

Simultaneous Determination of 18 Quinolone Residues in Marine and Livestock Products by Liquid Chromatography/Tandem Mass Spectrometry

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ABSTRACT

A simple and efficient LC/ESI/MS/MS method was developed to determine 18 (fluoro)quinolone (QNs) residues in milk, chicken, pork, fish and shrimp. This method is capable of screening and confirming the presence of 12 amphoteric QNs (marbofloxacin, norfloxacin, enrofloxacin, ciprofloxacin, desethylene ciprofloxacin, lomefloxacin, danofloxacin, sarafloxacin, difloxacin, ofloxacin, orbifloxacin and enoxacin) and 6 acidic QNs (oxolinic acid, nalidixic acid, flumequine, cinoxacin, piromidic acid and pipemidic acid). The drugs were extracted from matrix with acetonitrile containing 1% formic acid, diluted in 10% acetonitrile and defatted by extraction with hexane. The LC separation was conducted on an XDB C-8 (150 mm × 4.6 mm, 5 μm) column with gradient elution of 20 mM ammonium formate in 0.1% formic acid-acetonitrile as the mobile phase. Mass spectrum acquisition was completed in the positive ion mode by applying multiple reaction mode (MRM). The decision limit (CC α) and detection capability (CC β), stated in the Decision No. 2002/657/EC and the ISO standard No.11843, have been calculated in the case of nonauthorized substance. The values of CC α ranged from 0.18 to 0.68 ng/g and CC β ranged from 0.24 to 0.96 ng/g under specified conditions.

Key words: quinolones, residue, LC/MS/MS

INTRODUCTION

Quinolones (QNs) are broad-spectrum synthetic antimicrobial agents used in the treatment of bacterial infection of livestock and in aquaculture. They behave through inhibiting bacterial DNA-gyrase and topoisomerase IV enzyme activities⁽¹⁾. QNs can be grouped into acidic QNs and amphoteric QNs. Acidic QNs are represented by quinolones carboxylic acid, such as flumequine (FLU), oxolinic acid (OXO), nalidixic acid (NAL), cinoxacin (CIN), piromidic acid (PIR) and pipemidic acid (PIP). Amphoteric QNs (fluoroquinolones) contain fluorine at C-6 position and piperazinyl at C-7 position, such as marbofloxacin (MAR), norfloxacin (NOR), ciprofloxacin (CIP), lomefloxacin (LOM), danofloxacin (DAN),

enrofloxacin (ENR), sarafloxacin (SAR), difloxacin (DIF), ofloxacin (OFL), enoxacin (ENO) and orbifloxacin (ORB). The major metabolites of ENR were reported to be CIP, and desethylene ciprofloxacin (des-CIP)^(2,3). SAR (principal metabolite of DIF) is different only by methyl group at 7-(4-piperazinyl) position⁽²⁾. The widespread administration of fluoroquinolones to calves⁽⁴⁾, poultry and swine⁽⁵⁾ leads to the occurrence of residues in food products. These residues are of great concern for public health in the emergence of antibiotics-resistance to foodborne pathogens⁽⁶⁻⁸⁾. Delsol⁽⁶⁾ indicated that a single course of enrofloxacin treatment contributes directly to the emergence and persistence of fluoroquinolone resistance in *Campylobacter coli*. The resistance of bacteria to fluoroquinolones has increased significantly. Mutations in the genes that encode the topoisomerases (gyrA, gyrB, parC and parE) or in the genes that affect

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cell permeability or drug export are the main causes of quinolone resistance⁽⁷⁾. To ensure that food consumers are not exposed to residues at potentially harmful concentrations, maximum residue limits (MRLs) for residues of veterinary drugs in food animal tissues and products have been established. The MRLs for QNs among Taiwan⁽⁹⁾, Japan⁽¹⁰⁾ and the European Union^(3,11) are set, as shown in Table 1. Monitoring of QNs residues in animal products is very important to ensure that human food is entirely free of potentially harmful residues.

Many methods have been established for determining QNs in food-producing animals. Typically, these methods employed high-performance liquid chromatography (HPLC) with ultraviolet (UV) and fluorescence (FLD)⁽¹²⁻²⁵⁾ detection or mass spectrometric detection⁽²⁶⁻³⁷⁾. Although most of these methods can achieve the quantification of QNs at very low concentration levels, they are usually restricted to the numbers of QNs [maximum eleven^(17,26,33)] determined simultaneously. This is often due to the co-elution properties for compounds belonging to the same chemical family and also due to pKa differences between the acidic and the amphoteric QNs⁽³³⁾. Strict guidelines in European Union (EU)⁽³⁸⁾ for confirmatory techniques state that liquid chromatography/tandem mass spectrometry (LC/MS/MS) monitoring two or three transition-product ions gives sufficient data to confirm the identity of a residue. Other methods, such as LC with FLD or UV detection, must combined with another technique to meet the criteria. LC/MS/MS was the technique of choice, due to its high specificity and sensitivity, monitoring two transition-product ions gives sufficient data to confirm the identify of a substance found in the complex matrix. Consequently, a confirmatory method using LC/MS/MS would supplement the existing screening methods.

The study was aimed to develop a multi-residue method for screening and confirming a group of QNs in various food matrices by triple quadrupole mass spectrometry after a simple liquid-liquid extraction. Method validation was performed for each matrix and validation parameters including selectivity, accuracy, matrix effect, decision limit (CC α) and detection capability (CC β) were discussed.

MATERIALS AND METHODS

I. Chemicals

Norfloxacin (NOR, 100.0%), oxolinic acid (OXO, 99.9%), nalidixic acid (NAL, 99.5%), flumequine (FLU, 99.9%) and lomefloxacin (LOM, 99.1%) were purchased from Sigma (Steinheim, Germany). Marbofloxacin (MAR, 99.8%), ciprofloxacin (CIP, 100.0%), danofloxacin (DAN, 98.4%), sarafloxacin HCl (SAR, 99.7%) and difloxacin HCl (DIF, 98.4%) were from Riedel-de Haen (Seelze, Germany) Enrofloxacin (ENR, 98%) was from Fluka. Desethylene ciprofloxacin HCl (des-CIP, 98%)

was from Toronto Research (North York, Canada). The chemical structures of all 18 QNs are shown in Figure 1. Sodium hydroxide and ultra pure sodium sulfate anhydrate were from Nacalai Tesque (Kyoto, Japan). Formic acid and acetonitrile were of HPLC grade and purchased from J.T.Baker (Deventer, Holland). All other reagents were of analytical grade.

Table 1. Comparison of MRLs (ppm) for quinolones in Taiwan⁽⁹⁾, Japan⁽¹⁰⁾ and the EU⁽¹¹⁾

Quinolone	Species	Taiwan	Japan	EU
NAL	Bovine	—	0.03	—
FLU	Fish	0.05	0.5	0.15
	Bovine	0.05	0.5	0.1
	Porcine	0.05	0.5	0.1
	Sheep	0.05	—	0.1
	Chicken	0.05	0.5	0.1
OXO	Bovine	—	1.0	0.1
	Porcine	—	1.0	0.1
	Chicken	—	1.0	0.1
	Fish	0.05	—	0.3
NOR	Bovine	—	0.02	—
	Chicken	—	0.02	—
DAN	Bovine	0.2	0.2	0.2
	Chicken	0.2	0.2	0.2
	Porcine	0.1	0.1	—
	Milk	—	0.05	—
The sum of ENR and CIP	Bovine	—	0.05	0.1
	Chicken	—	0.05	0.1
	Porcine	—	0.05	0.1
	Milk	—	0.05	0.4
OFL	Chicken	—	0.05	—
MAR	Bovine	—	0.2	0.15
	Porcine	—	0.2	0.15
	Milk	—	0.08	0.075
SAR	Chicken	0.01	0.01	0.01
	Turkey	0.01	0.01	0.01
	Salmon	—	0.03	0.03
ORB	Bovine	—	0.02	—
	Chicken	—	0.02	—
	Porcine	—	0.02	—
	Milk	—	0.02	—
DIF	Bovine	—	0.4	0.4
	Chicken	—	0.3	0.3
	Porcine	—	0.02	0.4
	Fish	—	0.3	—

— : No MRL has been fixed.

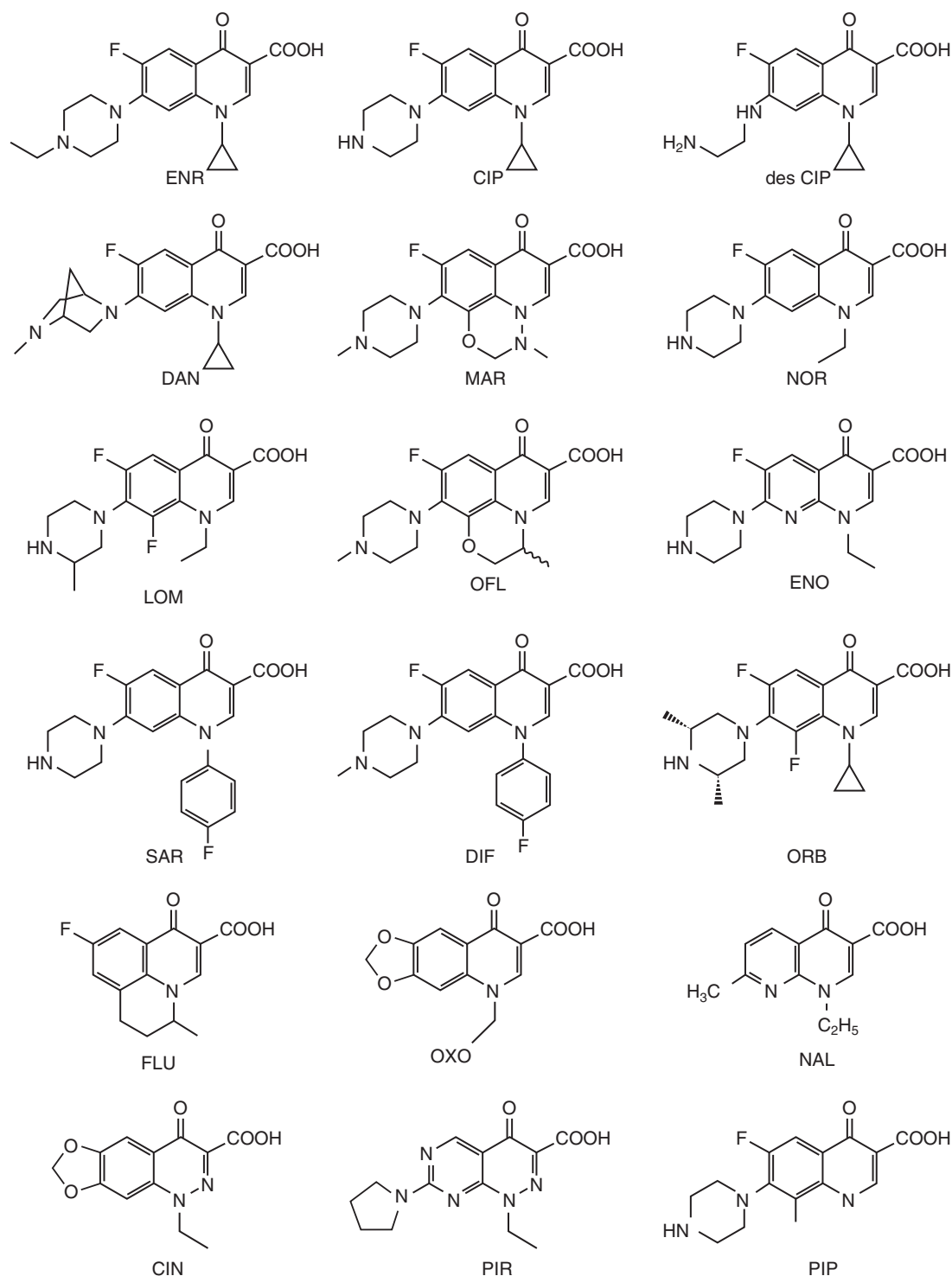


Figure 1. Chemical structures of 18 (fluoro)quinolones.

II. Instrumentation and Conditions

(I) Instrumentation

The instruments for sample preparation consisted of a shaker (model KM, IWAK, Japan), a centrifuge (model himac CF 16RX, Hitachi kok, Hitachinaka, Japan) and a

rotary evaporator (EYELA, Japan). Water was purified in a Milli-Q system (Millipore, Bedford, USA).

The HPLC system was Agilent 1100 series (Agilent Technologies, Palo Alto, USA) consisted of a quaternary pump (G1311A), a vacuum degasser (G1322A) and an automatic sampler (G1313A). The separation of QNs was achieved on a Zorbax Eclipse XDB-C8 (150 mm ×

4.6 mm, 5 μ m) analytical column (Agilent Technologies).

The triple-quadrupole API 4000 mass (Applied Biosystems/MDS Sciex, Foster, USA) was coupled to HPLC using an electrospray ionization interface in positive ionization mode (ESI+). Data acquisition was conducted using the Analyst 1.4 software (Applied Biosystems/MDS Sciex).

(II) LC and MS/MS Conditions

The injection volume was 20 μ L and the analysis was carried out with gradient elution using A eluent (20 mM ammonium formate in 0.1% formic acid) and B eluent (acetonitrile) as the mobile phase at a flow rate 0.7 mL/min. The program of gradient elution is listed in Table 2.

In order to establish the optimized Multiple reactions monitoring (MRM) conditions for individual compounds, the mass spectrometric conditions were optimized using infusion with a syringe pump and direct injection of each QNs individually at a concentration of 0.5 μ g/mL and at flow-rate of 0.01 mL/min to select the most suitable ion transitions.

Due to the presence of the amino group in most QNs

that is easily protonated in acidic medium, the turbo ion spray source was set in positive mode. The ESI/MS/MS conditions were set as ionspray voltage (IS) 5500 V and collision gas (CAD) 7 (arbitrary units). The nebulizer gas (synthetic air) and the curtain gas (nitrogen) were set at 45 and 12 (arbitrary units), respectively. The heater gas temperature (TEM) was set at 500°C. MS/MS product ions were produced by collision-activated dissociation (CAD) of selected precursor ions in the collision cell of

Table 2. Timetable of gradient elution program

Time (min)	A (%) [*]	B (%)
0	85	15
7	30	70
8	5	95
9	5	95
10	85	15
12	85	15

^{*}A: 20 mM ammonium formate in 0.1% formic acid, B: acetonitrile.

Table 3. Optimized MS/MS for MRM transitions selected for quantification and identification of the quinolones

Quinolone	Retention time (min)	Precursor ion (<i>m/z</i>)	Quantification	Identification
			(Collision energy) <i>m/z</i> (V)	(Collision energy) <i>m/z</i> (V)
NAL	6.83	233	233→215 (22)	233→187 (36)
FLU	7.27	262	262→244 (25)	262→202 (46)
OXO	6.13	262	262→244 (24)	262→216 (42)
CIN	5.60	263	263→217 (31)	263→245 (22)
PIR	7.38	289	289→271 (26)	289→243 (41)
PIP	3.57	304	304→286 (26)	304→217 (31)
des-CIP	4.25	306	306→288 (24)	306→268 (36)
NOR	4.35	320	320→302 (32)	320→276 (24)
ENO	4.11	321	321→303 (28)	321→234 (31)
CIP	4.50	332	332→314 (27)	332→288 (24)
LOM	4.58	352	352→308 (24)	352→265 (32)
DAN	4.52	358	358→340 (31)	358→283 (25)
ENR	4.76	360	360→342 (29)	360→316 (29)
OFL	4.35	362	362→318 (28)	362→261 (36)
MAR	4.23	363	363→345 (22)	363→320 (28)
SAR	5.24	386	386→368 (31)	368→342 (27)
ORB	4.93	396	396→352 (25)	396→295 (32)
DIF	5.42	400	400→382 (30)	400→356 (28)

triple-quadrupole mass spectrometer. Mass was determined in the second analyzer of the instrument. MRM experiments were performed using a dwell time of 80 ms and the collision energy transition (depending on the compound) as shown in Table 3. Two transitions were used for the identification each QNs but only one was used for quantification.

III. Preparation of Standard Solutions

For the QNs with good solubility in alkaline solution⁽¹⁵⁾, a 200 µg/mL stock standard solution was prepared for each of 18 QNs by dissolving the appropriate amount of standard in 50% acetonitrile containing 2% of 0.1 N NaOH. Working standard solutions were prepared by serial dilution of standard solutions with 10% acetonitrile. Stock and working standard solutions were stored at 4°C in brown volumetric flasks for at least 3 months without any degradation observed. Five working standard solutions ranging from 0.5 to 10 ng/mL were prepared for external standard calibration.

IV. Sample Preparation

Thawed tissue sample was ground first and 2.0 g of tissue was weighed and placed in a 50-mL polypropylene centrifuge tube. The working standard solution was then spiked into matrix samples to the desired concentrations (0.25, 0.5, 1.0, 2.5 and 5.0 ng/g). Then, the fortified sample was allowed to stand for 15 min at room temperature. Twenty milliliters acetonitrile containing 1% formic acid was added to the sample, which then was reversely shaking for 5 min at high setting. Two grams of sodium sulfate anhydrate was added to each tube, which then was reversely shaking again for another 5 min, followed by centrifugation for 10 min (6000 rpm, 4°C). The acetonitrile extract was evaporated in a rotary evaporator at 40°C. The residue was resuspended in 2 mL of 10% acetonitrile containing 0.1% formic acid, poured into a 16 × 12 mm glass tube and defatted by extraction with 4 mL hexane while mixing on a Vortex mixer in 15 sec twice. The mixture was centrifuged for 5 min (4000 rpm, 4°C) and the aqueous supernatant was transferred and filtered through 0.22 µm Nylon membrane before injection into LC/ESI/MS/MS system.

V. Matrix Effect and Recovery

The effect of matrix and recoveries of QNs were determined by comparing the responses obtained from standards injected in mobile phase (standard calibration curve, SCC), standards spiked into the fish homogenate before extraction (method matched calibration curve, MMCC), and standards spiked into the fish extracts followed by the extraction process (matrix calibration curves, MCC) as described by Durden *et al.*⁽²²⁾. Matrix effect can be estimated by dividing the MMCC slope

by the SCC slope. The recovery can be estimated by dividing the MMCC slope by the MCC slope.

VI. Decision Limit (CC α) and Detection Capability (CC β)

The new set of statistical performance limits recommended in the European Decision No. 657/2002/EC⁽³⁸⁾, i.e., the limit of decision CC α , and the capacity of detection CC β , were calculated from the lower portions of standard curves using the ISO 11843⁽³⁹⁾ and assuming Case 1, constant standard deviation and linear response. CC α is the lowest concentration at which a method can discriminate, with a statistical certainty of 1- α (α = 1% in the case of banned compound). At CC α , sample contains target analyte with probability of 0.99. The detection capability (CC β) is the concentration at which truly contaminated samples can be detected by the method with a statistical certainty of 1- β (β = 5% in the case of banned compounds). Negative matrix was fortified with analytes of 5 levels of concentration (0.25, 0.5, 1.0, 2.5 and 5.0 ng/g) and analyzed in 4 replicates at each level.

RESULTS AND DISCUSSION

I. MS Detection

Positive ESI-MS spectra gave intense signals characterized using prominent protonated molecule [MH]⁺ as precursor ion and the more intense transition was used to quantify QNs. Twelve amphoteric QNs (MAR, NOR, ENR, CIP, des-CIP, LOM, DAN, SAR, DIF, OFL, ORB and ENO) and 6 acidic QNs (OXO, NAL, FLU, CIN, PIR and PIP) were incorporated into this project. QNs are detected in the multiple reaction monitoring (MRM) mode which provides a high level of selectivity for targeting the analytes in a complex biological matrix. These transitions were listed in Table 3. The transition [MH]⁺ → [MH-H₂O]⁺ was used for quantification of NAL, FLU, OXO, CIN, PIR, PIP, des-CIP, NOR, ENO, CIP, DAN, ENR, SAR and DIF, while for quantification of LOM, OFL, MAR and ORB, the transition [MH]⁺ → [M+H-CO₂]⁺ was employed. Other less intense ions were used for the confirmation of QNs. The collision energies for each transition were optimized to maximize the product ion intensity. The retention times of the QNs are shown in Table 3. Although the compounds were not fully separated, monitoring two transition-product ions gave sufficient data to confirm the identity of QNs found in the complex matrix. The use of the precursor ion plus two products ions gave four identification points, which are acceptable proof of confirmation in conjunction with the chromatographic retention times⁽³⁷⁾.

MS/MS modes verified structural information of the compounds present in the unconfirmed samples. As an illustrative example, the interpretation of the product ions of NOR is shown in Figure 2 where an interpretation

of the main fragment ions (m/z 276, 302, 256, 233 and 219) of NOR is proposed.

As observed in Table 3, the same precursor ion (m/z 262) is obtained from OXO and FLU. These substances only generated two product ions and one of these product ions (m/z 244) is common to OXO and FLU. However, the OXO and FLU can be identified by different retention time and the transition $[MH]^+ \rightarrow [MH-OCH_2O]^+$ ($262 \rightarrow m/z$ 216) for OXO and $[MH]^+ \rightarrow [MH-OC_3H_8]^+$ (m/z 262 \rightarrow 202) for FLU.

Ion ratios were determined for each analyte from 4 replicate samples spiked with all the analytes at each of five levels (0.23–5.0 ng/g). Ion ratios were all within tolerance in the EU document 2002/657/EC⁽³⁸⁾. Suggested tolerances are based on EU guidelines and range from $\pm 20\%$ for peaks greater than 50% of the base peak and to $\pm 50\%$ for those less than or equal to 10% of the base peak. Ion ratios of QNs are shown in Table 4. We noticed a small change of ion ratio with concentration, paralleled in the standard curves and spikes.

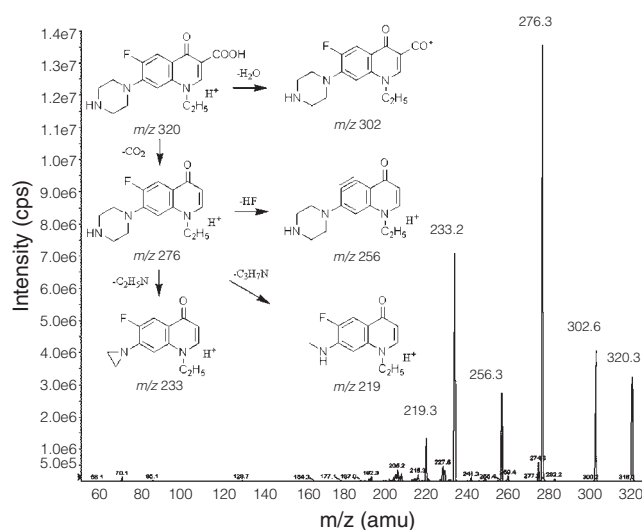


Figure 2. Product ion scan norfloxacin (m/z 320) mass spectrum at collision-induced dissociation in q2 at 30 V and proposed fragmentation pathway in norfloxacin.

Table 4. Ion ratios of two transition of 18 quinolones in standard and spiked samples

Quinolone	Ion ratios*						
	standard ^a	range ^b	Fish	Shrimp	Pork	Chicken	Milk
NAL	0.41	0.31-0.51	0.41-0.45	0.40-0.41	0.41-0.42	0.41-0.45	0.48-0.50
FLU	0.32	0.24-0.40	0.33-0.38	0.31-0.37	0.34-0.36	0.33-0.36	0.36-0.38
OXO	0.10	0.05-0.15	0.10-0.12	0.09-0.11	0.09-0.11	0.10-0.11	0.09-0.12
CIN	0.32	0.24-0.40	0.32-0.42	0.36-0.41	0.32-0.41	0.32-0.44	0.37-0.44
PIR	0.31	0.23-0.39	0.31-0.35	0.31-0.33	0.32-0.34	0.32-0.34	0.34-0.36
PIP	0.28	0.21-0.35	0.31-0.35	0.20-0.31	0.32-0.34	0.32-0.34	0.24-0.33
des-CIP	0.17	0.12-0.22	0.16-0.19	0.17-0.18	0.17-0.19	0.17-0.19	0.18-0.20
NOR	0.25	0.19-0.31	0.19-0.23	0.21-0.28	0.21-0.24	0.23-0.30	0.27-0.31
ENO	0.04	0.02-0.06	0.03-0.05	0.03-0.05	0.03-0.03	0.04-0.05	0.04-0.06
CIP	0.19	0.13-0.25	0.14-0.15	0.15-0.20	0.15-0.17	0.16-0.22	0.20-0.25
LOM	0.85	0.68-1.02	0.89-0.95	0.92-1.01	0.90-1.00	0.87-0.96	0.85-0.94
DAN	0.02	0.01-0.03	0.01-0.02	0.02	0.01-0.02	0.02	0.01-0.02
ENR	0.34	0.25-0.43	0.24-0.35	0.34-0.44	0.34-0.40	0.39-0.45	0.41-0.53
OFL	0.72	0.58-0.86	0.73-0.82	0.71-0.76	0.67-0.81	0.76-0.78	0.76-0.81
MAR	0.74	0.59-0.89	0.65-0.74	0.77-0.83	0.77-0.88	0.77-0.90	0.83-0.92
SAR	0.24	0.18-0.30	0.21-0.32	0.30-0.36	0.24-0.29	0.30-0.37	0.31-0.39
ORB	0.58	0.46-0.70	0.57-0.64	0.60-0.62	0.60-0.66	0.61-0.64	0.61-0.67
DIF	0.38	0.28-0.48	0.37-0.46	0.43-0.51	0.38-0.42	0.45-0.50	0.49-0.54

*Ion ratio: Identification ion peak area / Quantification ion peak area.

a: Ion ratios of standard solutions.

b: Maximum permitted tolerances according to Decision 2002/657/EC.

II. Sample Preparation

Analyses of residual QNs have been reported in many different matrices including fish^(16,20,25,32,37), milk^(17,24,26), egg^(28,29) and animal tissues such as muscle^(12,15,18,19,21,23), liver^(18,23) and kidney⁽³³⁻³⁶⁾. Rather time-consuming combination of liquid-liquid extraction (LLE) and SPE procedures for the extraction and cleanup were commonly used. However, due to the high selectivity of the MRM mode of LC/MS/MS for the determination of QNs in complex biological matrix⁽²³⁾, the sample treatment step was adapted to the simple LLE procedure in this study. The method has been applied to five different matrices (milk, fish, shrimp, pork and chicken) with simple and rapid extraction LLE method. Because acetonitrile has the best capability of extracting and removing protein⁽²⁶⁾, QNs are initially extracted with acidified (added 1% formic acid) acetonitrile and then dehydrated by anhydrous sodium sulfate. Hatano⁽²⁶⁾ demonstrated that acetonitrile containing either metaphosphoric acid, acetic acid or formic acid to maintain weak acidic condition may improve the recovery for QNs. He found that the formic acid used in acetonitrile achieved the best recovery.

III. Matrix Effect and Recovery

Matrix effect for Nalidixic acid was shown in Figure 3. The SCC was obtained from the analytical values of the beginning and end of the samples set. The individual values were calculated and gave a slope of 1.0514 (solid line with diamond marks, Figure 3). The MCC was compared to this. It can be seen that the matrix, in this case, enhanced the response by about 30 %, as the slope is greater than 1. Finally, the MMCC (solid line with diamond triangle marks, Figure 3) shows slope of 1.0076. By dividing the MMCC slope by the MCC slope it is possible to estimate the recovery, which in this case is about 76.8%. The matrix effect and recoveries for the QNs are shown in Table 5. The matrix effect enhances the responses from a low of 19.9% to a high of 49.3%. The absolute recoveries for CIN and PIR were lower than 60%, whereas the recoveries of the other QNs (NAL, ENO, CIP, ENR, OFL, SAR and DIF) were higher than 80%.

When using ESI ionization, the presence of matrix components (salts, proteins, lipids, carbohydrates, sugars, etc.) that affect the ionization of the target analytes may pose a significant problem, by either reducing or enhancing analytes response⁽³⁷⁾. This matrix effect can be minimized by an efficient clean-up step or different chromatographic conditions, but usually cannot be completely removed⁽⁴⁰⁾. Khong *et al.*⁽⁴⁰⁾ observed that a strong matrix effect of the quantification of tetracyclines (TCs) in honey extracts. Calibration curves of tetracycline from the LC/ESI/MS/MS of standard solution and in spiked honey matrices from different geographical and flower origins,

the matrix effect were quite different ranging from 163% (Swiss acacia) to 226% (Swiss multiflower). Samanidou *et al.*⁽³⁷⁾ accounted matrix effect of the quantification of QNs in seabream (fish). Relative responses ranged from 0.2 to 0.4 for DAN, ENR and FLU indicating signal suppression, 1.8 for CIP indicating signal enhancement, while no significant matrix effect was noticed for SAR, OXO and NAL.

Choice of internal standard (IS) can correct for matrix effects, recovery through the extraction, and any common loss. Toussaint *et al.*⁽³³⁾ used LOM as an internal standard for the quantitation of amphoteric QNs and cincophen for acidic QNs. Durden *et al.*⁽²⁹⁾ used

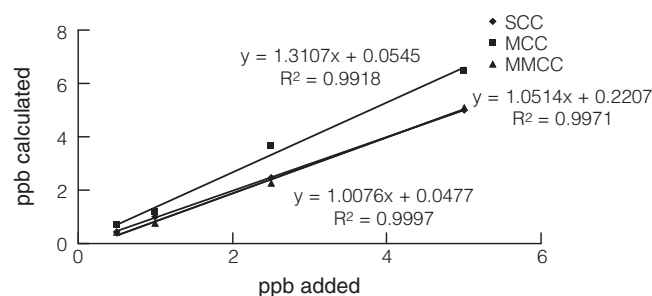


Figure 3. Matrix effect and absolute recovery in Tilapia fish for Nalidixic acid.

Table 5. Matrix effects and Absolute recoveries of 18 quinolones in Tilapia muscle (n = 4)

Quinolone	Matrix effect (%)	Absolute recovery (%)
NAL	124.7	95.8
FLU	81.2	67.0
OXO	89.0	72.9
CIN	92.0	57.3
PIR	91.1	51.5
PIP	107.0	61.2
des-CIP	82.3	66.5
NOR	88.5	76.6
ENO	149.3	95.1
CIP	94.9	80.1
LOM	83.6	68.7
DAN	87.8	82.3
ENR	97.8	88.8
OFL	95.1	84.7
MAR	119.1	64.3
SAR	97.2	89.2
ORB	80.1	65.1
DIF	94.8	91.2

NOR as the IS for CIP and the better IS for DAN, ENR, and SAR was LOM. Dufresne *et al.*⁽²⁸⁾ used LOM and PIP as the IS. Hermo *et al.*⁽³⁰⁾ chose NOR as the IS. It's difficult for us to select adequate internal standards for all QNs, therefore, an external mode without addition of

internal standards was employed in the current study. Quantification by matrix calibration curves (MCC) mode is strongly recommended when sample was detected certain levels of QNs.

Table 6. The different characteristics in CC α and CC β for 18 quinolones in the fish, shrimp, pork, chicken and milk

Quinolone	Fish		Shrimp		Pork		Chicken		Milk	
	CC α (ng/g)	CC β (ng/g)	CC α (ng/g)	CC β (ng/g)	CC α (ng/g)	CC β (ng/g)	CC α (ng/g)	CC β (ng/g)	CC α (ng/g)	CC β (ng/g)
NAL	0.45	0.61	0.49	0.66	0.59	0.80	0.43	0.59	0.18	0.24
FLU	0.44	0.59	0.67	0.91	0.43	0.59	0.48	0.66	0.36	0.49
OXO	0.48	0.66	0.42	0.58	0.50	0.69	0.56	0.77	0.60	0.82
CIN	0.41	0.56	0.51	0.70	0.35	0.47	0.51	0.69	0.51	0.69
PIR	0.39	0.53	0.64	0.87	0.43	0.59	0.53	0.72	0.39	0.54
PIP	0.53	0.73	0.57	0.77	0.68	0.92	0.51	0.70	0.41	0.56
des-CIP	0.60	0.83	0.65	0.88	0.37	0.50	0.55	0.75	0.48	0.66
NOR	0.54	0.75	0.70	0.96	0.59	0.81	0.52	0.71	0.62	0.84
ENO	0.56	0.76	0.59	0.81	0.42	0.57	0.51	0.70	0.61	0.84
CIP	0.50	0.69	0.58	0.79	0.70	0.95	0.40	0.55	0.65	0.89
LOM	0.62	0.84	0.58	0.79	0.60	0.81	0.52	0.71	0.42	0.57
DAN	0.49	0.67	0.64	0.88	0.56	0.76	0.50	0.68	0.56	0.76
ENR	0.54	0.74	0.64	0.87	0.57	0.77	0.39	0.53	0.51	0.70
OFL	0.31	0.43	0.60	0.81	0.57	0.77	0.64	0.88	0.41	0.56
MAR	0.47	0.64	0.53	0.73	0.50	0.61	0.40	0.55	0.41	0.56
SAR	0.50	0.68	0.35	0.48	0.42	0.58	0.38	0.52	0.50	0.68
ORB	0.47	0.56	0.43	0.59	0.28	0.38	0.41	0.56	0.48	0.65
DIF	0.34	0.46	0.47	0.65	0.52	0.70	0.47	0.65	0.50	0.69

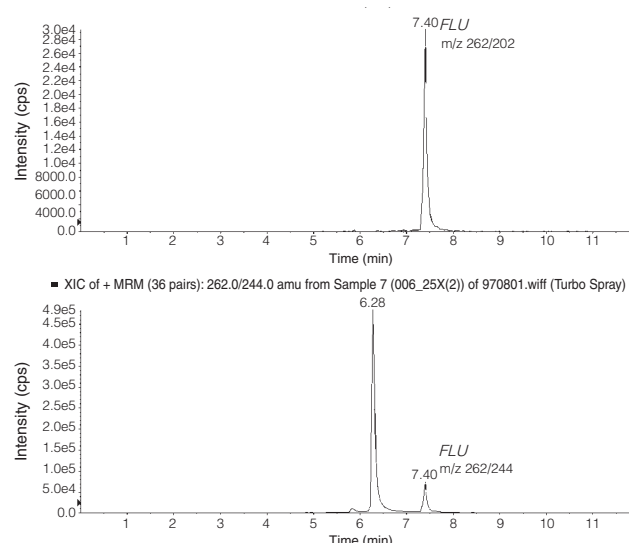
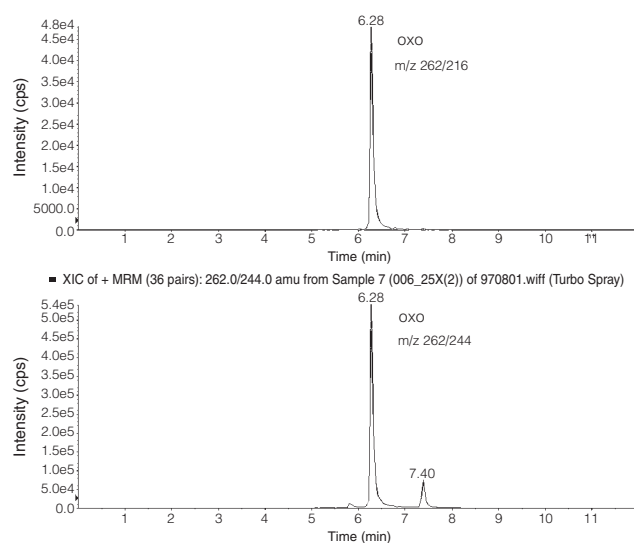


Figure 4. Ion reconstituted chromatographs obtained for a proficiency testing sample (fish muscle).

IV. Decision Limit ($CC\alpha$) and Detection Capability ($CC\beta$)

In Table 6, the $CC\alpha$ and $CC\beta$ are summarized for different QNs in each matrix. The values of $CC\alpha$ ranged from 0.18 to 0.68 ng/g and $CC\beta$ ranged from 0.24 to 0.96 ng/g. According to Taiwan's MRLs⁽⁹⁾, limits for residues of 4 QNs (OXO, FLU, SAR and DAN) were included in the regulation. The concentration range is wide (1/2 MRL to 2x MRL) owing to the authorized (MRL) substances. As a result, the calibration curve for each of QNs with LC-MS/MS method was hard to be linear in this range. In this study, for the four QNs that have permitted tolerance in Taiwan, those $CC\alpha$ and $CC\beta$ were also calculated as unauthorized substance.

V. Application to a Proficiency Testing Sample

A fish muscle sample of FAPAS proficiency test 02117⁽⁴¹⁾ was dispatched in July 2008, 56 laboratories took part in the exercise with test 02117, and the proposed method has been applied to determinate a proficiency fish muscle testing sample. In order to deal with the high concentration of FLU and OXO, the final extraction of the sample was diluted 25-fold. The assigned values for FLU and OXO are 90.8 and 286 ng/g, respectively, in the test 02117. Our sample has been tested with the values of FLU 88.6 ng/g and OXO 279 ng/g. Z-score of the two results sample are -0.1. Figure 4 shows the ion reconstituted chromatogram. The results of comparison between and LC/ESI/MS/MS method and HPLC-FLD⁽²⁵⁾ method are presented in Table 7.

CONCLUSIONS

The LC/ESI/MS/MS multi-residue method allows simultaneous determination 18 QNs within the ng/g level. To the best of our knowledge it is the first method for the determination of 18 QNs simultaneously. A simple and rapid extraction method allows the analysis conducted in 5 different food matrixes (chicken, pork, shrimp, fish and milk). The LC-ESI/MS/MS analysis procedure established gives $CC\alpha$ and $CC\beta$ down to 1 ng/g. The proposed method has been successfully applied to determine FLU and OXO in fish muscle from FAPAS proficiency testing

sample. Thus, the short extraction step with LC/MS/MS detection should easily support a successful switch from HPLC/FLD⁽²⁵⁾ method and could be applied for determining multi-quinolones residues.

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Table 7. The results of FAPAS proficiency testing (n = 3)

Quinolone	method			
	LC/MS/MS		LC/FLD ⁽²⁵⁾	
	(ng/g)	R.S.D. (%)	(ng/g)	R.S.D. (%)
FLU	88.6	5.2	86.9	4.3
OXO	279.0	9.7	281.3	9.5

Z-score for LC/MS/MS method was -0.1.

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