nutritional additivesNutritional additives 15. Uses

#### Calcium Citrate

Synonyms	INS No. 333(iii)
Definition	
Chemical names	Tricalcium citrate, tricalcium salt of 2-hydroxy-1,2,3- propanetricarboxylic acid,
	tricalcium salt of ß-hydroxy-tricarballylic acid
C.A.S. number	813-94-5
Chemical formula	$C_{12}H_{10}Ca_{3}O_{14}\cdot 4H_{2}O$
Structural formula	
	$\begin{bmatrix} cH_2 \cdot CO d^{\Theta} \\ HO_{-} \cdot C_{-} CO d^{\Theta} \\ cH_2 \cdot CO d^{\Theta} \\ cH_2 \cdot CO d^{\Theta} \end{bmatrix}_2 ca_3 \cdot 4H_2O$
Formula weight	570.51
Assay	Not less than 97.5% after drying
Description	Odourless, fine white powder
Characteristics	
Identification	
Solubility	Very slightly soluble in water. Insoluble in ethanol.
Test for citrate	Passes test
Test for calcium	Passes test
Purity	
Loss on drying	Not less than 10% and not more than 14% (150°C, 4 h)
Fluorides	Not more than 30 mg/kg
Free acid and alkali	Passes test
Oxalate	Dissolve 1 g of the sample in 5 ml of warm dilute hydrochloric acid TS and filter the
	solution if necessary.
Lead	Not more than 2 mg/kg
Category	Food additives category (7) (8)
Functional uses	Food quality improvement, fermentation and food processing agents; Nutritional
	additives.

### Zinc Citrate Trihydrate

Che	mical formula: C <sub>12</sub> H	$_{10}O_{14}Zn_3 \cdot 3H_2O$ Molecular weight: 628.4
1.	Assay	: Not less than 99.0% and not more than 102.5%.
2.	Description	: White powder. Insoluble in alcohol, slightly soluble in water, and soluble in diluted
		inorganic acid.
3.	Identification	: Solution S: Dissolve 2.5 g of the sample in dilute nitric acid and adjust the volume to 50
		mL using dilute nitric acid.
		Appearance of solution S: Transparent, clear, colorless.
		A. Test for citrate: Add 0.5 mL of sulphuric acid and 1 mL of potassium
		permanganate to 5 mL of solution S. Heat until until the color of the potassium
		permanganate remain unchanged. Add 0.5 mL of sodium nitroferric
		hydride/diluted sulfuric acid (100 g/L) solution and 4 g of sulfuric acid. Slowly
		add concentrated ammonia to alkalinize the solution until sulfuric acid is
		dissolved. Add an excess of concentrated ammonia to make the solution purple,
		then turn violet.
		B. Test for zinc: Add 0.2 mL of sodium hydroxide solution to 5 mL of solution S.
		A white precipitate forms. After adding 2 mL of sodium hydroxide solution, the
		precipitate will completely dissolve and the solution will be clear. After adding
		0.1 mL of sodium sulfide solution, a fluffy white precipitate forms.
4.	Chloride	: Not more than 200 ppm.
		Dilute 5 mL of solution S with water to 15 mL According to the limit test of chloride in
		Ph. Eur.
5.	Sulfate	: Not more than 200 ppm.
		Dilute 5 mL of solution S with water to 15 mL According to the limit test of sulfate in
		Ph. Eur.
		Preparation of standard solution: Mix 5 mL of sulphate standard solution (10 ppm SO <sub>4</sub> )
		and 10 mL of distilled water.
6.	Iron	: Not more than 100 ppm.
		Dilute 2 mL of solution S with water to 10 mL. According to the limit test of iron in Ph.
		Eur.
7.	Arsenic	: Not more than 3 ppm.
		Take 0.33 g of the sample, according to the test method for arsenic in Ph. Eur (A).

8.	Lead	: Not more than 20 ppm.	
		Interpret using Ph. Eur atomic absorption spectrometry (Method II). Use a lead hollow	
		cathode lamp as the radiation source and air-acetylene flame. Absorption wavelength	
		is 283.3 nm (or 217.0 nm, depending on the instrument).	
		Test solution: Dissolve 5.0 g of the sample in 24 mL of a solution of water and nitric acid	
		(lead-free) in equal proportions, then dilute to 100.0 mL with water.	
		Standard solution: Dilute lead standard solution (containing 0.1% of lead) with lead-free	
		nitric acid solution (3.5%).	
9.	Assay	: Dissolve 0.500 g of the sample in 5 mL of dilute acetic acid solution for complex metric	
	determination	titration of zinc. 1 mL Sodium Edetate 0.1 M is equivalent to 20.95 mg (as	
		$Cl)_2H_{10}O_{14}Zn_3\cdot 3H_2O.$	
10.	Category	: Food Additives Category 8	
11.	Uses	: Nutritional additivesNutritional additives	

Lycopene	(Synthetic)
	(~)

Synonyms	INS 160d(i)
Definition	Synthetic lycopene is produced by the Wittig condensation of synthetic intermediates
	commonly used in the production of other carotenoids used in food. Synthetic lycopene
	consists predominantly of all-trans-lycopene together with 5-cis-lycopene and minor
	quantities of other isomers.
Chemical names	ψ,ψ-carotene; all- <i>trans</i> -lycopene; (all-E)-lycopene; (all-E)-2,6,10,14,19,23,27,31-
	octamethyl-2,6,8,10,12,14,16,18,20, 22,24,26,30-dotriacontatridecaene
C.A.S. number	502-65-8
Chemical formula	C <sub>40</sub> H <sub>56</sub>
Structural formula	$H_3C$ $CH_3$
Formula weight	536.9
Assay	Not less than 96% total lycopenes; not less than 70% all-trans-lycopene
Description	Red crystalline powder
Characteristics	
Identification	
Solubility	Insoluble in water, freely soluble in chloroform
Test for carotenoids	The color of the solution of the sample in acetone disappears after successive additions
	of a 5% solution of sodium nitrite and 1N sulfuric acid
Solution in chloroform	A 1% solution is clear and has intensive red-orange color
Spectrophotometry	A solution in hexane shows an absorption maximum at approximately 470 nm
Purity	
Loss on drying	Not more than 0.5% (40°C, 4 h at 10 mmHg)
Lead	Not more than 1 mg/kg
Apo-12'-lycopenal	Not more than 0.15%
Triphenyl phosphine oxide	Not more than 0.01%

(TPPO)

CategoryFood additives category (08) (09)

Functional uses Nutritional additive; Colors.

### Calcium Gluconate

The same as § 07004

#### Synthetic Genistein

Chemical : 5, 7-dihydroxy-3-(4-hydroxyphenyl)chromen-4-one

names

Synonyms : Genistein; genisteol; sophoricol; CAS No. 446-72-0

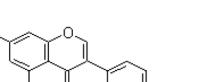
Chemical  $: C_{15}H_{10}O_5$ 

formula

Molecular : 270.2

weight

- HO I O OH ОH
- 1. : Not less than 98.5% Assay
- 2. : White or yellowish powder. Appearance
- : Not more than 0.2%. 3. Water
- 4. Sulfated ash : Not more than 0.1%.
- 5. Arsenic : Not more than 1 mg/kg.
- 6. Lead : Not more than 2 mg/kg.
- 7. : Not less than 10 mg/kg (as Pb). Heavy metals
- 8. : Food Additives Category 8 Category
- 9. : Nutritional additivesNutritional additives Uses



 $\beta$ – Carotene

The same as § 09014.

Ferrous Sulfate, Dried		
Definition	Ferrous sulfate, dried consists primarily of the monohydrate with smaller	
	amounts of the tetrahydrate.	
Molecular formula	: $FeSO_4$ ·H <sub>2</sub> O	
	FeSO <sub>4</sub> ·4H <sub>2</sub> O	
Formula weight	: Monohydrate: 169.91	
	Tetrahydrate: 223.91	
Assay	: Not less than 86% and not more than 89% of $FeSO_{4.}$	
<b>Appearance</b> : A greyish-white to buff-colored powder.		
Characteristics		
Identification		
Solubility	: Dissolves slowly in water; insoluble in ethanol.	
Iron	: Passes test	
Sulfate	: Passes test	
Purity		
Acid insoluble	: Not more than 0.05%.	
matter		
Lead	: Not more than 2 mg/kg.	
Mercury	: Not more than 1 mg/kg.	
Category	: Food Additives Category 8	
Uses	: Nutritional additivesNutritional additives	

### Sodium Molybdate (VI)

Synonyms : Sodium Molybdate Dihydrate, Disodium Molybdate Dihydrate CAS NO. 10102-40-6

Chem	ical formula: Na <sub>2</sub> Mo	Molecular weight: 241.9 (dihydrate)
1.	Purity	: Not less than 98.0% and not more than 100.5% on the dried basis.
2.	Appearance	: Whit or off-white powder or colorless crystal.
3.	Solubility	: Soluble in water.
4.	Identification	: Dissolve 0.2 g of the sample in 5 mL of nitric acid aqueous solution (nitrogen is
		mixed with water in equal volume). Add 0.1 g of ammonium chloride, 0.3 mL of
		disodium hydrogen phosphate solution and heat slowly at 50-60 $^\circ$ C to produce a
		yellow precipitate.
5.	Chloride	: Not more than 50 ppm of Cl.
6.	Phosphate	: Not more than 200 ppm of PO <sub>4</sub> .
7.	Ammonium salt	: Not more than 10 ppm, tested as 0.10 g (as NH <sub>4</sub> )
8.	Heavy metals	: Not more than 10 ppm (as Pb).
	Loss on drying	: 14.0%~16.0%. (1.000g, 140°C, 3hr)
10	Category	: Food Additives Category 8
11	Uses	: Nutritional additivesNutritional additives; Colors

### Potassium Dihydrogen Phosphate

The same as § 07026

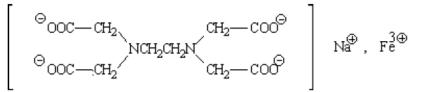
Sodium Dihydrogen Phosphate

The same as § 07029

#### L-Carnitine Tartrate

Synony	yms : L-ca	rnitine-L-tartrate (2: 1);
	Vit	amin BT-L-tartrate
	СА	S No. 36687-82-8
Chemi	cal formula: C <sub>18</sub> H <sub>36</sub> N	N <sub>2</sub> O <sub>12</sub> Molecular weight: 472.49
1.	Assay	: 67.2~69.2% (as L-carnitine)
		30.8~32.8% (as L-tartaric acid)
2.	Appearance	: White crystalline powder .
3.	Water	: Not more than 0.5%
4.	Solubility	: Soluble in water, not less than 1000 g/L (at 20°C).
5.	Specific rotation	: $[\alpha] 20 D = -11.0 \sim -9.5^{\circ}.$
6.	Identification	: This product is completely soluble in water, so it can be measured by L-carnitine and
		L-tartaric acid.
7.	Lead	: Not more than 1 ppm.
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Residue on	
	ignition	: Not more than 0.1% (600°C, 2h).
10.	Category	: Food Additives Category 8
11.	Uses	: Nutritional additivesNutritional additives

#### Ferric Sodium EDTA, EDTA FeNa



Synonyms

1.

Assay

: Ferric Sodium Edetate, Ferric Sodium EDTA Trihydrate, Sodium Feredetate, Sodium Iron EDTA,

Sodium Iron (iii) ethylenediaminetetraacetate, trihydrate

CAS No. 18154-32-0 (CAS anhydrous 15708-41-5)

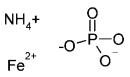
Chemical formula:  $C_{10}H_{12}FeN_2NaO_8\cdot 3H_2O$ 

 $M_2NaO_8 \cdot 3H_2O$  Molecular weight: 421.09 (trihydrate) : Not less than 99% (w/w).

		2		
	2.	Appearance	: Odorless, light yellow to tan powder.	
	3.	Solubility	: Miscible with water.	
	4.	Assay	: 12.5%~13.5% (calculated on trihydrate basis)	
	5.	EDTA	: 65.5%~70.5% (calculated on trihydrate basis)	
,	6.	pH	: 3.5~5.5 (1% solution).	
	7.	Water-insoluble	: Not more than 0.1%.	
	matter		. Not more than 0.176.	
	8.	Nitrilotriacetic acid	: Not more than 0.1%.	
	9.	Arsenic	: Not more than 1 mg/kg.	
	10.	Lead	: Not more than 1 mg/kg.	
	11.	Category	: Food Additives Category 8	
	12.	Uses	: Nutritional additivesNutritional additives	

#### Ferrous ammonium phosphate

Molecular weight: 168.85 (anhydrous)



#### Synonyms: Iron(II) ammonium phosphate; Phosphoric acid, ammonium iron (II) salt

CAS No. 10101-60-7

Chemical formula:

#### FeNH<sub>4</sub>PO<sub>4</sub>

: Not less than 24% and not more than 30% (as Iron(II)) 1. Assay 2. : Greyish green powder. Appearance 3. Solubility : Insoluble in water, soluble in diluted inorganic acids. 4. Fluorine : Not more than 50 mg/kg. 5. Iron (III) : Not more than 7%. 6. Water : Not more than 3%. 7. : Not more than 1 mg/kg. Mercury 8. Cadmium : Not more than 1 mg/kg. 9. Arsenic : Not more than 3 mg/kg. 10. Lead : Not more than 2 mg/kg. 11. : Food Additives Category 8 Category

#### **Potassium Fluoride**

CAS NUMBER		7789-23-3
Chemical formula		KF
Molecula	ar weight	58.1
Assay		Not less than 98.0% on the dried basis.
Appeara	ince	White crystal or powder.
Charact	eristics	
	Solubility	Slightly soluble in water.
	Free acid	Not more than 0.1% of HF.
	Free alkali	Not more than $0.15\%$ of $K_2CO_3$ .
	Chloride	Not more than 500 ppm of Cl.
	Sulfate	Not more than 100 ppm of SO <sub>4</sub> .
	SiF <sub>6</sub>	Not more than 0.1%.
	Sodium	Not more than 0.2%.
	Iron	Not more than 20 ppm.
	Heavy metals	Not more than 30 ppm (as Pb).
	Loss on ignition	Not more than 1% (500°C, 1h).
Category		: Food Additives Category 8
Uses		: Nutritional additivesNutritional additives

### Sodium Fluoride

Assay	Not less than 98.0% on the dried basis
Molecular weight	41.99
Chemical formula	NaF
CAS NUMBER	7681-49-4

### Characteristics

	Water-insoluble matter	Not more than 0.5%.
	Free acid	Not more than 0.25% of HF.
	Free alkali	Not more than 0.2% of Na <sub>2</sub> CO <sub>3</sub> .
	Chloride	Not more than 200 ppm of Cl.
	Sulfate	Not more than 300 ppm of SO4.
	SiF <sub>6</sub>	Not more than 0.1%.
	Iron	Not more than 50 ppm.
	Heavy metals	Not more than 30 ppm (as Pb).
	Loss on drying	Not more than 1% (150°C, 4 hr).
Category		Food Additives Category 8
Uses		Nutritional additivesNutritional additives

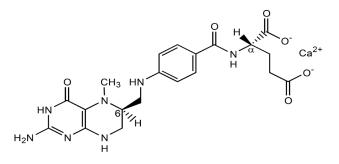
### Lycopene from Blakeslea trispora

Synonyms	INS 160d(iii)	
Definition	Lycopene from Blakeslea trispora is extracted from the fungal biomass and	
	purified by crystallization and filtration. It consists predominantly of all-	
	trans-lycopene. It also contains minor quantities of other carotenoids.	
	Isopropanol and isobutyl acetate are the only solvents used in the	
	manufacture.	
Chemical names	ψ,ψ-carotene; all- <i>trans</i> -lycopene; (all-E)-lycopene; (all-E)-	
	2,6,10,14,19,23,27,31-octamethyl-2,6,8,10,12,14,16,18,20, 22,24,26,30-	
	dotriacontatridecaene	
C.A.S. number	502-65-8	
Chemical formula	C <sub>40</sub> H <sub>56</sub>	
Structural formula	H <sub>3</sub> C CH <sub>3</sub>	
	CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> H <sub>3</sub> C CH <sub>3</sub>	
Formula weight	536.9	
Assay	Not less than 95% total lycopenes; not less than 90% all-trans-lycopene	
Description	Red crystalline powder	
Characteristics		
Identification		
Solubility	Insoluble in water, freely soluble in chloroform	
Test for carotenoids	The colour of the solution of the sample in acetone disappears after	
	successive additions of a 5% solution of sodium nitrite and 1N sulfuric acid	
Solution in chloroform	A 1% solution is clear and has intensive red-orange colour	
Spectrophotometry	A solution in hexane shows an absorption maximum at	
	approximately 470 nm	

Purity

Other carotenoids	Not more than 5%
Loss on drying	Not more than 0.5% (40°C, 4 h at 20 mmHg)
Lead	Not more than 1 mg/kg
Residual solvents	Isopropanol: Not more than 0.1%
	Isobutyl acetate: Not more than 1.0%
Category	Food additives category (08) (09)
Functional uses	Nutritional additive; Colors.

	Calcium L-5-Methyltetrahydrofolate
Synonyms	L-5-Methyltetrahydrofolic acid, calcium salt
	L-Methyltetrahydrofolate, calcium salt
	L-Methylfolate, calcium
	L-5-MTHF-Ca
Definition	Calcium L-5-methyltetrahydrofolate (L-5-MTHF-Ca) is a synthetic
	derivative of folic acid, the predominant, naturally occurring form of
	folate. It is synthesized by reduction of folic acid to tetrahydrofolic
	acid followed by methylation and diastereoselective crystallization (in
	water) of L-5-MTHF as its calcium salt. The product contains variable
	amounts of water of crystallization.
Chemical name	N-{4-[[((6S)-2-amino-3,4,5,6,7,8-hexahydro-5-methyl-4-oxo-6-
	pteridinyl)methyl]amino]benzoyl}-L-glutamic acid, calcium salt
C.A.S. number	151533-22-1
Chemical formula	C20H23CaN7O6 (anhydrous form)
Structural formula	(anhydrous form)



Formula weight	497.5 (anhydrous form)
Assay	95.0 – 102.0% (anhydrous basis)
Description	White to light yellowish, almost odourless, crystalline powder.
Characteristics	
Identification	
Solubility	Sparingly soluble in water and very slightly soluble or insoluble in
	most organic solvents; soluble in alkaline solutions.
Infrared absorption	The infrared absorption spectrum of a potassium bromide dispersion
	of the sample corresponds to that of a L-5-MTHF-Ca standard.

Calcium	Dilute 30 g of acetic acid (glacial) to 100 mL with water. Dissolve 5.3
	g of K4Fe(CN)6 in 100 mL of water. To 5 mL of the acetic acid
	solution, add 20 mg of the sample and then 0.5 mL of the potassium
	ferrocyanide solution. Mix and add 50 mg of ammonium chloride. A
	white crystalline precipitate is formed.
Liquid chromatography	Retention time matches that of a L-5-MTHF-Ca standard used in test.
Purity	
Water	Not more than 17.0% (Karl Fischer method)
	(Note: Allow sufficient time (15 min) for release of bound water.)
Calcium	7.0 - 8.5% (anhydrous basis)
Other folates and	Not more than 2.5%
related substances	
D-5-Methylfolate	Not more than 1.0%
Lead	Not more than 1 mg/kg
Cadmium	Not more than 0.5 mg/kg
Mercury	Not more than 1 mg/kg
Arsenic	Not more than 1.5 mg/kg
Category	Food Additives Category 8
Uses	Nutritional additives

### Category 9 Colors

Ponceau 4R		
Synonyms	CI Food Red 7; Cochineal Red A; New Coccine; Brilliant Scarlet; CI (1975) No. 16255;	
	INS No. 124	
Definition	Ponceau 4R consists essentially of trisodium 2-hydroxy-1-(4-sulfonato-1- naphthylazo)-	
	6,8-naphthalenedisulfonate, and subsidiary colouring matters together with sodium	
	chloride and/or sodium sulfate as the principal uncoloured components.	
	May be converted to the corresponding aluminium lake.	
Chemical names	Trisodium 2-hydroxy-1-(4-sulfonato-1-naphthylazo)-6,8- naphthalenedisulfonate	
C.A.S. number	2611-82-7	
Chemical formula	$C_{20}H_{11}N_2Na_3O_{10}S_3$	
Structural formula	NaO <sub>3</sub> S-N=N-N-N-N-NaO <sub>3</sub> S-NaO <sub>3</sub> S-NaO <sub>3</sub> S-NaO <sub>3</sub> Na	
Formula weight	604.48	
Assay	Not less than 80% total colouring matters	
Description	Reddish powder or granules	
Characteristics		
Identification		
Solubility	Soluble in water; sparingly soluble in ethanol	
Spectrophotometry	Maximum wave length: Between 505 and 510 nm	
Purity		
Loss on drying	Not more than 20% at 135° together with chloride and sulfate calculated as sodium salts	
Water-insoluble matter	Not more than 0.2%	
Subsidiary colouring matters	Not more than 1%	

Organic compounds other	4-amino-1-naphthalenesulfonic acid > 7-hydroxy-1,3-naphthalenedisulfonic acid > 3-
than colouring matters	hydroxy-2,7-naphthalened isulfonic acid $\smallsetminus$ 6-hydroxy-2-naphthalenes ulfonic acid $\searrow$ 7-
	hydroxy-1,3,6-naphthalenetrisulfonic acid: Not more than 0.5%
Unsulfonated primary	Not more than 0.01% calculated as aniline
aromatic amines	
Ether-extractable matter	Not more than 0.2%
Lead	Not more than 2 mg/kg
Category	Food additives category (09)
Functional uses	Colors

### Erythrosine (Food Red No.7)

Gene	ral name : Ery	/throsion	
Chemical names : Disodium salt of 9– (o– carboxyphenyl)– 6– hydroxy– 2, 4, 5, 7– tetraiodo– 3– isoxanthone			
Chen	Chemical formula: C <sub>20</sub> H <sub>6</sub> O <sub>5</sub> I <sub>4</sub> Na <sub>2</sub> ·H <sub>2</sub> O Molecular weight: 897.91		
1.	Assay	: Not less than 85%.	
2.	Appearance	: Red to red-brown powder or granules. Odorless.	
3.	Solution	: When 0.1 g of the sample is dissolved in 100 mL of water, the solution should be clear.	
4.	Water-insoluble	: Not more than 13%.	
	matter		
5.	pН	: 6.5~10 (1 in 100 soln)	
6.	Chloride and sulfat	te : Not more than 2% in total (as Cl, SO <sub>4</sub> ).	
7.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).	
8.	Heavy metals	: Not more than 500 ppm of iron.	
		Not more than 25 ppm of chromium.	
		Not more than 200 ppm of zinc.	
		Not more than 20 ppm (as Pb).	
9.	Subsidiary	: Not more than 3%.	
	coloring matter		
	content		
10.	Loss on drying	: Dry it at 135° for 6 hours: it loses not more than 12% of its weight.	
11.	Category	: Food Additives Category 9	
12.	Uses	: Colors	

### Erythrosine Aluminum Lake

Definition	Aluminium lakes are prepared under aqueous conditions by reacting aluminium oxide
	with colouring matter complying with purity criteria set out in the appropriate
	specification monograph. Undried aluminium oxide is usually freshly prepared by
	reacting aluminium sulfate or aluminium chloride with sodium carbonate or sodium
	bicarbonate or aqueous ammonia. Following lake formation, the product is filtered,
	washed with water and dried. Unreacted aluminium oxide may also be present in the
	final product.
Assay	Not less than 10% of Erythrosine
Characteristics	
Identification	
Solubility	Insoluble in water
Purity	
Water-soluble chlorides and	Not more than 2.0% calculated as sodium salts
sulfates	
Hydrochloric acid-insoluble	Not more than 0.5%
matters	
Ether-extractable matter	Not more than 0.2%
Arsenic	Not more than 3 mg/kg
Lead	Not more than 5 mg/kg
Category	Food additives category 9
Functional uses	Colors

### Tartrazine (Food Yellow No.4)

Gene	ral name : Tar	trazine
Chemical names : Trisodium salt of 3– carboxy– 5– hydroxy– 1– (p– sulfophenyl)– 4– (p– sulfophenylazo)–		
	р	yrazole
Chen	nical formula: C <sub>16</sub> H <sub>9</sub> C	D <sub>9</sub> N <sub>4</sub> S <sub>2</sub> Na <sub>3</sub> Molecular weight: 534.38
1.	Assay	: Not less than 85%.
2.	Appearance	: Yellowish orange to orange powder or granules, odorless.
3.	Solution	: When 0.1 g of the sample is dissolved in 100 mL of water, the solution should be clear.
4.	Water-insoluble	: Not more than 13%.
	matter	
5.	Chloride and sulfat	e : Not more than 6% in total (as Cl, SO <sub>4</sub> ).
6.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
7.	Heavy metals	: Not more than 500 ppm of iron.
		Not more than 25 ppm of chromium.
		Not more than 200 ppm of zinc.
		Not more than 20 ppm (as Pb).
8.	Subsidiary	: Not more than 3%.
	coloring matter	
	content	
9.	Loss on drying	: Not more than 10% (135°C, 6 h).
10.	Category	: Food Additives Category 9
11.	Uses	: Colors

#### Tartrazine Aluminum Lake

Definition	Aluminium lakes are prepared under aqueous conditions by reacting aluminium oxide
	with colouring matter complying with purity criteria set out in the appropriate
	specification monograph. Undried aluminium oxide is usually freshly prepared by
	reacting aluminium sulfate or aluminium chloride with sodium carbonate or sodium
	bicarbonate or aqueous ammonia. Following lake formation, the product is filtered,
	washed with water and dried. Unreacted aluminium oxide may also be present in the
	final product.
Assay	Not less than 10% of Tartrazine
Characteristics	
Identification	
Solubility	Insoluble in water
Purity	
Water-soluble chlorides and	Not more than 2.0% calculated as sodium salts
sulfates	
Hydrochloric acid-insoluble	Not more than 0.5%
matters	
Ether-extractable matter	Not more than 0.2%
Arsenic	Not more than 3 mg/kg
Lead	Not more than 5 mg/kg
Category	Food additives category 9
Functional uses	Colors

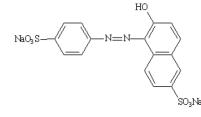
#### Sunset Yellow FCF (Food Yellow No.5)

Chemical names : Disodium 6-hydroxy-5-(4-sulfonatophenylazo)-2-naphthalene- sulfonate

Synonyms : CI Food Yellow 3; FD&C Yellow No. 6; Crelborange S; CI (1975) No. 15985; INS No. 110; CAS No. 2783-94-0.

 $Chemical \ formula \ : C_{16}H_{10}N_2Na_2O_7S_2$ 

Molecular weight: 452.38



- 1. Definition
   : Sunset Yellow FCF consists principally of the disodium salt of 6- hydroxy-5-[(4-sulfophenyl)azo]-2-naphthalenesulfonic acid and subsidiary coloring matters together

   with sodium chloride and/or sodium sulfate as the principal uncolored components.
- 2. Assay : Not less than 85% total coloring matters.
- 3. Appearance : Orange-red powder or granules
- 4. Solubility : Soluble in water; sparingly soluble in ethanol
- 5. Loss on drying : Not more than 15% together with chloride and sulfate calculated as sodium salts
- 6. Water-insoluble : Not more than 0.2%.
  - matter

7.

Subsidiary : Not more than 5%.

coloring matter Not more than 2% shall be colors other than trisodium 2-hydroxy-1- (4-

content sulfonatophenylazo)naphthalene-3,6-disulfonate.

- 8. Organic compounds other than coloring matters
   Not more than 0.5%, sum of the: monosodium salt of 4aminobenzenesulfonic acid, disodium salt of 3-hydroxy-2,7naphthalenedisulfonic acid, monosodium salt of 6-hydroxy-2naphthalenesulfonic acid, disodium salt of 7-hydroxy-1,3naphthalenedisulfonic acid, disodium salt of 4,4'-diazoaminobisbenzenesulfonic acid, and disodium salt of 6,6'-oxybis-2naphthalenesulfonic acid.
- Unsulfonated primary aromatic : Not more than 0.01%, calculated as aniline.
   amines
- 10. Ether-extractable : Not more than 0.2%.

matter

- 11. Lead : Not more than 2 mg/kg.
- 12. Arsenic : Not more than 2 mg/kg.
- 13. Category : Food Additives Category 9
- 14. Uses : Colors

### Sunset Yellow FCF Aluminum Lake

Definition	Aluminium lakes are prepared under aqueous conditions by reacting aluminium oxide
	with colouring matter complying with purity criteria set out in the appropriate
	specification monograph. Undried aluminium oxide is usually freshly prepared by
	reacting aluminium sulfate or aluminium chloride with sodium carbonate or sodium
	bicarbonate or aqueous ammonia. Following lake formation, the product is filtered,
	washed with water and dried. Unreacted aluminium oxide may also be present in the
	final product.
Assay	Not less than 10% of Sunset Yellow FCF
Characteristics	
Identification	
Solubility	Insoluble in water
Purity	
Water-soluble chlorides and	Not more than 2.0% calculated as sodium salts
sulfates	
Hydrochloric acid-insoluble	Not more than 0.5%
matters	
Ether-extractable matter	Not more than 0.2%
Arsenic	Not more than 3 mg/kg
Lead	Not more than 5 mg/kg
Category	Food additives category 9
Functional uses	Colors

### Fast Green FCF (Food Green No.3)

Gene	ral name : Fas	st Green FCF
Chen	nical names : Dis	sodium salt of 4- {[4- (N- ethyl- m- sulfobenzyl- amino)- phenyl]- (4- hydroxyl- 2-
	S	sulfophenyl)– methylene}– [1– (N– ethyl– N– m– sulfoniumbenzyl)– $\Delta^{2,5}$ –
	c	cyclohexadienimine]
Chemical formula: C <sub>37</sub> H <sub>34</sub> O <sub>10</sub> N <sub>2</sub> S <sub>3</sub> Na <sub>2</sub> Molecular weight: 808.88		
1.	Assay	: Not less than 85%.
2.	Appearance	: Dark green powder or granules with metallic glossy, odorless.
3.	Solution	: When 50 mg of the sample is dissolved in 100 mL of water, the solution should be
		clear.
4.	Water-insoluble	: Not more than 13%.
	matter	
5.	Chloride and sulfa	te : Not more than 5% in total (as Cl, SO <sub>4</sub> ).
6.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 500 iron.
		Not more than 50 ppm of chromium.
		Not more than 200 ppm of zinc.
		Others: Not more than 20 ppm (as Pb).
8.	Subsidiary	: Not more than 5%.
	coloring matter	
	content	
9.	Loss on drying	: Not more than 10% (135°C, 6 h).
10.	Category	: Food Additives Category 9
11.	Uses	: Colors

### Fast Green FCF Aluminum Lake

Definition	Aluminium lakes are prepared under aqueous conditions by reacting aluminium oxide
	with colouring matter complying with purity criteria set out in the appropriate
	specification monograph. Undried aluminium oxide is usually freshly prepared by
	reacting aluminium sulfate or aluminium chloride with sodium carbonate or sodium
	bicarbonate or aqueous ammonia. Following lake formation, the product is filtered,
	washed with water and dried. Unreacted aluminium oxide may also be present in the
	final product.
Assay	Not less than 10% of Fast Green FCF
Characteristics	
Identification	
Solubility	Insoluble in water
Purity	
Water-soluble chlorides and	Not more than 2.0% calculated as sodium salts
sulfates	
Hydrochloric acid-insoluble	Not more than 0.5%
matters	
Ether-extractable matter	Not more than 0.2%
Arsenic	Not more than 3 mg/kg
Lead	Not more than 5 mg/kg
Category	Food additives category 9
Functional uses	Colors

### Brilliant Blue FCF

Chemical names : Disodium salt of 4- {[4- (N- ethyl- m- sulfobenzyl- amino)- phenyl]- (2-			
sulfoniumphenyl)- methylene}- $[1-(N-ethyl-N-m-sulfobenzyl)-\Delta^{2,5}-cyclo-$			
	h	nexadienimine]	
Chemical formula: C <sub>37</sub> H <sub>34</sub> N <sub>2</sub> Na <sub>2</sub> O <sub>9</sub> S <sub>3</sub> Molecular weight: 792.87			
1.	Assay	: Not less than 85%.	
2.	Appearance	: Purple powder or granular with metallic luster, odorless.	
3.	Solution	: When 50 mg of the sample is dissolved in 100 mL of water, the solution should be	
		clear.	
4.	Water-insoluble	: Not more than 13%.	
	matter		
5.	5. Chloride and sulfate : Not more than 4% in total.		
6.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
7.	Heavy metals	: Not more than 500 ppm of iron.	
		鉻 Not more than 50 ppm	
		Not more than 200 ppm of zinc.	
		Others: Not more than 20 ppm (as Pb).	
8.	Subsidiary	: Not more than 5%.	
	coloring matter		
	content		
9.	Loss on drying	: Not more than 10% (135°C, 6 h).	
10.	Category	: Food Additives Category 9	
11.	Uses	: Colors	

### Brilliant Blue FCF Aluminum Lake

Definition	Aluminium lakes are prepared under aqueous conditions by reacting aluminium oxide
	with colouring matter complying with purity criteria set out in the appropriate
	specification monograph. Undried aluminium oxide is usually freshly prepared by
	reacting aluminium sulfate or aluminium chloride with sodium carbonate or sodium
	bicarbonate or aqueous ammonia. Following lake formation, the product is filtered,
	washed with water and dried. Unreacted aluminium oxide may also be present in the
	final product.
Assay	Not less than 10% of Brilliant Blue FCF
Characteristics	
Identification	
Solubility	Insoluble in water
Purity	
Water-soluble chlorides and	Not more than 2.0% calculated as sodium salts
sulfates	
Hydrochloric acid-insoluble	Not more than 0.5%
matters	
Ether-extractable matter	Not more than 0.2%
Arsenic	Not more than 3 mg/kg
Lead	Not more than 5 mg/kg
Category	Food additives category 9
Functional uses	Colors

		Food Blue No.2
General name : Indigo		ligotine (Indigo Carmine)
Chen	nical names : Dis	sodium salt of 5–5',– indigotin– disulfonic acid
Chemical formula: C <sub>16</sub> H <sub>8</sub> O <sub>8</sub>		D <sub>8</sub> N <sub>2</sub> S <sub>2</sub> Na <sub>2</sub> Molecular weight: 466.37
1.	Assay	: Not less than 85%.
2.	Appearance	: Dark purple to dark purplish brown powder or granules, odorless.
3.	Solution	: When 50 mg of the sample is dissolved in 100 mL of water, the solution should be
		clear.
4.	Water-insoluble	: Not more than 13%.
	matter	
5.	Chloride and sulfat	te : Not more than 7% in total (as Cl and SO <sub>4</sub> )
6.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 1,000 ppm of iron.
		Not more than 25 ppm of chromium.
		Not more than 200 ppm of zinc.
		Others: Not more than 20 ppm (as Pb).
8.	Subsidiary	: Not more than 3%.
	coloring matter	
	content	
9.	Loss on drying	: Not more than 10% (135°C, 6 h).
10.	Category	: Food Additives Category 9
11.	Uses	: Colors

### Indigo Carmine Aluminum Lake

Definition	Aluminium lakes are prepared under aqueous conditions by reacting aluminium oxide
	with colouring matter complying with purity criteria set out in the appropriate
	specification monograph. Undried aluminium oxide is usually freshly prepared by
	reacting aluminium sulfate or aluminium chloride with sodium carbonate or sodium
	bicarbonate or aqueous ammonia. Following lake formation, the product is filtered,
	washed with water and dried. Unreacted aluminium oxide may also be present in the
	final product.
Assay	Not less than 10% of indigo carmine
Characteristics	
Identification	
Solubility	Insoluble in water
Purity	
Water-soluble chlorides and	Not more than 2.0% calculated as sodium salts
sulfates	
Hydrochloric acid-insoluble	Not more than 0.5%
matters	
Ether-extractable matter	Not more than 0.2%
Arsenic	Not more than 3 mg/kg
Lead	Not more than 5 mg/kg
Category	Food additives category 9
Functional uses	Colors

## $\beta$ – Carotene

Synonyms	From Blakeslea trispora: CI Food Orange 5; INS No. 160a(iii)
	Other sources: CI Food Orange 5; INS No. 160a(i); CI (1975) No. 40800
Definition	These specifications consist predominantly of all-trans-\beta-carotene and may also contain
	minor amounts of <i>cis</i> -isomers and other carotenoids. Commercial preparations of $\beta$ -
	carotene intended for use in food are prepared from $\beta$ - carotene meeting these
	specifications and are formulated as suspensions in edible oils or water-dispersible
	powders. These preparations may have different ratio of trans/cis isomers. The colour is
	also obtained by a fermentation process using the fungus Blakeslea trispora.
Chemical names	From <i>Blakeslea trispora</i> : β-Carotene, β, β-carotene
	Other sources: $\beta$ -Carotene, $\beta$ , $\beta$ -carotene
	1,1'-(3,7,12,16-tetramethyl-1,3,5,7,9,11,13,15,17-octadecanonaene-1,18-diyl)bis[2,6,6-
	trimethylcyclohexene]
C.A.S. number	7235-40-7
Chemical formula	$C_{40}H_{56}$
Structural formula	All- <i>trans</i> -β-carotene (main compound)
	$H_3C$ $CH_3$ $CH_3$ $CH_3$ $H_3C$ $H_3C$ $H_3C$ $H_3C$ $CH_3$
Formula weight	536.88
Assay	Not less than 96.0% of total colouring matter (expressed as $\beta$ -carotene)
Description	Red to brownish-red crystals or crystalline powder; sensitive to oxygen and light and
	should therefore be kept in a light-resistant container under inert gas.
Characteristics	
Identification	
Solubility	Insoluble in water; practically insoluble in ethanol; slightly soluble in vegetable oils.

Test for carotenoids	The colour of a solution of the sample in acetone disappears after successive additions of
	a 5% solution of sodium nitrite and 0.5 M sulfuric acid.
Spectrophotometry	From Blakeslea trispora:
	Determine the absorbance of the diluted sample solution used in the Method of Assay at
	455 nm and 483 nm. The ratio is between 1.14 and 1.19.
	Determine the absorbance of the diluted sample solution used in the Method of Assay at
	455 nm and 340 nm. The ratio is not lower than 0.75.
	Other sources:
	Determine the absorbance of the diluted sample solution used in the Method of Assay at
	455 nm and 483 nm. The ratio $A_{455}/A_{483}$ is between 1.14 and 1.19.
	Determine the absorbance of the diluted sample solution used in the Method of Assay at
	455 nm and 340 nm. The ratio $A_{455}/A_{340}$ is not lower than 15.

### Purity

Sulfated ash	From Blakeslea trispora: Not more than 0.2%
	Other sources: Not more than 0.1%
Subsidiary colouring matters	Carotenoids other than $\beta$ -carotene: Not more than 3% of total colouring matters.
Residual solvent	From Blakeslea trispora:
	Ethanol and Ethyl acetate: Not more than 0.8% singly or in combination
	Isopropanol: Not more than 0.1%
	Isobutyl acetate: Not more than 1.0%
	Other sources:-
Lead	Not more than 2 mg/kg
Category	Food additives category (08) (09)
Functional uses	Nutritional additive; Colors.

$\beta$ – Apo– 8'– Carotenal			
Chemical formula: C <sub>30</sub> H <sub>40</sub> O		O Molecular weight: 416.65	
1.	Assay	: Not less than 98% (vacuum desiccator over sulfuric acid, 4 h)	
2.	Appearance	: Grayish purple to purple black crystalline powder, with a characteristic aroma.	
3.	Solution	: When 0.1 g of the sample is dissolved in 20 mL of chloroform, the solution should be	
		clear.	
4.	Decomposition	: 136∼140°C.	
	temperature		
5.	Absorbance	: The maximum absorption band of the cyclohexane solution of the sample is around 461	
		nm and 488 nm.	
6.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
7.	Heavy metals	: Not more than 20 ppm (as Pb).	
8.	Loss on drying	: Not more than 1% (vacuum desiccator over sulfuric acid, 4 h).	
9.	Residue on	: Not more than 0.1%.	
	ignition		
10.	Category	: Food Additives Category 9	
11.	Uses	: Colors	

$\beta$ - Apo- 8'- Carotenoate, Ethyl			
Chemical formula: C <sub>32</sub> H <sub>44</sub> O <sub>2</sub> Molecular weight: 460.70			
1.	Assay	: Not less than 98% (vacuum desiccator over sulfuric acid, 4 h)	
2.	Appearance	: Zinc brown crystalline powder.	
3.	Solution	: When 0.1 g of the sample is dissolved in 10 mL of chloroform, the solution should be	
		clear.	
4.	Decomposition	: 134~138°C	
	temperature		
5.	Absorbance	: The maximum absorption band of the cyclohexane solution of the sample is around 449	
		nm and 475 nm.	
6.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
7.	Heavy metals	: Not more than 20 ppm (as Pb).	
8.	Loss on drying	: Not more than 1% (vacuum desiccator over sulfuric acid, 4 h).	
9.	Residue on	: Not more than 0.1%.	
	ignition		
10.	Category	: Food Additives Category 9	
11.	Uses	: Colors	

Canthaxanthin			
Chemical formula: C40H52O2		O <sub>2</sub> Molecular weight: 564.86	
1.	Assay	: Not less than 98% (vacuum desiccator over sulfuric acid, 4 h)	
2.	Appearance	: Deep violet crystals or crystalline powder; slightly aromatic.	
3.	Solution	: When 0.1 g of the sample is dissolved in 10 mL of chloroform, the solution should be	
		clear.	
4.	Decomposition	: 210°C	
	temperature		
5.	Absorbance	: Canthaxanthin in cyclohexane has an absorbance maximum of 470 nm.	
6.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).	
7.	Heavy metals	: Not more than 20 ppm (as Pb).	
8.	Loss on drying	: Not more than 1% (vacuum desiccator over sulfuric acid, 4 h).	
9.	Residue on	: Not more than 0.1%.	
	ignition		
10.	Category	: Food Additives Category 9	
11.	Uses	: Colors	

## Laccaic Acid

1.	Appearance	: Red powder
2.	Solution	: Dissolve 0.1g of the sample in 30ml of 1% sodium carbonate, and the solution should
		be "clear".
3.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
4.	Heavy metals	: Not more than 30 ppm (as Pb).
5.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 14% of its weight.
6.	Residue on	: Not more than 0.8%
	ignition	
7.	Absorbance	: Dissolve 50 mg of the sample in 5 to 10 mL of 1% sodium carbonate solution, and add
		water to make it 50 mL. Take 10 mL of the solution, add water and dilute
		hydrochloric acid to make a pH 4 and 20 mL solution. When the absorbance is
		measured at a wavelength of 485 nm, its $E_{1sh}^{1\%}$ ould be 107.5 or more.
		E <sup>1%</sup> = <u>波長485nm</u> 測得之吸光度 檢體量(g)
		檢體量 $(g)$

- 8. Category : Food Additives Category 9
- 9. Uses : Colors

#### Copper Chlorophyll

- 1. Appearance : Dark green flakes, scale, powder or viscous fluid, with a characteristic odor.
- Specific : Dissolve 10 mg of the sample, accurately weighed, in 50 mL of ether. Mix and shake absorbance
   with sodium hydroxide, 2 mL of 1 in 100 methanol solution. Heat in water bath for 30 minutes with a reflux condenser. After cooling, extract three times with 10 mL of water each time, and add phosphate buffer (pH 7.5) to 200 mL. When the absorbance is measured at a wavelength of 405 nm, its E<sup>1%</sup><sub>1 cm</sub> should be 62 or more.
- 3. Arsenic : Not more than 4 ppm (as  $As_2O_3$ ).
- Salt-based coal tar
   Mix 5 mL of 1 in 100 ether solution of the sample with 1 mL of acetic acid and 5
   mL of water, filter with filter paper wetted by water. When the ether layer is left
   on the filter paper, the filtrate should be colorless.
- 5. Acidic coal tar dye, inorganic iron and copper salt
   5. Acidic coal tar dye, inorganic iron and copper salt
   5. Acidic coal tar dye, inorganic iron and copper salt
   5. The test methods of "acid coal tar dye" and "inorganic iron salt and copper salt" of "sodium copper chlorophyllin" are applicable. Test with 0.01 mL of 1 in 300 acetone solution of the sample.
- 6. Loss on drying : Dry it at 105°C for 2 hours: it loses not more than 1.5% of its weight.
- 7. Residue on : Dry it at  $105^{\circ}$ C for 2 hours: it loses not more than 2.2% of its weight.
- ignition
- 8. Category : Food Additives Category 9
- 9. Uses : Colors

#### Sodium Copper Chlorophyllin

- 1. Appearance : Blue/black powder, odorless or having not more than a slight characteristic odor.
- 2. pH : 9.5∼10.7 (1 in 100 soln)
- Specific : Dry the sample at 105°C for 1 hour. Dissolve 0.1 g of the sample, accurately weighed, absorbance in water to make a 1000 mL solution. Add phosphate buffer (pH 7.5) to 10 mL of the solution to make a 100 mL solution. When the absorbance is measured at a wavelength of 405 nm, its E<sup>1%</sup><sub>1 cm</sub> should be 508 to 568.
- 4. Arsenic : Not more than 4 ppm (as  $As_2O_3$ ).
- Sodium Copper : Add 1 mL of sodium hydroxide solution (1→50) and 50 mL of ether to 5 mL of the 1 in 200 solution of the sample, shake and mix to remove the water layer. Wash the ether layer twice with 15 mL of sodium hydroxide solution (1 50), and mix with 5 mL of acetic acid solution (150) andhe aqueous layer should be colorless.
- 6. Acidic coal tar dye : Test 0.01 mL of the aqueous solution (1→300) of the sample by the second method of filter paper chromatography. Use No. 1 filter paper, and a mixture of n-butanol, glacial acetic acid and water (4: 1: 2). Expand to 5 cm from the center point, take the filter paper off and dry it. When observed under natural light, there must be no other spot than a single circular ribbon.
- 7. Inorganic iron and : When the dried filter paper obtained by the acidic coal tar dye test is sprayed with potassium ferrocyanide solution (1→1,000) and sodium diethyldithiocarbamate solution (1→1,000). No blue or light brown spot should appear.
- 8. Loss on drying : Dry it at 105°C for 2 hours: it loses not more than 5% of its weight.
- 9. Residue on : Dry it at 105°C for 1 hour: it loses not more than 37% of its weight. ignition
- 10. Category : Food Additives Category 9
- 11. Uses : Colors

## Sodium Iron Chlorophyllin

1.	Appearance	: Dark green crystalline powder, odorless or with a slightly characteristic odor.
2.	pН	: 9.6~10.0 (1 in 100 soln)
3.	Specific	: Dissolve 0.1 g of the sample in water to make a 1,000 mL solution. Add phosphate
	absorbance	buffer (pH 7.5) to 10 mL of the solution to make a 100 mL solution. When the
		absorbance is measured at a wavelength of 398 nm, its $E_{1 \text{ cm}}^{1\%}$ should be 400 to 465.
4.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
5.	Salt-based coal tar	: The test method of "salt-based coal tar colors" of "copper chlorophyll sodium" is
	colors	applicable.
6.	Acidic coal tar dye	: The test method of "acidic coal tar dye" of "copper chlorophyll sodium" is
		applicable.
7.	Inorganic iron and	: The test method of "inorganic iron and copper salt " of "copper chlorophyll
	copper salt	sodium" is applicable.
8.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 5% of its weight.
9.	Residue on	: Not more than 30% (for material used as an anticaking agent) (105°, 1 h)
	ignition	
10.	Category	: Food Additives Category 9
11.	Uses	: Colors

## Iron Oxides

Chemical formula: FeO(OH)·xH <sub>2</sub> O		DH)∙xH <sub>2</sub> O Color: Y	ellow	Molecular weight: 88.85 (anhydrous)
	Fe <sub>2</sub> O <sub>3</sub>		Ed to red-brown	159.70
	FeO·Fe <sub>2</sub> O <sub>3</sub>		Black	231.55
1.	Assay	: Not less than 60% of F	Se.	
2.	Description	: The sample is an anhy	drous or hydrated forn	n of iron oxide. Yellow, red, brown, or
		black powder. Insolu	ble in water and organ	ic solvents, soluble in concentrated
		inorganic acids.		
3.	Water soluble	: Not more than 1.0%.		
	substances			
4.	Arsenic	: Not more than 3 ppm (	as As).	
5.	Lead	: Not more than 10 ppm		
6.	Mercury	: Not more than 1 ppm.		
7.	Barium	: Not more than 50 ppm		
8.	Cadmium	: Not more than 10 ppm		
9.	Chromium	: Not more than 100 ppr	n.	
10.	Copper	: Not more than 50 ppm		
11.	Nickel	: Not more than 100 ppr	n.	
12.	Zinc	: Not more than 100 ppr	n.	
13.	Category	: Food Additives Catego	ory 9	
14.	Uses	: Colors		

## Iron Sesquioxide

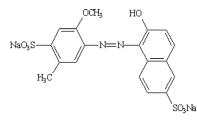
Chemical formula: Fe<sub>2</sub>O<sub>3</sub>

Molecular weight: 159.69

1.	Assay	: Not less than 98%
2.	Appearance	: Red to red-brown powder
3.	Water soluble	: Not more than 0.75%.
	substances	
4.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
	1 Hoeme	. 100 more than 2 ppm (ds / 15203).
5.	Heavy metals	: Not more than 30 ppm (as Pb).
5. 6.		

#### Allura Red AC(Food Red No. 40)

Chemical names: Disodium 6-hydroxy-5-(2-methoxy-5-methyl-4-sulfonato- phenylazo)-2-naphthalenesulfonateSynonyms: CI Food Red 17; FD&C Red No.40; CI (1975) No.16035; INS No. 129; CAS No. 25956-17-6.Chemical formula: C18H14N2Na2O8S2Molecular weight: 496.43



1.	Definition	: Mixture of disodium 6-hydroxy-5-(2-methoxy-5-methyl-4- sulfonato-phenylazo)-2-
		naphthalene-sulfonate, other coloring matter, sodium chloride and/or sodium sulfate.
2.	Assay	: Not less than 85% total coloring matters.
3.	Appearance	: Dark red to red-brown powder.
4.	Solubility	: Soluble in water, insoluble in ethanol.
5.	Loss on drying	: Not more than 15% together with chloride and sulfate calculated as sodium salts
6.	Water-insoluble	: Not more than 0.2%.
	matter	
7.	Subsidiary	: Not more than 3%.
	coloring matter	
	content	
8.	Organic compound	s other than : Not more than 0.3% of sodium 6-hydroxy-2-naphtalene sulfonate; not
	coloring matters	more than 0.2% of 4-amino-5-methoxy-2-methyl-benzene-sulfonic
		acid; not more than 1.0% of disodium 6,6'-oxybis(2-naphthalene-
		sulfonate).
9.	Unsulfonated prima	ry aromatic : Not more than 0.01%, calculated as aniline.
	amines	
10.	Ether-extractable	: Not more than 0.2%.
	matter	
11.	Lead	: Not more than 2 mg/kg.
12.	Arsenic	: Not more than 3 mg/kg.
13.	Category	: Food Additives Category 9
14.	Uses	: Colors

Riboflavin (Vitamin B2)

The same as § 08008

## Riboflavin Phosphate Sodium

The same as § 08009

#### Titanium Dioxide

Synonyms: Titania; CI Pigment white 6; CI (1975) No.77891; INS No. 171; CAS No. 13463-67-7.Chemical formula: TiO2Molecular weight: 79.88

- 1. Definition : Titanium dioxide is produced by either the sulfate or the chloride process. Processing conditions determine the form (anatase or rutile structure) of the final product. In the sulfate process, sulfuric acid is used to digest ilmenite (FeTiO3) or ilmenite and titanium slag. After a series of purification steps, the isolated titanium dioxide is finally washed with water, calcined, and micronized. In the chloride process, chlorine gas is reacted with a titanium containing mineral under reducing conditions to form anhydrous titanium tetrachloride, which is subsequently purified and converted to titanium dioxide either by direct thermal oxidation or by reaction with steam in the vapor phase. Alternatively, concentrated hydrochloric acid can be reacted with the titanium-containing mineral to form a solution of titanium tetrachloride, which is then further purified and converted to titanium dioxide by hydrolysis. The titanium dioxide is filtered, washed, and calcined. Titanium dioxide may be coated with small amounts of alumina and/or silica to improve the technological Characteristics of the product. 2. Assay : Not less than 99.0% on the dried basis (on an aluminum oxide and silicon dioxide-free basis)
- 3. Appearance : White to slightly colored powder.
- Solubility : Insoluble in water, hydrochloric acid, dilute sulfuric acid, and organic solvents.
   Dissolves slowly in hydrofluoric acid and hot concentrated sulfuric acid.
- 5. Identification : Add 5 mL sulfuric acid to 0.5 g of the sample, heat gently until fumes of sulfuric acid appear, then cool. Cautiously dilute to about 100 mL with water and filter. To 5 mL of this clear filtrate, add a few drops of hydrogen peroxide; an orange-red color appears immediately.
- 6. Loss on drying : Not more than 0.5% (105°C, 3 hr).
- 7. Loss on ignition : Not more than 1.0% (800°C, on the dried basis).
- Aluminum oxide and/or : Not more than 2%, either singly or combined.
   silicon dioxide
- Acid-soluble : Not more than 0.5%. Not more than 1.5% for products containing alumina or silica. substances

10. Water soluble : Not more than 0.5%. substances

11.	Impurities soluble in 0.5 :		
	N hydrochloric acid		
	(1) Antimony	: Not more than 2 mg/kg.	
	(2)Arsenic	: Not more than 1 mg/kg.	
	(3)Cadmium	: Not more than 1 mg/kg.	
	(4)Lead	: Not more than 10 mg/kg.	
12.	Mercury	: Not more than 1 mg/kg.	
13.	Category	: Food Additives Category 9	
14.	Uses	: Colors	

## Allura Red AC Aluminum Lake

Definition	Aluminium lakes are prepared under aqueous conditions by reacting aluminium oxide
	with colouring matter complying with purity criteria set out in the appropriate
	specification monograph. Undried aluminium oxide is usually freshly prepared by
	reacting aluminium sulfate or aluminium chloride with sodium carbonate or sodium
	bicarbonate or aqueous ammonia. Following lake formation, the product is filtered,
	washed with water and dried. Unreacted aluminium oxide may also be present in the
	final product.
Assay	Not less than 10% of Allura Red AC
Characteristics	
Identification	
Solubility	Insoluble in water
Purity	
Water-soluble chlorides and	Not more than 2.0% calculated as sodium salts
sulfates	
Hydrochloric acid-insoluble	Not more than 0.5%
matters	
Ether-extractable matter	Not more than 0.2%
Arsenic	Not more than 3 mg/kg
Lead	Not more than 5 mg/kg
Category	Food additives category 9
Functional uses	Colors

## Gold (Metallic)

Synoi	onyms : Pigment Metal 3, Aurum, Au, INS No. 175, Color Index No. 77480		
Mole	cular : 197.0.		
W	eight		
1.	Assay	: Not less than 90%	
2.	Description	: Golden powder or flakes.	
3.	Silver content	: Not more than 7.0% (Determined after complete dissolution)	
4.	Copper content	: Not more than 4.0% (Determined after complete dissolution)	
5.	Category	: Food Additives Category 9	
6.	Uses	: Colors	

Lutein

The same as § 08133

Lycopene (Synthetic)

The same as § 08139

		Potassium aluminum silicate-based pearlescent pigments		
Synonyms		: Mica-based pearlescent pigments		
1.	Description	: The sample is obtained by the reaction of mica and titanium salt at high temperature that produces titanium dioxide on mica. Different pearlescent colors can be produced by different particle size and combination with titanium dioxide.		
2.	Titanium dioxide content	: 10~65% (weight percentage)		
3.	Aluminum silicate potassium content	: 35~90% (weight percentage)		
4.	Particle size	: 3~82µm.		
5.	Loss on drying	: Not more than 0.5% (105°C, 2 hr).		
6.	pH	: 4-11.		
7.	Antimony	: Not more than 3 mg/kg.		
8.	Zinc	: Not more than 25 mg/kg.		
9.	Barium	: Not more than 25 mg/kg.		
10.	Chromium	: Not more than 100 mg/kg.		
11.	Copper	: Not more than 25 mg/kg.		
12.	Nickel	: Not more than 50 mg/kg.		
13.	Arsenic	: Not more than 3 mg/kg.		
14.	Mercury	: Not more than 1 mg/kg.		
15.	Cadmium	: Not more than 1 mg/kg.		
16.	Lead	: Not more than 4 mg/kg.		
17.	Category	: Food Additives Category 9		
18.	Uses	: Colors		

#### Caramel Colors

Synonyms : Caramel colors are divided into four classes. The synonyms for each class are: Class I: Plain caramel; INS No.150a

Class II: Sulfite caramel; INS No.150b

Class III: Ammonia caramel; INS No.150c

Class IV: Sulfite ammonia caramel; INS No.150d

Definition : Complex mixtures of compounds, some of which are in the form of colloidal aggregates, manufactured by heating carbohydrates either alone or in the presence of food-grade acids, alkalis or salts; classified according to the reactants used in their manufacture as follows:

Class I: Prepared by heating carbohydrates with or without acids or alkalis; no ammonium or sulfite compounds are used.

Class II: Prepared by heating carbohydrates with or without acids or alkalis in the presence of sulfite compounds; no ammonium compounds are used.

Class III: Prepared by heating carbohydrates with or without acids or alkalis in the presence of ammonium compounds; no sulfite compounds are used.

Class IV: Prepared by heating carbohydrates with or without acids or alkalis in the presence of both sulfite and ammonium compounds.

In all cases the carbohydrate raw materials are commercially available food-grade nutritive sweeteners consisting of glucose, fructose and/or polymers thereof. The acids and alkalis are food grade sulfuric or citric acids and sodium, potassium or calcium hydroxides or mixtures thereof.

Where ammonium compounds are used they are one or any of the following: ammonium hydroxide,

ammonium carbonate and ammonium hydrogen carbonate, ammonium phosphate, ammonium sulfate,

ammonium sulfite and ammonium hydrogen sulfite.

Where sulfite compounds are used they are one or any of the following: sulfurous acid, potassium, sodium and ammonium sulfites and hydrogen sulfites.

Food-grade anti-foaming agents may be used as processing aids during manufacture.

- 1. Appearance : Dark brown to black liquids or solids having an odor of burnt sugar.
- 2. Solubility : Miscible with water.
- Classification : Class I: Not more than 50% of the color is bound by DEAE Cellulose and not more than 50% of the color is bound by Phosphoryl Cellulose.

308

Class II: More than 50% of the color is bound by DEAE Cellulose and it exhibits an Absorbance Ratio of more than 50. Class III: Not more than 50% of the color is bound by DEAE Cellulose and more than 50% of the color is bound by Phosphoryl Cellulose. Class IV: More than 50% of the color is bound by DEAE Cellulose and it exhibits an Absorbance Ratio of not more than 50. 4. Solid content : Class I: 62-77% Class II: 65-72% Class III: 53-83% Class IV: 40-75% : 0.1% (w/v) caramel pigment solution in 1 cm colorimetric tube. The absorbance 5. Color intensity value is tested at 610 nm. Class I: 0.01-0.12 Class II: 0.06-0.10

Class III: 0.08-0.36

Class IV: 0.10-0.60

6.	Total nitrogen content	: Tested by Kjeldahl Method
		Class I: Not more than 0.1%
		Class II: Not more than 0.2%
		Class III: 1.3-6.8%
		Class IV: 0.5-7.5%
7.	Total sulfur content	: Class I: Not more than 0.3%
		Class II: 1.3-2.5%
		Class III: Not more than 0.3%
		Class IV: 1.4-10.0%
8.	Sulfur Dioxide	: Class I: -
		Class II: Not more than 0.2%
		Class III: -
		Class IV: Not more than 0.5%
9.	銨鹽氮	: Class I: -
		Class II: -
		Class III: Not more than 0.4%
		Class IV: Not more than 2.8%
10.	4-Methylimidazole,4-MEI	: Class I: -
		Class II: -
		Class III: Not more than 200 mg/kg on equivalent color basis.
		Class IV: Not more than 250 mg/kg on equivalent color basis.
		Equivalent color basis: Color intensity is expressed in units of
		0.1.

11.	2-Acetyl-4-	: Class I: -
	tetrahydroxy-	Class II: -
	butylimidazole (THI)	Class III: Not more than 25 mg/kg on equivalent color basis.
		Class IV: -
		Equivalent color basis: Color intensity is expressed in units of 0.1.
12.	Arsenic	: Not more than 1 mg/kg.
13.	Lead	: Not more than 2 mg/kg.
14.	Category	: Food Additives Category 9
15.	Uses	: Colors

Lycopene from Blakeslea trispora

The same as § 08322

# Category 10 Flavoring Agents AgentsFlavoring Agents

# § 10001

## Ethyl Acetate

Molecular weight: 88.11

1.	Assay	: Not less than 98%
2.	Appearance	: Colorless liquid, with a fruity odor.
3.	Density	: 0.897~0.906.
4.	Index of	$: n_{\rm D}^{20} = 1.370 \sim 1.375.$
	refraction	
5.	Acid value	: Not more than 0.1.
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Readily	: Cool 5 mL of the sample to $10^{\circ}$ C, mix with 5 mL of sulfuric acid slowly. Do not raise
	carbonizable	the temperature. Stand at $10^{\circ}$ C for 5 minutes, the color of solution can not be thicker
	substance	than the colorimetric solution (add water to 0.3 mL of 0.01 N iodine solution to make
		a 10 mL solution).
8.	Distillation range	: Not less than 95% v/v distils between 74°C and 78°C.
9.	Non-volatile	: Not more than 2 mg when drying 50 g of the sample at 105°C for 2 hours.
	residue	
10.	Category	: Food Additives Category 10
11.	Uses	: Flavoring Agents

## Butyl Acetate

Chemi	cal formula: C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>	Molecular weight: 116.16
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 0.880~0.887.
4.	Index of	$: n_{\rm D}^{20} = 1.392 \sim 1.397.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 122°C and 127°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

## Benzyl Acetate

Chemi	cal formula: C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	Molecular weight: 150.18
1.	Assay	: Not less than 96%.
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 1.055~1.058.
4.	Index of	$: n_{\rm D}^{20} = 1.501 \sim 1.504.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear
6.	Acid value	: Not more than 1.
7.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## Phenylethyl Acetate

Chemical formula: C <sub>10</sub> H <sub>12</sub> O		D2Molecular weight: 164.21
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 1.033~1.038.
4.	Index of	$: n_{\rm D}^{20} = 1.497 \sim 1.501.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 6 mL of 60 v/v% ethanol, the solution should
		be clear
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## Terpinyl Acetate

Chemic	cal formula: C <sub>12</sub> H <sub>20</sub> O <sub>2</sub>	Molecular weight: 196.29
1.	Assay	: Not less than 97%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.956~0.966.
4.	Index of	$: n_{\rm D}^{20} = 1.464 \sim 1.467.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 7 mL of 70 v/v% ethanol, the solution should
		be clear
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## Cinnamyl Acetate

Chemical formula: C <sub>11</sub> H <sub>12</sub> O <sub>2</sub>		D2Molecular weight: 176.22
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 1.053~1.057.
4.	Index of	$: n_{\rm D}^{20} = 1.539 \sim 1.543.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 6 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## Geranyl Acetate

Chemical formula: C <sub>12</sub> H <sub>20</sub> O <sub>2</sub>		D2Molecular weight: 196.29
1.	Assay	: Not less than 90%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.900~0.917.
4.	Index of	$: n_{\rm D}^{20} = 1.457 \sim 1.464.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 4 mL of 80 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## Citronellyl Acetate

Chemic	al formula: C <sub>12</sub> H <sub>22</sub> O <sub>2</sub>	Molecular weight: 198.31
1.	Assay	: Not less than 95%.
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 0.891~0.897.
4.	Index of	$: n_{\rm D}^{20} = 1.442 \sim 1.450.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 5 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## Linalyl Acetate

Chemic	cal formula: C <sub>12</sub> H <sub>20</sub> O <sub>2</sub>	Molecular weight: 196.29
1.	Assay	: Not less than 90%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.902~0.917.
4.	Index of	$: n_{\rm D}^{20} = 1.450 \sim 1.460.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 7 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not less than 2
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## Isoamyl Acetate

Chemical formula: C <sub>7</sub> H <sub>14</sub> O <sub>2</sub>		Molecular weight: 130.19
1.	Assay	: 98~101%.
2.	Appearance	: Colorless, clear liquid, having a characteristic fruit-like odor.
3.	Density	$: 0.872 \sim 0.878.$
4.	Index of	$: n_{\rm D}^{20} = 1.397 \sim 1.402.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 130°C and 143°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

## Cyclohexyl Acetate

Chemical formula: C <sub>8</sub> H <sub>14</sub> O <sub>2</sub>		Molecular weight: 142.20
1.	Assay	: Not less than 96%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.969~0.972.
4.	Index of	$: n_{\rm D}^{20} = 1.439 \sim 1.442.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## *l*-Menthyl Acetate

Chem	ical formula: C <sub>12</sub> H <sub>22</sub> C	D2Molecular weight: 198.30
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow, clear liquid, with a characteristic odor.
3.	Density	: 0.924~0.928.
4.	Index of	$: n_{\rm D}^{20} = 1.445 \sim 1.448.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 7 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Specific rotation	$: \alpha_{\rm D}^{20} = -70 \sim -75^{\circ}.$
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Ethyl Vanillin

Chemical formula: C <sub>9</sub> H <sub>10</sub> O <sub>3</sub>		Molecular weight: 166.18
1.	Appearance	: White to light yellow scale crystal or crystalline powder, with vanilla extract odor.
2.	Melting range	: 75.5~78.0°C.
3.	Solution	: When 1 g of the sample is dissolved in 10 mL of 60 v/v% ethanol, the solution should
		be clear.
4.	Loss on drying	: Not more than 0.5% (over sulfuric acid, 4 h)
5.	Residue on	: Not more than 0.05%.
	ignition	
6.	Category	: Food Additives Category 10
7.	Uses	: Flavoring Agents

#### Ethyl Acetoacetate

Chem	ical formula: C <sub>6</sub> H <sub>10</sub> O <sub>3</sub>	Molecular weight: 130.15
1.	Assay	: Not less than $98 \sim 102\%$ .
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 1.029~1.032.
4.	Index of	$:n_{\rm D}^{20} = 1.415 \sim 1.422.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 3 mL of 30 v/v% ethanol, the solution should
		be clear.
6.	Free acid	: Add 25 mL of fresh boiled and cooled water to 5 g of the sample, stir and mix for 2
		minutes, then allow it to stand. Add 2 drops of phenolphthalein TS and 3.4 mL of 0.1
		N potassium hydroxide solution to 10 mL of the water layer taken from the solution.
		Red color should appear.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### Eugenol

Chemical formula: C <sub>10</sub> H <sub>12</sub> O		D2Molecular weight: 164.21
1.	Assay	: Not less than 98 v/v%
2.	Appearance	: Colorless or pale tan, clear liquid; having a specific aroma.
3.	Density	: 1.065~1.071.
4.	Index of	$: n_{\rm D}^{20} = 1.539 \sim 1.542.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Category	: Food Additives Category 10
7.	Uses	: Flavoring Agents

### Butyric Acid

Chemical formula: C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>		Molecular weight: 88.11
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless liquid, with a stimulating odor.
3.	Density	: 0.958~0.961.
4.	Index of	$: n_{\rm D}^{20} = 1.398 \sim 1.401.$
	refraction	
5.	Sulfate	: Not more than 19.2 ppm of SO <sub>4</sub> .
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Distillation range	: Not less than 95% v/v distils between 160°C and 165°C.
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### Ethyl Butyrate

Chem	ical formula: C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>	Molecular weight: 116.16
1.	Assay	: Not less than 98% and not more than 102%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.874~0.882.
4.	Index of	$: n_{\rm D}^{20} = 1.390 \sim 1.395.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 112°C and 123°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Butyl Butyrate

Chemi	cal formula: C <sub>8</sub> H <sub>16</sub> O <sub>2</sub>	Molecular weight: 144.21
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.866~0.871.
4.	Index of	$: n_{\rm D}^{20} = 1.404 \sim 1.408.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 160°C and 169°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Isoamyl Butyrate

Chem	ical formula: C <sub>9</sub> H <sub>18</sub> O <sub>2</sub>	Molecular weight: 158.24
1.	Assay	: 98∼101%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.863~0.867.
4.	Index of	$: n_{\rm D}^{20} = 1.409 \sim 1.413.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 5 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 175°C and 182°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Cyclohexyl Butyrate

Chem	ical formula: $C_{10}H_{18}C_{10}$	D2Molecular weight: 170.25
1.	Assay	: Not less than 97%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.941~0.945.
4.	Index of	$: n_{\rm D}^{20} = 1.441 \sim 1.444.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 5 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

#### Undecalactone

Chemical formula: C <sub>11</sub> H <sub>20</sub> O <sub>2</sub>		D2Molecular weight: 184.28
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.944~0.948.
4.	Index of	$: n_{\rm D}^{20} = 1.450 \sim 1.455.$
	refraction	
5.	Acid value	: Not more than 5.
6.	Solution	: When 2 mL of the sample is dissolved in 6 mL of 70 v/v% ethanol, the solution should
		be clear.
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

### Anisaldehyde

Chemical formula: C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>		Molecular weight: 136.15
1.	Assay	: Not less than 95%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 1.123~1.128.
4.	Index of	$: n_{\rm D}^{20} = 1.571 \sim 1.576.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 5 mL of 60 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 10.
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

### Ethyl Caproate

Chem	ical formula: C <sub>8</sub> H <sub>16</sub> O <sub>2</sub>	Molecular weight: 144.22
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.869~0.875.
4.	Index of	$: n_{\rm D}^{20} = 1.404 \sim 1.409.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 8 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 163°C and 170°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Allyl Caproate

Chem	ical formula: C <sub>9</sub> H <sub>16</sub> O <sub>2</sub>	Molecular weight: 156.23
1.	Assay	: 98∼102%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.887~0.893.
4.	Index of	$: n_{\rm D}^{20} = 1.422 \sim 1.426.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 7 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

#### Nonalactone

Chem	ical formula: C <sub>9</sub> H <sub>16</sub> O <sub>2</sub>	Molecular weight: 156.23
1.	Assay	: Not less than 97%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.965~0.970.
4.	Index of	$: n_{\rm D}^{20} = 1.445 \sim 1.450.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 5.
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

### Geranyl Formate

Chemi	ical formula: C <sub>11</sub> H <sub>18</sub> C	Molecular weight: 182.26
1.	Assay	: Not less than 80%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.909~0.920.
4.	Index of	$:n_{\rm D}^{20}=1.460\sim1.466.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 4 mL of 80 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1 (Titrate under cold water until light red color lasting for 10 seconds).
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

#### Isoamyl Formate

Chem	ical formula: C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>	Molecular weight: 116.61
1.	Assay	: Not less than 95%.
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 0.880~0.886.
4.	Index of	$: n_{\rm D}^{20} = 1.395 \sim 1.400.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1 (Titrate under cold water until light red color lasting for 10 seconds).
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 92% v/v distils between 120°C and 124.5°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Citronellyl Formate

Chemical formula: C <sub>11</sub> H <sub>20</sub> O		Molecular weight: 184.28
1.	Assay	: Not less than 86%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 0.891~0.900.
4.	Index of	$: n_{\rm D}^{20} = 1.444 \sim 1.450.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 3 mL of 80 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1 (Titrate under cold water until light red color lasting for 10 seconds)
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### Methyl Salicylate

Chemical formula: C <sub>8</sub> H <sub>8</sub> O <sub>3</sub>		Molecular weight: 152.15
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 1.180~1.187.
4.	Index of	$:n_{\rm D}^{20}=1.534\sim 1.538.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 8 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1 (Use phenolphthalein TS as indicator)
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### Ethyl Propionate

Chemi	cal formula: C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Molecular weight: 102.14
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	$: 0.891 \sim 0.894.$
4.	Index of	$: n_{\rm D}^{20} = 1.382 \sim 1.385.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 3 mL of 50 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 95°C and 100°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Benzyl Propionate

Chem	ical formula: $C_{10}H_{12}C$	D2Molecular weight: 164.21
1.	Assay	: Not less than 93%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 1.032~1.037.
4.	Index of	$: n_{\rm D}^{20} = 1.495 \sim 1.501.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 5 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

#### Isoamyl Propionate

Chem	cal formula: C <sub>8</sub> H <sub>16</sub> O <sub>2</sub>	Molecular weight: 144.22
1.	Assay	: 98~101%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.868~0.873.
4.	Index of	$: n_{\rm D}^{20} = 1.404 \sim 1.409.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 156°C and 165°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Methyl<sub>β</sub>- Naphthyl Ketone

Chem	ical formula: C <sub>12</sub> H <sub>10</sub> O	Molecular weight: 170.21
1.	Appearance	: White to pale yellow crystals or crystalline powders, with a characteristic aroma.
2.	Melting range	: 51.5~54°C.
3.	Solution	: Dissolve 0.1 g of the sample in 10 mL of 70 v/v% ethanol by heating to 30°C, the
		solution should be clear.
4.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
5.	Nitrobenzene	: Mix 20 mg of the sample with 10 mL of dilute ethanol, add 2 mL of hydrochloric acid,
		2 mL of water and 0.5 g of zinc powder. Cover with a glass dish and slightly heat in a
		water bath. Add 10 mL of water, cool with ice and filter. Mix the filtrate with sodium
		nitrite solution (1f hydrochlshake for 1 minute. After standing for 10 minutes, add 1
		mL of p-aminobenzenesulphonic acid solution (1ding, mix and shake for 10 minutes.
		Then, add 2 mL of N-(1-naphthyl)-N-(diethyl)-bis-aminoethylene dihydrochloride
		solution (1 $\rightarrow$ ne d. When placed for 10 minutes, the liquid color should not be thicker
		than the matching fluid s.
6.	Loss on drying	: Not more than 0.5% (over sulfuric acid, 4 h)
7.	Residue on	: Not more than 0.05%.
	ignition	
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### Methyl N – Methyl Anthranilate

Chem	ical formula: C <sub>9</sub> H <sub>11</sub> N	O <sub>2</sub> Molecular weight: 165.20
1.	Assay	: Not less than 98% and not more than 102%.
2.	Appearance	: Colorless to pale yellow, clear liquid or crystal, with a characteristic aroma. The liquid
		form has a distinctive violet fluorescence.
3.	Density	: 1.128~1.135.
4.	Index of	: n $_{\rm D}^{20} = 1.577 \sim 1.582.$
	refraction	
5.	Setting	: Not less than 10°C.
	temperature	
6.	Solution	: When 1 mL of the sample is dissolved in 10 mL of 70 v/v% ethanol, the solution
		should be clear.
7.	Acid value	: Not more than 1.
8.	Residue on	: Not more than 0.05%.
	ignition	
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Piperonal(Heliotropin)

Chemical formula: C <sub>8</sub> H <sub>6</sub> O <sub>3</sub>		Molecular weight: 150.14
1.	Assay	: Not less than 99% on the dried basis (over sulfuric acid, 4 h)
2.	Appearance	: White crystals or masses, having a characteristic aroma.
3.	Melting range	: 35.5~37°C.
4.	Solution	: When 1 g of the sample is dissolved in 8 mL of 70 v/v% ethanol, the solution
		should be clear.
5.	Loss on drying	: Not more than 0.5% (over sulfuric acid, 4 hr).
6.	Residue on	: Not more than 0.05%.
	ignition	
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

### Ethyl Oenanthate

Chem	ical formula: C <sub>9</sub> H <sub>18</sub> O <sub>2</sub>	Molecular weight: 158.24
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.869~0.874.
4.	Index of	$: n_{\rm D}^{20} = 1.411 \sim 1.416.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 5 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 185°C and 193°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

# Octyl Aldehyde

Chemical formula: C <sub>8</sub> H <sub>16</sub> O		Molecular weight: 128.22
1.	Assay	: Not less than 92%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.821~0.833.
4.	Index of	$: n_{\rm D}^{20} = 1.417 \sim 1.426.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 3 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Free acid	: 1 mL of the sample is dissolved in 10 mL of neutral ethanol. When adding 2 drops of
		phenolphthalein TS and 1.5 mL of 0.1 N sodium hydroxide solution, the solution
		should be red.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### Ethyl Caprylate

Chemical formula: C <sub>10</sub> H <sub>20</sub> O <sub>2</sub>		D2Molecular weight: 172.27
1.	Assay	: Not less than 98% and not more than 101%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.867~0.871.
4.	Index of	$: n_{\rm D}^{20} = 1.416 \sim 1.419.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 8 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

#### Linalool

Chem	cal formula: C <sub>10</sub> H <sub>18</sub> O	Molecular weight: 154.25
1.	Assay	: Not less than 92%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 0.864~0.876.
4.	Index of	$: n_{\rm D}^{20} = 1.461 \sim 1.465.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
7.	Acid value	: Not more than 1.
8.	Ester value	: Not less than 2
9.	Heavy metals	: Not more than 10 ppm (as Pb).
10.	Distillation range	: Not less than 85% v/v distils between 195°C and 205°C.
11.	Category	: Food Additives Category 10
12.	Uses	: Flavoring Agents

### Benzyl Alcohol

Chemical formula: C7H8O		Molecular weight: 108.14
1.	Appearance	: Colorless, clear liquid, with a faint, aromatic odor.
2.	Density	: 1.042~1.053.
3.	Solution	: When 1 mL of the sample is dissolved in 50 mL of water, the oil should not be
		separated immediately even though the solution is turbid.
4.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
5.	Free Acid and free	alkali : Dissolve 1 g of the sample in 10 mL of neutral ethanol. Add 2 drops of
		phenolphthalein TS. No red color should appear. Add a drop of 0.1 N
		sodium hydroxide. A red color should appear.
6.	Distillation range	: Not less than 90% v/v distils between 202.5 and 206.5°C.
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

### Benzaldehyde

Chemical formula: C <sub>7</sub> H <sub>6</sub> O		Molecular weight: 106.12
1.	Assay	: Not less than 97%
2.	Appearance	: Colorless with a characteristic, strong odor.
3.	Density	: 1.044~1.049.
4.	Index of	$: n_{\rm D}^{20} = 1.544 \sim 1.547.$
	refraction	
5.	Acid value	: Not more than 10.
6.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
7.	Hydrogen cyanide	: Add 5 mL of water to 0.5 mL of the sample, shake and mix. Add 0.5 mL of sodium
		hydroxide TS and 0.1 mL of ferrous sulfate TS, heat slowly. After cooling, make it
		weakly acidic with dilute hydrochloric acid. The solution should not be greenish
		blue or produce blue precipitate within 15 minutes.
8.	Nitrobenzene	: Dissolve 1 mL of the sample in 20 mL of ethanol, add water until it is slightly turbid.
		Add 3 g of zinc powder and 10 mL of dilute sulfuric acid to have it release hydrogen
		gas for 1 hour, and filter. After the filtrate is concentrated to 20 mL in water bath, add 2
		drops of potassium chromate TS and heat. The solution should not be purple.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

#### Acetophenone

Chemical formula: C <sub>8</sub> H <sub>8</sub> O		Molecular weight: 120.15
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless liquid or white crystals with characteristic odor.
3.	Melting point	: 18~20°C.
4.	Index of	$:n_{\rm D}^{20}=1.532\sim 1.534.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 4 mL of 60 v/v% ethanol, the solution should
		be clear.
6.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### Ethyl Phenyl Acetate

Chem	ical formula: $C_{10}H_{12}C$	D2Molecular weight: 164.21
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 1.031~1.036.
4.	Index of	: n $_{\rm D}^{20} = 1.496 \sim 1.500.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 3 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Isobutyl Phenyl Acetate

Chemic	al formula: C <sub>12</sub> H <sub>16</sub> O <sub>2</sub>	Molecular weight: 192.26
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 0.987~0.992.
4.	Index of	$: n_{\rm D}^{20} = 1.486 \sim 1.488.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 8 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Isoamyl Phenyl Acetate

Chem	ical formula: C <sub>13</sub> H <sub>18</sub> C	D2Molecular weight: 206.29
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 0.978~0.980.
4.	Index of	$: n_{\rm D}^{20} = 1.485 \sim 1.487.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 4 mL of 80 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

#### Citronellol

Chemic	cal formula: C <sub>10</sub> H <sub>20</sub> O	Molecular weight: 156.27
1.	Assay	: Not less than 94%
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 0.853∼0.864.
4.	Index of	$: n_{\rm D}^{20} = 1.453 \sim 1.462.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Ester value	: Not more than 4.
8.	Aldehydes	: Not more than 2% (as Citronellal)
9.	Heavy metals	: Not more than 10 ppm (as Pb).
10.	Category	: Food Additives Category 10
11.	Uses	: Flavoring Agents

#### Citronellal

Chemie	cal formula: C <sub>10</sub> H <sub>18</sub> O	Molecular weight: 154.25
1.	Assay	: Not less than 85%.
2.	Appearance	: Colorless liquid with a characteristic odor.
3.	Density	: 0.852~0.859.
4.	Index of	$: n_{\rm D}^{20} = 1.446 \sim 1.452.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 3.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

#### Geraniol

Chemio	cal formula: C <sub>10</sub> H <sub>18</sub> O	Molecular weight: 154.25
1.	Assay	: Not less than 85%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.870~0.885.
4.	Index of	$: n_{\rm D}^{20} = 1.469 \sim 1.478.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 3 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Ester value	: Not more than 6
8.	Aldehydes	: Not more than 2.3% (as Citronellal)
9.	Heavy metals	: Not more than 10 ppm (as Pb).
10.	Category	: Food Additives Category 10
11.	Uses	: Flavoring Agents

### Vanillin

Chemical formula: C <sub>8</sub> H <sub>8</sub> O <sub>3</sub>		Molecular weight: 152.15
1.	Assay	: 97~103%.
2.	Appearance	: White to slightly yellow crystal flakes with a vanilla odor.
3.	Melting range	: 81~83°C.
4.	Solution	: When 1 g of the sample is dissolved in 20 mL of water and heated to $80^{\circ}$ C, the
		solution should be clear.
5.	Arsenic	: Not more than 3 ppm (as $As_2O_3$ ).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Loss on drying	: Not more than 0.5% (over sulfuric acid, 4 hr).
8.	Residue on	: Not more than 0.05%.
	ignition	
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

## Cinnamic Aldehyde

Chemical formula: C <sub>9</sub> H <sub>8</sub> O	Molecular weight: 132.16
1. Assay	: Not less than 98%
2. Appearance	: Colorless to yellow liquid, with a strong cinnamon odor and stimulating sweet taste.
3. Density	$: 1.050 \sim 1.056.$
4. Index of	$: n_{\rm D}^{20} = 1.619 \sim 1.625.$
refraction	
5. Solution	: When 1 mL of the sample is dissolved in 10 mL of 60 v/v% ethanol, the solution
	should be clear.
6. Acid value	: Not more than 5.
7. Chloride	: The result of copper mesh flame test shall not be green.
compound	
8. Residue on	: Not more than 0.02%.
ignition	
9. Category	: Food Additives Category 10
10. Uses	: Flavoring Agents

### Cinnamyl Alcohol

Chemical formula: C <sub>9</sub> H <sub>10</sub> O		Molecular weight: 134.18
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to light yellow liquid or white to light yellow crystal, with special odor.
3.	Setting	: 28~33°C.
	temperature	
4.	Solution	: Add 3 mL of 50 v/v% ethanol to 1 g of the sample and dissolve it at 35°C. The
		solution should be "clear".
5.	Acid value	: Not more than 1.
6.	Cinnamaldehyde	: Not more than 1.5%.
7.	Residue on	: Not more than 0.03%.
	ignition	
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### Cinnamic Acid

Chemical formula: C <sub>9</sub> H <sub>8</sub> O <sub>2</sub>		Molecular weight: 148.16
1.	Assay	: Not less than 99% (over sulfuric acid, 3 h)
2.	Appearance	: White crystalline scales, with a characteristic odor.
3.	Melting range	: 132∼135°C.
4.	Solution	: When 1 g of the sample is dissolved in 7 mL of ethanol, the solution should be
		clear.
5.	Alkaline solution	: When 0.2 g of the sample is dissolved sodium carbonate TS and 3 mL of water, the
		solution should be "clear".
6.	Chloride	: Not more than 14 ppm of Cl.
	compound	
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Loss on drying	: Not more than 1% (over sulfuric acid, 3 h).
9.	Residue on	: Not more than 0.05%.
	ignition	
10.	Category	: Food Additives Category 10
11.	Uses	: Flavoring Agents

## Methyl Cinnamate

Chemical formula: C <sub>10</sub> H <sub>10</sub> O <sub>2</sub>		D2Molecular weight: 162.19
1.	Assay	: Not less than 98%
2.	Appearance	: White solid, with a characteristic odor.
3.	凝固點	: Not less than 33.8°C.
4.	Solution	: When 1 g of the sample is added to 3 mL of 70 v/v% ethanol and heated until
		dissolved, the solution should be clear.
5.	Acid value	: Not more than 1.
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

## Ethyl Cinnamate

Chemic	cal formula: C <sub>11</sub> H <sub>12</sub> O <sub>2</sub>	Molecular weight: 176.22
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 1.050~1.055.
4.	Index of	$: n_{\rm D}^{20} = 1.556 \sim 1.562.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 5 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## Decyl Aldehyde

Chemical formula: C <sub>10</sub> H <sub>20</sub> O		Molecular weight: 156.27
1.	Assay	: Not less than 93%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.826~0.835.
4.	Index of	$: n_{\rm D}^{20} = 1.427 \sim 1.435.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 6 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 10.
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

## Decyl Alcohol

Chemical formula: C <sub>10</sub> H <sub>22</sub> O		Molecular weight: 158.29
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Setting	: Not less than 5°C.
	temperature	
4.	Density	: 0.826~0.831.
5.	Index of	$: n_{\rm D}^{20} = 1.435 \sim 1.438.$
	refraction	
6.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
7.	Acid value	: Not more than 1.
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

## Eucalyptol (Cincol)

Chem	ical formula: C <sub>10</sub> H <sub>18</sub> C	Molecular weight: 154.24
1.	Assay	: Not less than 85%.
2.	Appearance	: Colorless to pale yellow liquid, with a camphor like odor.
3.	Density	: 0.915~0.929.
4.	Index of	$: n_{\rm D}^{20} = 1.454 \sim 1.462.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 6 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Resorcin	: Add 5 mL of water to 1 mL of the sample. Add 1 mL of the mixture of 1
		mL of mercury nitrate TS and 3 mL of water. Shake and heat in water
		bath for 2 minutes. Cool, add 1 drop of dilute sulfuric acid and 1 drop of
		sodium nitrite TS. When heated in water bath for 2 minutes, the water
		layer shall not appear yellow to tan.
7.	Specific rotation	$: \alpha_{D}^{20} = -3 \sim +10^{\circ}.$
8.	Phellandrene	: Dissolve 2.5 mL of the sample in 5 mL of petroleum benzene,
		then add 10 mL of sodium nitrite solution $(1\rightarrow 20)$ . Slowly
		add 5 mL of glacial acetic acid. No crystals form within 10
		minutes.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Isoeugenol

Chem	ical formula: C <sub>10</sub> H <sub>12</sub> C	D2Molecular weight: 164.21
1.	Assay	: Not less than 98 v/v%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 1.083~1.090.
4.	Index of	$: n_{D}^{20} = 1.572 \sim 1.577.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Category	: Food Additives Category 10
7.	Uses	: Flavoring Agents

## Ethyl Isovalerate

Chemi	cal formula: C7H14O2	Molecular weight: 130.19
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.866~0.870.
4.	Index of	$: n_{\rm D}^{20} = 1.393 \sim 1.398.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 6 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 130°C and 140°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Isoamyl Isovalerate

Chemical formula: C <sub>10</sub> H <sub>20</sub> O		Molecular weight: 172.27
1.	Assay	:98~101%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.857~0.862.
4.	Index of	$: n_{\rm D}^{20} = 1.410 \sim 1.415.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 8 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Distillation range	: Not less than 95% v/v distils between 185°C and 195°C.
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

## Allyl Isothiocyanate

Chemical formula: C <sub>4</sub> H <sub>5</sub> NS		S Molecular weight: 99.16
1.	Assay	: Not less than 97%
2.	Appearance	: Colorless to pale yellow oily liquid, with strong characteristic odor.
3.	Density	: 1.018~1.023.
4.	Index of	$: n_{\rm D}^{20} = 1.528 \sim 1.531.$
	refraction	
5.	Carbon disulfide, J	betroleum, essential oil and fatty : Cool 3 mL of the sample and add 4 mL of sulfuric
	acid	acid to it slowly. When oscillating and mixing,
		the solution shall not be red or turbid nor produce
		colorless oil droplet or oil layer.
6.	Phenols and thiocy	ranate compounds : Dissolve 1 mL of the sample in 5 mL of ethanol. When adding 1
		drop of ferric chloride TS, the solution shall not be blue or red.
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

### Maltol

Chemical formula: C <sub>6</sub> H <sub>6</sub> O <sub>3</sub>		Molecular weight: 126.11
1.	Appearance	: White to off-white crystalline powder having a characteristic odor.
2.	Melting range	: 160∼163°C.
3.	Solution	: When 0.1 mL of the sample is dissolved in 5 mL of 70 v/v% ethanol, the solution
		should be clear.
4.	Loss on drying	: Not more than 0.5% (over sulfuric acid, 4 hr).
5.	Residue on	: Not more than 0.05%.
	ignition	
6.	Category	: Food Additives Category 10
7.	Uses	: Flavoring Agents

## Methyl Anthranilate

Chemic	cal formula: C <sub>8</sub> H <sub>9</sub> O <sub>2</sub> N	Molecular weight: 151.17
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to light yellow liquid or crystalline block. The liquid form has a unique blue-
		violet fluorescent. With a characteristic aroma.
3.	Setting	: 22~24°C.
	temperature	
4.	Solution	: When 2 mL of the sample is dissolved in 6 mL of 70 v/v% ethanol at 30°C, the
		solution should be clear.
5.	Index of	$: n_{\rm D}^{20} = 1.580 \sim 1.585.$
	refraction	
6.	Acid value	: Not more than 1.
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

## Hydroxy Citronellal

Chem	ical formula: $C_{10}H_{20}C$	D2Molecular weight: 172.27
1.	Assay	: Not less than 90%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.921~0.928.
4.	Index of	$: n_{\rm D}^{20} = 1.447 \sim 1.450.$
	refraction	
5.	Solution	: When 2 mL of the sample is dissolved in 3 mL of 50 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 10.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

## Hydroxy Citronellal Dimethylacetal

Chemical formula: C <sub>12</sub> H <sub>26</sub> O <sub>2</sub>		Molecular weight: 218.34
1.	Assay	: Not less than 95%.
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.928~0.934.
4.	Index of refraction	$: n_{\rm D}^{20} = 1.441 \sim 1.444.$
5.	Solution	: When 2 mL of the sample is dissolved in 4 mL of 50 v/v% ethanol, the solution
		should be clear.
6.	Acid value	: Not more than 1.
7.	Hydroxycitronellal	: Accurately weigh 5 g of the sample, add 70 mL of hydroxylamine TS, shake, and let it
		stand for 1 hour. The excess hydroxylamine is titrated with 0.5 N hydrochloric acid.
		the required 0.5 N hydrochloric acid solution should be below 3 mL.
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

## *l*–Perillaldehyde

Chem	cal formula: C <sub>10</sub> H <sub>14</sub> O	Molecular weight: 150.22
1.	Assay	: Not less than 90%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.965~0.980.
4.	Index of	$:n_{\rm D}^{20} = 1.502 \sim 1.510.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 3 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 3.
7.	Specific rotation	$: \alpha_{\rm D}^{20} = -100 \sim -150^{\circ}.$
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Category	: Food Additives Category 10
10.	Uses	: Flavoring Agents

### Ionone

Chemic	cal formula: C <sub>13</sub> H <sub>20</sub> O	Molecular weight: 192.30
1.	Assay	: Not less than 90%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.930~0.948.
4.	Index of	$: n_{\rm D}^{20} = 1.497 \sim 1.522.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 4 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Residue on	: Not more than 0.05%.
	ignition	
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

### p-Methyl Acetophenone

Chemical formula: C <sub>9</sub> H <sub>10</sub> O		Molecular weight: 134.18
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 1.004~1.008.
4.	Index of	$:n_{\rm D}^{20}=1.532\sim 1.535.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 3 mL of 70 v/v% ethanol, the solution should
		be clear.
6.	Chloride	: The result of copper mesh flame test shall not be green.
	compound	
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### dl-Menthol

Chemical formula: C <sub>10</sub> H <sub>20</sub> O		Molecular weight: 156.27
1.	Appearance	: Colorless columnar or needle-like crystal, or white crystalline powder with a
		characteristic cool fragrance. Tastes burning at first and becomes cool afterward.
2.	Melting range	: 32∼38°C.
3.	Non-volatile	: Not more than 0.05% (heating and volatilizing on water bath)
	matter	
4.	Thymol	: 0.2 g of the sample should not be colored when added to the cold mixture of 2 mL of
		glacial acetic acid, 6 drops of sulfuric acid and 2 drops of nitrate acid.
5.	Nitromethane	: Add 1 mL of sodium hydroxide TS to 0.5 g of the sample, heat and boil for about 2
		minutes. After the solution cools, add 0.5 mL of sulfanilic acid· $\alpha$ -naphthylamine
		T.S., it should not be red for 5 minutes.
6.	Category	: Food Additives Category 10
7.	Uses	: Flavoring Agents

### *l*–Menthol

Chemical formula: C <sub>10</sub> H <sub>20</sub> O		Molecular weight: 156.27
1.	Appearance	: Colorless columnar or needle-like crystal, or white crystalline powder with a
		characteristic cool fragrance. Tastes burning at first and becomes cool afterward.
2.	Melting range	: 42~43°C.
3.	Specific rotation	: $(\alpha)_{D}^{20} = 45 \sim -51^{\circ}$ (Dissolve 2.5 g of the sample in ethanol to make a 25 mL
		solution)
4.	Non-volatile	: Not more than 0.05% (heating and volatilizing on water bath)
	matter	
5.	Thymol	: 0.2 g of the sample should not be colored when added to the cold mixture of 2 mL of
		glacial acetic acid, 6 drops of sulfuric acid and 2 drops of nitrate acid.
6.	Nitromethane	: Add 1 mL of sodium hydroxide TS to 0.5 g of the sample, heat and boil for about 2
		minutes. After the solution cools down, add 0.5 mL of sulfanilic acid $\alpha$ -
		naphthylamine T.S., the solution should not be red for 5 minutes.
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

## α–Amyl Cinnamic Aldehyde

Chem	ical formula: C <sub>14</sub> H <sub>18</sub> O	Molecular weight: 202.30
1.	Assay	: Not less than 98%
2.	Appearance	: Colorless to pale yellow liquid, with a characteristic odor.
3.	Density	: 0.967~0.972.
4.	Index of	$: n_{\rm D}^{20} = 1.554 \sim 1.559.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 5 mL of 80 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Residue on	: Not more than 0.05%.
	ignition	
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### Citral

Chemic	cal formula: C <sub>10</sub> H <sub>16</sub> O	Molecular weight: 152.24
1.	Assay	: Not less than 96%.
2.	Appearance	:Colorless to pale yellow liquid, having a strong lemon like odor.
3.	Density	: 0.888~0.894.
4.	Index of	$: n_{\rm D}^{20} = 1.486 \sim 1.490.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 10 mL of 60 v/v% ethanol, the solution
		should be clear.
6.	Acid value	: Not more than 5.
7.	Category	: Food Additives Category 10
8.	Uses	: Flavoring Agents

## Allyl Cyclohexyl Propionate

Chemical formula: C <sub>12</sub> H <sub>20</sub> O <sub>2</sub>		Molecular weight: 196.29
1.	Assay	: Not less than 98%
2.	Appearance	: Pale yellow, clear liquid having a characteristic odor.
3.	Density	: 0.948~0.953.
4.	Index of	$: n_{\rm D}^{20} = 1.457 \sim 1.462.$
	refraction	
5.	Solution	: When 1 mL of the sample is dissolved in 4 mL of 80 v/v% ethanol, the solution should
		be clear.
6.	Acid value	: Not more than 1.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Category	: Food Additives Category 10
9.	Uses	: Flavoring Agents

### d–Borneol

Chemical formula: C <sub>10</sub> H <sub>18</sub> O		Molecular weight: 154.25
1.	Appearance	: White crystal, crystalline powder or block, with a characteristic aroma.
2.	Melting range	: 207~210°C.
3.	Specific rotation	: $(\alpha)_{D}^{20} = +16 \rightarrow +37^{\circ}$ (Dissolve 1 g of the sample in ethanol to make a 10 mL solution)
4.	Non-volatile	: Not more than 0.05% (After heating and sublimating on water bath, dry at 105°C for 1
	matter	hour)
5.	Category	: Food Additives Category 10
6.	Uses	: Flavoring Agents

L-Cystein Monohydrochloride

The same as § 03012.

Monosodium L-Aspartate

Chemical names	
C.A.S. number	

Monosodium (2S)-2-aminobutanedioate monohydrate

3792-50-5

 $C_4H_6\ NNa\ O_4\boldsymbol{\,\cdot\,} H_2O$ 

Chemical formula

Structural formula

Formula weight	173.10
Content	Monosodium L-Aspartate, when calculated on the dried basis, contains not less
	than 98% of monosodium L-aspartate
Description	Monosodium L-Aspartate occurs as colorless to white prisms or as a white
	crystalline powder. It has a characteristic taste.
Identification	(1) To 5 mL of a solution of Monosodium L-Aspartate (1 in 1000), add 1 mL of
	ninhydrin solution (1 in 1000), and heat for 3 minutes. A purple color
	develops.
	(2) Monosodium L-Aspartate responds to all tests for Sodium Salt in the
	Qualitative Tests.
Specific Rotation	$\left( \alpha \right) {}_{\mathrm{D}}^{20}$ : +18~+21°
pН	6.0 ~ 7.5 (1.0 g, water 20 mL)
Purity	
Clarity and color	Colorless and clear (1.0 g, water 10 mL)
of solution	
Chloride	Not more than 0.041% as Cl
Arsenic	Not more than 2 mg/kg as As <sub>2</sub> O <sub>3</sub>
Lead	Not more than 2 mg/kg
Loss on Drying	Not more than 0.3%
Category	Food additives category 11
Functional uses	Seasoning Agents

### Fumaric Acid

Chemical formula: C <sub>4</sub> H <sub>4</sub> O		O <sub>4</sub> Molecular weight: 116.08
1.	Assay	: Not less than 99%
2.	Appearance	: Odorless, white crystalline powder or granules, having a characteristic sour taste.
3.	Melting range	: 287 $\sim$ 302°C (closed capillary, rapid heating after drying at 105°C for 3 hours)
4.	Solution	: When 0.5 g of the sample is dissolved in 10 mL of sodium hydroxide TS, the
		solution should be clear.
5.	Sulfate	: Not more than 0.01% of SO <sub>4</sub> .
6.	Heavy metals	: Not more than 20 ppm (as Pb).
7.	Residue on	: Not more than 0.05%.
	ignition	
8.	Category	: Food Additives Category 11
9.	Uses	: Seasoning Agents.

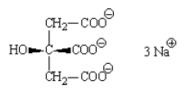
Sodium Fumarate		
Synonyms	Monosodium fumarate; INS No. 365	
Definition		
Chemical names	Monosodium fumarate, monosodium trans-butenedioic acid, monosodium trans-	
	1,2-ethylenedicarboxylate; monosodium trans-1,2-ethylenedicarboxylic acid	
C.A.S. number	7704-73-6	
Chemical formula	C <sub>4</sub> H <sub>3</sub> NaO <sub>4</sub>	
Structural formula	H COO <sup>©</sup> I Na <sup>⊕</sup> HOOC H	
Formula weight	138.06	
Assay	Not less than 98.0% and not more than 102.0% on the dried basis	
Description	Odourless, white crystalline powder	
Characteristics		
Identification		
Solubility	Soluble in water	
рН	3 - 4 (1 in 30 solution)	
1,2-Dicarboxylic acid	Place 50 mg of the sample in a test tube, add 2 to 3 mg of resorcinol and 1 ml of	
	sulfuric acid, shake, heat at 130° for 5 min. and cool. Dilute with water to 5 ml and	
	add sodium hydroxide solution (2 in 5) dropwise to render the solution alkaline,	
	cool and dilute with water to 10 ml. A greenish blue fluorescence is observed under	
	an ultraviolet lamp.	
Test for double bond	Add 10 ml of water to 0.5 g of the sample and dissolve by boiling. Add 2 or 3 drops	
	of bromine TS to the hot solution. The colour of bromine TS disappears.	
Test for sodium	Passes test	
Purity		
Loss on drying	Not more than 0.5% (120°, 4 h) 390	

Sulfates	Not more than 0.01%
Maleic acid	Not more than 0.05%
Lead	Not more than 2 mg/kg
Category	Food additives category (11)
Functional uses	Seasoning Agents

## Citric Acid; Citric Acid, Anhydrous

Cher	nical names : 2-h	ydroxy-1,2,3-propanetricarboxylic acid		
Synonyms : INS		No. 330; CAS No. 77-92-9 (anhydrous); CAS No. 5949-29-1 (monohydrate).		
Cher	nical : C61	H807 (anhydrous)Molecular weight: 192.13		
form	ula C	210.14 210.14		
	CH	2-COOH CH2-COOH		
	HO <b>≖</b> -Ç-	-СООН НОССООН Н <sub>2</sub> О		
	I CH	$\begin{array}{c} c_{2}-cooh & c_{H_{2}}-cooh \\ \hline -cooh & ho - c_{}cooh & h_{2}o \\ \hline c_{2}-cooh & c_{H_{2}}-cooh \end{array}$		
	Anhydrous	Monohydrate		
1.	Assay	: Not less than 99.5% and not more than 100.5% on the anhydrous basis.		
2.	Appearance	: White or colorless, odorless, crystalline solid. The monohydrate form effloresces		
		in dry air.		
3.	Solubility	: Very soluble in water; freely soluble in ethanol; slightly soluble in ether.		
4.	Water	: Anhydrous: Not more than 0.5%.		
		Monohydrate: Not less than 7.5% and not more than 8.8%.		
5.	Sulfated ash	: Not more than 0.05%.		
6.	Oxalate	: Not more than 100 mg/kg.		
7.	Sulfate	: Not more than 150 mg/kg.		
8.	Readily	: Heat 1.0 g of sample with 10 mL of 98% sulfuric acid in a water bath at $90\pm1^{\circ}C$		
	carbonizable	for 60 min. No color darker than Matching Fluid K (25°C) should be produced (not		
	substance	more than 0.5 absorbance units at 470 nm in a 10 mm cell).		
9.	Lead	: Not more than 0.5 mg/kg.		
10.	Heavy metals	: Not more than 5 mg/kg (as Pb).		
11.	Category	: Food Additives Category 11		
12.	Uses	: Seasoning Agents.		

#### Sodium Citrate



Synonyms: Trisodium citrate; INS No.331(iii)

Chemical names: trisodium salt of 2-hydroxy-1,2,3- propanetricarboxylic acid, trisodium salt of βhydroxy-tricarballylic acid

Molecular weight: 258.07 (Anhydrous)

Chemical formula:

Anhydrous: C<sub>6</sub>H<sub>5</sub>Na<sub>3</sub>O<sub>7</sub>

### Hydrated: C<sub>6</sub>H<sub>5</sub>Na<sub>3</sub>O<sub>7</sub>·nH<sub>2</sub>O (n=2 or 5)

1.	Assay	: Not less than 99.0% calculated on the dried basis (180°C, 2 h)
2.	Appearance	: Colorless crystal or white crystalline powder, odorless.
3.	Description	: 1. Soluble in water, insoluble in ethanol.
		2. Passes test for citrate and test for sodium.
4.	Loss on drying	: Anhydrous: not more than 1% (180°C to constant weight) Dihydrate: not
		more than 13% (180°C to constant weight) Pentahemihydrate: not more
		than 30% (180°C to constant weight)
5.	Alkalinity	: A 1 in 20 solution of the sample is alkaline to litmus. After the addition of
		0.2 mL of 0.1 N sulfuric acid and 1 drop of phenolphthalein TS to 10 mL
		of the solution no pink color is produced.
6.	Oxalate	: To 10 mL of a 1 in 10 solution of the sample add 5 drops of dilute acetic
		acid TS and 2 mL of calcium chloride TS. No turbidity develops within 1
		h.
7.	Lead	: Not more than 2 mg/kg.
8.	Category	: Food Additives Category 11
9.	Uses	: Seasoning Agents.

### Succinic Acid

Chemical formula: C <sub>4</sub> H <sub>6</sub> C		D <sub>4</sub> Molecular weight: 118.09
1.	Assay	: Not less than 99%
2.	Appearance	: Colorless to white crystal or white crystalline powder, odorless, having a
		characteristic sour taste.
3.	Melting range	: 184~189°C.
4.	Heavy metals	: Not more than 20 ppm (as Pb).
5.	Readily	: Dissolve 1 g of the sample in 25 mL of water 25 mL of dilute sulphuric acid, and
	oxidizable matter	maintain at 20°C. When 4 mL of 0.1 N potassium permanganate solution is
		added, the solution color should not disappear within 3 minutes.
6.	Residue on	: Not more than 0.05%.
	ignition	
7.	Category	: Food Additives Category 11
8.	Uses	: Seasoning Agents.

### Monosodium Succinate

Chemical formula: C <sub>4</sub> H <sub>5</sub> O <sub>4</sub> Na		D <sub>4</sub> Na Molecular weight: 140.08
1.	Assay	: Not less than 98 and not more than 102%.
2.	Appearance	: Colorless to white crystal or white crystalline powder, odorless, with a
		characteristic odor.
3.	pН	: 4.3~5.3 (1 in 20 soln)
4.	Chloride	: Not more than 0.1% of Cl.
5.	Sulfate	: Not more than 0.02% of SO <sub>4</sub> .
6.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
7.	Heavy metals	: Not more than 20 ppm (as Pb).
8.	Readily	: Dissolve 2 g of the sample in 25 mL of water 25 mL of dilute sulphuric acid and
	oxidizable matter	maintain at 20°C. When 4 mL of 0.1 N potassium permanganate solution is
		added, the solution color should not disappear within 3 minutes.
9.	Residue on	: 49.5~51.5%.
	ignition	
10.	Category	: Food Additives Category 11
11.	Uses	: Seasoning Agents.

### **Disodium Succinate**

Chemical formula: C4H4O4Na2·6H2O		D <sub>4</sub> Na <sub>2</sub> ·6H <sub>2</sub> O Molecular weight: 270.15
1.	Assay	: Not less than 98% and not more than 102% of $C_4H_4O_4Na_2(120^\circ\!C,2~h)$
2.	Appearance	: Colorless to white crystal or white crystalline powder, odorless, with a
		characteristic odor.
3.	pН	$:7 \sim 9 (1 \text{ in } 10 \text{ soln})$
4.	Sulfate	: Not more than 0.02% of $SO_{4}$
5.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
6.	Heavy metals	: Not more than 20 ppm (as Pb).
7.	Readily	: Dissolve 2 g of the sample in 20 mL of water 30 mL of dilute sulphuric acid and
	oxidizable matter	maintain at 20°C. When 4 mL of 0.1 N potassium permanganate solution is
		added, the solution color should not disappear within 3 minutes.
8.	Loss on drying	: Not more than 41% (120°C, 2 h)
9.	Residue on	: 51.5~53.5%.
	ignition	
10.	Category	: Food Additives Category 11
11.	Uses	: Seasoning Agents.

### L-Glutamic Acid

Chemical formula: C5H9O4N		D4N Molecular weight: 147.14
1.	Assay	: Not less than 99.0% on the dried basis (105°C, 3 h)
2.	Appearance	: Colorless or white crystals or crystalline powder, with a characteristic sour odor.
3.	Solution	: When 1 g of the sample is dissolved by heating in 50 mL of water, the solution
		should be clear.
4.	pН	$: 3.0 \sim 3.5$ (saturated solution)
5.	Specific rotation	: $(\alpha)_{D}^{20} = +30.7 \sim +32.7^{\circ} (105^{\circ}C, 3 \text{ h, then } 10\%(\text{w/v}) \text{ soln in 2N hydrochloric})$
		acid)
6.	Chloride	: Not more than 0.2% of Cl.
7.	Ammonium salt	: Not more than 0.02% of NH <sub>4</sub> .
8.	Arsenic	: Not more than 3 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Other amino	: Test for other amino acid of "L-aspartate" is applicable.
	acids	
11.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 0.2% of its weight
12.	Residue on	: Not more than 0.2%.
	ignition	
13.	Category	: Food Additives Category 11
14.	Uses	: Seasoning Agents.

#### Monosodium L-Glutamate

<b>S</b>	
Synonyms	Sodium glutamate, MSG, INS No. 621
Definition	
Chemical names	Monosodium L-glutamate monohydrate, glutamic acid monosodium salt
	monohydrate
C.A.S. number	142-47-2
Chemical formula	$C_5H_8NNaO_4 \cdot H_2O$
Structural formula	
	HO $O$ $O$ $Na^+$ $H_2O$ $NH_2$
Formula weight	187.13
Assay	Not less than 99.0% on the dried basis
Description	White, practically odourless crystals or crystalline powder
Characteristics	
Identification	
Solubility	Freely soluble in water; sparingly soluble in ethanol; practically insoluble in ether
Test for glutamate	Passes test
Test for sodium	Passes test
Purity	
Loss on drying	Not more than 0.5% (98°C, 5 h)
pH	6.7 - 7.2 (1 in 20 soln)
Specific rotation	$[\alpha]_{D}^{20}$ =+24.8~+25.3° (10% (w/v) solution in 2N hydrochloric acid)
Chlorides	Not more than 0.2%
Pyrrolidone carboxylic acid	l Passes test
Lead	Not more than 1 mg/kg
Category	Food additives category (11)

**Functional uses** 

Seasoning agents.

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### dl-Tartaric Acid

Chemical formula: C <sub>4</sub> H <sub>6</sub> O <sub>6</sub>		D <sub>6</sub> Molecular weight: 150.09
1.	Assay	: Not less than 99.7% on the dried basis (over sulfuric acid, 3 h)
2.	Appearance	: Colorless crystal or white crystalline powder, odorless, having a smelly sour
		taste.
3.	Melting range	: 200~206°C.
4.	Sulfate	: Not more than $0.05\%$ of SO <sub>4</sub> .
5.	Heavy metals	: Not more than 10 ppm (as Pb).
6.	Calcium	: The test for calcium of "d-tartaric acid" is applicable.
7.	Readily	: Dissolve 1 g of the sample in 25 mL of water 25 mL of dilute sulphuric acid.
	oxidizable matter	When 4 mL of 0.1 N potassium permanganate solution is added at 20°C, the
		solution color should not disappear within 3 minutes.
8.	Loss on drying	: Not more than 0.5% (over sulfuric acid, 3 hr).
9.	Residue on	: Not more than 0.1%.
	ignition	
10.	Category	: Food Additives Category 11
11.	Uses	: Seasoning Agents.

### d-Tartaric Acid

Chemical formula: C <sub>4</sub> H <sub>6</sub> O <sub>6</sub>		Molecular weight: 150.09
1.	Assay	: Not less than 99.7% on the dried basis (over sulfuric acid, 3 h).
2.	Appearance	: Colorless or translucent crystals, or white, fine to granular, crystalline powder;
		odorless; having a sour taste.
3.	Specific rotation	: ( $\alpha$ ) $_{D}^{20}$ =+11.5~+13.5° (over sulfuric acid, 3 h. Then dissolve 10 g of the
		product to form a 50 mL solution)
4.	Sulfate	: Not more than $0.05\%$ of SO <sub>4</sub> .
5.	Oxalate	: Nearly neutralize 10 mL of a 1 in 10 solution of the sample with ammonia TS,
		and add 10 mL of calcium sulfate TS. No turbidity is produced.
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Calcium	: Dissolve 1 g of the sample in 10 mL of water, neutralize with ammonia TS.
		Then, add 1 mL of ammonium oxalate TS. The solution should not be turbid.
8.	Loss on drying	: Not more than 0.5% (over sulfuric acid, 3 hr).
9.	Residue on	: Not more than 0.1%.
	ignition	
10.	Category	: Food Additives Category 11
11.	Uses	: Seasoning Agents.

### Sodium *dl*-Tartrate

Chen	nical formula: C <sub>4</sub> H <sub>4</sub> C	D <sub>6</sub> Na <sub>2</sub> Molecular weight: 194.06
1.	Assay	: Not less than 98.5% on the dried basis (105°C, 4 h)
2.	Appearance	: Colorless crystal or white crystalline powder.
3.	Solution	: Dissolve 1 g of the sample in 20 mL of water, the solution should be less than
		"almost clear".
4.	Free Acid and free	: Test method of "5.Free Acid and free alkali" for "d-sodium tartrate" is
	alkali	applicable.
5.	Sulfate	: Not more than $0.02\%$ of SO <sub>4.</sub>
6.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Readily	: Dissolve 2 g of the sample in 20 mL of water 30 mL of dilute sulphuric acid.
	oxidizable matter	When 4 mL of 0.1 N potassium permanganate solution is added at 20°C, the
		solution color should not disappear within 3 minutes.
9.	Loss on drying	: Dry it at 105°C for 4 hours: it loses not more than 0.5% of its weight
10.	Category	: Food Additives Category 11
11.	Uses	: Seasoning Agents.

#### Sodium *d*-Tartrate

Chen	nical formula: C4H4O	Molecular weight: 230.09
1.	Assay	: Not less than 98.5% on the dried basis (105°C,4 h).
2.	Appearance	: Colorless crystal or white crystalline powder.
3.	Solution	: Dissolve 1 g of the sample in 20 mL of water, the solution should be less than
		"almost clear".
4.	Specific rotation	: $(\alpha)_{D}^{20} = +25.0 \sim +27.5^{\circ}$ (Dissolve 5 g of the sample in water to form a 50 mL
		solution)
5.	Free Acid and free	: Dissolve 2 g of the sample in 20 mL of freshly boiled and cooled water.
	alkali	Add 2 drops of phenolphthalein TS. If a colorless solution is produced, not
		more than 0.1 mL of 0.1 N sodium hydroxide should be required to give a
		red color. If a red color is produced, not more than 0.1 mL of 0.1 N
		hydrochloric acid should be required to discharge it.
6.	Sulfate	: Not more than 0.02% of $SO_{4}$ .
7.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Loss on drying	: 14~16% (150°C, 4 hr).
10.	Category	: Food Additives Category 11
11.	Uses	: Seasoning Agents.

		Lactic Acid
Synony	ms	INS No. 270
Definiti	on	Obtained by the lactic fermentation of sugars or is prepared
		synthetically; may contain condensation products such as lactic acid,
		lactate and dilative.
		Common products of commerce are 50-90% solutions. Solid products
		containing about 100-125% of treatable lactic acid also exist.
		(Note: Lactic acid is hygroscopic and when concentrated by boiling or
		by distillation it forms condensation products which hydrolyze to lactic
		acid on dilution and heating in water).
Chemic	al names	Lactic acid,
		2-hydroxypropanoic acid,
		2-hydroxypropionic acid
CAS NU	JMBER	50-21-5
		(L-: 79-33-4;
		D-: 10326-41-7;
		DL-: 598-82-3)
Chemic	al formula	$C_3H_6O_3$
Molecul	ar weight	90.08
Assay		Not less than 95.0% and not more than 105.0% of the labelled
		concentration. (as C <sub>3</sub> H <sub>6</sub> O <sub>3</sub> )
Appear	ance	Colorless, syrupy liquid or white to light yellow solid or powder.
Charac	teristics	
	Identification	
	Solubility	Liquid: Soluble in water and in ethanol.
		Solid: Sparingly soluble in water, soluble in acetone.
	Test for acid	A 1 in 10 solution or dispersion of the sample is acid to litmus paper.
	Test for lactate	Passes test.
	Purity	
	Sulfated ash	Not more than 0.1%
	Chloride	Not more than 0.2% of Cl.
	Sulfate	Not more than $0.25\%$ of SO <sub>4</sub> .

	Iron	Not more than 10 mg/kg.
	Cyanide	Passes test (limit approx. 1 mg/kg)
	Citric, oxalic,	Dilute 1 g of the sample to 10 mL with water, add 40 mL of calcium
	phosphoric or tartaric	hydroxide TS, and boil for 2 min. No turbidity is produced.
	acid	
	Sugars	Add 5 drops of the 40% sample solution (Concentration of lactic acid is
		calculated from the indicated concentration. No dilution required if the
		indicated concentration is less than 40%) to 10 mL of hot alkaline cupric
		tartrate TS. No red precipitate is formed.
	Readily carbonizable	Superimpose carefully 5 mL of the sample solution kept at $15^{\circ}$ C on 5
	substance	mL of sulfuric acid TS kept at 15°C. No deep grey color is produced
		within 15 min at the contact zone of the two liquids.
	Lead	Not more than 2 mg/kg.
Category		Food Additives Category 11
Uses		Seasoning Agents.

### Sodium Lactate (solution)

SynonymsINS No. 325DefinitionSodium lactate, sodium 2-hydroxypropanoateChemical namesSodium lactate, sodium 2-hydroxypropanoateC.A.S. number72-17-3Chemical formula $C_3H_5NaO_3$ Structural formula $H_3C$ H\_3CONa

ÓН

Formula weight	112.06 (anhydrous)
Assay	Not less than 95% and not more than 110% of the labelled amount. This
	specification is based on a 60% w/w solution of $C_3H_5NaO_3$ in water.
Description	Colourless, transparent, liquid. Odourless, or with a slight, characteristic odour
Characteristics	
Identification	
Ignition	Ignite to an ash. The ash is alkaline, and an effervescence occurs when acid is
	added
Colour reaction	Overlay 2 ml of the sample on 5 ml of a 1 in 100 solution of catechol in sulfuric
	acid. A deep red colour is produced at the zone of contact
Test for lactate	Passes test
Test for sodium	Passes test
Purity	
Acidity	Neutralization of 1 g of sample shall require not more than 0.5 ml of 0.1 N sodium
	hydroxide solution, using phenolphthalein TS
pН	6.5 - 7.5 (1 in 5 soln)

Lead Not more than 2 mg/kg

**Category** Food additives category (11)

Functional uses Seasoning Agents

#### Sodium Lactate Solution

Chemical formula : C <sub>3</sub> H <sub>5</sub> O <sub>3</sub> Na		D <sub>3</sub> Na Formula weight : 112.06
1.	Assay	: 50~60% (as C <sub>6</sub> H <sub>5</sub> O <sub>3</sub> Na)
2.	Appearance	: Colorless, transparent, liquid. Odorless, or with a slight, characteristic odor.
3.	Density	: 1.26~1.32.
4.	pН	$: 6.5 \sim 7.5 (1 \text{ in } 5 \text{ soln})$
5.	Chloride	: Not more than 70 ppm of Cl.
6.	Sulfate	: Not more than 0.012% SO <sub>4</sub> .
7.	Malate, tartrate and	citrate : Add 1 mL of the sample to form a 50 mL solution.
		When adding 3 to 5 drops of lead acetate solution to
		5 mL of the solution, the turbidity should be below
		"slightly turbidity".
8.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Iron	: Not more than 10 ppm.
11.	Volatile fatty acids	: When 5 mL of the sample is heated with 2 mL of dilute sulfuric acid on
		water bath, strong odor of fatty acid does not occur.
12.	Methanol	: Not more than 0.2%.
13.	Category	: Food Additives Category 11
14.	Uses	: Seasoning Agents.

#### Acetic Acid

1.	Assay	: This product contains $29 \sim 31\%$ of $C_2H_4O_2$ .
2.	Appearance	: Colorless liquid, with a stimulating odor and sour taste.
3.	Chloride	: Not more than 1.46 ppm of Cl.
4.	Sulfate	: Not more than 6.5 ppm of SO <sub>4</sub> .
5.	Heavy metals	: Not more than 10 ppm (as Pb).
6.	Readily	: Add 0.3 mL of 0.1 N potassium permanganate solution to 20 mL of the sample,
	oxidizable matter	the color of the solution should not disappear within 30 minutes.
7.	Non-volatile	: Not more than 0.01%.
	residue	
8.	Formaldehyde	: Add 0.1 g of chromo tropic acid and 5 mL of sulfuric acid to 1 mL of the sample.
		After heated in water bath for 30 minutes, cool it down. Add cold water to make
		a 250 mL solution, the liquid color should not be thicker than that of the 1 mL of
		formaldehyde standard solution that is treated with the same process. (1
		mL=HCHO 0.01 mg)
9.	Category	: Food Additives Category 11
10.	Uses	: Seasoning Agents.

#### Acetic Acid, Glacial

Synonyms	INS No. 260
Definition	Acetic acid is manufactured by aerial oxidation of C5-C6 fractions of aliphatic
	hydrocarbons, and separation of the various acids by distillation. Also by oxidation of
	acetaldehyde, methanol and of butane or as the reaction product of methanol and
	carbon dioxide.
Chemical names	Acetic acid, ethanoic acid
C.A.S. number	64-19-7
Chemical formula	CH₃COOH
Structural formula	H <sub>3</sub> C OH
Formula weight	60.05
Assay	Not less than 99.5%
Description	Colourless liquid, having a pungent characteristic odour
Characteristics	
Characteristics	
Identification	
	Miscible with water, ethanol, glycerol and diethyl ether
Identification	Miscible with water, ethanol, glycerol and diethyl ether 1 in 3 aqueous solution is acidic
Identification Solubility	
Identification Solubility Test for acid	1 in 3 aqueous solution is acidic
Identification Solubility Test for acid Test for acetate	1 in 3 aqueous solution is acidic
Identification Solubility Test for acid Test for acetate Purity	1 in 3 aqueous solution is acidic Passes test
Identification Solubility Test for acid Test for acetate Purity Solidification point	1 in 3 aqueous solution is acidic Passes test Not lower than 15.6°C
Identification Solubility Test for acid Test for acetate Purity Solidification point	1 in 3 aqueous solution is acidic Passes test Not lower than 15.6°C Not more than 0.01% after evaporation of 20 g of the sample and holding at 100°C for
Identification Solubility Test for acid Test for acetate Purity Solidification point Non-volatile residue	1 in 3 aqueous solution is acidic Passes test Not lower than 15.6°C Not more than 0.01% after evaporation of 20 g of the sample and holding at 100°C for 2 h.
Identification Solubility Test for acid Test for acetate Purity Solidification point Non-volatile residue Readily oxidizable	<ul> <li>1 in 3 aqueous solution is acidic</li> <li>Passes test</li> <li>Not lower than 15.6°C</li> <li>Not more than 0.01% after evaporation of 20 g of the sample and holding at 100°C for 2 h.</li> <li>Dilute 2 ml of the sample in a glass-stoppered container with 10 ml of water and add</li> </ul>
Identification Solubility Test for acid Test for acetate Purity Solidification point Non-volatile residue Readily oxidizable	<ul> <li>1 in 3 aqueous solution is acidic</li> <li>Passes test</li> <li>Not lower than 15.6°C</li> <li>Not more than 0.01% after evaporation of 20 g of the sample and holding at 100°C for 2 h.</li> <li>Dilute 2 ml of the sample in a glass-stoppered container with 10 ml of water and add 0.1 ml of 0.1 N potassium permanganate. The pink color does not disappear within 30</li> </ul>
Identification Solubility Test for acid Test for acetate Purity Solidification point Non-volatile residue Readily oxidizable substances	<ul> <li>1 in 3 aqueous solution is acidic</li> <li>Passes test</li> <li>Not lower than 15.6°C</li> <li>Not more than 0.01% after evaporation of 20 g of the sample and holding at 100°C for 2 h.</li> <li>Dilute 2 ml of the sample in a glass-stoppered container with 10 ml of water and add 0.1 ml of 0.1 N potassium permanganate. The pink color does not disappear within 30 min.</li> </ul>

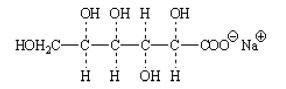
		DL-Malic Acid
Synonyms		2-Hydroxybutanedioic acid; INS No. 296
Definition		
Chemical nan	nes	dl-Malic acid, 2-Hydroxybutanedioic acid, Hydroxysuccinic acid
CAS NUMBE	ER	6915-15-7
Chemical form	mula	$C_4H_6O_5$
Molecular we	right	134.1
Assay		Not less than 99.0%
Appearance		White or off-white crystalline powder or particle.
Characterist	ics	
Ider	ntification	
Solu	ıbility	Very soluble in water, soluble in ethanol.
Mel	ting range	127 - 132°C
		Passes test.
Test	t for malate	Test 5 mL of a 1 in 20 solution of the sample, neutralized with
		ammonia TS.
Pur	ity	
Fum	naric and maleic acid	Not more than 1.0% of fumaric acid and not more than 0.05% of
		maleic acid.
Lead	d	Not more than 2 mg/kg.
Category		Food Additives Category 11
Uses		Seasoning Agents.

	Sodium DL-Malate
Synonyms	Malic acid sodium salt; INS No. 350(ii)
Definition	
Chemical names	Disodium DL-malate, hydroxybutanedioic acid disodium salt
CAS NUMBER	676-46-0
Chemical formula	Hemihydrate: C <sub>4</sub> H <sub>4</sub> Na <sub>2</sub> O <sub>5</sub> · 1/2 H <sub>2</sub> O
	Trihydrate: C <sub>4</sub> H <sub>4</sub> Na <sub>2</sub> O <sub>5</sub> · 3 H <sub>2</sub> O
Molecular weight	Hemihydrate: 187.1
	Trihydrate: 232.1
Assay	Not less than 98% and not more than 102% on the dried basis.
Appearance	Odorless white crystalline powder or lumps.
Characteristics	
Identification	
Solubility	Soluble in water
Positive test for	Passes test
sodium	
Test for malate	Passes test (Test a 1 in 20 solution)
Purity	
Loss on drying	Hemihydrate: Not more than 7% (130°C, 4 hr)
	Trihydrate: 20.5% - 23.5% (130°C, 4 hr)
Alkalinity	Not more than $0.2\%$ as Na <sub>2</sub> CO <sub>3</sub> .
	Dissolve 1 g of the sample in 20 mL of freshly boiled and cooled water, and
	add 2 drops of phenolphthalein TS. If a pink color is produced, add 0.4 mL
	of 0.1 N sulfuric acid. The color of the solution disappears.
Fumaric and maleic	Not more than $1.0\%$ of fumaric acid and not more than $0.05\%$ of maleic
acid	acid.
Lead	Not more than 2 mg/kg
Category	Food Additives Category 11
Uses	Seasoning Agents.

Sodium Gluconate

Chemical formula: C<sub>6</sub>H<sub>11</sub>NaO<sub>7</sub>

Molecular weight: 218.14



- 1. Assay : Not less than 98%.
- 2. Appearance : White to tan, granular to fine, crystalline powder.
- 3. Solubility : Very soluble in water, slightly soluble in ethanol.
- 4. Reducing : Not more than 1.0% calculated as D-glucose.
  - substances
- 5. Lead : Not more than 2 mg/kg.
- 6. Category : Food Additives Category 11
- 7. Uses : Seasoning Agents.

#### $Glucono-\delta-Lactone$

Chemical formula:	$C_6H_{10}O_6$
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Molecular weight: 178.15

1.	Assay	: Not less than 99.0% on the dried basis (105°C, 2 h)
2.	Appearance	: White, odorless or nearly odorless crystals or crystalline powder. The taste is
		sweet at first and then becomes sour.
3.	Solution	: Dissolve 1 g of the sample in 10 mL of water, the solution should be less than
		"almost clear" and colorless.
4.	Chloride	: Not more than 0.035% of Cl.
5.	Sulfate	: Not more than $0.024\%$ of SO <sub>4</sub> .
6.	Arsenic	: Not more than 1 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Calcium	: Dissolve 1 g of the sample in 10 mL of water, neutralize with ammonia TS and
		add 1 mL of ammonium oxalate TS. No precipitate occurs.
9.	Iron	: Dissolve 0.5 g of the sample in 5 mL of water, mix with 1 drop of hydrochloric
		acid. Then add 3 drops of potassium ferrocyanide TS. The solution should
		not appear blue immediately.
10.	Sucrose or reducin	g :Boil 0.5 g of the sample, 10 mL of water and 2 mL of dilute hydrochloric acid
	sugar	for 2 minutes. Add 5 mL of sodium carbonate test solution after cooling.
		After standing for 5 minutes, add water to make the solution 20 mL. Add 2
		mL of Fehling's reagent to 5 mL of the sample and boil for 1 minute.
		Orange-red precipitate does not form immediately.
11.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 1% of its weight.
12.	Residue on	: Not more than 0.1%.
	ignition	
13.	Category	: Food Additives Category 11
14.	Uses	: Seasoning Agents.

# Glycine

Synonyms	Aminoacetic acid, Aminoethanoic acid, Glycocoll
Definition	
Chemical names	Glycine
CAS NUMBER	56-40-6
Molecular formula	C <sub>2</sub> H <sub>5</sub> O <sub>2</sub> N
Molecular weight	75.07
Assay	Not less than 98.5% on the dried basis.
Appearance	White crystal or crystalline powder.
Characteristics Identification Infra-red absorption spectrum analysis	The maximum absorption wavelength of the sample should be the same as the standard product.
Purity	
Residue on ignition Loss on drying Arsenic	Not more than 0.1%. Dry it at 105°C for 3 hours: it loses not more than 0.2% of its weight. Not more than 3 mg/kg.
Lead	Not more than 5 mg/kg.
Category	Food Additives Category 11
Uses	Seasoning Agents.

DL-Alanine

Chemical formula: C <sub>3</sub> H <sub>7</sub> C		D <sub>2</sub> N Molecular weight: 89.10
1.	Assay	: Not less than 98.5% on the dried basis (105°C, 3 h)
2.	Appearance	: Colorless to white crystalline powder, with a sweet taste.
3.	Solution and pH	: When 1 g of the sample is dissolved in 10 mL of water, the solution should be
		clear and the pH is $5.5 - 7.0$ .
4.	Chloride	: Not more than 0.02% of Cl.
5.	Ammonium salt	: Not more than 0.03% of $NH_4$ .
6.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
7.	Heavy metals	: Not more than 20 ppm (as Pb).
8.	Other amino	: Test for other amino acid of "L-aspartate" is applicable.
	acids	
9.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 0.3% of its weight
10.	Residue on	: Not more than 0.2%.
	ignition	
11.	Category	: Food Additives Category 11
12.	Uses	: Seasoning Agents.

#### Sodium 5'- Inosinate

Chemical formula: C10H11O8N4PNa2

Molecular weight: 392.19

1.	Assay	: 97~102%.
2.	Appearance	: Colorless to white crystal or white crystalline powder, having a characteristic
		odor.
3.	Solution	: Dissolve 0.5 g of the sample in 10 mL of water, the solution should be less than
		"almost clear" and colorless.
4.	pН	: 7.0~8.5 (1 in 20 soln)
5.	Ammonium salt	: Not more than 0.02% of NH <sub>4</sub> .
6.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 20 ppm (as Pb).
8.	Amino acid	: Not detectable.
9.	Absorbance ratio	: The ratio A250/A260 is between 1.55 and 1.65, and the ratio A280/260 is
		between 0.20 and 0.30 in a 1 in 50,000 solution of the sample in 0.01 N
		hydrochloric acid.
10.	Other nucleotifes	: Dissolve 1 g of the sample in water to make a 100 mL solution. Take 0.01
		mL of the solution to perform filter paper chromatography. When the
		developing solution drops about 30 cm from the origin, stop unfolding.
		Air-dry the filter paper, observe the back of the filter paper with
		ultraviolet light at a wavelength of about 250 nm in the dark. Only one
		spot should exist.
	Suspension	: A 160:3:40 mixture of saturated ammonium sulfate solution, tert-butyl alcohol,
		and 0.025 N ammonium, respectively.
	Filter paper	: Chromatography filter paper No.2
11.	Water	: Not more than 28.5% (Karl Fischer Method)
12.	Category	: Food Additives Category 11
13.	Uses	: Seasoning Agents.

Sodium 5'- Guanylate

Cher	nical formula: C <sub>10</sub> H <sub>12</sub> O <sub>8</sub> N	5PNa <sub>2</sub> Molecular weight: 407.20
1.	Assay	: Not less than 97.0% and not more than 102.0% calculated on the dried basis.
		(120°C, 4 h)
2.	Appearance	: Colorless to white crystal or white crystalline powder, having a
		characteristic odor.
3.	Solution	: Dissolve 0.1 g of the sample in 10 mL of water, the solution should be less
		than "almost clear" and colorless.
4.	pН	$: 7.0 \sim 8.5 (1 \text{ in } 20 \text{ soln})$
5.	Ammonium salt	: Not more than 0.02% of NH <sub>4</sub> .
6.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 20 ppm (as Pb).
8.	Amino acid	: Not detectable.
9.	Spectrophotometry	: A 1 in 50,000 solution of the sample in 0.01 N hydrochloric acid exhibits an
		absorbance maximum at 256±2nm. The ratio A250/A260 is between 0.95 and
		1.03, and the ratio A280/260 is between 0.36 and 0.71.
10.	Other nucleic acid resol	lvent : Test method for other nucleic acid resolvent of "5'-hypoxanthine
		phosphate disodium" is applicable.
11.	Loss on drying	: Dry it at $120^{\circ}$ for 4 hours: it loses not more than $25\%$ of its weight.
12.	Category	: Food Additives Category 11
13.	Uses	: Seasoning Agents.

# Phosphoric Acid

Chen	nical formula: H <sub>3</sub> PO <sub>4</sub>	Molecular weight: 98.00
1.	Assay	: Not less than 85%.
2.	Appearance	: Clear, colorless, odorless, viscous liquid.
3.	Density	: Not less than 1.69.
4.	Solution	: Dissolve 4 mL of the sample in 6 mL of ethanol. The solution should be less than
		"almost clear" and colorless.
5.	Sulfate	: Not more than 0.03% of SO <sub>4</sub> .
6.	Arsenic	: Not more than 3 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Readily	: Dissolve 7 g of the sample in 5 mL of water, add0.2 mL of 0.1 N potassium
	oxidizable matter	permanganate solution. The red color does not disappear within 10 minutes
		when heated on water bath.
9.	Category	: Food Additives Category 11
10.	Uses	: Seasoning Agents.

Potassium Chloride

The same as § 08063

#### Calcium 5'-Ribonucleotide

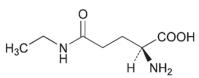
1.	Description	: A mixture of calcium inosine-5'-monophosphate, calcium guanosine-5'-
		monophosphate, calcium cytidine-5'-monophosphate and calcium uridine-5'-
		monophosphate, or a mixture of calcium inosine-5'-monophosphate and
		calcium guanosine-5'-monophosphate. Odorless, white or off-white crystals or
		powder, having a characteristic taste.
2.	Assay	: The anhydrous form of the sample contains not less than 97% and not more than
		102% of calcium 5'-ribonucleotide. Calcium inosine-5'-monophosphate and
		calcium guanosine-5'-monophosphateaccounts for more than 95%.
3.	рН	: Add 200 mL of water to 0.1 g of the sample and heat in a water bath until it
		dissolves. After cooling, the pH of the solution is 7.0-8.0 (1 in 2,000 soln).
4.	Arsenic	: Not more than 4 ppm (as As <sub>2</sub> O <sub>3</sub> ).
5.	Heavy metals	: Not more than 20 ppm (as Pb).
6.	Water soluble	: Not more than 8%.
	substances	
7.	Water	: Not more than 23%.
8.	Category	: Food Additives Category 11
9.	Uses	: Seasoning Agents.

### Caffeine

Chem	nical formula: C <sub>8</sub> H <sub>10</sub>	N <sub>4</sub> O <sub>2</sub> Molecular weight: 194.19
1.	Assay	: Not less than 98.5% and not more than 101.0% on the dried basis
2.	Appearance	: The sample is anhydrous or monohydrated white powder or white flash needle,
		odorless and bitter. The solution is neutral on litmus paper. The hydrated form
		can be weathered in the air.
3.	Description	: 1 g of the hydrated sample is soluble in about 50 mL of water, 75 mL of alcohol,
		60 mL of chloroform and 600 mL of ether.
4.	Identification	: (1) Take 5 mg of the sample in a magnetic dish and dissolve it in 1 mL of
		hydrochloric acid. Then add 50 mg of potassium chlorate and evaporate to
		dryness in a steam bath. Place the magnetic dish on a vessel containing a
		few drops of ammonia TS. The residue is purple. Add lye, and the purple
		color disappears.
		(2) Add tannic acid TS to the saturated solution of the sample, and the precipitate
		forms. However, the precipitate disappears after adding an excess of tannic
		acid TS.
		(3) Add 5 drops of iodine TS to 5 mL of the saturated solution of the sample. No
		precipitation should occur. However, a reddish brown precipitate can be
		produced by adding 3 drops of the diluted hydrochloric acid TS. If a little
		excess of sodium hydroxide TS is added, the precipitate dissolves.
5.	Melting range	: 235~237.5°C.
6.	Other alkali	: Add a few drops of 5 mL of potassium iodide TS that contains mercury to 5 mL
		of aqueous solution of the sample (1 g of the sample in 50 mL of water). No
		precipitation forms.
7.	Arsenic	: Not more than 3 ppm (as As).
8.	Lead	: Not more than 10 ppm.
9.	Heavy metals	: Not more than 20 ppm (as Pb).
10.	Readily	: Dissolve 500 gm of the sample in 5 mL of sulfuric acid TS. The color produced
	carbonizable	should not be darker than the Matching Fluid D.
	substance	
11.	Water	: Anhydrous: not more than 0.5%. Hydrated: not more than 8.5%.

- 12. Residue on : Not more than 0.1%. ignition
- 13. Category : Food Additives Category 11
- 14. Uses : Seasoning Agents.

L-Theanine



		H H NH <sub>2</sub>		
Chemical formula: C <sub>7</sub> H <sub>14</sub> N <sub>2</sub> O <sub>3</sub> Molecular weight: 174.2				
Chemical names: (2S)-2-Amino-4-(N-ethylcarbamoyl)butanoic acid				
C.A.S. number: 3081-61-6				
1.	Assay	: Not less than 98.0% and not more than 102.0% on the dried basis		
2.	Appearance	: White crystalline powder, odorless, with a slightly characteristic and sweet taste.		
3.	Identification	: (1) To 5 mL of a solution of L-Theanine (1 in 1000), add 1 mL of ninhydrin solution (1		
		in 1000), and heat for 3 minutes. A purple color develops.		
		(2) Dissolve about 1 g of L-Theanine in 10 mL of diluted hydrochloric acid (1 in 2),		
		put in a water bath that is equipped with a reflux cooler for 6 hours, and add water to		
		20 mL. Put 5 mL of the solution into a test tube and add 2 g of sodium hydroxide. Take		
		another piece of red litmus paper to cover the mouth of the test tube after wetting it		
		with water. After the tube is bathed for 5 minutes, the color of the litmus paper turns		
		blue.		
4.	Specific rotation	: $\left[\alpha\right]_{D}^{20} = +7.7 \sim +8.5$ (Add 50 mL of water to 2.5 g of the sample)		
5.	Solution state	: Dissolve 1 g of the sample in 20 mL of water. The solution should be colorless and		
		almost clear.		
6.	рН	: 5.0~6.0 (1 in 100 soln)		
7.	Chloride	: Not more than 0.021% (as Cl).		
8.	Lead	: Not more than 2 ppm.		
9.	Arsenic	: Not more than 3 ppm (as $As_2O_3$ ).		
10.	Loss on drying	: Dry it at 105°C for 3 hours: it loses not more than 0.5% of its weight.		
11.	Residue on	: Not more than 0.2%.		
	ignition			
12.	Category	: Food Additives Category (11)		
13	Uses	· Seasoning Agents		

13. Uses : Seasoning Agents.

# Sodium Dihydrogen Citrate

Synonyms	: Monosodium citrate; sodium citrate monobasic; INS No. 331(i)
Definition	
Chemical names	: Monosodium citrate; monosodium salt of 2-hydroxy-1,2,3-
	propanetricarboxylic acid
Molecular formula	$: C_6H_7NaO_7$
Structural formula	СНСООН
	$H_2 - COOH$ HO $-C - COO^{\Theta}$ Na $^{\oplus}$ $H_2 - COOH$
Molecular weight	: 214.11
Assay	: Not less than 99.0% and not more than 101.0%.
Appearance	: White, odorless crystals or crystalline powder.
Characteristics	
Identification	
Solubility	: Freely soluble in water; practically insoluble in ethanol.
pH	: 3.4-3.8 (1: 10 soln)
Test for citrate	: Passes test
Positive test for	: Passes test
sodium	
Purity	
Loss on drying	: Not more than 0.4% (105°C, 4 hr).
Oxalate	: To 10 mL of a 1 in 10 solution of the sample add 5 drops of dilute acetic acid
	TS and 2 mL of calcium chloride TS. No turbidity develops within 1 h.
Lead	: Not more than 2 mg/kg.
Category	: Food Additives Category 11
Uses	: Seasoning Agents.

# Category 11-1 Sweeteners

# § 11-1-001

§ 07087

### D-Sorbitol

	D Sololiol
Synonyms	INS No. 420(i); D-Glucitol, D-sorbitol, sorbit, sorbol
Definition	
Chemical names	D-Sorbitol
CAS NUMBER	50-70-4
Molecular formula	$C_6H_{14}O_6$
Molecular weight	182.17
Assay	Not less than 97.0% of $C_6H_{14}O_6$ of total glycitols and not less than
	91.0% of D-sorbitol on the anhydrous basis. The term glycitols refers
	to compounds with the structural formula CH2OH-(CHOH)n-CH2OH,
	where n is an integer less than or equal to 4.
Appearance	White hygroscopic powder, crystalline powder, flakes or granules.
Characteristics	
Identification	
Solubility	Very soluble in water, slightly soluble in ethanol.
Melting range	88~102°C
Thin layer	Passes test
chromatography	
Purity	
Water	Not more than 1% (Karl Fischer Method).
Sulfated ash	Not more than 0.1%.
Chloride	Not more than 50 mg/kg.
Sulfate	Not more than 100 mg/kg.
Nickel	Not more than 2 mg/kg.
Reducing sugar	Not more than 0.3%.
Total sugars	Not more than 1% (as glucose).
Lead	Not more than 1 mg/kg.
Category	Food Additives Category 11-1, 7.

Sweeteners, Food quality improvement, fermentation and food

processing agents

Uses

Xylitol

<u>Synonyms</u>	<u>INS No. 967</u>
<b>Definition</b>	
Chemical names	Xylitol
C.A.S. number	<u>87-99-0</u>
Chemical formula	C5H12O5
Structural formula	
	$\begin{array}{c} CH_{2}OH \\ H \longrightarrow C \longrightarrow OH \\ HO \longrightarrow C \longrightarrow H \\ H \longrightarrow C \longrightarrow OH \\ H \longrightarrow C \longrightarrow OH \\ CH_{2}OH \end{array}$
Formula weight	<u>152.15</u>
Assay	Not less than 98.5% and not more than 101.0% on the anhydrous basis
Description	White, crystalline powder, practically odourless
Description Characteristics	White, crystalline powder, practically odourless
-	White, crystalline powder, practically odourless
Characteristics	White, crystalline powder, practically odourless Very soluble in water, sparingly soluble in ethanol
Characteristics Identification	
Characteristics Identification Solubility	Very soluble in water, sparingly soluble in ethanol
Characteristics Identification Solubility Melting range	<u>Very soluble in water, sparingly soluble in ethanol</u> <u>92 - 96°C</u>
Characteristics Identification Solubility Melting range	Very soluble in water, sparingly soluble in ethanol 92 - 96°C The infrared spectrum of a potassium bromide dispersion of the sample corresponds
Characteristics Identification Solubility Melting range Infrared absorption	Very soluble in water, sparingly soluble in ethanol 92 - 96°C The infrared spectrum of a potassium bromide dispersion of the sample corresponds
Characteristics Identification Solubility Melting range Infrared absorption	Very soluble in water, sparingly soluble in ethanol 92 - 96°C The infrared spectrum of a potassium bromide dispersion of the sample corresponds with the reference infrared spectrum below
Characteristics Identification Solubility Melting range Infrared absorption Purity Water	Very soluble in water, sparingly soluble in ethanol 92 - 96°C The infrared spectrum of a potassium bromide dispersion of the sample corresponds with the reference infrared spectrum below Not more than 0.5% (Karl Fischer Method)
Characteristics Identification Solubility Melting range Infrared absorption Purity Water Sulfated ash	Very soluble in water, sparingly soluble in ethanol 92 - 96°C The infrared spectrum of a potassium bromide dispersion of the sample corresponds with the reference infrared spectrum below Not more than 0.5% (Karl Fischer Method) Not more than 0.1%

LeadNot more than 1 mg/kgCategoryFood additives category (7)(11-1)Functional usesFood quality improvement, fermentation and food processing agents; Sweeteners.

# § 11-1-004

# Glycyrrhizin

Assay	: Not less than 95% on the dried basis (80°C, 4 h)
Appearance	: Colorless to white crystals or powder with a strong sweet taste.
Solubility	: Soluble in hot water, less soluble in cold water.
pН	: 4.5~6.5 (1 in 100 soln)
Sulfate	: Not more than 0.014% of SO <sub>4</sub> .
Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
Heavy metals	: Not more than 20 ppm (as Pb).
Loss on drying	: Dry it at $80^{\circ}$ for 4 hours: it loses not more than 5% of its weight.
Residue on	: Not more than 8%.
ignition	
Category	: Food Additives Category 11-1
Uses	: Sweetening agent.
	Appearance Solubility pH Sulfate Arsenic Heavy metals Loss on drying Residue on ignition Category

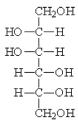
§ 11-1-006

#### D-Mannitol

Synonyms : Mannite; INS No. 421; CAS No. 69-65-8

Chemical formula: C<sub>6</sub>H<sub>14</sub>O<sub>6</sub>

Molecular weight: 182.17



1.	Assay	: Not less than 96.0% and not more than 102.0% on the dried basis.
2.	Appearance	: White, odorless, crystalline powder, with a sweet and cool taste.
3.	Solubility	: Soluble in water, very slightly soluble in ethanol; practically insoluble in
		ether
4.	Melting range	: 164 ~ 169°C.
5.	Thin layer	: Passes test
	chromatography	
6.	рН	: 5~8 (Add 0.5 mL of a saturated solution of potassium chloride to 10 mL of
		a 10% w/v solution of the sample, then measure the pH)
7.	Specific rotation	: $\left[\alpha\right]_{D}^{20} = +23 \sim +25^{\circ}$ (Borate solution)
8.	Chloride	: Not more than 70 mg/kg of Cl.
9.	Sulfate	: Not more than 100 mg/kg (SO <sub>4</sub> ).
10.	Nickel	: Not more than 2 mg/kg.
11.	Lead	: Not more than 1 mg/kg.
12.	Reducing sugar	: Not more than 0.3% (as glucose).
13.	Total sugars	: Not more than 1% (as glucose).
14.	Loss on drying	: Dry it at 105°C for 4 hours: it loses not more than 0.3% of its weight.
15.	Residue on ignition	: Not more than 0.1%.
16.	Category	: Food Additives Category 11-1, 7.
17.	Uses	: Sweeteners, Food quality improvement, fermentation and food processing
		agents

# § 11-1-007

#### Saccharin

Chemical formula: C7H5N		IO <sub>3</sub> S Molecular weight: 183.19
1.	Assay	: Not less than 98% on the dried basis (105°C, 2 h)
2.	Appearance	: Colorless or white crystals or a white, crystalline efflorescent powder, odorless
		or with a slight aroma. Has an intensely sweet taste that 10,000 times aqueous
		solution is still sweet.
3.	Melting range	: 226~230°C.
4.	Solution	: 1 g of the sample is dissolved in 30 mL hot water and 35 mL ethanol
		respectively. Both solution should be colorless and clear.
5.	o-Toluenesulfonam	ide : Add 3 drops of ferric chloride TS to 15 mL of a hot water and 0.5 g of the
		sample. No precipitate or violet color appears.
6.	o-Toluenesulfonam	ide : Not more than 100 ppm.
7.	Heavy metals	: Not more than 10 ppm (as Pb).
8.	Readily	: Dissolve 0.2 g of the sample in 5 mL of sulfuric acid TS. Keep at 48°C to 50°C
	carbonizable	for 10 min. The color should not be darker than a very light brownish-yellow
	substance	(Matching Fluid A).
9.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 1% of its weight.
10.	Category	: Food Additives Category 11-1
11.	Uses	: Sweetening agent.

### Saccharin Sodium

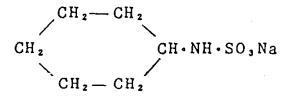
### Chemical formula: $C_7H_4O_3NNaS\cdot O\sim 2H_2O$

1.	Assay	: Not less than 98% of C7H4O3NNaS (dried at 120°C for 4 hours)
2.	Appearance	: White crystals or a white, crystalline efflorescent powder, with an intensely
		sweet taste that 10,000 times aqueous solution is still sweet.
3.	Solution	: 1 g of this product (powder) is dissolved in 1.5 mL of water and 70 mL of
		ethanol respectively. The solution should be colorless and clear.
4.	Free Acid and free	alkali : Dissolve 1 g of the sample in 10 mL of freshly boiled and cooled water.
		Add a drop of phenolphthalein TS. No red color should appear. Add a drop
		of 0.1 N sodium hydroxide. A red color should appear.
5.	Benzoic and salicy	ic : 0.5 g of this product is dissolved in 10 mL of water. Add 5 drops of acetic
	acid	acid and 3 drops of ferric chloride test solution. No precipitate or violet
		color appears.
6.	o-Toluenesulfonam	ide : Not more than 100 ppm.
7.	Arsenic	: Not more than 3 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Readily	: Dissolve 0.2 g of the sample in 5 mL of sulfuric acid TS. Keep at $48^{\circ}$ C to $50^{\circ}$ C
	carbonizable	for 10 min. The color should not be darker than the Matching Fluid A.
	substance	
10.	Loss on drying	: Dry it at 105°C for 4 hours: it loses not more than 15% of its weight.
11.	Category	: Food Additives Category 11-1
12.	Uses	: Sweetening agent.

### Sodium Cyclamate

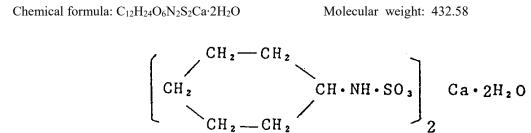
Chemical formula: C<sub>6</sub>H<sub>12</sub>NNaO<sub>3</sub>S

Molecular weight: 201.23



- 1. Description
   : White, odorless crystals or crystalline powder with a sweet taste. Soluble in water, practically insoluble in ethanol, ether, chloroform and benzene. The pH of its 10% solution is 5.5 to 7.5.
- 2. Assay :  $C_6H_{12}NNaO_3S$  Not less than 98.0%
- 3. Sulfate : Not more than 0.024% of SO<sub>4</sub>.
- 4. Chloride : Not more than 0.014% of Cl.
- 5. Arsenic : Not more than 2 ppm (as  $As_2O_3$ ).
- 6. Heavy metals : Not more than 20 ppm (as Pb).
- 7. Loss on drying : Dry it at 105°C for 2 hours: it loses not more than 1.0% of its weight.
- 8. Category : Food Additives Category 11-1
- 9. Uses : Sweetening agent.

### Calcium Cyclamate



- 1. Description
   : White, odorless, sweet crystals or crystalline powder. Soluble in water, sparingly soluble in ethanol, insoluble in chloroform, ether and benzene. The pH of its 10% solution is 5.5-7.5.
- 2. Assay :  $C_{12}H_{24}O_6N_2S_2Ca$  Not less than 98.0%
- 3. Calcium :  $9.9 \sim 10.3\%$ .
- 4. Sulfate : Not more than 0.024% of SO<sub>4</sub>.
- 5. Chloride : Not more than 0.035% of Cl
- 6. Arsenic : Not more than 2 ppm (as  $As_2O_3$ ).
- 7. Heavy metals : Not more than 20 ppm (as Pb).
- 8. Loss on drying : Dry it at 130° for 4 hours: it loses not more than 9.5% of its weight.
- 9. Category : Food Additives Category 11-1
- 10. Uses : Sweetening agent.

### Aspartame

Chen	nical names : N–I	L-α-Aspartyl-L-Phenylalanine 1-Methyl Ester; APM
Chen	nical formula: C <sub>14</sub> H <sub>1</sub>	Molecular weight: 294.31
		$H = CH_2$ $H = CONH = COOCH_3$ $H = H = H = H = H = H = H = H = H = H =$
1.	Description	: White, odorless, crystalline powder, having a strong sweet taste. Slightly soluble
		in water and in ethanol. The pH of its 0.8% solution is 4 - 6.5.
2.	Identification	A. Dissolve 2 g of ninhydrin in 75 mL of dimethylsulfoxide, add 62 mg of
		hydrindantin, dilute to 100 mL with 4 M lithium acetate buffer solution
		(pH 9), and filter. Transfer about 10 mg of the sample to a test tube, add 2 $$
		mL of the reagent solution, and heat. A dark purple color is formed.
		B. Dissolve about 20 mg in 1 mL of methanol, add 0.5 mL of methanol
		saturated with hydroxylamine hydrochloride, mix, and then add 0.3 mL
		of 5 N potassium hydroxide in methanol. Heat the mixture to boiling,
		then cool, adjust the pH to between 1 and 1.5 with hydrochloric acid TS,
		and add 0.1 mL of ferric chloride TS. A burgundy color is produced.
3.	Assay	: $C_{14}H_{18}N_2O_5 98.0\% \sim 102.0\%$ (on the dried basis; 105°C, 4 hr).
4.	5–Benzyl–3,6–dio	xo-2-piperazineacetic Acid: Not more than 1.5%.
5.	Specific rotation	: $(\alpha)_{\rm D}^{20} = +12.5^{\circ} \sim +17.5^{\circ}$ (on the dried basis; 105°C, 4 hr).
6.	Transmittance	: Moderate.
7.	Arsenic	: Not more than 3 ppm (as As).
8.	Heavy metals	: Not more than 10 ppm (as Pb).
9.	Loss on drying	: Not more than 4.5%.
10.	Residue on	: Not more than 0.2%.
	ignition	
11.	Category	: Food Additives Category 11-1

12. Uses : Sweetening agent.

# Steviol glycosides

# Synonyms

Steviol Glycosides from	INS No. 960a
Stevia Rebaudiana Bertoni	
Enzyme modified Steviol	-
Glycosides	
Definition	Steviol glycosides consist of a mixture of compounds containing a steviol backbone
	conjugated to any number or combination of the principal sugar moieties (glucose,
	rhamnose, xylose, fructose, arabinose, galactose and deoxyglucose) in any of the
	orientations occurring in the leaves of Stevia rebaudiana Bertoni.
Steviol Glycosides from	The product is obtained from the leaves of Stevia rebaudiana Bertoni. The leaves
Stevia Rebaudiana Bertoni	are extracted with hot water and the aqueous extract is passed through an
	adsorption resin to trap and concentrate the component steviol glycosides. The resin
	is washed with a solvent alcohol to release the glycosides and the product is
	recrystallized from methanol or aqueous ethanol. Ion exchange resins may be used
	in the purification process. The final product may be spray-dried.
Enzyme modified Steviol	The product is obtained from the enzymatic treatment of purified steviol glycosides
Glycosides	extracted from the leaves of Stevia rebaudiana Bertoni. The purified leaf extract is
	treated with enzymes produced by non-toxigenic non-pathogenic strains of Pichia
	pastoris and Escherichia coli that have been genetically modified with genes from
	multiple donor organisms (listed below) to produce glucosyltransferase and sucrose
	synthase. The resulting material is heated and filtered to denature and remove the
	enzymes. The raw product is concentrated using resin adsorption/desorption or
	solid/liquid filtration, followed by purification and preparation of the product of
	commerce using processes that may include decolourization, crystallization, and
	spray drying.
	Enzyme production organism Gene source

	Pichia pastoris	Horedum vulgare L
		Stevia rebaudiana Bertoni
		Vigna radiate
	Escherichia coli	Acidithiobacillus caldus
		Arapidopsis thaliana
		Solanum tuberosum
		Stevia rebaudiana Bertoni
Chemical names	See Appendix 1	
C.A.S. number	See Appendix 1	
Chemical formula	See Appendix 1	
Structural formula	CH <sub>3</sub> COO-R1	
	Steviol $(R1 = R2 = H)$ is the aglycone of the	e steviol glycosides.
	Glc, Rha, Fru, deoxyGlc, Gal, Ara and Xyl	represent, respectively, glucose,
	rhamnose, fructose, deoxyglucose xylose, g	alactose, arabinose and xylose sugar
	moieties.	
Assay	Not less than 95% on the dried basis.	
Description	White to light yellow powder, odourless or	having a slight characteristic odour.
	About 200 - 300 times sweeter than sucross	2.
Characteristics		
Identification		
Solubility	Very slightly soluble to freely soluble in wa	ater; slightly soluble to freely soluble in
	a mixture of ethanol and water (50:50 v/v).	
HPLC chromatographic	Correspond to steviol glycoside standards	
profile		
nH	Between 4.5 and 7.0 (1 in 100 solution)	

Between 4.5 and 7.0 (1 in 100 solution)

Purity

рΗ

Total ash	Not more than 1%
Loss on drying	Not more than 6% (105°C, 2 h)
Residual solvents	Not more than 200 mg/kg methanol and not more than 5000 mg/kg ethanol
Arsenic	Not more than 1 mg/kg
Lead	Not more than 1 mg/kg
Microbiological criteria	Total (aerobic) plate count: Not more than 1,000 CFU/g
	Yeasts and moulds: Not more than 200 CFU/g
	<i>E. coli</i> : Negative in 1 g
	Salmonella: Negative in 25 g
Category	Food additives category (11-1)
Functional uses	Sweeteners

# <Appendix 1>

Summary of Formula, Molecular weight, steviol equivalent and sugar moieties in Identified Steviol Glycosides from the Leaves of Stevia rebaudiana Bertoni.

#	Common Name	CAS	Trivial	Mol.	Steviol	R1	R2	Reference
		Number	Formula	Wt	Equivalent			
1. Ste	1. Steviol + Glucose (SvGn)							
1.01	Steviolmonoside		SvG1	481	0.66	Н	Glcβ1-	Ohta et al. (2010)
1.02	Steviolmonoside A		SvG1	481	0.66	Glcβ1-	Н	Gardana et al. (2010)
1.03	Rubusoside	64849-39-4	SvG2	643	0.49	Glcβ1-	Glcβ1-	Ohta et al. (2010)
1.04	Steviolbioside	41093-60-1	SvG2	643	0.49	Н	Glcβ(1–2)Glcβ1-	Kohda et al. (1976)
1.05	Stevioside	57817-89-7	SvG3	805	0.40	Glcβ1-	Glcß(1–2)Glcβ1-	Bridel and Lavielle
								(1931)
1.06	Stevioside A		SvG3	805	0.40	Glcß(1-2)Glcβ1-	Glcβ1-	Wu et al. (2012)
1.07	Rebaudioside B	58543-17-2	SvG3	805	0.4	Н	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Kohda et al. (1976)
1.08	Rebaudioside G		SvG3	805	0.4	Glcβ1-	Glcβ(1-3)Glcβ1-	Ohta et al. (2010)
1.09	Stevioside B		SvG3	805	0.4	Glcß(1–3)Glcβ1-	Glcβ1-	Chaturvedula and
								Zamora (2014)
1.10	Rebaudioside E	63279-14-1	SvG4	967	0.33	Glcβ(1–2)Glcβ1-	Glcβ(1–2)Glcβ1-	Sakamoto et al. (1977a)
1.11	Rebaudioside A	58543-16-1	SvG4	967	0.33	Glcβ1-	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Kohda et al. (1976)
1.12	Rebaudioside A2		SvG4	967	0.33	Glcβ1-	Glcβ(1-6)Glcβ(1-2)Glcβ1-	Chaturvedula and
								Prakash (2011a)
1.13	Rebaudioside D	63279-13-0	SvG5	1129	0.28	Glcβ(1–2)Glcβ1-	Glcβ(1–2)[Glcβ(1–3)]Glcβ1-	Sakamoto et al. (1977a)
1.14	Rebaudioside I		SvG5	1129	0.28	Glcβ(1–3)Glcβ1-	Glcβ(1–2)[Glcβ(1–3)]Glcβ1-	Ohta et al. (2010)
1.15	Rebaudioside L		SvG5	1129	0.28	Glcβ1-	Glcβ(1-6)Glcβ(1-2)[Glcβ(1-3)]	Ohta et al. (2010)
							Glcβ1-	
1.16	Rebaudioside Q2		SvG5	1129	0.28	Glcα(1–2)Glcα(1–4)	Glcβ(1–2)Glcβ1-	Chaturvedula and
						Glcβ1-		Prakash (2011b)
1.17	Rebaudioside Q		SvG5	1129	0.28	Glcβ1-	Glcα(1-4)Glcβ(1-2)[Glcβ(1-3)]	ı

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Summary of Formula, Molecular weight, steviol equivalent and sugar moieties in Identified Steviol Glycosides from the Leaves of

1.18     Rebaudioside I2       1.19     Rebaudioside Q       1.20     Rebaudioside I3       1.21     Rebaudioside AI       1.21     Rebaudioside AI	Rebaudioside I2 Rebaudioside Q3		LU, C				Glcβ1-	
	dioside I2 dioside Q3		10.0				-	
	dioside Q3		SVG5	1129	0.28	Glcβ1-	Glcα(1–3)Glcβ(1–2)[Glcβ(1–3)]	Chaturvedula et al.
	dioside Q3						Glcβ1-	(2011a)
			SvG5	1129	0.28	Glcβ1-	$Glc\alpha(1-4)Glc\beta(1-3)[Glc\beta(1-2)]$	Chaturvedula et al.
							Glcβ1-	(2011a)
	aloside 13		SvG5	1129	0.28	Glcβ(1-2)[Glcβ(1-	Glcβ(1-2)Glcβ1-	Chaturvedula et al.
						6)] Glcβ1-		(2011a)
	Rebaudioside AM	222580-	SvG5	1129	0.28	Glcβ(1-2)[Glcβ (1-	Glcβ(1-2)Glcβ1-	Prakash and Ma (2018)
		26-7				3)] Glcβ1-		
	Rebaudioside M	1220616-	SvG6	1291	0.25	Glcβ(1-2)[Glcβ (1-	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Ohta et al. (2010)
		44-3				3)] Glcβ1-		
1.23 Rebau	Rebaudioside 1h		SvG7	1453	0.22	Glcβ(1–3)Glcβ(1–	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Prakash and Ma (2018)
						2)[Glcβ (1–3)]Glcβ1-		
1.24 Rebau	Rebaudioside IX		SvG9	1778	0.18	Glcβ(1–2)[Glcβ(1–	Glcβ(1–3){Glcβ(1–3) [Glcβ(1–	Prakash and Ma (2018)
						3)] Glcβ1-	2)] Glcα (1–6)Glcβ(1–2)} Glcβ1-	
2. Steviol + R	2. Steviol + Rhamnose + Glucose (SvR1Gn)	se (SvR1Gn)						
2.01 Dulcoside A	side A	64432-06-0	SvR1G2	789	0.40	Glcβ1-	Rhaα(1–2)Glcβ1-	Kobayashi et al. (1977)
2.02 Dulcoside B	side B		SvR1G2	789	0.40	Н	Rha $\alpha(1-2)$ [Glc $\beta(1-3)$ ]Glc $\beta1$ -	Ohta et al. (2010)
2.03 Rebau	Rebaudioside C	63550-99-2	SvR1G3	951	0.33	Glcβ1-	Rhaα(1–2)[Glcβ(1–3)]Glcβ1-	Sakamoto et al. (1977b)
2.04 Rebau	Rebaudioside C2		SvR1G3	951	0.33	Rhaα(1–2)Glcβ1-	Glcβ(1–2)Glcβ1-	Purkayastha et al. (2019)
2.05 Rebau	Rebaudioside S		SvR1G3	951	0.33	Rhaα(1–2)Glcβ1-	Glcα (1–2)Glcβ1-	Ibrahim et al. (2016)
2.06 Rebau	Rebaudioside H		SvR1G4	1113	0.29	Glcβ1-	Glcβ(1–3)Rhaα(1–2)[Glcβ(1–3)]	Ohta et al. (2010)
							Glcβ1-	
2.07 Rebau	Rebaudioside K		SvR1G4	1113	0.29	Glcβ(1–2)Glcβ1-	Rhaα(1–2)[Glcβ(1–3)]Glcβ1-	Ohta et al. (2010)
2.08 Rebau	Rebaudioside K2		SvR1G4	1113	0.29	Glcβ(1–6)Glcβ1-	Rha $\alpha(1-2)$ [Glc $\beta(1-3)$ ]Glc $\beta1$ -	Purkayastha et al. (2019)
2.09 Rebau	Rebaudioside J		SvR1G4	1113	0.29	Rhaα(1–2)Glcβ1-	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Ohta et al. (2010)
2.10 Rebau	Rebaudioside N	1220616-	SvR1G5	1275	0.25	Rha $\alpha(1-2)$ [Glc $\beta(1-$	Glcβ(1–2)[Glcβ(1–3)]Glcβ1-	Ohta et al. (2010)

<Appendix 1>

Summary of Formula, Molecular weight, steviol equivalent and sugar moieties in Identified Steviol Glycosides from the Leaves of Stevia rehaudiana Bertoni

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	Purkayastha
•	from
	[Adapted

Ste	Stevia rebaudiana Bertoni	i.					[Adapted from Purkayastha & Kwok (2020)	stha & Kwok (2020)]
		46-5				3)] Glcβ1-		
2.11	Rebaudioside N2		SvR1G5	1275	0.25	Glcβ(1–2)[Glcβ (1– 3)] Glcβ1-	Rhaα(1–2)[Glcβ(1–3)]Glcβ1-	Prakash and Ma (2018)
C1 C	Pahandiosida NG		CUB1GE	175	0.75	GlrR(1_2)Bhar(1_2)	Glc8(1-2)Glc81-	Drakach and Ma (2018)
71.7			COTUAC	C / 7T	C7.0			
						-rdnin[/c_r)dnin]		
2.13	Rebaudioside O	1220616-	SvR1G6	1437	0.22	Glcβ(1–3)Rhaα(1–2)	Glcβ(1–2)[Glcβ(1–3)]Glcβ1-	Ohta et al. (2010)
		48-7				[Glcβ(1–3)]Glcβ1-		
2.14	Rebaudioside O2		SvR1G6	1437	0.22	Glcβ(1–4)Rhaα(1–2)	Glcβ(1–2)[Glcβ(1–3)]Glcβ1-	Purkayastha (2016)
						[Glcβ(1–3)]Glcβ1-		
2.15	Rebaudioside O5		SvR1G6	1437	0.22	Glcβ(1-2)[Glcβ (1-	Glcβ(1–3)Rhaα(1–2)[Glcβ(1–3)]	Prakash and Ma (2018)
						3)] Glcβ1-	Glcβ1-	
2.16	Rebaudioside O6		SvR1G7	1600	0.20	Glc $\beta(1-3)$ Rha $\alpha(1-2)$	Glcβ(1-6)Glcβ(1-3)[Glcβ(1-2)]	Prakash and Ma (2018)
						[Glcβ(1–3)]Glcβ1-	Glcβ1-	
2.17	Rebaudioside O7		SvR2G6	1584	0.20	Glcβ(1–3)Rhaα(1–2)	$Glc\beta(1-3)Rha\alpha(1-2)[Glc\beta(1-3)]$	Prakash and Ma (2018)
						[Glcβ(1–3)]Glcβ1-	Glcβ1-	
3. Ste	<ol> <li>Steviol + Xylose + Glucose (SvX1Gn)</li> </ol>	vX1Gn)						
3.01	Stevioside F		SvX1G2	775	0.41	Glcβ1-	Xylβ(1–2)Glcβ1-	Chaturvedula and
								Prakash (2011c)
3.02	Rebaudioside F	438045-89- 7	SvX1G3	937	0.34	Glcβ1-	Xylβ(1–2)[Glcβ(1–3)]Glcβ1	Starratt et al. (2002)
3.03	Rebaudioside F2		SvX1G3	937	0.34	Glcβ1-	Glcβ(1–2)[Xylβ(1–3)]Glcβ1-	Chaturvedula and
								Prakash (2011c)
3.04	Rebaudioside F3		SvX1G3	937	0.34	Xylβ(1–6)Glcβ1-	Glcβ(1–2)Glcβ1-	Chaturvedula et al.
								(2011b)
3.05	Rebaudioside R		SvX1G3	937	0.34	Glcβ1-	Glcβ(1-2)[Glcβ(1-3)] Xylβ1-	Ibrahim et al. (2016)
3.06	Rebaudioside U		SvX1G4	1099	0.29	Xylβ(1–2)Glcβ1-	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Purkayastha et al. (2019)
3.07	Rebaudioside U2		SvX1G4	1099	0.29	Xylβ(1–2)[Glcβ(1–3)]	Glcβ(1–2)Glcβ1-	Purkayastha (2016)

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Summary of Formula, Molecular weight, steviol equivalent and sugar moieties in Identified Steviol Glycosides from the Leaves of

Ste	Stevia rebaudiana Bertoni						[Adapted from Purkayastha & Kwok (2020)]	stha & Kwok (2020)]
						Glcβ1-		
3.08	Rebaudioside U3		SvX1G4	1099	0.29	Xylβ(1-2)[Glcβ(1-4)]	Glcβ(1–2)Glcβ1-	Purkayastha et al. (2019)
						Glcβ1-		
3.09	Rebaudioside V		SvX1G5	1261	0.25	Glcβ(1-2)[Glcβ(1-	Glcβ(1-2)[Xylβ(1-3)]Glcβ1-	Purkayastha et al. (2019)
						3)] Glcβ1-		
3.10	Rebaudioside V2		SvX1G5	1261	0.25	Xylβ (1-2)[Glcβ(1-	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Prakash and
						3)] Glcβ1-		Chaturvedula (2013)
4. Ste	4. Steviol + Arabinose + Glucose (SvA1Gn)	(SvA1Gn)						
4.01	Rebaudioside W		SvA1G4	1098	0.29	Glc $\beta(1-2)$ [Ara $\beta(1-$	Glcß(1-2)Glcß1-	Purkayastha (2016)
						3*)] Glcβ1		
4.02	Rebaudioside W2		SvA1G4	1098	0.29	Araβ(1–2*)Glcβ1	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Purkayastha (2016)
4.03	Rebaudioside W3		SvA1G4	1098	0.29	Araβ(1–6)Glcβ1-	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Purkayastha et al. (2019)
4.04	Rebaudioside Y		SvA1G5	1260	0.25	Glcβ(1–2)[Araβ(1–	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Purkayastha et al. (2019)
						3*)] Glcβ1		
5. Ste	5. Steviol + Fructose + Glucose (SvF1Gn)	SvF1Gn)						
5.01	Rebaudioside A3		SvF1G3	967	0.33	Glcβ1-	Glcβ(1-2)[Fruβ(1-3)]Glcβ1-	Chaturvedula et al.
								(2011c)
6. Ste	6. Steviol + Galactose + Glucose (SvGa1Gn)	(SvGa1Gn)						
6.01	Rebaudioside T		SvGa1G4	1129	0.28	Glcβ(1–2)Glcβ1-	Galβ(1–2)[Glcβ(1–3)]Glcβ1-	Purkayastha (2016)
7. Ste	7. Steviol + Deoxyglucose + Glucose (SvdG1Gn)	cose (SvdG1G	(-					
7.01	Stevioside D		SvdG1G2	789	0.40	Glcβ1-	6-deoxyGlcβ(1–2)Glcβ1-	Chaturvedula and
								Prakash (2011d)
7.02	Stevioside E		SvdG1G3	951	0.33	Glcβ1-	6-deoxyGlcβ(1-2)[Glcβ(1-3)]	Chaturvedula and
							Glcβ1-	Prakash (2011d)
7.03	Stevioside E2		SvdG1G3	951	0.33	6-deoxyGlcβ1-	Glcβ(1-2)[Glcβ(1-3)]Glcβ1-	Chaturvedula et al.
								(2011d)

### Licorice Extracts

1.	Description	: Extract of Glycyrrhiza glabra L. or other plants' roots and stems. The main ingredient
		of its sweet taste is glycyrrhizin. The sample is dark brown crystal, powder, granule,
		liquid, cream, scale or block, etc., with a unique sweet taste, odorless or with a
		characteristic odor.
2.	Methanol	: Not detectable.
3.	Arsenic	: Not more than 3 ppm (as As).

- 4. Lead : Not more than 10 ppm.
- 5. Heavy metals : Not more than 50 ppm (as Pb).
- 6. Category : Food Additives Category 11-1
- 7. Uses : Sweetening agent.

	Acesulfame Potassium
Synonyms	Acesulfame K; INS No. 950
Definition	
Chemical names	Potassium salt of 6-methyl-1,2,3-oxathiazine-4(3H)-one-2,2-dioxide;
	potassium salt of 3,4-dihydro-6-methyl-1,2,3-oxathiazine-4-one-2,2-dioxide
CAS NUMBER	55589-62-3
Chemical formula	C <sub>4</sub> H <sub>4</sub> KNO <sub>4</sub> S
Molecular weight	201.24
Assay	99.0%-101.0% on the dried basis
Appearance	Odorless, white crystalline powder.
Characteristics	
Identification	
Solubility	Freely soluble in water, very slightly soluble in ethanol
Spectrophotometry	Dissolve 10 mg of the sample in 1,000 mL of water. The solution shows an
	absorbance maximum at 227±2 nm
Test for potassium	Passes test
	(Test the residue obtained by igniting 2 g of the sample)
Precipitation test	Add a few drops of a 10% solution of sodium cobaltinitrite to a solution of
	0.2~g of the sample in 2 mL of acetic acid TS and 2 mL of water. A yellow
	precipitate is produced.
Purity	
Loss on drying	Dry it at 105°C for 2 hours: it loses not more than 1.0% of its weight.
pH	5.5 - 7.5 (1% soln)
Organic impurities	Passes test for 20 mg/kg of UV active components
Fluoride	Not more than 3 mg/kg
Lead	Not more than 1 mg/kg
Category	Food Additives Category 11-1
Uses	Sweetening agent.

# Ammoniated Glycyrrlizinate

1.	Ash	: Not more than 2.5%.
2.	Arsenic	: Not more than 3 ppm (as As).
3.	Heavy metals	: Not more than 40 ppm (as Pb).
4.	Category	: Food Additives Category 11-1
5.	Uses	: Sweetening agent.

### Maltitol

Chemical formula: C <sub>12</sub> H <sub>24</sub>		O <sub>11</sub> Molecular weight: 344.31
1.	Description	: White crystalline masses. Very soluble in water, slightly soluble in ethanol.
2.	Assay	: D-maltitol: Not less than 98.0%.
3.	Melting range	: 148~151°C.
4.	Specific rotation	: $(\alpha)_{D}^{20} = +105.5 \sim +108.5^{\circ}$ (Dissolve 5 g of this product in 100 mL of water)
5.	Water	: Not more than 1% (Karl Fischer Method)
6.	Sulfated ash	: Not more than 0.1%.
7.	Reducing sugar	: Not more than 0.1%.
8.	Chloride	: Not more than 50 ppm.
9.	Sulfate	: Not more than 100 ppm.
10.	Nickel	: Not more than 2 ppm.
11.	Lead	: Not more than 1 ppm.
12.	Heavy metals	: Not more than 10 ppm (as Pb).
13.	Category	: Food Additives Category 11-1, 7.
14.	Uses	: Sweeteners, Food quality improvement, fermentation and food processing agents

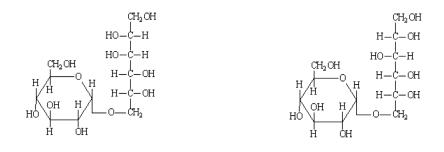
# Maltitol Syrup

<u>Synonyms</u>	Hydrogenated high maltose-content glucose syrup, hydrogenated glucose syrup, dried
	maltitol syrup, maltitol syrup powder; INS No. 965(ii)
<b>Definition</b>	A mixture consisting of mainly maltitol with sorbitol and hydrogenated oligo- and
	polysaccharides. It is manufactured by the catalytic hydrogenation of high maltose-
	content glucose syrup. The article of commerce is typically supplied as a syrup. It may
	also be dried and supplied as a solid product
Assay	Not less than 99.0% of total hydrogenated saccharides on the anhydrous basis and not
	less than 50.0% of maltitol on the anhydrous basis
<b>Description</b>	Colourless and odourless, clear viscous liquids or white crystalline masses
<b>Characteristics</b>	
Identification	
Solubility	Very soluble in water, slightly soluble in ethanol
Thin layer	Passes test
chromatography	
<u>Purity</u>	
Water	Not more than 31% (Karl Fischer)
Sulfated ash	Not more than 0.1%
Chloride	Not more than 50 mg/kg
Sulfate	Not more than 100 mg/kg
Nickel	Not more than 2 mg/kg
Reducing sugars	Not more than 0.3%
Lead	Not more than 1 mg/kg
Category	Food additives category (07) (11-1)
Functional uses	Food quality improvement, fermentation and food processing agents; Sweeteners.

### Isomalt (Hydrogenated Palatinose)

Chemical names	: A mixture of 6-O-alpha-D-glucopyranosyl-D-sorbitol (1,6-GPS) and	1-O-alpha-D-
	glucopyranosyl-D-mannitol dihydrate (1,1-GPM).	
Synonyms	: Hydrogenated isomaltulose; INS No. 953; CAS No. 64519-82-0	
Chemical	$: C_{12}H_{24}O_{11}(1,6-GPS)$	Molecular weight:
formula	C <sub>12</sub> H <sub>24</sub> O <sub>11</sub> ·2H <sub>2</sub> O (1,1-GPM)	344.32

380.32



6-O-alpha-D-glucopyranosyl-D-sorbitol

1-O-alpha-D-glucopyranosyl-D-mannitol (without molecules of crystal water)

1.	Assay	: Not less than 98% of hydrogenated mono- and disaccharides and not less than 86% of
		the mixture of 1,6-GPS and 1,1-GPM.
2.	Description	: Odorless, white, crystalline slightly hygroscopic substance.
3.	Solubility	: Soluble in water, very slightly soluble in ethanol.
4.	Identification	: 0.5 g of the sample is soluble in 100 mL of water. Spot on a silicone plate with a
		thickness of 0.2 mm and a length of about 12 cm (Kieselgel 60 F254 or the same size).
		The sample can be expanded to get 1,6-GPS and 1,1-GPM
5.	Water	: Not more than 7.0% (Karl Fischer Method)
6.	Sulfated ash	: Not more than 0.05%.
7.	D-Mannitol	: Not more than 3%.
8.	D-Sorbitol	: Not more than 6%.
9.	Reducing sugar	: Not more than 13%.
10.	Nickel	: Not more than 2 mg/kg.
11.	Lead	: Not more than 1 mg/kg.
12.	Heavy metals	: Not more than 10 mg/kg.
13.	Category	: Food Additives Category 11-1, 7.
14.	Uses	: Sweeteners, Food quality improvement, fermentation and food processing agents

### Lactitol

Chemical formula: C <sub>12</sub> H <sub>24</sub> O <sub>11</sub>		O <sub>11</sub> Molecular weight: 344.32
1.	Assay	: Not less than 95% and not more than 102% on the dried basis.
2.	Description	: Sweet tasting crystalline powders or colorless solutions. Very soluble in water.
3.	Specific rotation	: $(\alpha)_{D}^{25} = 13 \sim 15^{\circ} (10\% \text{ w/v aqueous solution})$
4.	Water content	: Not more than 10.5% (Karl Fischer Method)
5.	Other polyols	: Not more than 2.5% (on the dried basis)
6.	Reducing sugar	: Not more than 0.2% (on the dried basis, as glucose)
7.	Chloride	: Not more than 100 ppm of Cl.
8.	Sulfate	: Not more than 200 ppm of SO <sub>4.</sub>
9.	Sulfated ash	: Not more than 0.1%.
10.	Nickel	: Not more than 2 ppm of Ni.
11.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
12.	Lead	: Not more than 1 ppm (as Pb).
13.	Heavy metals	: Not more than 10 ppm (as Pb).
14.	Category	: Food Additives Category 11-1, 7.
15.	Uses	: Sweeteners, Food quality improvement, fermentation and food processing agents

# Monoglucuronyl Glycyrrhetic Acid

1.	Assay	: glycyrrhizic acid $40 \sim 45\%$ ;
		monoglucuronyl glycyrrhetic acid $15 \sim 20\%$ .
2.	Description	: The sample is obtained by hydrolyzing glycyrrhizin, extract of root or stem of
		Glycyrrhiza glabra L. or other plants of the same genus, by a glucuronic acid. Yellow
		powder, odorless, with a special sweet taste. Soluble in hot water.
3.	Solution	: When 1 g of the sample is dissolved in 10 mL of 50%(v/v) ethanol, the solution should
		be clear.
4.	pH	$: 5.0 \sim 6.0 (1 \text{ in } 100 \text{ soln})$
5.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Loss on drying	: Not more than 6.0% (80°C, 3 h)
8.	Residue on	: Not more than 16%
	ignition	
9.	Category	: Food Additives Category 11-1
10.	Uses	: Sweetening agent.

### Thaumatin

1.	Assay	: Not less than 16.0% nitrogen
2.	Description	: Obtained by aqueous extraction of the arils of the fruit of Thaumatococcus daniellii
		(Benth); odorless, cream-colored powder with a sweet taste, odorless. Very soluble in
		water; insoluble in acetone.
3.	Carbohydrates	: Not more than 3.0% on the dried basis.
4.	Specific	: The specific absorption, A1% 1 cm at the wavelength of maximum absorption (about
	absorbance	279 nm) shall be not less than 12.0 and not more than 12.5 determined on the dried basis
		and using a 1 in 100 w/v solution of the sample in water at pH 2.7.
5.	Aluminum	: Not more than 100 ppm.
6.	Lead	: Not more than 10 ppm.
7.	Arsenic	: Not more than 3 ppm (as As).
8.	Sulfated ash	: Not more than 2.0% on the dried basis.
9.	Less en durine	Not more than $0.00/(105\%)$ to constant weight
9.	Loss on drying	: Not more than 9.0% (105°C to constant weight).
10.	Category	: Food Additives Category 11-1
11.	Uses	: Sweetening agent.

# Erythritol

Chemical formula: C <sub>4</sub> H <sub>10</sub>	O <sub>4</sub> Molecular weight: 122.12
1. Assay	: Not less than 99.5%
2. Description	: Colorless to white, odorless crystals or crystalline powder, having a sweet taste. Freely
	soluble in water, slightly soluble in ethanol, insoluble in diethyl ether.
3. Reducing sugar	: Not more than 13%.
4. Lead	: Not more than 1 ppm.
5. Heavy metals	: Not more than 5 ppm (as Pb).
6. Loss on drying	: Not more than 0.2%.
7. Residue on	: Not more than 0.1% (Sulfated ash)
ignition	
8. Category	: Food Additives Category 11-1, 7.
9. Uses	: Sweeteners, Food quality improvement, fermentation and food processing agents

### Sucralose

Gene	eral name : Suc	ralose
Chemical names : 1,6–1		$-Dichloro-1, 6-dideoxy-\beta-D-fructofuranosyl-4-chloro-4-deoxy-\alpha-D-galactopyranoside;$
	4,	1–Dichloro–1,6–dideoxy–β–D–fr
Cher	nical formula: C <sub>12</sub> H <sub>19</sub> O	Cl <sub>3</sub> O <sub>8</sub> Molecular weight: 397.64
1.	Description	: White to off-white, practically odorless crystalline powder with a sweet taste. Freely
		soluble in water, methanol and ethanol; slightly soluble in ethyl acetate.
2.	Identification	: (1) The infrared absorption spectrum (absorption intensity may vary) obtained by the
		KBr disc method of the sample should be the same as the standard product.
		(2) The retention time of the major peak (except the solvent peak) in the chromatogram
		of the Assay Preparation is the same as that of the Standard Preparation.
		(3) The major spot in the test solution of the thin layer chromatography has the same Rf
		value as that of the main spot of Standard Solution.
3.	Assay	: Not less than 98.0 and not more than 102.0% (as $Cl$ ) <sub>2</sub> H <sub>19</sub> Cl <sub>3</sub> O <sub>8</sub> on the dried basis.
4.	Heavy metals	: Not more than 10 mg/kg (as Pb).
5.	Arsenic	: Not more than 3 mg/kg (as As).
6.	Hydrolysis products	: Passes test (Not more than 0.1%).
7.	Methanol	: Not more than 0.1%.
8.	Related substances	: Passes test (Not more than 0.5%).
9.	Residue on	: Not more than 0.7%.
	ignition	
10.	Specific rotation	: $\left[\alpha\right]_{D}^{20} = +84.0 \sim +87.5$ (10% w/v solution, on the dried basis).
11.	Water	: Not more than 2.0%.
12.	Category	: Food Additives Category 11-1

Chemical formula: C<sub>20</sub>H<sub>30</sub>N<sub>2</sub>O<sub>5</sub>

### Neotame

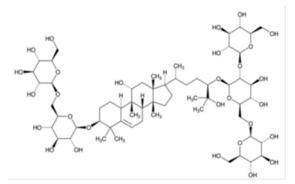
 $Chemical \ names \qquad : N-[N-(3,3-Dimethylbutyl)-L-\alpha-aspartyl]-L-phenylalanine \ 1-methyl \ ester$ 

1.	Definition	: This product is anufactured from aspartame and 3,3-dimet	hylbutyraldehyde. Equimolar
		amounts (as As)partame and 3,3-dimethylbutyraldehyde are	e reacted with hydrogen gas in
		methanol. The product is obtained by isolation and purificat	tion.
2.	Assay	: Not less than 97.0 and not more than 102.0% on the dried	basis.
3.	Description	: White to off-white powder.	
4.	Identification	: (1) Solubility: Slightly soluble in water, soluble in ethanol	
		(2) Infrared spectrum: The product is identified by compar	ring the infrared absorption
		spectrum of a potassium bromide dispersion with that of a r	eference standard.
5.	pН	: 5.0–7.0 (0.5% soln).	
6.	Melting range	: 81–84°C.	
7.	Water	: Not more than 5.0% (Sampling 25±5 mg, Karl Fischer).	
8.	N-[N-(3,3-Dimethy	/lbutyl)-α- aspartyl]-L-phenylalanine	: Not more than 1.5%.
9.	Other related	: Not more than 2.0%.	
	substances		
10.	Sulfated ash	: Not more than 0.2%.	
11.	Specific rotation	: $\left[\alpha\right]_{D}^{20} = -40.0^{\circ} \sim -43.3^{\circ} (0.5\% \text{ soln, on the dried basis})$	
12.	Lead	: Not more than 1 mg/kg.	
13.	Category	: Food Additives Category 11-1	
14.	Uses	: Sweetening agent.	

### **Mogroside Extract**

DEFINITION :Mogroside Extract is obtained from the fruits of the luohanguo plant Siraitia grosvenorii (Swingle) C. Jeffrey ex A. M. Lu & Zhi Y. Zhang (Momordica grosvenori Swingle) and consists mainly of mogrosides.

Content :Mogroside Extract, when dried, contains not less than 20% of mogroside V ( $C_{60}H_{102}O_{29} = 1287.43$ ).



Mogroside V

**Description** : light yellow to light brown powder having a sweet taste.

### CHARACTERISTICS

### **IDENTIFICATION**

- Color reaction : To 5–10 mg of Mogroside Extract, previously dried, add 2 mL of acetic anhydride, warm for 2 minutes, and slowly add 0.5 mL of sulfuric acid. The boundary surface turns red-brown.
- Chromatography : Major component corresponds with the Mogroside V.

Purity

Lead	: Not more than 1.0 mg/kg.
Cadmium	: Not more than 1.0 mg/kg.
Arsenic	: Not more than 0.5 mg/kg.
Loss in Drying	: Not more than 6.0% (105°C , hours).
Residue on Ignition	: Not more than 2.0%.
Category	: Food additives category (11-1).
Functional uses	: Sweeteners.

§ 12001

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Sodium Alginate
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	C
Synonyms	INS No. 401
Definition	Sodium salt of alginic acid
C.A.S. number	9005-38-3
Chemical formula	(C <sub>6</sub> H <sub>7</sub> NaO <sub>6</sub> )n
Structural formula	$ \begin{array}{c} & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & $
	sequence of the Mannuronate and Glucuronate residues shown above vary in the
	naturally occurring alginate. The water molecules associated with the alginate molecule
	are not shown in the above structural formula.
Formula weight	Structural unit : 198.11 (theoretical), 222 (actual average) Macromolecule : 10,000 -
	600,000 (typical average)
Assay	Yields, on the dried basis, not less than 18.0% and not more than 21.0% of carbon
	dioxide (CO <sub>2</sub> ), equivalent to not less than 90.8% and not more than 106.0% of sodium
	alginate (C <sub>6</sub> H <sub>7</sub> NaO <sub>6</sub> )n.
Description	White to yellowish brown filamentous, grainy, granular or powdered forms
Characteristics	
Identification	
Solubility	Dissolves slowly in water, forming a viscous solution; insoluble in ethanol and ether
Precipitate formation with	To a 0.5% solution of the sample in sodium hydroxide TS add one-fifth of its volume of
calcium chloride	a 2.5% solution of calcium chloride. A voluminous, gelatinous precipitate is formed.
Precipitate formation with	To a 0.5% solution of the sample in sodium hydroxide TS add one-half of its volume of a
ammonium sulphate	saturated solution of ammonium sulfate. No precipitate is formed.
Test for alginate	Passes test
	459

Test for sodium	Passes test
Purity	
Loss on drying	Not more than 15% (105°, 4 h)
Water-insoluble matter	Not more than 2% on the dried basis
Arsenic	Not more than 3 mg/kg
Lead	Not more than 5 mg/kg
Microbiological criteria	Total plate count: Not more than 5,000 colonies per gram.
	Yeasts and moulds: Not more than 500 colonies per gram
	Coliforms: Negative by test
	Salmonella: Negative by test
Category	Food additives category (12)
Functional uses	Pasting Agent

# § 12002

# Propylene Glycol Alginate

1.	Appearance	: Occurs as white to yellowish brown filamentous, grainy, granular or powdered forms. It
		is almost odorless.
2.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
3.	Heavy metals	: Not more than 20 ppm (as Pb).
4.	Degree of	: Not less than 75%.
	esterification	
5.	Insoluble ash	: Not more than 1.5%.
6.	Loss on drying	: Dry it at 105°C for 4 hours: it loses not more than 15% of its weight.
7.	Category	: Food Additives Category 12
8.	Uses	: Pasting Agent.

### Casein

1.	Assay	: Not less than 14.7% and not more than 16.0% of nitrogen on the dried basis (110°C, 3
		h)
2.	Appearance	: White or pale yellow granules or powder; tasteless, odorless or having a specific aroma.
3.	Solution	: Dry the sample for 4 hours in reduced pressure sulfuric acid dryer, and ground the
		sample into a fine powder. Shake and mix 0.1 g of the sample and 30 mL of water.
		After standing for 10 minutes, add 2 mL of 0.1 N sodium hydroxide solution. While
		heating at 60 $^{\circ}$ C for 1 hour, it was mixed by shaking to dissolve it. When cold water is
		added to make a 100 mL solution, the solution should be colorless and the turbidity
		should be below "slightly turbid".
4.	pН	: Add 1.5 g of the sample to 30 mL of water, shake and mix for 10 minutes, then filter.
		The pH of the filtrate should be $3.7 \sim 6.5$ .
5.	Water soluble	: Not more than 1%.
	substances	
6.	Fat	: Not more than 1.5%.
7.	Arsenic	: Not more than 1 ppm (as As <sub>2</sub> O <sub>3</sub> ).
8.	Heavy metals	: Not more than 20 ppm (as Pb).
9.	Loss on drying	: Dry it at 100° for 3 hours: it loses not more than 12% of its weight.
10.	Residue on	: Dry it at 100° for 3 hours: it loses not more than 2.5% of its weight.
	ignition	
11.	Category	: Food Additives Category 12
12.	Uses	: Pasting Agent.

# § 12004

### Sodium Caseinate

1.	Assay	: Not less than 14.5% and not more than 15.8% of nitrogen after drying 3 hours at 100°C.
2.	Annoaranaa	: White or pale yellow granules or powder; practically odorless or having a specific
2.	Appearance	aroma.
3.	Solution	: The test in "3. Solution" of Casein is applicable.
4.	рН	: 6.0~7.5. 6.5 - 7.5 (1 in 50 soln)
5.	Fat	: Not more than 1.5%.
6.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
7.	Heavy metals	: Not more than 20 ppm (as Pb).
8.	Loss on drying	: Dry it at 100° for 3 hours: it loses not more than 15% of its weight.
0	Residue on	: Not more than 6% on the ignited basis (100°C, 3 h)
9.	ignition	
10.	Category	: Food Additives Category 12
11.	Uses	: Pasting Agent.

### Sodium Carboxmethyl Cellulose

Chemical names : Sodium salt of carboxymethyl ether of cellulose

Synonyms : Sodium cellulose glycolate, Na CMC, CMC, Cellulose Gum, INS No. 466, C.A.S. No. 9004-32-4

Chemical formula:  $[C_6H_7O_2(OH)_x(OCH_2COONa)_v]_n$ 

n is the degree of polymerization

x=1.50~2.80

y=0.2~1.50

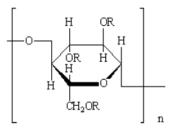
x+y=3.0

y=degree of substitution

Molecular weight: Structural unit with a degree of substitution of 0.20: 178.14

Structural unit with a degree of substitution of 1.50: 282.18

Macromolecules: greater than about 17,000 (n about 100)



where R=H or CH<sub>2</sub>COONa

- 1. Assay : Not less than 99.5% of sodium carboxymethyl cellulose, calculated on the dried basis.
- Appearance : White or slightly yellowish, almost odorless hygroscopic granules, powder or fine fibers.
- 3. Solubility : Yield viscous colloidal solution with water; insoluble in ethanol.
- Identification : (1) Foam test: Vigorously shake a 0.1% solution of the sample. No layer of foam appears. This test distinguishes sodium carboxymethyl cellulose from other cellulose ethers and from alginates and natural gums.

(2)Precipitate formation: To 5 mL of an 0.5% solution of the sample add 5 mL of a 5% solution of copper sulfate or of aluminum sulfate. A precipitate appears. (This test permits the distinction of sodium carboxymethyl cellulose from other cellulose ethers, and from gelatine, carob bean gum and tragacanth gum).

(3)Color reaction: Add 0.5 g of powdered carboxymethylcellulose sodium to 50 mL of water, while stirring to produce a uniform dispersion. Continue the stirring until a clear solution is produced. To 1 mL of the solution, diluted with an equal volume of water, in a small test tube, add 5 drops of 1-naphthol TS. Incline the test tube, and carefully

		introduce down the side of the tube 2 mL of sulfuric acid so that it forms a lower layer. A
		red-purple color develops at the interface.
5.	Loss on drying	: Not more than 12.0% (105°Cto constant weight).
6.	pН	: 6.0 - 8.5 (1 in 100 soln)
7.	Sodium	: Not more than 12.4% on the dried basis Determine total sodium content by Atomic
		Absorption Spectroscopy or Flame Photometry.
8.	Sodium chloride	: Not more than 0.5% on the dried basis.
9.	Free glycolate	: Not more than 0.4% calculated as sodium glycolate on the dried basis.
10.	Degree of	: Not less than 0.20 and not more than 1.50.
	substitution	
11.	Lead	: Not more than 2 mg/kg on the dried basis. Determine using an atomic absorption
		technique appropriate to the specified level.
12.	Heavy metals	: Not more than 20 mg/kg (as Pb).
13.	Category	: Food Additives Category 12
14.	Uses	: Pasting Agent.

# § 12007

# Calcium Carboxmethyl Cellulose

1.	Appearance	: White to off-white powder, odorless.
2.	pH	: Add 50 mL of freshly boilded and cooled water to 1 g of this product. After fully
		oscillating and mixing, the pH should be below 8.0.
3.	Chloride	: Not more than 0.6% of Cl.
4.	Sulfate	: Not more than $0.9\%$ of SO <sub>4</sub> .
5.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
6.	Heavy metals	: Not more than 30 ppm (as Pb).
7.	Starch	: Add 10 mL of water to 0.1g of the sample, and dissolved it by heating. When adding 2
		drops of iodine test solution after cooling, it should not be blue to violet.
8.	Loss on drying	: Dry it at 105°C for 4 hours: it loses not more than 10% of its weight.
9.	Residue on	: $10 \sim 20\%$ (ignited at 105°C for 4 hours)
	ignition	
10.	Category	: Food Additives Category 12
11.	Uses	: Pasting Agent.

	Food Starch, Modified
Synonyms	Acid treated starch: INS No. 1401
	Alkaline treated starch: INS No. 1402
	Bleached starch: INS No. 1403
	Oxidized starch: INS No. 1404
	Monostarch phosphate: INS No. 1410
	Distarch phosphate: INS No. 1412
	Phosphated distarch phosphate: INS No. 1413
	Acetylated distarch phosphate: INS No. 1414
	Starch acetate: INS No. 1420
	Acetylated distarch adipate: INS No. 1422
	Hydroxypropyl starch: INS No. 1440
	Hydroxypropyl distarch phosphate: INS No. 1442
	Starch sodium octenylsuccinate: INS No. 1450
Definition	: Food starches which have one or more of their original characteristics altered by treatment
	in accordance with good manufacturing practice by one of the procedures listed in Table 1.
	In the case of starches treated with heat in the presence of acid or with alkali, the alteration
	is a minor fragmentation. When the starch is bleached, the change is essentially in the color
	only. Oxidation involves the deliberate production of carboxyl groups. Acetylation results in
	substitution of hydroxyl groups with acetyl esters. Treatment with reagents such as
	orthophosphoric acid results in partial substitution in the 2, 3- or 6- position of the
	anhydroglucose unit unless the 6-position is occupied for branching. In cases of cross-
	linking, where a polyfunctional substituting agent, such as phosphorus oxychloride, connects
	two chains, the structure can be represented by: Starch-O-R-O-Starch, where $R = cross$ -
	linking group and Starch refers to the linear and/or branched structure.
C.A.S. number	Starch acetate: 9045-28-7
	Acetylated distarch adipate: 68130-14-3
	Hydroxypropyl starch: 9049-76-7
	Hydroxypropyl distarch phosphate: 53124-00-8
	Starch sodium octenylsuccinate: 66829-29-6
Description	Most modified starches are white or off-white, odourless powders. According to the drying
	method these powders can consist of whole granules having the appearance of the original
	native starch, or aggregates consisting of a number of granules (pearl starch, starch-grits) or,
	if pre-gelatinized, of flakes, amorphous powder or coarse particles.
Characteristics	

Identification

Solubility	Insoluble in cold water (if not pre-gelatinized); forming typical colloidal solutions with
	viscous properties in hot water; insoluble in ethanol.
Microscopy	Modified starches which have not been pre-gelatinized retain their granular structure and can
	be identified as starches by microscopic observation. Shape, size and sometimes striations
	are characteristics of the botanical origin. In polarized light under cross nicol prisms the
	typical polarization cross will be observed
Iodine stain	Add a few drops of 0.1 N potassium tri-iodide to an aqueous suspension of the sample.
	These starches stain with iodine in the same way as native starches. The colour can range
	from dark blue to red.
Copper	Place about 2.5 g of the sample previously washed with water, in a boiling flask, add 10 ml
reduction	of dilute hydrochloric acid (3%) and 70 ml of water, mix, reflux for about three hours and
	cool. Add 0.5 ml of the resulting solution to 5 ml of hot alkaline cupric tartrate TS. A
	copious red precipitate
	is produced.
Differentiation	Passes test for type of starch
test	1. Hypochlorite oxidized starch
	2. Specific reaction for acetyl groups
	3. Positive test for ester groups
Purity	
Sulfur dioxide	: Not more than 50 mg/kg for modified cereal starches
	Not more than 10 mg/kg for other modified starches unless otherwise specified in Table 1
Lead	: Not more than 2 mg/kg
Additional purity	See Table 1
specifications for	
individual chemically	
modified starches	
Category	: Food additives category (12)
Functional uses	: Pasting Agents

Table 1. Additional purity specifications for individual chemically modified starches (All percentages calculated on dry substance)

Modification	Process limitations	End-product specifications
Acid treated starch	Treatment with hydrochloric acid or ortho-phosphoric acid or sulfuric acid	Final pH 4.8 – 7.0

Gelatinized starch (Alkaline treated starch)	Treatment with sodium hydroxide or potassium hydroxide	Final pH 5.0 – 7.5
Bleached starch	Treatment with peracetic acid and/or hydrogen peroxide, or sodium hypochlorite or sodium chlorite, or sulfur dioxide or alternative permitted forms of sulfites, or potassium permanganate or ammonium persulfate	Added carbonyl group not more than 0.1% No residual reagent Residual sulfur dioxide not more than 50 mg/kg Residual manganese not more than 50 mg/kg
Oxidized starch	Treatment with sodium hypochlorite	Carboxyl groups not more than 1.1% Residual sulfur dioxide not more than 50 mg/kg
Monostarch phosphate	Esterification with orthophosphoric acid, or sodium or potassium ortho- phosphate, or sodium tripolyphosphate	Phosphate calculated as phosphorus not more than 0.5% for potato or wheat, and not more than 0.4% for other starches
Distarch phosphate	Esterification with sodium trimetaphosphate or phosphorus oxychloride	Phosphate calculated as phosphorus not more than 0.5% for potato and wheat, and not more than 0.4% for other starches
Phosphated distarch phosphate	Combination of treatments for Monostarch phosphate and Distarch phosphate	Phosphate calculated as phosphorus not more than 0.5% for potato and wheat, and not more than 0.4% for other starches
Acetylated distarch phosphate	Esterification with sodium trimetaphosphate or phosphorus oxychloride combined with esterification with acetic anhydride or vinyl acetate	Acetyl groups not more than 2.5%; phosphate calculated as phosphorus not more than 0.14% for potato and wheat, and 0.04% for other starches; and vinyl acetate not more than 0.1 mg/kg

Starch acetate	Esterification with acetic anhydride or vinyl acetate	Acetyl groups not more than 2.5%
Acetylated distarch adipate	Esterification with acetic anhydride and adipic anhydride	Acetyl groups not more than 2.5% and adipate groups not more than 0.135%
Hydroxypropyl starch	Etherification with propylene oxide	Hydroxypropyl groups not more than 7.0%; propylene chlorohydrin not more than 1 mg/kg
Hydroxypropyl distarch phosphate	Esterification with sodium trimetaphosphate or phosphorus oxychloride combined with etherification by propylene oxide	Hydroxypropyl groups not more than 7.0%; propylene chlorohydrin not more than 1 mg/kg; and residual phosphate calculated as phosphorus not more than 0.14% for potato and wheat, and not more than 0.04% for other starches
Starch sodium octenylsuccinate	Esterification with octenylsuccinic anhydride	Octenylsuccinyl groups not more than 3%; and residual octenylsuccinic acid not more than 0.3%
Oxidized Hydroxypropyl Starch	Chlorine, as sodium hypochlorite, not to exceed 5.5% of dry starch; active oxygen obtained from hydrogen peroxide, not to exceed 0.45%; and propylene oxide, not to exceed 25%	Propylene chlorohydrin not more than 5 mg/kg
Starch Aluminum Octenyl Succinate	Octenyl succinic anhydride, not to exceed 2%, and aluminium sulfate, not to exceed 2%	

Starch Sodium Succinate	Succinic anhydride, not to exceed 4%	_
Distarchoxy Propanol	Acrolein not to exceed 0.6%	
Modification	Process limitations	End-product specifications
Acid treated starch	Treatment with hydrochloric acid or ortho-phosphoric acid or sulfuric acid	Final pH 4.8 – 7.0
Gelatinized starch (Alkaline treated starch)	Treatment with sodium hydroxide or potassium hydroxide	Final pH 5.0 – 7.5
Bleached starch	Treatment with peracetic acid and/or hydrogen peroxide, or sodium hypochlorite or sodium chlorite, or sulfur dioxide or alternative permitted forms of sulfites, or potassium permanganate or ammonium persulfate	Added carbonyl group not more than 0.1% No residual reagent Residual sulfur dioxide not more than 50 mg/kg Residual manganese not more than 50 mg/kg
Oxidized starch	Treatment with sodium hypochlorite	Carboxyl groups not more than 1.1% Residual sulfur dioxide not more than 50 mg/kg
Monostarch phosphate	Esterification with orthophosphoric acid, or sodium or potassium ortho- phosphate, or sodium tripolyphosphate	Phosphate calculated as phosphorus not more than 0.5% for potato or wheat, and not more than 0.4% for other starches
Distarch phosphate	Esterification with sodium trimetaphosphate or phosphorus oxychloride	Phosphate calculated as phosphorus not more than 0.5% for potato and wheat, and not more than 0.4% for other starches

Phosphated distarch phosphate	Combination of treatments for Monostarch phosphate and Distarch phosphate	Phosphate calculated as phosphorus not more than 0.5% for potato and wheat, and not more than 0.4% for other starches
Acetylated distarch phosphate	Esterification with sodium trimetaphosphate or phosphorus oxychloride combined with esterification with acetic anhydride or vinyl acetate	Acetyl groups not more than 2.5%; phosphate calculated as phosphorus not more than 0.14% for potato and wheat, and 0.04% for other starches; and vinyl acetate not more than 0.1 mg/kg
Starch acetate	Esterification with acetic anhydride or vinyl acetate	Acetyl groups not more than 2.5%
Acetylated distarch adipate	Esterification with acetic anhydride and adipic anhydride	Acetyl groups not more than 2.5% and adipate groups not more than 0.135%
Hydroxypropyl starch	Etherification with propylene oxide	Hydroxypropyl groups not more than 7.0%; propylene chlorohydrin not more than 1 mg/kg
Hydroxypropyl distarch phosphate	Esterification with sodium trimetaphosphate or phosphorus oxychloride combined with etherification by propylene oxide	Hydroxypropyl groups not more than 7.0%; propylene chlorohydrin not more than 1 mg/kg; and residual phosphate calculated as phosphorus not more than 0.14% for potato and wheat, and not more than 0.04% for other starches

Starch sodium octenylsuccinate	Esterification with octenylsuccinic anhydride	Octenylsuccinyl groups not more than 3%; and residual octenylsuccinic acid not more than 0.3%
Oxidized Hydroxypropyl	Chlorine, as sodium hypochlorite, not to exceed 5.5% of dry starch; active	Propylene chlorohydrin not more than 5 mg/kg
Starch	oxygen obtained	
	from hydrogen peroxide, not to	
	exceed 0.45%; and propylene oxide,	
	not to exceed 25%	
Starch Aluminum	Octenyl succinic anhydride, not to	_
Octenyl Succinate	exceed 2%, and aluminium sulfate,	

	not to exceed 2%
Starch Sodium Succinate	Succinic anhydride, not to exceed 4% —
Distarchoxy Propanol	Acrolein not to exceed 0.6% —

## Methyl Cellulose

Chemical formula:	$[C_6H_7O_2(OH)_x(OCH_3)_y]_n$	Molecular weight:	
		Unsubstituted	162.14
		Monosubstituted	176.17
		Disubstituted	190.20
		Trisubstituted	204.23
		Low Polymer(n about 200)	
		High Polymer(n about 900)	40,000
			180,000

1.	Assay	: Not less than 25% and not more than 33% of methoxyl groups.
2.	Appearance	: White or off-white, odorless fine granules, filaments or powder.
3.	Transparency	: Prepare two solutions. (1) Dissolve 1 g of the sample in 100 mL of water at about 70°C.
		After stirring, cool while shaking and mixing. Then place the solution in a cold place
		until a homogeneous paste is formed. (2) Add water to 4 mL of 0.01 N sulfuric acid
		solution, 1 mL of dilute hydrochloric acid, 5 mL of ethanol to make a solution 50 mL,
		then add 2 mL of cesium chloride TS and shake, place for 10 minutes. The
		transparency of the solution (1) should be higher.
4.	Chloride	: Not more than 0.14% of Cl.
5.	Sulfate	: Not more than $0.024\%$ of SO <sub>4</sub> .
6.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 20 ppm (as Pb).
8.	Sodium carboxyme	thyl : Add 1 g of the sample to 100 mL of water at 70°C. After stirring, cool while
	cellulose	shaking and mixing, then place in a cold place to form a uniform paste. Take
		40 mL of the paste and heat it while shaking and mixing. After boiling, filter
		with an insulated funnel. After cooling, add 1 mL of copper sulfate TS to 3 mL
		of filtrate. No precipitation forms.
9.	Starch	: Add 1 drop of iodine TS to the filtrate obtained in "8. Sodium carboxymethyl
		cellulose". The solution not turn blue.
10.	Loss on drying	: Not more than 8% (105°C, 4 hr).
11.	Residue on	: Not more than 1% (105°C, 4 hr).
	· ·	

ignition

- 12. Category : Food Additives Category 12
- 13. Uses : Pasting Agent.

## Sodium Polyacrylate

1.	Appearance	: powder, odorless, tasteless.
2.	Transparency	: Transparency test method for "methyl cellulose" is applicable.
3.	Free alkali	: Dissolve 0.2 g of the sample in 60 mL of water, stir. Add 3 mL of calcium chloride TS,
		heat on water bath for about 20 minutes, filter after cool. Wash the residue on the
		filter paper with water, combine the washing solution with the filtrate, and add water
		to make it 100 mL. Add 2 drops of phenolphthalein TS to 50 mL of the solution. The
		solution should not turn red.
4.	Sulfate	: Not more than $0.49\%$ of SO <sub>4</sub> .
5.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
6.	Heavy metals	: Not more than 20 ppm (as Pb).
7.	Monomer	: Not more than 1%.
8.	Low complex	: Not more than 5%.
9.	Starch	: Add 10 mL of water to 0.1 g of the sample to form a paste. When adding 2 drops of
		iodine TS, it should not be blue.
10.	Loss on drying	: Dry it at 105°C for 4 hours: it loses not more than 10% of its weight.
11.	Residue on	: Not more than 76% (105°C, 4 h)
	ignition	
12.	Category	: Food Additives Category 12
13.	Uses	: Pasting Agent.

#### Carrageenan

Synonyms	Irish moss gelose (from Chondrus spp.); Eucheuman (from Eucheuma spp.); Iridophycan
	(from Iridaea spp.); Hypnean (from Hypnea spp.); Furcellaran or Danish agar (from
	Furcellaria fastigiata); INS No. 407.
Definition	A substance with hydrocolloid properties obtained from certain members of the class
	Rhodophyceae (red seaweeds).
	The principal commercial sources of carrageenans are the following families and genera
	of the class of <i>Rhodophyceae</i> :
	Furcellariacaea such as Furcellaria
	Gigartinaceae such as Chondrus, Gigartina, Iridaea
	Hypnaeceae such as Hypnea
	Phyllophoraceae such as Phyllophora, Gynmogongrus, Ahnfeltia
	Solieriaceae such as Eucheuma, Anatheca, Meristotheca.
	Carrageenan is a hydrocolloid consisting mainly of the ammonium, calcium, magnesium,
	potassium and sodium sulfate esters of galactose and 3, 6-anhydrogalactose
	polysaccharides. These hexoses are alternately linked $\alpha$ -1,3 and $\beta$ -1,4 in the copolymer.
	The relative proportions of cations existing in carrageenan may be changed during
	processing to the extent that one may become predominant.
	The prevalent polysaccharides in carrageenan are designated as kappa-, iota-, and
	lambda-carrageenan. Kappa-carrageenan is mostly the alternating polymer of D-
	galactose-4-sulfate and 3, 6-anhydro-D-galactose; iota-carrageenan is similar, except that
	the 3,6-anhydrogalactose is sulfated at carbon 2. Between kappa-carrageenan and iota-
	carrageenan there is a continuum of intermediate compositions differing in degree of
	sulfation at carbon 2. In lambda-carrageenan, the alternating monomeric units are mostly
	D-galactose-2-sulfate (1,3-linked) and D-galactose-2,6- disulfate (1,4-linked).
	Carrageenan is obtained by extraction from seaweed into water or aqueous dilute alkali.
	Carrageenan may be recovered by alcohol precipitation, by drum drying, or by
	precipitation in aqueous potassium chloride and subsequent freezing. The alcohols used

	during recovery and purification are restricted to methanol, ethanol, and isopropanol.
	Articles of commerce may include sugars for standardization purposes, salts to obtain
	specific gelling or thickening characteristics, or emulsifiers carried over from drum
	drying processes.
C.A.S. number	9000-07-1
Description	Yellowish or tan to white, coarse to fine powder that is practically odourless.
Characteristics	
Identification	
Solubility	Insoluble in ethanol; soluble in water at a temperature of about 80°C, forming a viscous
	clear or slightly opalescent solution that flows readily; disperses in water more readily if
	first moistened with alcohol, glycerol, or a saturated solution of glucose or sucrose in
	water.
Test for sulfate	Dissolve a 100-mg sample in 20 ml of water (with heating if necessary), and add 3 ml of
	barium chloride TS and 5 ml of hydrochloric acid, dilute TS; filter if a precipitate forms.
	Boil the solution or the filtrate for 5 min. A white, crystalline precipitate appears.
Test for galactose and	Galactose and 3, 6-anhydrogalactose should be present.
anhydrogalactose	
Identification of	Add 4 g of sample to 200 ml of water, and heat the mixture in a water bath at 80°C, with
hydrocolloid and	constant stirring, until dissolved. Replace any water lost by evaporation, and allow the
predominant type of	solution to cool to room temperature. It becomes viscous and may form a gel. To 50 ml
copolymer	of the solution or gel add 200 mg of potassium chloride, then reheat, mix well, and cool.
	A short textured ("brittle") gel indicates a carrageenan of a predominantly kappa type,
	and a compliant ("elastic") gel indicates a predominantly iota type. If the solution does
	not gel, the carrageenan is of a predominantly lambda type.
Infrared absorption	Passes test
Purity	
Loss on drying	Not more than 12% (105°C to constant weight)
pH	Between 8 and 11 (1 in 100 suspension)
Viscosity	Not less than 5 cp at 75°C (1.5% solution)

Sulfate	Not less than 15% and not more than 40% (as $SO_4^{2-}$ ) on the dried basis
Total ash	Not less than 15% and not more than 40% on the dried basis
Acid-insoluble ash	Not more than 1%
Acid-insoluble matter	Not more than 2%
Residual solvents	Not more than 0.1% of ethanol, isopropanol, or methanol, singly or in combination
Microbiological criteria	Initially prepare a 10 <sup>-1</sup> dilution by adding a 50-g sample to 450 ml of Butterfield's
	phosphate-buffered dilution water and homogenising the mixture in a high-speed
	blender.
	Total (aerobic) plate count: Not more than 5000 cfu/g
	Salmonella spp.: Negative per test
	E. coli: Negative in 1 g
Arsenic	Not more than 3 mg/kg
Lead	Not more than 5 mg/kg
Cadmium	Not more than 2 mg/kg
Mercury	Not more than 1 mg/kg
Category	Food additives category (12)
Functional uses	Pasting Agent.

#### Xanthan Gum

1.	Description	: A high molecular weight polysaccharide gum produced by a pure-culture fermentation
		of a carbohydrate with Xanthomonas campestris, purified by recovery with ethanol or
		isopropanol, dried and milled; contains D-glucose and D-mannose as the dominant
		hexose units, along with D-glucuronic acid and pyruvic acid, and is prepared as the
		sodium, potassium or calcium salt; its solutions are neutral.
2.	Appearance	: Cream-colored powder
3.	Assay	: Yields, on the dried basis, not less than 4.2% and not more than 5.4% of carbon dioxide
5.	1000	(CO2), corresponding to between 91.0% and 117.0% respectively of xanthan gum.
4.	Solubility	: Soluble in water; insoluble in ethanol.
т.	boluolinty	. Soluble in water, insoluble in chanoi.
5.	Identification	: To 300 mL of water, previously heated to 80°C and stirred rapidly with a mechanical
		stirrer in a 400-mL beaker, add, at the point of maximum agitation, a dry blend of 1.5 g
		of the sample and 1.5 g of carob bean gum. Stir until the mixture goes into solution, and
		then continue stirring for 30 min longer. Do not allow the water temperature to drop
		below 600 during stirring. Discontinue stirring, and allow the mixture to cool at room
		temperature for at least 2 h. A firm rubbery gel forms after the temperature drops below
		40oC, but no such gel forms in a 1% control solution of the sample prepared in the same
		manner but omitting the carob bean gum.
6.	Loss on drying	: Dry it at 105°C for 2.5 hours: it loses not more than 15% of its weight.
7.	Ash	: Not more than 16%
8.	Pyruvic acid	: Not less than 1.5%
9.	Nitrogen	: Not more than 1.5%. Proceed according to the Kjeldahl method.
10.	Ethanol and	: Not more than 500 mg/kg, singly or in combination.
	isopropanol	
11.	Lead	: Not more than 2 ppm.
12.	Total plate count	: Not more than 5000 cfu/g.
13.	E. coli	: Negative by test.
14.	Salmonella	: Negative by test.
15.	Yeasts and	: Not more than 500 cfu/g.
	moulds	

- 16. Category : Food Additives Category 12
- 17. Uses : Pasting Agent.

## Alginic Acid

Chemical formula: (C <sub>6</sub> H <sub>8</sub> O <sub>6</sub> ) <sub>n</sub>		O <sub>6</sub> ) <sub>n</sub> Formula weight: 176.13 (theoretical)
		200.00 (actual average)
1.	Description	: Alginic acid is a naturally occurring hydrophilic colloidal polysaccharide obtained from
		the various species of brown seaweed (Phaeophyceae). It is a linear copolymer consisting
		mainly of residues of b -1,4-linked D-mannuronic acid and a -1,4-linked L-glucuronic
		acid. These monomers are often arranged in homopolymeric blocks separated by regions
		approximating an alternating sequence of the two acid monomers. Occurs as white to
		yellowish brown filamentous, grainy, granular or powdered forms. Odorless and
		tasteless. Insoluble in water and organic solvents, but soluble in alkaline solution. pH of
		its (3 ess and tasteless.s of brown seaweed (P
2.	Identification	: A. To 5 mL of 1 in 150 solution of the sample add 0.1 N sodium hydroxide. A
		voluminous, gelatinous precipitate is formed.
		B.Add 1 mL of dilute sulfuric acid TS to 5 mL of the solution obtained in test A to form
		a viscous gelatinous precipitate.
		C. Put 5 mg of the sample in a test tube, add 5 mL of water, 1 mL of freshly dispensed
		naphtholresorcinol/ethanol solution and 5 mL of hydrochloric acid (1 $\rightarrow$ 100), mix and
		boil for 3 minutes, then cool to about 15°C. Transfer the contents of this tube to a 30 mL
		separatory funnel, and rinse the tube with 5 mL of water. Then extract with 15 mL of
		isopropyl ether. Compared with the blank test, the isopropyl ether extract layer of the
		experimental case has a deeper purple color.
3.	Assay	: Yields, on the dried basis not less than 20.0% and not more than 23.0% of carbon
		dioxide (CO <sub>2</sub> ).
4.	Ash	: Not more than 4% on the dried basis
5.	Arsenic	: Not more than 3 ppm (as As).
6.	Lead	: Not more than 10 ppm.
7.	Heavy metals	: Not more than 0.004% (as Pb).
8.	Loss on drying	: Not more than 15%.
9.	Category	: Food Additives Category 12
10.	Uses	: Pasting Agent.

## Potassium Alginate (Algin)

Chemical formula: (C <sub>6</sub> H <sub>7</sub> O <sub>6</sub> K) n		O <sub>6</sub> K) <sub>n</sub> Formula weight: 241.22 (theoretical); 238.00
		(actual average)
1.	Description	: The sample is a potassium salt of alginic acid (see the specification of alginic acid),
		which is white to yellowish fibrous or granular powder, nearly odorless and tasteless.
		The sample forms a viscous colloidal solution when dissolved in water. Insoluble in
		alcohol and solution having more than 30% of alcohol. The sample is also insoluble
		in chloroform, ether and acid with a pH below 3.
2.	Identification	: A. To 5 mL of 1 in 100 solution of the sample add 1 mL of calcium chloride TS. A
		voluminous, gelatinous precipitate is formed.
		B.To 10 mL of 1 in 100 solution of the sample add 1 mL of dilute sulphuric acid TS. A
		voluminous, gelatinous precipitate is formed.
		C. Same as the identification test C of alginic acid.
		D. Dip the "ash" of the sample with dilute hydrochloric acid TS, filter. The filtrate
		passes test for potassium.
3.	Assay	: Yields, on the dried basis not less than 16.5% and not more than 19.5% of carbon
		dioxide (CO <sub>2</sub> ).
4.	Ash	: Not more than $22 \sim 33\%$ on the dried basis.
5.	Arsenic	: Not more than 3 ppm (as As).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Loss on drying	: Not more than 15%.
8.	Category	: Food Additives Category 12
9.	Uses	: Pasting Agent.

## Calcium Alginate (Algin)

Chemical formula: $(C_6H_7O_6)_2Ca$ ] n		$H_7O_6)_2Ca ]_n$	Formula weight: 195.16 (theoretical); 219.00
		(actua	al average)
1.	Description	: The sample is a calcium salt of algin	ic acid (see the specification of alginic acid), which
		is white to slightly yellowish fibro	us or granular powder, nearly odorless and
		tasteless. Insoluble in water and or	ganic solvents, but soluble in alkaline solutions or
		solutions of compounds that bind t	to calcium.
2.	Identification	: A. Same as the identification test C o	of alginic acid.
		B. Dip the "ash" of the sample with c	dilute hydrochloric acid TS, filter. The filtrate
		passes test for potassium.	
3.	Assay	: Not more than $18 \sim 21\%$ on the dried	l basis.
4.	Ash	: Not more than $12 \sim 18\%$ on the dried	l basis.
5.	Arsenic	: Not more than 3 ppm (as As).	
6.	Lead	: Not more than 10 ppm.	
7.	Heavy metals	: Not more than 0.004% (as Pb).	
8.	Loss on drying	: Not more than 15%.	
9.	Category	: Food Additives Category 12	
10.	Uses	: Pasting Agent.	

## Ammonium Alginate(Algin)

Chemical formula: (C <sub>6</sub> H <sub>7</sub> O <sub>6</sub> NH <sub>4</sub> ) <sub>n</sub>			Formula weight: 193.16(theoretical); 217.00 (actual
			average)
1.	Description	: Ammonium alginate is the an	nmonium salt of alginic acid. Occurs as white to yellowish
		brown filamentous, grainy, gra	nular or powdered forms. Dissolves slowly in water
		forming a viscous solution; ins	oluble in alcohol and aqueous solutions with an alcohol
		content of more than 30%, chlo	proform, ether and acid with pH below 3.
2.	Identification	: A. To 5 mL of 1 in 100 soluti	on of the sample add 1 mL of calcium chloride TS. A
		voluminous, gelatinous precipi	tate is formed.
		B. To 10 mL of 1 in 100 solut	tion of the sample add 1 mL of dilute sulphuric acid TS. A
		voluminous, gelatinous precipi	tate is formed.
		C. Same as the identification	test C of alginic acid.
		D. Heat 1 g of the sample wit	h 5 mL of sodium hydroxide TS in a tube. Swirl the
		mixture, and the odor o	f ammonia appears.
3.	Assay	: Not more than $18 \sim 21\%$ on the second se	he dried basis.
4.	Ash	: Not more than 4% on the drie	ed basis
5.	Arsenic	: Not more than 3 ppm (as As).	
6.	Lead	: Not more than 10 ppm.	
7.	Heavy metals	: Not more than 0.004% (as Pb	).
8.	Loss on drying	: Not more than 15%.	
9.	Category	: Food Additives Category 12	
10.	Uses	: Pasting Agent.	

## Hydroxypropyl Cellulose

Hydroxypropyl Methylcellulose

(Propylene Glycol Ether of Methylcellulose)

Polydextrose

#### Curdlan

1.	Description	: Curdlan is produced by fermentation from <i>Alcaligemes faecalis var. myxogenes</i> .
		Odorless or almost odorless, white to nearly white powder.
		Adhesive strength should be in above of 500 g/cm.
2.	Identification	: (1) Heat a 2% aqueous suspension of the sample (produced by adding 10 mL water to
		200 mg of the sample) in a boiling water bath for 10 min and cool. A firm gel forms.
		(2) Add 10 mL of water to 200 mg of the sample to form a suspension. Then add 10 mL
		of 10 N sulfuric acid, and heat in a boiling water bath for 20 minutes. After cooling,
		neutralize with cesium carbonate, centrifuge. Take 1 mL of the upper layer, and add 2
		mL of Fehling's test solution to it. Reddish brown cuprous oxide precipitate forms after
		heating.
3.	pН	: 6.0~8.5. 6.0 - 8.5 (1% aqueous suspension for 30 min)
4.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
5.	Heavy metals	: Not more than 20 ppm (as Pb).
6.	Loss on drying	: Not more than 10% (Not more than 10 mmHg, 60°C, 5 h).
7.	Residue on	: Not more than 7%.
	ignition	
8.	ignition Category	: Food Additives Category 12

#### Gellan Gum

Synonyms	INS No. 418
Definition	Gellan gum is a high molecular weight polysaccharide gum produced by a pure culture
	fermentation of a carbohydrate by Pseudomonas elodea, purified by recovery with
	isopropyl alcohol, dried, and milled. The high molecular weight polysaccharide is
	principally composed of a tetrasaccharide repeating unit of one rhamnose, one
	glucuronic acid, and two glucose units, and is substituted with acyl (glyceryl and
	acetyl) groups as the O-glycosidically-linked esters. The glucuronic acid is neutralized
	to a mixed potassium, sodium, calcium, and magnesium salt. It usually contains a small
	amount of nitrogen containing compounds resulting from the fermentation procedures.
CAS NUMBER	71010-52-1
Molecular weight	Approximately 500,000.
Assay	Yields, on the dried basis, not less than 3.3% and not more than 6.8% of carbon dioxide
	(CO2).
Appearance	Off-white powder.
Characteristics	
Identification	
Solubility	Soluble in water, forming a viscous solution; insoluble in ethanol.
Gel test with calcium	Add 1.0 g of the sample to 99 mL of water, and stir for about 2 h, using a motorized
ion	stirrer having a propeller-type stirring blade. Draw a small amount of this solution into
	a wide bore pipet and transfer into a 10% solution of calcium chloride. A tough worm-
	like gel will be formed immediately.
Gel test with sodium	To the 1% solution of the sample prepared for the previous test, add 0.50 g of sodium
ion	chloride, heat to $80^{\circ}$ with stirring, and hold at $80^{\circ}$ for 1 min. Allow the solution to cool
	to room temperature. A firm gel is formed.
Purity	
Loss on drying	Dry it at 105°C for 2.5 hours: it loses not more than 15% of its weight.
Residual solvent	Not more than 50 mg/kg of ethanol; not more than 750 mg/kg of isopropyl alcohol.
Nitrogen	Not more than 3%.
Microbiological	Total plate count: Not more than 10,000 colonies per gram.
criteria	E. coli: Negative by test.
	Salmonella: Negative by test.
	Yeasts and moulds: Not more than 400 colonies per gram.

LeadNot more than 2 mg/kg.CategoryFood Additives Category 12UsesPasting Agent.

Gelatinized Starch (Alkaline Treated Starch)

## Hydroxypropyl Distarch Phosphate

## Oxidized Hydroxypropyl Starch

Bleached Starch

Oxidized Starch

Starch Acetate

## Acetylated Distarch Adipate

#### Monostarch Phosphate

Starch Sodium Octenyl Succinate

#### Distarch Phosphate

#### Phosphated Distarch Phosphate

## Acetylated Distarch Phosphate

## Hydroxypropyl Starch

## Starch Aluminum Octenyl Succinate

### Starch Sodium Succinate

### Distarchoxy Propanol

# Pectins

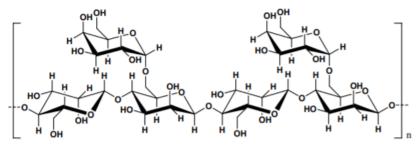
SYNONYMS	INS No. 440
Definition	Consists mainly of the partial methyl esters of polygalacturonic acid and
	their sodium, potassium, calcium and ammonium salts; obtained by
	extraction in an aqueous medium of appropriate edible plant material,
	usually citrus fruits or apples; no organic precipitants shall be used other
	than methanol, ethanol and isopropanol; in some types a portion of the
	methyl esters may have been converted to primary amides by treatment
	with ammonia under alkaline conditions. Sulfur dioxide may be added as
	a preservative.
C.A.S. number	9000-69-5
DESCRIPTION	White, yellowish, light greyish or light brownish powder.
Characteristics	
IDENTIFICATION	
Test for pectins	Passes test.
Test for amide group	Passes test (amidated pectins only).
Loss on drying	Not more than 12% (105°C, 2 h).
Sulfur dioxide	Not more than 50 mg/kg.
Residual solvents	Not more than 1% methanol, ethanol and 2-propanol, singly or in
	combination.
Acid-insoluble ash	Not more than 1%.
Total insolubles	Not more than 3%.
Nitrogen content	Not more than 2.5% after washing with acid and ethanol.
Galacturonic acid	Not less than 65% calculated on the ash-free and dried basis.
Degree of amidation	Not more than 25% of total carboxyl groups of pectin.
Lead	Not more than 2 mg/kg.
Category	Food additives category (12) (16).
Functional uses	Pasting Agent; Emulsifiers.

### Guar Gum

SYNONYMSGum cyamopsis, guar flour; INS No. 412DefinitionPrimarily the ground endosperm of the seeds from Cyamopsis<br/>tetragonolobus (L.) Taub. (Fam. Leguminosae) mainly consisting of high<br/>molecular weight (50,000-8,000,000) polysaccharides composed of<br/>galactomannans; the mannose:galactose ratio is about 2:1. The seeds are<br/>crushed to eliminate the germ, the endosperm is dehusked, milled and<br/>screened to obtain the ground endosperm (native guar gum). The gum<br/>may be washed with ethanol or isopropanol to control the microbiological<br/>load (washed guar gum).

C.A.S. number 9000-30-0

Structural formula



White to yellowish-white, nearly odourless, free-flowing powder.

#### **Characteristics**

DESCRIPTION

IDENTIFICATION

Solubility Insoluble in ethanol.

Gel formation Add small amounts of sodium borate TS to an aqueous dispersion of the sample; a gel is formed.

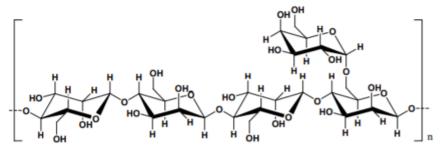
ViscosityTransfer 2 g of the sample into a 400-ml beaker and moisten thoroughly<br/>with about 4 ml of isopropanol. Add 200 ml of water with vigorous<br/>stirring until the gum is completely and uniformly dispersed. An<br/>opalescent, viscous solution is formed. Transfer 100 ml of this solution<br/>into another 400-ml beaker, heat the mixture in a boiling water bath for

	about 10 min and cool to room temperature. There is no substantial
	increase in viscosity (differentiating guar gums from carob bean gums).
Gum constituents	Use galactose and mannose as reference standards. These constituents
	should be present.
Microscopic	Place some ground sample in an aqueous solution containing 0.5% iodine
examination	and 1% potassium iodide on a glass slide and examine under a
	microscope. Guar gum shows close groups of round to pear formed cells,
	their contents being yellow to brown.
PURITY	
Loss on drying	Not more than 15.0% (105°C, 5 h).
Borate	Disperse 1 g of the sample in 100 ml of water. The dispersion should
	remain fluid and not form a gel on standing. Mix 10 ml of dilute
	hydrochloric acid with the dispersion, and apply one drop of the resulting
	mixture to turmeric paper. No brownish red colour is formed.
Total ash	Not more than 1.5% (800°C, 3-4 h).
Acid-insoluble matter	Not more than 7.0%.
Protein	Not more than 10.0%.
Residual solvents	Not more than 1% of ethanol or isopropanol, singly or in combination.
Lead	Not more than 2 mg/kg.
Category	Food additives category (12) (16).
Functional uses	Pasting Agent; Emulsifiers.

#### **Carob Bean Gum**

SYNONYMS	Locust bean gum, INS No. 410
Definition	Carob bean gum, also known as locust bean gum, is a galactomannan
	polysaccharide obtained from the seeds of Ceratonia siliqua (L.) Taub.
	(Fam. Leguminosae). The ground endosperm of the seeds consists mainly
	of high molecular weight (approximately 50,000-3,000,000)
	polysaccharides composed of galactomannans with a mannose:galactose
	ratio of about 4:1. The seeds are dehusked by treating the seeds with
	dilute sulfuric acid or with thermal mechanical treatments, elimination of
	the germ followed by milling and screening of the endosperm to obtain
	native carob bean gum. The gum may be washed with ethanol or
	isopropanol to control the microbiological load (washed carob bean gum).
C.A.S. number	9000-40-2

Structural formula



### DESCRIPTION

White to yellowish white, nearly odourless powder.

### **Characteristics**

**IDENTIFICATION** 

Solubility Insoluble in ethanol.

Gel formation Add small amounts of sodium borate TS to an aqueous dispersion of the sample; a gel is formed.

Viscosity Transfer 2 g of the sample into a 400-mL beaker and moisten thoroughly with about 4 mL of isopropanol. Add 200 mL of water with vigorous stirring until the gum is completely and uniformly dispersed. An opalescent, slightly viscous solution is formed. Transfer 100 mL of this

solution into another 400-mL beaker. Heat the mixture in a boiling water bath for about 10 min and cool to room temperature. There is an appreciable increase in viscosity (differentiating carob bean gums from guar gums).

Gum constituents Use galactose and mannose as reference standards. These constituents should be present.

MicroscopicDisperse a sample of the gum in an aqueous solution containing 0.5%examinationiodine and 1% potassium iodide on a glass slide and examine under a<br/>microscope. Carob bean gum contains long stretched tubiform cells,<br/>separated or slightly interspaced. Their brown contents are much less<br/>regularly formed than in Guar gum.

### PURITY

Loss on drying	Not more than 14% (105°C, 5 h).
Total ash	Not more than 1.2% (800°C, 3-4 h).
Acid-insoluble matter	Not more than 4.0%.
Protein	Not more than 7.0%.
Starch	To a 1 in 10 dispersion of the sample add a few drops of iodine TS; no
	blue colour is produced.
Residual solvents	Not more than 1% of ethanol or isopropanol, singly or in combination.
Lead	Not more than 2 mg/kg.
Category	Food additives category (12) (16).
Functional uses	Pasting Agent; Emulsifiers.

#### Category 13 Coagulating Agents

#### § 13002

5.

#### Tetrasodium Pyrophosphate, Decahydrate

Sodium Pyrophosphate

#### Tetrasodium Diphsophate, Decahydrate

Molecular weight: 446.06

Chemical formula: Na<sub>4</sub>P<sub>2</sub>O<sub>7</sub>·10H<sub>2</sub>O

Assay : Not less than 95% of Na<sub>4</sub>P<sub>2</sub>O<sub>7</sub> on the dried basis.
 Description : Colorless or white crystals, or a white crystalline or granular powder. Soluble in water;

insoluble in ethanol.

- 3. pH : 9.9 10.7 (1 in 100 soln)
- Phosphate : Dissolve 1 g of the sample in 100 mL of water. Add a few drops of silver nitrate TS to
   1 mL of the solution. No obvious yellow precipitate should be formed. If a white or
   yellowish precipitate is formed, the precipitate can be redissolved in dilute nitric acid.
  - Sodium : The 1 in 20 solution of the sample passes test for sodium.
- 6. Fluoride : Not more than 10 mg/kg.
- 7. Arsenic : Not more than 3 mg/kg (as As).
- 8. Lead : Not more than 5 mg/kg.
- 9. Heavy metals : Not more than 10 mg/kg (as Pb).
- 10. Loss on drying : Not more than 38~42% (dried at 105°C for 4 hours, and then dried at 800°C for 30 minutes)

,

- 11.Water-insoluble matter: Not more than 0.2%.
- 12. Category : Food Additives Category 13
- 13. Uses : Coagulating agent

Tetrasodium Pyrophosphate, Anhydrous

Sodium Pyrophosphate, Anhydrous

Tetrasodium Diphsophate, Anhydrous

Chen	nical formula: Na <sub>4</sub> P <sub>2</sub>	D <sub>7</sub> Molecular weight: 265.94
1.	Assay	: Not less than 95% of $Na_4P_2O_7$ on the dried basis
2.	Description	: Colorless or white crystals, or a white crystalline or granular powder. Soluble in water;
		insoluble in ethanol.
3.	pH	: 9.9 - 10.7 (1 in 100 soln)
4.	Phosphate	: Dissolve 1 g of the sample in 100 mL of water. Add a few drops of silver nitrate TS to 1
		mL of the solution. No obvious yellow precipitate should be formed. If a white or
		yellowish precipitate is formed, the precipitate can be redissolved in dilute nitric acid.
5.	Sodium	: The 1 in 20 solution of the sample passes test for sodium.
6.	Fluoride	: Not more than 10 mg/kg.
7.	Arsenic	: Not more than 3 mg/kg (as As).
8.	Lead	: Not more than 5 mg/kg.
9.	Heavy metals	: Not more than 10 mg/kg (as Pb).
10.	Loss on drying	: Not more than 0.5% (dried at 105°C for 4 hours, and then dried at 800°C for 30
		minutes)
11.	Water-insoluble m	atter : Not more than 0.2%.
12.	Category	: Food Additives Category 13
13.	Uses	: Coagulating agent

## Potassium Polyphosphate

## Sodium Polyphosphate

### Potassium Metaphosphate

### Sodium Metaphosphate

## Potassium Dihydrogen Phosphate

### Sodium Phosphate, Monobasic

Sodium Phosphate, Monobasic, Anhydrous

### Potassium Phosphate, Dibasic

# Sodium Phosphate, Dibasic

Sodium Phosphate, Dibasic, Anhydrous

## Potassium Phosphate, Tribasic

## Trisodium Phosphate

Sodium Phosphate, Tribasic, Anhydrous

# Category 14 Chemicals for food industry

## § 14001

# Sodium Hydroxide

Chemical formula: NaOH	Molecular weight: 40.00
1. Assay	: Not less than 95%.
2. Appearance	: White or nearly white pellets, flakes, sticks, fused masses or other forms.
3. Solution	: Dissolve 50 g of the sample in 250 mL of fresh boiled and cooled water. Mix 5 mL of
	the solution with 20 mL of water. The solution should be colorless and the turbidity
	should be below "almost clear".
4. Sulfate	: Not more than $0.2\%$ of SO <sub>4</sub> .
5. Sodium	: Not more than 2%.
carbonate	
6. Arsenic	: Not more than 3 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7. Heavy metals	: Not more than 30 ppm (as Pb).
8. Mercury	: Not more than 0.1 ppm.
9. Potassium	: When the flame reaction is viewed from cobalt glass, it should not be continuous
	purple.
10. Category	: Food Additives Category 14
11. Uses	: Chemicals for food industry

## Potassium Hydroxide

Chemical formula: KOH		Molecular weight: 56.11
1.	Assay	: Not less than 85.0% of total alkali calculated as KOH.
2.	Appearance	: White or nearly white pellets, flakes, sticks, fused masses or other forms.
3.	Solubility	: Very soluble in water; freely soluble in ethanol
4.	Test for alkali	: A 1 in 100 solution of the sample is strongly alkaline.
5.	Water-insoluble	: A 1 in 20 solution of the sample is complete, clear, and colorless.
	matter	
6.	Carbonate	: Not more than $3.5\%$ of $K_2CO_3$ .
7.	Lead	: Not more than 2 mg/kg.
8.	Category	: Food Additives Category 14
9.	Uses	: Chemicals for food industry

### Sodium Hydroxide Solution

1.	Assay	: The sample contains 95% to 120% of indicated NaOH content.
2.	Appearance	: Colorless or slightly colored liquid.
3.	Solution	: Add freshly boiled and cooled water to the sample to make 20 w/v% solution of NaOH
		according to the indicated amount. Mix 5 mL of the solution and 20 mL of water. The
		solution should be colorless and the turbidity should be below "almost clear".
4.	Sulfate	: Not more than $0.2\%$ of SO <sub>4</sub> .
5.	Sodium	: Not more than 2%.
	carbonate	
6.	Arsenic	: Not more than 3 ppm (as As <sub>2</sub> O <sub>3</sub> ).
7.	Heavy metals	: Not more than 30 ppm (as Pb).
8.	Potassium	: When the flame reaction is viewed from cobalt glass, it should not be continuous purple.
9.	Mercury	: Not more than 0.1 ppm.
10.	Category	: Food Additives Category 14
11.	Uses	: Chemicals for food industry

## Hydrochloric Acid

Chem	ical formula: HCl	Molecular weight: 36.46
1.	Assay	: The sample contains 90% to 120% of indicated HCl content.
2.	Appearance	: Colorless to pale yellow liquid, with a stimulating odor.
3.	Sulfate	: Not more than 0.012 w/v% of SO <sub>4</sub> .
4.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
5.	Heavy metals	: Not more than 5 ppm (as Pb).
6.	Mercury	: Not more than 0.1 ppm.
7.	Iron	: Not more than 30 ppm.
8.	Residue on	: Add 2 drops of sulfuric acid to 100 g of the sample. When it is evaporated to a constant
	ignition	amount on a sand bath, the amount of residue should be not more than 20 mg.
9.	Category	: Food Additives Category 14
10.	Uses	: Chemicals for food industry

### Sulfuric Acid

Chem	ical formula: H <sub>2</sub> SO <sub>4</sub>	Molecular weight: 98.08
1.	Assay	: Not less than 94%
2.	Appearance	: Clear, colorless or slightly brown, very corrosive oily liquid.
3.	Chloride	: Not more than 50 ppm of Cl.
4.	Nitric acid	: Not more than 20 ppm.
5.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
6.	Heavy metals	: Not more than 10 ppm (as Pb).
7.	Iron	: Not more than 0.01%.
8.	Selenium	: Not more than 1 ppm.
9.	Readily	: Dissolve 4 g of the sample in 10 mL of water. After cooling, add 3 drops of 0.1 N $$
	oxidizable	potassium permanganate solution. The color should not disappear within 5 minutes.
	matter	
10.	Residue on	: Not more than 0.02%.
	ignition	
11.	Category	: Food Additives Category 14
12.	Uses	: Chemicals for food industry

### Oxalic Acid

Chemical formula: C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> ·2H <sub>2</sub> O		A-2H <sub>2</sub> O Molecular weight: 126.07
1.	Assay	: 99.5∼101.0%.
2.	Appearance	: Colorless crystal, odorless.
3.	Solution	: When 1 g of the sample is dissolved and boiled in 20 mL of water, the solution should
		be colorless and the turbidity should be below "almost clear".
4.	Sulfate	: Not more than $0.08\%$ of SO <sub>4</sub> .
5.	Arsenic	: Not more than 2 ppm (as $As_2O_3$ ).
6.	Heavy metals	: Not more than 20 ppm (as Pb).
7.	Residue on	: Not more than 13%.
	ignition	
8.	Category	: Food Additives Category 14
9.	Uses	: Chemicals for food industry

#### Ion-Exchange Resin

- : Black, brown, pale auburn or white ball, block or powder, almost odorless. Appearan 1
- ce
- : (1) Cation exchange resin: Take 25 mL of the sample into a chromatography tube with an inner 2 Solid
  - diameter of about 3 cm. After eluting with 4% hydrochloric acid at a rate of 15 to 20 mL per matter minute, wash with the purified water at the same rate until the chlorine content in 10 mL of the solution is less than that of 0.3 mL of 0.1 N hydrochloric acid. Use it as a standard sample (H type).
    - (2) Anion exchange resin: Take 25 mL of the sample into a chromatography tube with an inner diameter of about 3 cm. After eluting with 1,000 mL of 4% hydrochloric acid at a rate of 15 to 20 mL per minute, wash with the purified water at the same rate until the solution is not red to phenolphthalein TS. Use it as a standard sample (OH type).

Fully impregnate the standard sample with purified water and absorb attached

water by the filter paper. Weigh 10 g. If it is a cation exchange resin, dry at

100°C for 12 hours. If it is an anion exchange resin, dry in a vacuum dryer at

40°C and 30 mmHg for 12 hours. The amount should be not less than 25%.

- Water : Fully impregnate the standard sample with purified water and absorb attached 3
- soluble water by the filter paper. Weigh 10 g into a cylindrical filter paper with an inner diameter of 28 mm and a length of 100 mm, hang it in 1,000 mL of purified water, and shake it often for 5 hours. substance After extracting 50 mL of the extract, dry it at 110°C for 3 hours. The residue should be not more s than 0.5%. The same method must be used as a control test.
- Arsenic : Fully impregnate the standard sample with purified water and absorb attached 4 water by the filter paper. Put 2 g in the decomposition bottle, add 5 mL of sulfuric acid and 20 mL of nitric acid, slowly heat and add 2 to 3 mL of nitric acid until the liquid becomes colorless to pale yellow. After cooling, add 5 mL of saturated ammonium oxalate solution and heat to produce white smoke. After cooling, add water to make it 25 mL. Take 5 mL as the test liquid, and the arsenic content should not be more than 3 ppm.
- Total 5

: (1) Cation exchange resin: Fully impregnate the standard sample with purified water and absorb attached water by the filter paper. Accurately weigh 5 g, immerse it in 500 mL of 0.2 N exchange capacity sodium hydroxide solution and shake it often for 12 hours. Take 10 mL of clear liquid and titrate with 0.1 N sulfuric acid (3 drops of methyl orange TS as indicator). Perform a

controlled test. When calculating the total exchange capacity according to the following formula, the value should be 1.0 meq/g or more.

Total exchange capacity=

 $\frac{0.1N\,sulfate\,titration\,required\,for\,control\,experiments\,(mL)-\,0.1N\,sulfate\,titration\,required\,for\,control\,experimen}{Standard\,sample\,volume\,(g)*\frac{Solid\,matter(\%)}{100}}$ 

 $*5(\frac{meq}{g})$ 

- (2) Anion exchange resin: The test method and calculation are the same as "cation exchange resin". However, 0.2 N hydrochloric acid solution is used instead of 0.2 N sodium hydroxide solution, and 0.1 N sodium hydroxide solution is used instead of 0.1 N sulfuric acid solution.
- 6 Category : Food Additives Category 14
- 7 Uses : Chemicals for food industry

#### Potassium Carbonate

### Sodium Carbonate, Anhydrous

# Category 15 Carrier

# § 15001

# Propylene Glycol

Chen	nical formula: C <sub>3</sub> H <sub>8</sub> O	2 Molecular weight: 76.10
1.	Appearance	: Clear, colorless, hygroscopic, viscous liquid
2.	Density	: 1.036~1.040.
3.	<b>Boiling Point</b>	: 183∼195°C.
4.	Free acid	: Add 1 mL of phenol red TS to 50 mL water, then add 0.01N sodium hydroxide until
		solution remains red for 30 sec. To this solution add about 10 mL of the sample. Titrate
		with 0.2 m of 0.1 N sodium hydroxide until the original red color returns and remains for
		30 sec.
5.	Chloride	: Not more than 70 ppm of Cl.
6.	Heavy metals	: Not more than 5 ppm (as Pb).
7.	Glycerin and	: Accurately weigh 1 g of the sample and add water to make a 1000 mL solution.
	ethylene glycol	Add 0.2 g of potassium periodate, 1 mL of sulfuric acid and 50 mL of water to 13
		mL of the solution, and distill at a rate of 3 to 5 mL per minute until the residue is
		about 1 mL (the receiver of the distillate should be placed in iced water). The
		distillate is added with water to make it 500 mL. Add 0.1 g of chromic acid and 5
		mL of sulfuric acid to 1 mL of distillate solution, heat in a water bath for 30
		minutes, then cool. When adding water to make it 250 mL, the liquid color should
		not be thicker than 1 mL of formaldehyde standard solution undergone the same
		process.
8.	Residue on	: Not more than 0.05%.
	ignition	
9.	Category	: Food Additives Category 15
10.	Uses	: Carrier.

Glycerol

Glycerin Fatty Acid Ester

Synonyms	Mono- And Diglycerides: Glyceryl monostearate, glyceryl monopalmitate, glyceryl			
	monooleate, etc; monostearin, monopalmitin, monoolein, etc.; GMS (for glyceryl			
	monostearate); INS No. 471			
	Acetic and Fatty Acid Esters of Glycerol: Acetic acid esters of mono- and diglycerides,			
	acetoglycerides, acetylated mono- and diglycerides, INS No.472a			
Definition	Mono- And Diglycerides: A mixture of mono- and diglyceryl esters of long chain,			
	saturated and unsaturated fatty acids that occur in food fats; contain not less than 30% of			
	alpha-monoglycerides and may also contain other isomeric monoglycerides, as well as			
	di- and triglycerides, free glycerol, free fatty acids, soap and moisture; usually			
	manufactured by the glycerolysis of edible fats and oils, but may also be prepared by			
	esterification of fatty acids with glycerol, with or without molecular distillation of the			
	product.			
	Acetic and Fatty Acid Esters of Glycerol: Mixed glycerol esters of acetic acid and fatty			
	acids of food fats. Contains mono- and diesters of fatty acids with glycerol which is itself			
	partially acetylated; may also contain free glycerol and free fatty acids.			
Structural formula	al formula Mono- And Diglycerides:			
	alfa-mono-	beta-mono-	alfa, beta-di-	alfa, alfa-di-
	CH200CR	сн <sub>2</sub> он	CH2COCR	CH2OOCR
	снон	CHOOCR	CHOCCR	снон
	 Сн <sub>2</sub> он	 CH₂OH	CH2OH	I CH2OOCR
	where -OCR represents the fatty acid moiety			

Acetic and Fatty Acid Esters of Glycerol:

	CH2-OR1
	$CH - OR_2$
	$CH_2 - OR_1$   CH - OR_2   CH_2 - OR_3
	where R1, R2 and R3 each may be a fatty acid moiety, - $COCH_3$ or H
Formula weight	Mono- And Diglycerides:
	Glyceryl monostearate: 358.6
	Glyceryl distearate: 625.0
	These are two major components of commercial products
	Acetic and Fatty Acid Esters of Glycerol: -
Description	Mono- And Diglycerides: White or cream coloured hard fats of waxy appearance, plastic
	products or viscous liquids.
	Acetic and Fatty Acid Esters of Glycerol: From liquid to solid in consistency; white to
	pale yellow, may have the odour of acetic acid
	The article of commerce may be further specified as the saponification value, acid value,
	free fatty acid content, solidification point of the free fatty acids, Reichert-Meissl value,
	iodine value and free glycerol content.
Characteristics	
Identification	
Solubility	Mono- And Diglycerides: Insoluble in water; soluble in ethanol, chloroform and benzene Acetic and Fatty Acid Esters of Glycerol: Insoluble in water; soluble in ethanol
Infrared absorption	Mono- And Diglycerides: The infrared spectrum of the sample is characteristic of a
	partial fatty acid ester of a polyol
	Acetic and Fatty Acid Esters of Glycerol: -
Tests for fatty acids	Passes tests
Test for glycerol	Passes tests
Purity	
Water	Mono- And Diglycerides: Not more than 2.0% (Karl Fischer Method)
	Acetic and Fatty Acid Esters of Glycerol: -
	541

Acid value	Mono- And Diglycerides: Not more than 6
	Acetic and Fatty Acid Esters of Glycerol: -
Free glycerol	Mono- And Diglycerides: Not more than 7%
	Acetic and Fatty Acid Esters of Glycerol: -
Soap	Mono- And Diglycerides: Not more than 6%, calculated as a sodium oleate
	Add 10.00 g of the sample to a mixture of 60 ml of acetone and 0.15 ml of bromophenol
	blue solution (0.5%), previously neutralized with 0.1 N hydrochloric acid or 0.1 N
	sodium hydroxide. Warm gently on a water bath until solution is complete, and titrate
	with 0.1 N hydrochloric acid until the blue colour is discharged. Allow to stand for 20
	min, warm until any solidified matter has re-dissolved and, if the blue colour reappears,
	continue the titration. Each ml of 0.1 N hydrochloric acid is equivalent to 0.0304 g of
	$C_{18}H_{33}O_2Na.$
	Acetic and Fatty Acid Esters of Glycerol: -
Lead	Not more than 2 mg/kg
Category	Food additives category (16)
Functional uses	Emulsifiers

## Sucrose Fatty Acid Ester

1.	Appearance	: Stiff gels, soft solids or white to slightly greyish white powders.
2.	Acid value	: Not more than 5.
3.	Nitrogen	: Not more than 0.02%.
4.	Arsenic	: Not more than 1 ppm (as As <sub>2</sub> O <sub>3</sub> ).
5.	Heavy metals	: Not more than 20 ppm (as Pb).
6.	Free sucrose	: Not more than 10%.
7.	Loss on drying	: Not more than 4% (vacuum desiccator over sulfuric acid, 4 hr).
8.	Residue on	: Not more than 1.5%.
	ignition	
9.	Category	: Food Additives Category 16
10.	Uses	: Emulsifier

### Sorbitan Fatty Acid Ester

1.	Appearance	: White to tan liquid or wax.
2.	Acid value	: Not more than 14.
3.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
4.	Heavy metals	: Not more than 20 ppm (as Pb).
5.	Polyoxyethylene	: Heat and oscillate 1 g of the sample and 20 mL of water. After cooling, add ammonium
		thiocyanate, 10 mL of cobalt nitrate TS, and mix thoroughly by shaking. Add 10 mL
		of chloroform, shake and mix. The chloroform layer should not be blue when placed.
6.	Residue on	: Not more than 1.5%.
	ignition	
7.	Category	: Food Additives Category 16
8.	Uses	: Emulsifier

### Propylene Glycol Fatty Acid Ester

1.	Appearance	: White or cream colored solids of waxy appearance, plastic products or viscous liquids.
		Odorless and tasteless.
2.	Acid value	: Not more than 8.
3.	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
4.	Heavy metals	: Not more than 20 ppm (as Pb).
5.	Polyoxyethylene	: The test method for "5.Polyoxyethylene" of "fatty acid sorbitan ester" is applicable.
6.	Residue on	: Not more than 1.5%.
	ignition	
7.	Category	: Food Additives Category 16
8.	Uses	: Emulsifier

Diace	tyl Tartaric Acid Esters of Mono- and Diglycerides
Synonyms	Diacetyltartaric acid esters of mono- and diglycerides; DATEM;
	Tartaric, acetic and fatty acid esters of glycerol, mixed; Mixed acetic
	and tartaric acid esters of mono and diglycerides of fatty acids; INS
	No. 472e.
Definition	The product consists of mixed glycerol esters of mono- and
	diacetyltartaric acid and fatty acids from edible fats and oils. It is
	made by the interaction of diacetyltartaric anhydride and monoand
	diglycerides of fatty acids in the presence of acetic acid, or by
	interaction of acetic anhydride and mono- and diglycerides of fatty
	acids in the presence of tartaric acid.
	Because of inter- and intramolecular acyl-group exchange, both
	methods of production lead to the same essential components, the
	distribution of which depends on the relative proportions of the
	basic raw materials, on temperature, and on reaction time. The
	product may contain small amounts of free glycerol, free fatty acids,
	and free tartaric and acetic acids.
C.A.S. numbers	308068-42-0
	100085-39-0
Structural formula	$CH_2 - OR_1$   $CH - OR_2$
	CH2-OR3
	in which one or two of the R groups is a fatty acid moiety and the
	other R groups are either:
	- diacetylated tartaric acid moiety
	- monoacetylated tartaric acid moiety
	- tartaric acid moiety
	- acetic acid moiety
	- hydrogen
Description	Liquid, paste, or wax-like solid
Characteristics	
Identification	

Solubility	Dispersible in cold and hot water; soluble in methanol, ethanol,
	acetone, and ethyl acetate.
1,2-diols	To a solution of 500 mg in 10 ml methanol, add dropwise, lead
	acetate TS. A white, flocculent, insoluble precipitate is formed.
Fatty acids	Passes test
Acetic acid	Passes test
Tartaric acid	Passes test
Glycerol	Passes test
Purity	
Acids	Acids other than acetic, tartaric and fatty acids, shall not be
	detectable.
Sulfated ash	Not more than 0.5% (determined at 800±25°C)
Acid value	40-130
Total acetic acid	8%-32% after hydrolysis
Total tartaric aci	d 10%-40% after saponification
Total glycerol	11%-28% after saponification
Free glycerol	Not more than 2.0%
Lead	Not more than 2 mg/kg
Category	Food Additives Category 8
Uses	Nutritional additives

#### Sodium Aluminum Phosphate, Basic

1.	Description	: An autogenous mixture of an alkaline sodium aluminum phosphate (approximately
		Na <sub>8</sub> Al <sub>2</sub> (OH) <sub>2</sub> (PO <sub>4</sub> ) <sub>4</sub> ) with about 30% dibasic sodium phosphate. White, odorless powder.
		Soluble in hydrochloric acid; the sodium phosphate moiety is soluble in water, whereas
		the sodium aluminum phosphate moiety is only sparingly soluble in water.
2.	Identification	: Test a 1 in 10 solution in dilute hydrochloric acid (1 in 2). Passes test for sodium,
		aluminum and phosphate.
3.	Assay	: Not less than 9.5% and mot more than 12.5% of Al2O3, on the ignited basis.
4.	Fluoride	: Not more than 0.0025%.
5.	Arsenic	: Not more than 3 ppm (as As).
6.	Lead	: Not more than 10 ppm.
7.		
1.	Heavy metals	: Not more than 0.004% (as Pb).
7. 8.	Heavy metals Loss on ignition	<ul><li>: Not more than 0.004% (as Pb).</li><li>: Not more than 9%.</li></ul>
	•	

### Polysorbate 20

## (Polyoxyethylene (20) Sorbitan Monolaurate)

1.	Assay	: Not less than 70.0 and not more than 74.0% of oxyethylene groups (-C2H4O-),
		equivalent to not less than 97.3 and not more than 103.0% of polyoxyethylene (20)
		sorbitan monolaurate calculated on the anhydrous basis.
2.	Description	: Consists of a mixture of the partial esters of sorbitol and sorbitol anhydride with lauric
		acid and condensed with approximately 20 moles of ethylene oxide (C2H4O) per mole of
		sorbitol and its anhydrides.
		Lemon to amber colored oily liquid, with a faint characteristic odor and bitter taste.
		Soluble in water, methanol, ethanol, ethyl acetate and dioxane; insoluble in mineral oil,
		mineral essential oil and petroleum ether.
3.	Identification	: To 5 mL of a 5% (w/v) aqueous solution of the sample add 5 mL sodium hydroxide TS.
		Boil for a few min, cool, and acidify with dilute hydrochloric acid. The solution is
		strongly opalescent.
4.	Acid value	: Not less than 2
5.	Hydroxy value	$:96 \sim 108.$
6.	Saponification value	e : $40 \sim 50$ .
7.	Lauric acid	: 15~17%.
8.	Water	: Not more than 3%.
9.	1, 4-dioxane	: Not more than 10 mg/kg.
10.	Arsenic	: Not more than 3 ppm (as As).
11.	Heavy metals	: Not more than 10 ppm (as Pb).
12.	Residue on	: Not more than 0.25%.
	ignition	
13.	Category	: Food Additives Category 16
14.	Uses	: Emulsifier

### Polysorbate 60

## (Polyoxyethylene (20) Sorbitan Monostearate)

1.	Assay	: Not less than 65.0 and not more than 69.5% of oxyethylene groups, equivalent to not
		less than 97.0 and not more than 103.0% of polyoxyethylene (20) sorbitan monostearate,
		on the anhydrous basis.
2.	Description	: Consists of a mixture of the partial esters of sorbitol and its mono- and dianhydrides
		(which have an acid value below 10 and a water content below $0.2\%$ ) with the stearic
		acid and condensed with approximately 20 moles of ethylene oxide ( $C_2H_4O$ ) per mole of
		sorbitol and its anhydrides. Yellow to orange colored oily liquid or semi-gel, with a faint
		characteristic odor and slightly bitter taste. Soluble in water, ethyl acetate, aniline and
		toluene; insoluble in mineral oil and vegetable oils.
3.	Identification	: (1) To 5 mL of a 5% (w/v) aqueous solution of the sample add 5 mL sodium hydroxide
		TS. Boil for a few min, cool, and acidify with dilute hydrochloric acid. The solution is
		strongly opalescent.
		(2) A mixture of 60 parts by volume of the sample and 40 parts of water yields a
		gelatinous mass at or below 25°C.
4.	Acid value	: Not less than 2
5.	Hydroxy value	: 81~96.
6.	Saponification value	ie : $45 \sim 55$ .
7.	Stearic acid	: 24~26%.
	and palmitic acid	. 24/~20/0.
8.	Water	: Not more than 3%.
9.	1, 4-dioxane	: Not more than 10 mg/kg.
10.	Arsenic	: Not more than 3 ppm (as As).
11.	Heavy metals	: Not more than 10 ppm (as Pb).
12.	Residue on	: Not more than 0.25%.
	ignition	. INOU IIIOFO UIDII 0.2070.
13.	Category	: Food Additives Category 16
14.	Uses	: Emulsifier

### Polysorbate 65

## (Polyoxyethylene (20) Sorbitan Tristearate)

1.	Assay	: Not less than 46.0 and not more than 50.0% of oxyethylene groups, equivalent to not
		less than 96.0 and not more than 104.0% of polyoxyethylene (20) sorbitan tristearate on
		the anhydrous basis.
2.	Description	: Consists of a mixture of the partial esters of sorbitol and sorbitol dianhydrides with
		stearic acid and condensed with approximately 20 moles of ethylene oxide (C2H4O) per
		mole of sorbitol and its anhydrides. Tan colored, waxy solid, with a faint characteristic
		odor and bitter taste. Dispersible in water and in carbon tetrachloride; soluble in mineral
		oil, vegetable oils, petroleum ether, acetone, ether, dioxane, ethanol and methanol.
3.	Identification	: To 5 mL of a 5% (w/v) aqueous solution of the sample add 5 mL sodium hydroxide TS.
		Boil for a few min, cool, and acidify with dilute hydrochloric acid. The solution is
		strongly opalescent.
4.	Acid value	: Not less than 2
5.	Hydroxy value	$:40\sim\!60.$
6.	Saponification valu	ie : $88 \sim 98$ .
7.	Stearic acid and pa	Imitic acid : $42 \sim 44\%$ .
8.	Water	: Not more than 3%.
9.	1, 4-dioxane	: Not more than 10 mg/kg.
10.	Arsenic	: Not more than 3 ppm (as As).
11.	Heavy metals	: Not more than 10 ppm (as Pb).
12.	Residue on	
	ignition	: Not more than 0.25%.
13.	Category	: Food Additives Category 16
14.	Uses	: Emulsifier

### Polysorbate 80

## (Polyoxyethylene (20) Sorbitan Monooleate)

1.	Assay	: Not less than 65.0 and not more than 69.5% of oxyethylene groups, equivalent to not
		less than 96.5 and not more than 103.5% of polyoxyethylene (20) sorbitan monooleate,
		calculated on the anhydrous basis.
2.	Description	: Consists of a mixture of the partial esters of sorbitol and sorbitol anhydride with oleic
		acid and condensed with approximately 20 moles of ethylene oxide (C <sub>2</sub> H <sub>4</sub> O) per mole of
		sorbitol and its anhydrides. Lemon to amber colored oily liquid, with a faint characteristic
		odor and a little bitter taste. Soluble in water, forming an odorless and almost colorless
		solution. Soluble in ethanol, methanol, ethyl acetate and toluene; insoluble in mineral oil
		and petroleum ether.
3.	Identification	: (1) To 5 mL of a 5% (w/v) aqueous solution of the sample add 5 mL sodium hydroxide
		TS. Boil for a few min, cool, and acidify with dilute hydrochloric acid. The solution is
		strongly opalescent.
		(2) To a solution of the sample (1 in 20) add bromine TS dropwise. The bromine is
		decolorized.
		(3) A mixture of 60 parts by volume of the sample and 40 parts of water yields a
		gelatinous mass at or below 25°C.
4.	Acid value	: Not less than 2
5.	Hydroxy value	: 65~80.
6.	Saponification valu	e : 45~55.
7.	Oleate	: 22~24%.
8.	Water	: Not more than 3%.
9.	1, 4-dioxane	: Not more than 10 mg/kg.
10.	Arsenic	: Not more than 3 ppm (as As).
11.	Heavy metals	: Not more than 10 ppm (as Pb).
12.	Residue on	: Not more than 0.25%.
	ignition	
13.	Category	: Food Additives Category 16
14.	Uses	: Emulsifier

### Hydroxypropyl Cellulose

Hydroxypropyl Methylcellulose

(Propylene Glycol Ether of Methylcellulose)

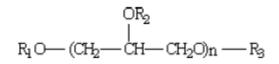
#### Polyglycerol Esters of Fatty Acids

where the average value of n is about 3 and R1, R2 and R3 each may be a fatty acid moiety or hydrogen.

- 1. Definition : Mixed partial esters formed by reacting polymerized glycerols with edible fats, oils, or fatty acids; minor amounts of mono-, di-, and triglycerides, free glycerol and polyglycerols, free fatty acids, and sodium salts of fatty acids may be present; degree of polymerization varies, and is specified by a number (such as tri-) that is related to the average number of glycerol residues per polyglycerol molecule. A specified polyglycerol consists of a distribution of molecular species characteristic of its nominal degree of polymerization. By varying the proportions as well as the nature of the fats or fatty acids to be reacted with the polyglycerols, a large and diverse class of products may be obtained. The article of commerce may be further specified as to saponification value, solidification point of the free fatty acids, iodine value, hydroxyl value and ash content. 2. Appearance : Light yellow to amber, oily to very viscous liquids; light tan to medium brown, plastic or soft solids; and light tan to brown, hard, waxy solids. 3. Identification : (1)Tests for fatty acids: Passes test. (2)Test for glycerol and polyglycerols: Spot 5 to 20  $\mu$ l of the aqueous layer obtained in the test for fatty acids Identification tests for funtional groups alongside control spots of glycerol on paper such as Whatman No. 3 and develop using descending chromatography for 36 h with isopropanol: water (90:10). The glycerol spot moves 40 cm and the polyglycerols are revealed in succession below that for glycerol when the paper is sprayed with either permanganate in acetone or ammoniacal silver nitrate. 4. Solubility : From very hydrophilic to very lipophilic, but as a class tend to be dispersible in water and soluble in organic solvents and oils. 5. Acids : Acids other than fatty acids shall not be detectable.
- 6. Polyglycerols : The polyglycerol moiety shall be composed of not less than 75% of di-, triand tetraglycerols and shall contain not more than 10% of polyglycerols equal to or higher than heptaglycerol.
- 7. Lead : Not more than 2 mg/kg.
- 8. Category : Food Additives Category 16
- 9. Uses : Emulsifier

Polyglycerol Esters of Interesterified Ricinoleic Acid

The major components have the general structure:



where the average value of n is about 3 and R1, R2 and R3 each may be hydrogen or a linear condensation polymer of ricinoleic acid with itself thus:

where the average value of m is between 5 and 8.

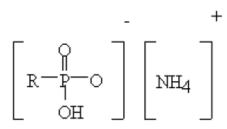
- 1. Definition : Prepared by the esterification of polyglycerol with condensed castor oil fatty acids.
- 2. Appearance : Highly viscous liquids
- 3. Identification : (1) Tests for fatty acids: Passes test.

(2)Test for ricinoleic acid: The fatty acids liberated in test for fatty acids Identification tests for funtional groups should have a Hydroxyl value corresponding to that for castor oil fatty acids (about 150 to 170).

(3)Test for glycerol and polyglycerols: Spot 5 to 20  $\mu$ l of the aqueous layer obtained in the test for fatty acids under Identification test for functional groups alongside control spots of glycerol on paper such as Whatman No. 3 and develop using descending chromatography for 36 h with isopropanol : water, 90:10. The glycerol spot moves 40 cm and the polyglycerols are revealed in succession below that for glycerol when the paper is sprayed with either permanganate in acetone or ammoniacal silver nitrate.

- Solubility : Insoluble in water and in ethanol; soluble in ether, hydrocarbons and halogenated hydrocarbons.
- 5. Polyglycerols : The polyglycerol moiety shall be composed of not less than 75% of di-, triand tetraglycerols and shall contain not more than 10% of polyglycerols equal to or higher than heptaglycerol.
- 6. Lead : Not more than 2 mg/kg.
- 7. Category : Food Additives Category 16
- 8. Uses : Emulsifier

#### Ammonium Phosphatide



# where R may be a mono- or di-glyceride moiety

Synonyms : Ammonium salts of phosphatidic acid; Mixed ammonium salts of phosphorylated glycerides INS No. 442

1.	Definition	: The product consists essentially of a mixture of the ammonium compounds of
		phosphatidic acids derived from the edible fat (usually partially hardened rapeseed
		oil). A mono- or diglyceride moiety may be attached to phosphorus. Moreover, two
		phosphorus esters may be linked together as phosphatidyl phosphatides.
2.	Assay	: The phosphorus content is not less than 3.0% and not more than 3.4% by weight;
		the ammonium N content is not less than 1.2% and not more than 1.5%.
3.	Appearance	: Unctuous semisolid
4.	Solubility	: Insoluble in water, partially soluble in ethanol and in acetone, soluble in fats.
5.	Lead	: Not more than 2 mg/kg.
6.	Category	: Food Additives Category 16
7.	Uses	: Emulsifier

## Pectins

#### Guar Gum

#### **Carob Bean Gum**

## Category 17 Others

## § 17001

## Piperonyl Butoxide

Chemical formula: C <sub>19</sub> H <sub>30</sub>		O <sub>5</sub> Molecular weight: 338.45
1.	Appearance	: An oily colorless to slightly yellow liquid. Odorless or with a faint characteristic odor.
2.	Color	: The color should not be thicker than that of 1.4 mL of the cobalt chloride colorimetric standard stock solution, 4.3 mL of the ferric chloride colorimetric standard stock solution and 0.3 mL of the copper sulfate colorimetric standard stock solution.
3.	Density	: 1.05~1.07.
4.	Index of	20
	refraction	$:n_{\rm D} = 1.497 \sim 1.512.$
5.	Chloride	: Not more than 0.2% (as 6-Propyl Piperonyl Chloride)
	compound	
6.	Heavy metals	: Not more than 20 ppm of Ni.
7.	Distillation range	: Put 25 g of the sample in a 100 mL flask. Under the decompression of 4 mm mercury
		column, when distilling to 194°C, the residue in the bottle should be more than 85% of
		the sample amount. When distilled to 203°C, the residue in the bottle should be less
		than 5% of the sample amount.
8.	Category	: Food Additives Category 17
9.	Uses	: Others.

### Polyvinyl Acetate

1	Appearan	: Colorless to light yellow particles or glassy blocks.
	ce	
2	Arsenic	: Not more than 2 ppm (as As <sub>2</sub> O <sub>3</sub> ).
3	Heavy	: Not more than 4 ppm (as Pb).
	metals	
4	Free acid	: Not more than 0.2% (as CH <sub>3</sub> COOH)
5	Residual	: Accurately weigh 1 g of the sample, add 100 mL of 75% acetic acid solution to it. After placed for
	monovinyl	24 hours to dissolve, add 4 mL of bromine glacial acetic acid TS and 2 mL of potassium iodide TS.
		Titrate free iodine with 0.1 N sodium thiosulfate (indicator: starch test solution). In another
		controlled test, when the amount of residual vinyl monomer is calculated as follows, the amount
		should be less than 1%.
		Residual monovinyl=
		0.0043  imes (0.1N sodium thiosulf ate solution required for control test-0.1N sodium thiosulf ate solution required for
		Sample volume (g)
		100%
6	Residue	: Not more than 0.05%.
	on	
	ignition	
7	Category	: Food Additives Category 17
8	Uses	: Others.

#### Silicon Resin

#### (polydimethyl siloxane)

- 1. Appearance : Clear, light grey and almost transparent, viscous liquid. Almost odorless.
- 2. Density  $: 0.98 \sim 1.02$ .

3. Viscosity and index of : 15 g of the sample is placed in a Soxhelt extractor and extracted with 150 mL of refraction carbon tetrachloride for 3 hours. Evaporated the extract on a water bath to form a viscous liquid. Its viscosity should be 100~1,100 Centistokes at 25°C and the

refringence be n 
$$\frac{25}{D} = 1.400 \sim 1.410.$$

- Silica : The extract residue of "3. Viscosity and index of refraction" should range from 0.45 g to
   2.25 g after drying at 100°C for 1 hour.
- 5. Category : Food Additives Category 17
- 6. Uses : Others.

#### Diatomaceous Earth

#### Sodium Oleate

- 1. Appearance : White crystals or yellow amorphous granules, having a characteristic odor.
- Solution : Mix and dissolve 0.5 mg of the sample in 20 mL of water. The liquid should be "almost clear".
- Free alkali
  Accurately weigh 5.0 gm of the sample powder, add 100 mL of neutral ethanol, heat to dissolve, and filter. Wash with warm neutral ethanol until the wash is colorless. Combine the wash with the filtrate. After cooling, titrate with 0.1 N sulfuric acid. Its consumption is expressed as a mL. Wash the residue with 10 mL of hot water for five times and combine the washings. After cooling, add 3 drops of bromophenol blue indicator dropwise and titrate with 0.1 N sulfuric acid. The consumption is expressed in b mL. When the amount of free alkali is determined according to the following formula, the amount should be less than 0.5%.

Free alkali =

$$\frac{0.004 \times a + 0.0053 \times b}{Sample (gm)} \times 100(\%)$$

- 4. Arsenic : Not more than 2 ppm (as  $As_2O_3$ ).
- 5. Heavy metals : Not more than 40 ppm (as Pb).
- 6. Residue on  $:22 \sim 25\%$ .

ignition

- 7. Category : Food Additives Category 17
- 8. Uses : Others.

#### Oxyethylene Higher Aliphatic Alcohol

- 1. Appearance : White to pale yellow powder, flakes, or waxy lumps, odorless, tasteless or with a slight odor.
- 2. Solution : When 0.5 g of the sample is dissolved in 10 mL of Dioxan, the solution should be "almost clear".
- 3. Hydroxy value : Accurately weigh 1.5 g of the sample into a long neck round bottom flask, add anhydrous acetic acid, 5 mL of pyridine TS. Place a funnel on the mouth of the flask and heat in a 95-100°C oil pan for 1 hour. After cooling, dissolve the attached matter on the funnel with 1 mL of water, thoroughly shake and mix in the flask. Then heat in an oil pan for 10 minutes. After cooling, wash the funnel with 5 mL of neutral ethanol and burn the bottleneck, which is titrated with 0.5 N potassium hydroxide ethanol solution (indicator: 1 mL of phenolphthalein test solution) Perform a blank test with same process. When calculating the hydroxyl value according to the following formula, the value should be 155 to 187. (Acid value is regarded as 0).

Hydroxy value =

 $\frac{0.5N \text{ potassium hydroxide ethanol solution consumption (ml)} \times 28.05}{\text{Sample } (g)}$ 

4.	Heavy metals	Not more than 40 ppm (as Pb).
5.	Polyoxyethylene	: Heat and oscillate 0.5 g of the sample and 20 mL of water. Filter
		after cooling, add 10 mL of Ammonium thiocyanate nitrate TS,
		and mix thoroughly by shaking. Add 10 mL of chloroform, shake
		and mix. The chloroform layer should not be blue when placed.
6.	Residue on	Not more than 0.5%.
	ignition	
7.	Category	Food Additives Category 17

8. Uses : Others.

#### Shellac

Syno	onyms : IN	S No. 904
1.	Description	: Shellac is a polyester resin obtained from lac, the resinous secretion of the insect
		Laccifer (Tachardia) lacca Kerr (Fam. Coccidae). Bleached shellac is obtained by
		dissolving the lac in aqueous sodium carbonate, followed by bleaching with sodium
		hypochlorite, precipitation of the bleached lac with dilute sulfuric acid solution, and
		drying; wax-free bleached shellac is prepared by further treatment whereby the wax is
		removed by filtration.
2.	Appearance	: Bleached shellac: off-white to tan, amorphous granular resin; wax-free bleached
		shellac: light yellow, amorphous, granular resin.
3.	Color reaction	: To 50 mg of the sample add a few drops of a solution of 1 g ammonium molybdate in 3
		mL of sulfuric acid. A green color is produced, changing to lilac when the solution is
		neutralized with 6 N ammonium hydroxide.
4.	Solubility	: Insoluble in water; freely (though very slowly) soluble in ethanol; slightly soluble in
		acetone and ether.
5.	Acid value	: 60~89.
6.	Loss on drying	: Not more than 6.0% (40°C, 4 h, then room temperature over silica gel, 15 h)
7.	Rosin	: Dissolve 2 g of the sample in 10 mL of dehydrated ethanol, and add slowly, with
		shaking, 50 mL of solvent hexane. Transfer to a separator, wash with two 50-mL
		portions of water, and discard the washings. Filter the solvent layer, evaporate it to
		dryness, and to the residue add 2 mL of a mixture of 1 volume of liquefied phenol and 2
		volumes of methylene chloride. Stir and transfer a portion of the mixture to a cavity of a
		color-reaction plate. Fill an adjacent cavity with a mixture of 1 volume of bromine and
		four volumes of methylene chloride, and cover both cavities with an inverted watch
		glass. No purple or deep indigo blue color is produced in or above the liquid containing
		the sample residue.
8.	Wax	: Bleached shellac: not more than 5.5%; wax-free bleached shellac, not more than 0.2%.
9.	Lead	: Not more than 2 mg/kg
10.	Category	: Food Additives Category 17
11.	Uses	: Others

Petroleum Wax (Refined Paraffin Wax, Microcrystalline Wax)

### Petroleum Wax, Synthetic

## (Synthetic Wax, Ethylene Polymer)

1.	Description	: The sample is obtained by catalytic polymerization of ethylene. It is a paraffinic,
		refined solid hydrocarbon mixture. Amber to off-white. The melting point is between
		$85^{\circ}$ C and $116^{\circ}$ C (185 ~ 240°F). Soluble in aromatic hydrocarbons, but not soluble in
		ketones, esters and alcohols.
2.	Identification	: The infrared absorbance spectrum of the sample melted and prepared on a caesium or
		potassium bromide plate corresponds to the
		infrared absorption spectrum and the standard map.
3.	Average molecular	500 1 000
	weight	: 500∼1,200.
4.	Ultraviolet absorba	nce : $280 \sim 289$ nm, not more than 0.15.
	limits	$290 \sim 299$ nm, not more than 0.12.
		$300 \sim 359$ nm, not more than 0.08.
		$360 \sim 400$ nm, not more than 0.02.
5.	Arsenic	: Not more than 3 ppm (as As).
6.	Lead	: Not more than 3 ppm.
7.	Heavy metals	: Not more than 0.002% (as Pb).
8.	Color, melting poir	it and
	odor	: Should meet the seller's own specifications.
9.	Category	: Food Additives Category 17
10.	Uses	: Others.

### Polyethylene Glycols

Chen	nical formula: HOCH	I <sub>2</sub> (CH <sub>2</sub> OCH <sub>2</sub> ) <sub>n</sub> CH <sub>2</sub> OH	Molecular weight: 200~9500
1.	Solution	: Dissolve 5 g of the sample in 5	0 mL of water. Those with a molecular weight of 700 or
		less shall be colorless and clear	or colorless and slightly ambiguous liquid; those with a
		molecular weight of 700 to 900	should be semi-solid; those with a molecular weight
		greater than 1000 should be mill	xy white, waxy solid or powder with no fluidity.
2.	Arsenic	: Not more than 3 ppm (as As).	
3.	1, 4–Dioxane	: Not more	than 10 ppm.
4.	Ethylene Glycol an	nd Diethylene Glycol	: Not more than 0.25% separately or in
			total.
5.	Ethylene Oxide	: Not more	than 0.02%.
6.	Heavy metals	: Not more than 10 ppm (as Pb).	
7.	pH	: Dissolve 5 g of the sample in 1	00 mL of water without $CO_2$ that contains 0.3 mL of
		saturated potassium chloride s	solution. The pH of the solution should be 4.5-7.5.
8.	Residue on	No.4	
	ignition	: Not more than 0.1%.	
9.	Viscosity	: Should meet the criteria specifi	ed in the appendix.
10.	Acidity	: Not more than 0.05% w/w (as a	acetic acid).
11.	Category	: Food Additives Category 17	
12.	Uses	: Others.	

#### Tannic Acid

Syno	nyms : Tai	nnins (food grade), Gallotannic acid .INS No. 181
1.	Definition	: Consists of gallotannins obtained by solvent extraction from certain natural sources; the
		substance is not an acid in the chemical sense. The common name "Tannic acid" has
		been adopted to distinguish the commercial substance from other tannins, such as
		condensed tannins. These specifications relate only to hydrolysable gallotannins, i.e.,
		those which yield gallic acid on hydrolysis. These specifications do not apply to many
		other kinds of tannins which occur in nature, including condensed (nonhydrolysable)
		tannins and hydrolysable ellagitannins. Hydrolysable gallotannins may be obtained from
		nutgalls, the excrescences which form on young twigs of various Quercus species, e.g.,
		Q. infectoria; these include Chinese and Aleppo tannins. They may also be obtained from
		various Sumac species, e.g. Rhus corieria, R. galabra, R. thypia; these include Sicilian
		and American sumacs. All of these consist essentially of polydigalloyl esters of glucose.
		A further source of hydrolysable gallotannins is the seed pods of Tara (Caesalpinia
		spinosa); these tannins consist essentially of the polydigalloyl esters of quinic acid.
2.	Assay	: Not less than 96% (on the dried basis)
3.	Appearance	: Amorphous powder, glistening scales or spongy mass, varying in color from yellowish
		white to light brown; odorless or with a characteristic odor.
4.	Solubility	: Soluble in water, acetone and ethyl alcohol; insoluble in benzene, chloroform and ether;
		1 g dissolves in about 1 mL of warm glycerin.
5.	Color reaction	: To a 1 in 10 solution add a small quantity of ferric chloride TS. A bluish black color or
		precipitate forms.
6.	Precipitate	: A solution of the sample when added to a solution of either albumin or gelatin produces
	formation	a precipitate.
7.	Loss on drying	: Not more than 7% (105°C, 2 h).
8.	Sulfated ash	: Not more than 1%.
9.	Gums or dextrin	: Dissolve 1 g in 5 mL of water, filter, and to the filtrate add 10 mL of alcohol; no
		turbidity is produced within 15 minutes.
10.	Resinous	: Dissolve 1 g in 5 mL of water, filter and dilute the filtrate to 15 mL; no turbidity is

substances produced.

11.	Condensed	: Not more than 0.5%.	
	tannins	. Not more than 0.5 %.	
12.	Residual solvent	: Not more than 25 mg/kg acetone or ethyl acetate, singly or in combination.	
13.	Lead	: Not more than 2 mg/kg.	
14.	Category	: Food Additives Category 17	
15.	Uses	: Clarifying agent	

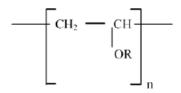
### Quillaia Extracts

Synonyms	Quillaja extract, Soapbark extract, Quillay bark extract, Bois de Panama, Panama bark
	extract, Quillai extract; INS No. 999
Definition	Quillaia extract is obtained by aqueous extraction (Type 1) or either by chromatographic
	separation or ultrafiltration of the aqueous extraction (Type 2) of the milled inner bark or
	of the wood of pruned stems and branches of Quillaja saponaria Molina (family
	Rosaceae). It contains triterpenoid saponins (quillaia saponins, QS) consisting
	predominantly of glycosides of quillaic acid. Polyphenols and tannins are major
	components and some sugars and calcium oxalate will be present.
	Quillaia extract is available commercially as liquid product or as spray-dried powder that
	may contain carriers such as lactose, maltitol or maltodextrin. The liquid product is
	usually preserved with sodium benzoate or ethanol.
C.A.S. number	68990-67-0
Formula weight	Monomeric saponins range from ca. 1800 to ca. 2300, consistent with a triterpene with
	8-10 monosaccharide residues
Assay	Saponin content:
	Type 1: not less than 20% and not more than 26% on the dried basis
	Type 2: not less than 65% and not more than 90% on the dried basis
Description	Type 1: Red-brownish liquid or light brown powder with a pink tinge
	Type 2: Light red-brownish liquid or powder
Characteristics	
Identification	
Solubility	Very soluble in water, insoluble in ethanol, acetone, methanol and butanol
Foam	Dissolve 0.5 g of powder extract in 9.5 g of water or 1 ml of liquid extract in 9 ml of
	water. Add 1 ml of this mixture to 350 ml of water in a 1000-ml graduated cylinder.
	Cover the cylinder, vigorously shake it 30 times, and allow settling. Record the foam
	level (ml) after 30 min. Typical values:
	Type 1: 150 ml.

	Type 2: 260 ml.
Chromatography	The retention time of major peak of the sample corresponds to the major saponin peak
	(QS-18) of the standard.
Colour and turbidity	Powder form only: Dissolve 0.5 g in 9.5 g of water. The solution is not turbid. Determine
	the absorbance of the solution against water at 520 nm.
	Type 1: The absorbance is less than 1.2.
	Type 1: The absorbance is less than 0.7.
Purity	
Water	Powder form: not more than 6% (Karl Fischer Method)
Loss on drying	Liquid form: (2 g, 105°C, 5 h)
	Type 1: 50 to 80%
	Type 2: 50 to 90%
pH	3.7 -5.5 (4 % solution)
Ash	Type 1: Not more than 14% on a dried basis
	Type 2: Not more than 5% on a dried basis
	Use 1.0 g for powder samples; for liquid samples, use the residue from loss on drying.
Tannins	Not more than 8% on a dried basis
Lead	Not more than 2 mg/kg.
Category	Food additives category (17)
Functional uses	Others.

#### Polyvinyl Alcohol

Chemical formula: (C<sub>2</sub>H<sub>3</sub>OR) where R=H or COCH3 (randomly distributed)



Where R=H or COCH3 (randomly distributed)

1.	Definition	: Polyvinyl alcohol is a synthetic resin prepared by the polymerization of vinyl acetate,
		followed by partial hydrolysis of the ester in the presence of an alkaline catalyst. The
		physical characteristics of the product depend on the degree of polymerization and the
		degree of hydrolysis.
2.	Description	: Odorless, translucent, white or cream-colored granular powder.
3.	Identification	: (1) Solubility: Soluble in water, insoluble in ethanol.
		(2)pH: 5.0-6.5 (1 in 5).
		(3) Infrared spectrum: The infrared absorption spectrum of a potassium bromide
		dispersion of the sample corresponds to that of a polyvinyl alcohol standard.
		(4) Color reaction A: Dissolve 0.01 g of the sample in 100 mL of water with warming
		and let the solution cool to room temperature. To 5 mL of the solution, add one drop of
		iodine TS and a few drops of boric acid solution (1 in 25). A blue color is produced.
		(5) Color reaction B: Dissolve 0.5 g of the sample in 10 mL of water with warming and
		let the solution cool to room temperature. Add 1 drop of iodine TS to 5 mL of solution
		and allow to stand. A dark red to blue color is produced.
		(6) Precipitation reaction: Add 10 mL of ethanol to the remaining 5 mL of solution
		prepared for Color Reaction B. A white, turbid or flocculent precipitate is formed.
4.	Loss on drying	: Not more than 5.0% (105°, 3 h)
5.	Residue on	: Not more than 1.0%.
	ignition	
6.	Water-insoluble	: Not more than 0.1%.
	matter	
7.	Category	: Food Additives Category 17
8.	Uses	: Others.

### Magnesium silicate(synthetic)

Synonyms : INS No. 553(i)

1.	Definition	: Magnesium silicate (synthetic) is manufactured by the precipitation reaction between
		sodium silicate and a soluble magnesium salt. The aqueous suspension of the precipitate
		is filtered and the collected solid washed, dried, classified for particle size and packaged.
		The molar ratio of MgO to SiO2 is approximately 2:5.
2.	Appearance	: Very fine, white, odorless powder, free from grittiness.
3.	Assay	: Not less than 15% of MgO and not less than 67% of SiO2, calculated on the ignited
		basis.
4.	Identification	: (1) Solubility: Insoluble in water and in ethanol but soluble in mineral acids.
		(2) pH: 7.0-10.8 (1 in 10 soln).
		(3) Magnesium test: Mix about 0.5 g of the sample with 10 mL of $2.7$ N dilute
		hydrochloric acid TS, filter, and neutralize the filtrate to litmus paper with 6 N ammonia
		TS. The neutralized filtrate gives a positive test for magnesium.
		(4) Silicate test: Prepare a bead by fusing a few crystals of sodium ammonium
		phosphate on a platinum loop in the flame of a Bunsen burner. Place the hot, transparent
		bead in contact with the sample, and again fuse. Silica floats about in the bead,
		producing, upon cooling, an opaque bead with a web-like structure.
5.	Loss on drying	: Dry it at 105°C for 2 hours: it loses not more than 15% of its weight.
6.	Loss on ignition	: Not more than 15% on the dried basis. Weigh to the nearest 0.1 mg, 1 g of the dried
		sample in a tared platinum crucible provided with a cover. Gradually apply heat to the
		crucible at first, then strongly ignite at $900/1000^{\circ}$ for 20 min. Cool, weigh and calculate
		as percentage.
7.	Soluble salts	: Not more than 3%. Boil 10 g of the sample with 150 mL of water for 15 min. Cool to
		room temperature, and add water to restore the original volume. Allow the mixture to
		stand for 15 min, and filter until clear. Retain 20 mL of the filtrate for the test Free alkali.
		Evaporate 75 mL of the filtrate, representing 5 g of the sample in a tared platinum dish
		on a steam bath to dryness, and ignite gently to constant weight. Cool, weigh and
		calculate as percentage (the weight of the residue should not exceed 150 mg).
8.	Free alkali	: Not more than 1% (as NaOH) Add 2 drops of phenolphthalein TS to 20 mL of dilute
		filtrate prepared in the test for Soluble salts, representing 1 g of the sample. Not more

than 2.5 mL of 0.1 N hydrochloric acid should be required to discharge the pink color produced.

- 9. Arsenic : Not more than 0.1 ppm (as As).
- 10. Fluoride : Not more than 10 mg/kg. Determine using Fluoride Limit Test.
- 11. Lead : Not more than 5 mg/kg. Determine using an AAS/ICP-AES.
- 12. Category : Food Additives Category 17
- 13. Uses : Anticaking agent, filtering aid